Coal Processing Plants for Hydrogen Production with CO<sub>2</sub> Capture

Workshop on Production of Hydrogen from Fossil Fuels with Carbon Sequestration

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# Background

- Preparing conceptual designs, systems analysis and economics for hydrogen plants based on Oak Ridge hydrogen separation membranes
- Objective is to economically produce separate streams of hydrogen and CO<sub>2</sub>
- Results are compared with conventional production of hydrogen and CO<sub>2</sub> from coal and natural gas

Membranes for Hydrogen Separation

- Proton Conductive Ceramic Membrane
- Electrochemical Conversion Membrane
- Palladium-Based Membrane
- Inorganic Membrane
  - Eastern Tennessee Technology Park Oak Ridge, Tennessee

#### **ETTP Inorganic Membrane Characteristics**

#### Porous Ceramic

- $Al_2O_3$
- 5 Angstrom
- Knudson diffusion
- Separation Factor
  - ◆ H<sub>2</sub> transport relative to retenate
  - ♦ Gas purity = 1 1/SF
- H<sub>2</sub> Transport Proportional to Partial Pressure Differential

#### The Inorganic Membrane Separation Concept



# Hydrogen Separation Device Design Assumptions

- Hydrogen Separation Factor
  - ~ 200 @ 99.5% hydrogen
- Hydrogen Transport Rate
  - ◆ 0.1 cm<sup>3</sup>/minute/cm<sup>2</sup>/cmHg<sub>△</sub>P
- Operating Temperature and Pressure
  - 600°C, 65 bar exit gas conditions
  - 300°C, 65 bar exit gas conditions
- **Cost** 
  - **\$1,076/m<sup>2</sup> membrane**
- Design Configuration
  - Tube side pressurization
- Tube Dimensions
  - 16 mm OD x 2.8 m long

#### Hydrogen Separation Device Concept



#### **Design Basis for Advanced Hydrogen Plant**

Plant Size	H <sub>2</sub> production from 2,270 metric tpd dry coal feed Excess power sold offsite
Gasifier	Oxygen-blown entrained bed 1040°C, outlet pressure 68 bar
Gas Cleanup	593°C desulfurization with transport reactor Sulfur recovery as sulfuric acid Ceramic candle particulate filter
Hydrogen Separation	95% separation, 99.5% pure H <sub>2</sub>
Hydrogen Utilization	Compress to 24 bar
Retenate Gas	5% of fuel value remaining @ 65 bar Fire with oxygen and expand to 1.4 bar

# Hydrogen Plant with Hot Gas Desulfurization and Conventional Turbine Expander



# **Conventional Hydrogen Plant** with Maximum CO<sub>2</sub> Removal



# Steam Reforming Natural Gas with CO<sub>2</sub> Removal



### **Plant Summaries**

I	Coal to Hydrogen with 600°C norganic Membrane	Conventional Coal to Hydrogen w/CO <sub>2</sub> Recovery	Natural Gas- Steam Reforming w/CO <sub>2</sub> Recovery
Feedstock Rate	100,620 kg/h	100,620 kg/h	1.708 MM nm <sup>3</sup> /day
Oxygen Feed (95%)	101,932 kg/h	128,714 kg/h	N/A
Hydrogen Production	16,300 kg/h	12,025 kg/h	16,600 kg/h
■ CO <sub>2</sub> Recovery (% of total)	240,421 kg/h (94%)	235,818 kg/h (92%)	98,707 kg/h (71%)
Gross Power Production	84 MW	64 MW	<b>0 MW</b>
Auxiliary Power Requirem	nent 77 MW	52 MW	6 MW
Net Power Production	7 MW	12 MW	(6 MW)
Effective Thermal Efficier	ncy 80.4%	60.1%	78.6%
Capital Cost, \$1,000 (Yr 2	000) \$362,994	\$374,906	\$142,370
Feedstock Cost Delivered	\$0.95/GJ	\$0.95/GJ	\$3.00/GJ
Hydrogen Product Cost	\$4.80/GJ	\$6.55/GJ	\$5.62/GJ

#### Hydrogen Plant with Wyodak Coal/Biomass



# **Performance and Cost Comparisons Pittsburgh No. 8 and Wyodak/Biomass**

	Inorganic Membrane <u>Pittsburgh No. 8</u>	Inorganic Membrane <u>Wyodak/Biomass</u>
Coal Feed	100,620 kg/h	128,860 kg/h
<b>Biomass Feed</b>	N/A	14,317 kg/h
Oxygen to Gasifier	75,281 kg/h	84,739 kg/h
Oxygen to Retenate Combustor	26,650 kg/h	11,486 kg/h
Hydrogen Product Stream	16,300 kg/h	15,135 kg/h
Sulfuric Acid Byproduct	8,845 kg/h	2,296 kg/h
Net Power Production	7 MW	14 MW
Effective Thermal Efficiency	80.4%	79.8%
Capital Cost, \$1,000	\$359,791	\$365,662
Feedstock Cost, Delivered	\$0.95/GJ	\$0.62/GJ
Hvdrogen Product Cost	\$4.95/GJ	\$4.80/GJ

# CO<sub>2</sub> Emissions Comparisons Pittsburgh No. 8 and Wyodak/Biomass

	Inorganic Membrane <u>Pittsburgh No. 8</u>	Inorganic Membrane <u>Wyodak/Biomass</u>
<b>Total CO<sub>2</sub> Produced</b>	255,897 kg/h	262,491 kg/h
Biomass Credit	N/A	26,143 kg/h
Net CO <sub>2</sub> Produced	255,897 kg/h	236,348 kg/h
■ CO <sub>2</sub> Recovered	240,421 kg/h	235,845 kg/h
<b>CO<sub>2</sub> Emissions</b>	15,476 kg/h	503 kg/h
Ton CO, Emissions/Ton H,	0.95	0.0331

#### **Conclusions**

#### **Hydrogen Fuel:**

- Can be produced from coal in a cost-competitive manner with advances in materials and separation technologies
- Can be produced while efficiently capturing CO<sub>2</sub>
- Will play an important role in meeting market and environmental challenges
- Can be made available to meet regional challenges for sustainable growth and economic prosperity