

Radiation Litmus Paper

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Funding Provided by:

DOE Office of Nonproliferation and National Security,
Office of Research and Development (NA-22)

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Project Goals:

- 1) Improve detection limits of chemical-based radiation measurements
- 2) Incorporate these reactions into a generally useable form
 - Adjustable Sensitivity
 - Lightweight and Portable - Similar to a TLD
 - No Power/Cooling Requirements
 - Require Little/No Training
 - Inexpensive
 - Extended Shelf-Life

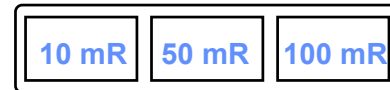
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Detect Radiation by a Color Change

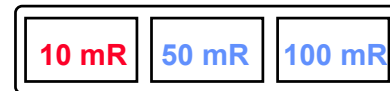
Potential Uses:

- “First Responder” Badge
- Personal Dosimetry

- Form and Use Similar to Other Detection Equipment



Before Exposure



After Exposure



Anachemia M256A1 Chemical Test Kit

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Background Numbers: *How Little Can You See ?*

A person can see a 2% change in color

Assume a good colorant ($\epsilon=50,000$), 1 cm path length, 1 cc volume

$$\text{Log } (I_0/I) = \epsilon \times \text{path length} \times [\text{colorant}]$$

$$\text{Log } (100/98) = .009 = 50,000 \times 1 \times [\text{colorant}]$$

$$[\text{colorant}] = 1.7 \times 10^{-7} \text{ moles/liter}$$

$$= 1.7 \times 10^{-10} \text{ moles/mL}$$

$$= 10^{14} \text{ molecules}$$

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Background Numbers: *Chemical Reactions from Radiation*

Radiation produces ion or radical pairs

$$\mathbf{G = reactions / 100 eV}$$

Commonly G=3 for gasses, liquids (G=30 for solids)

For a 100 keV hv: $10^5 \text{ eV} \times 3 \text{ rxns}/100 \text{ eV} = 3,000 \text{ rxns/hv}$

For a $^{10}\text{B}(n,\alpha) \text{}^7\text{Li}$ reaction: $2.4 \text{ MeV} \times 3 \text{ rxns}/100 \text{ eV} = 72,000 \text{ rxns/n}$

Radiation can produce $\sim 10^3$ - 10^5 reactions per interaction

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Historical Problems with Colorimetric Detection: Sensitivity

Radiation ionizes materials: $\sim 10^3$ (γ) to 10^5 (n_{th}) molecules per event

Common chain reactions give a 10-100-fold amplification

A person can see $\sim 10^{14}$ molecules of a colorant

So need 10^7 - 10^{10} captured events to see

Example: 100 keV γ : Assume a 1 mm thick hydrocarbon detector

= $(10^{14} \text{ rxns}) / (10^3 \text{ rxns per } \gamma) \times (0.1\% \text{ capture efficiency})$

= $10^{14} \text{ h}_v / \text{cm}^2$ to see = $\sim 5,000 \text{ R}$

Example: Neutrons: Assume a 1 mm thick hydrocarbon detector, 10% ^{10}B

= $(10^{14} \text{ rxns}) / (10^5 \text{ rxns per } n_{th}) \times (30\% \text{ capture efficiency})$

= 10^8 - $10^9 n_{th} / \text{cm}^2$ to see

however, thermal neutrons (n_{th}) = 1% of initial flux (n_f)

so to get $10^8 n_{th}$, need initially $10^{10} n_f / \text{cm}^2$

which is a initial dose of 300-400 rem

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Historical Problems with Colorimetric Detection: Stability

Chain reactions can increase the chemical yield from captured radiation

Common chain reactions include decomposition of halogenated materials (RX), including CCl_4 , CHCl_3 , PVC

RX decomposition can yield acid or oxidizers

Can achieve chain lengths of 10-100 (realistically closer to 10)

Problem: *RX is usually unstable to light and heat*

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Methods to Enhance Signal: Autocatalysis

Ionization Creates Catalyst

Product of Catalyzed Reaction is Second Catalyst Molecule

Issues:

Sensitivity vs Stability

Stability to Environment (heat, light)

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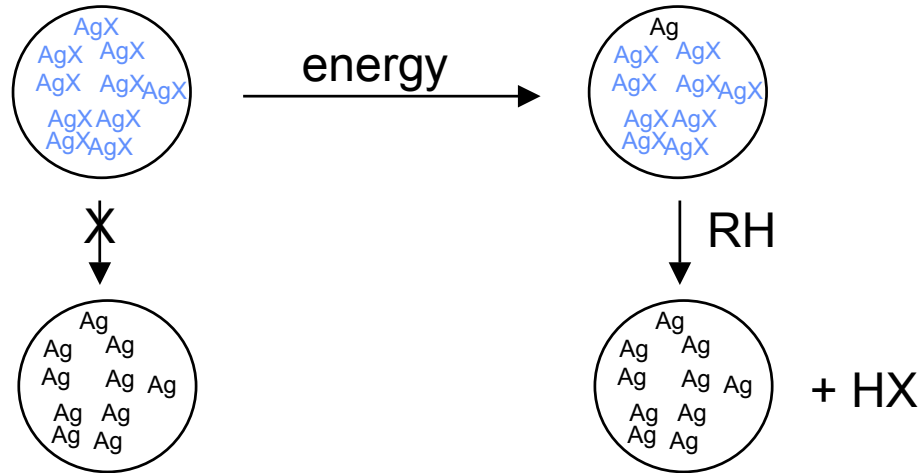
Underlying Chemistry: 10^8 - 10^{10} Amplification

$(\text{AgX})_n + \text{RH} \longrightarrow$ no reaction - *device is dormant*

$(\text{AgX})_n + \text{radiation} \longrightarrow \text{AgX}_{n-m} \text{Ag}_m + \text{X}_{m/2}$

$(\text{AgX})_{n-m} (\text{Ag})_m + n\text{RH} \longrightarrow \text{Ag}_n + \text{R}_{n-m} + \text{HX}_{n-m}$

$\text{HX}_{n-m} + \text{Indicator} \longrightarrow \text{Indicator} \cdot \text{HX}$ - *color change*

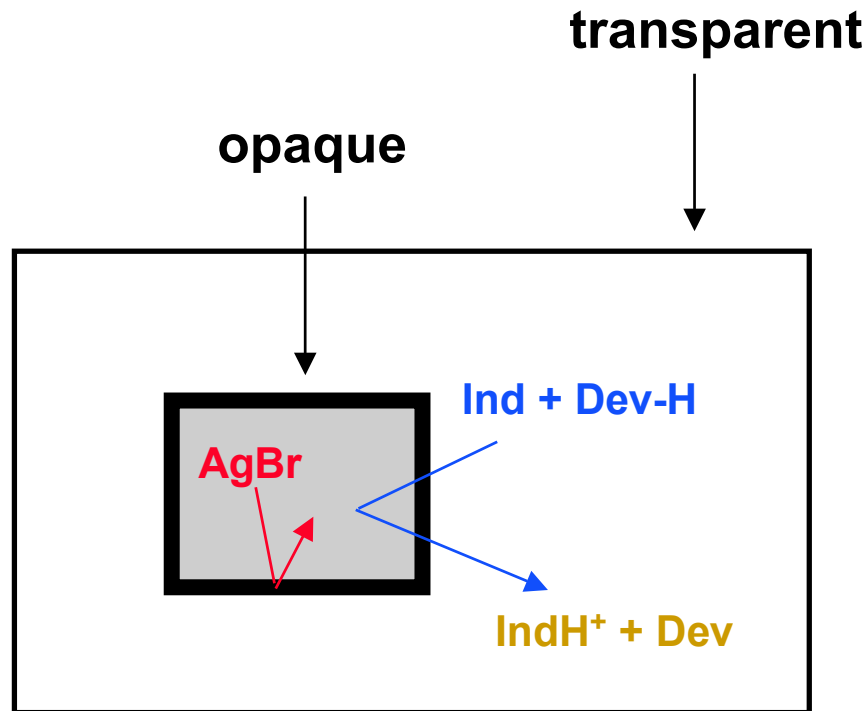


Need *four* atoms of Ag^0 to activate grain

Typical grains are .25-1 micron diameter = 10^8 - 10^{10} atoms Ag/grain

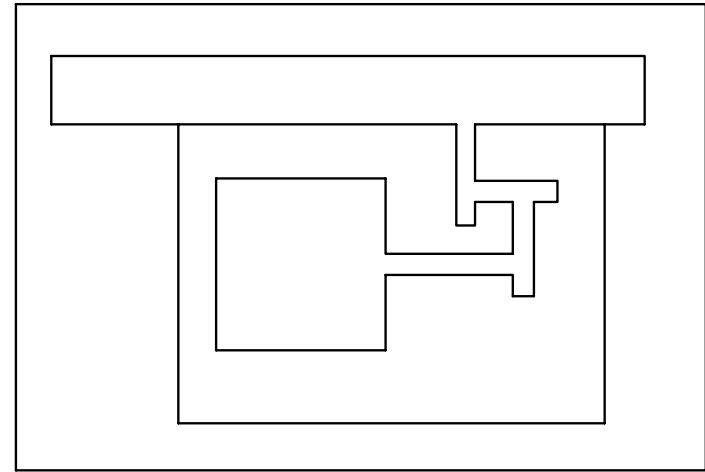
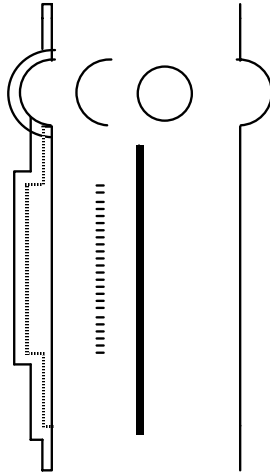
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General Schematic Drawing



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Schematic:



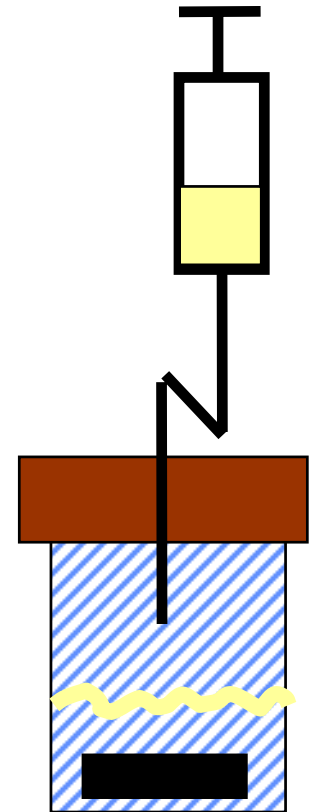
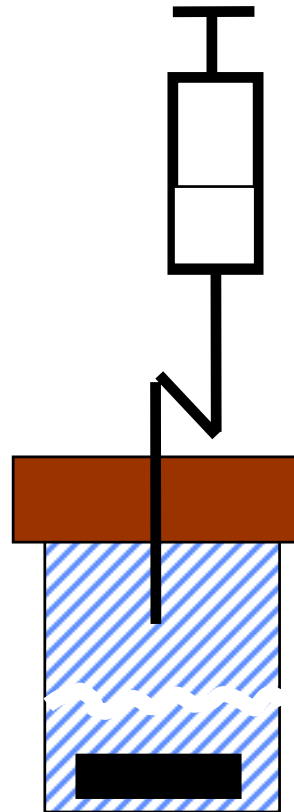
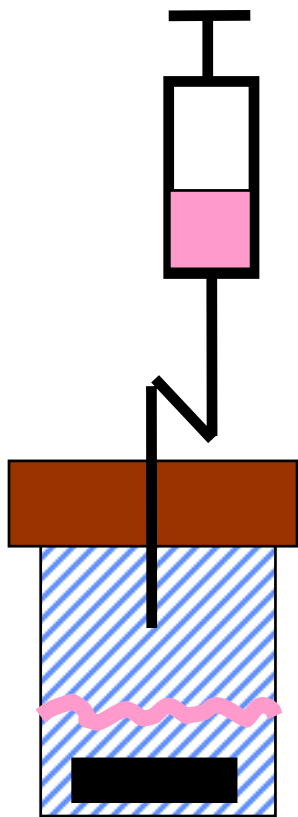
**Fieldable form of
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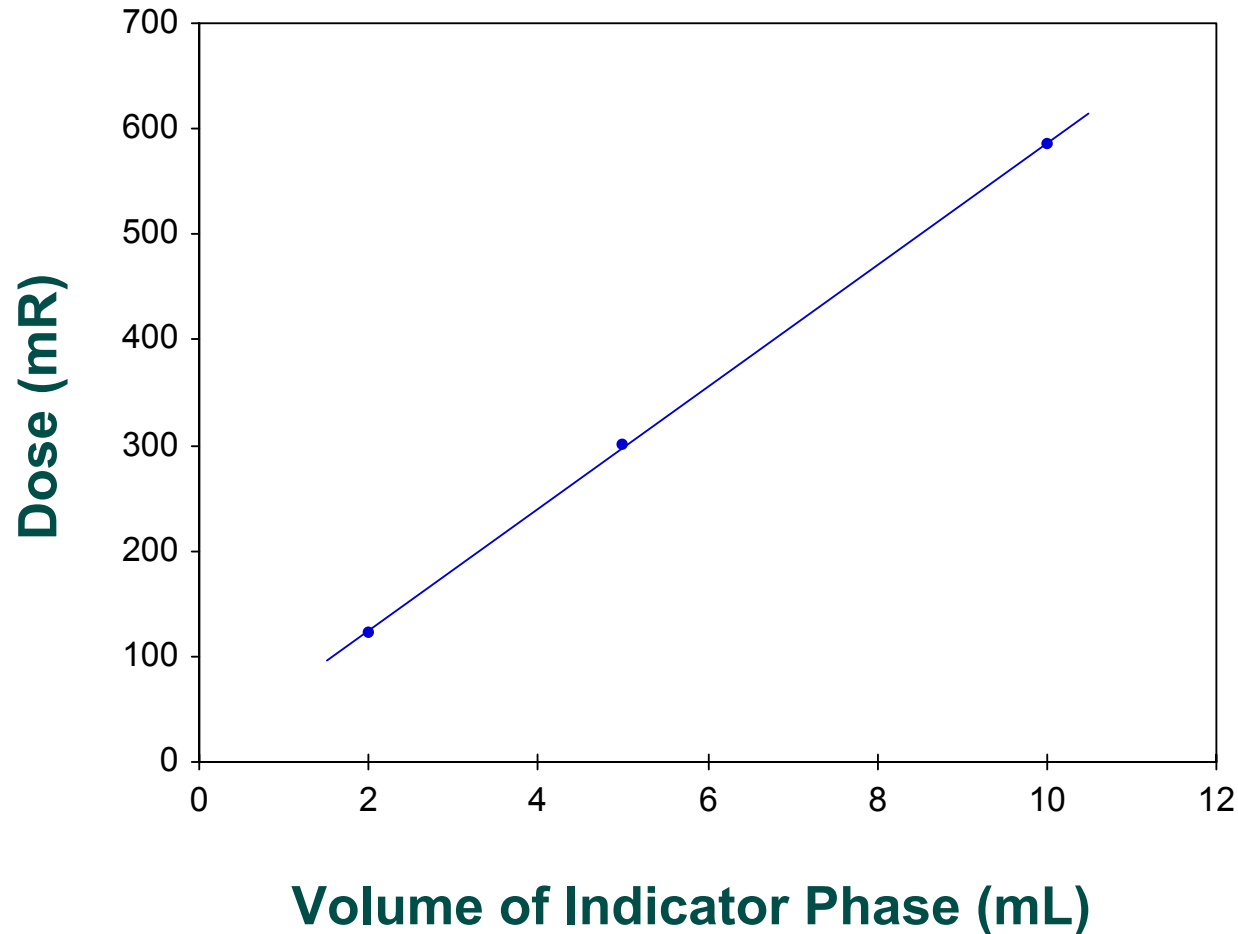
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Prototype For Testing Chemistry:



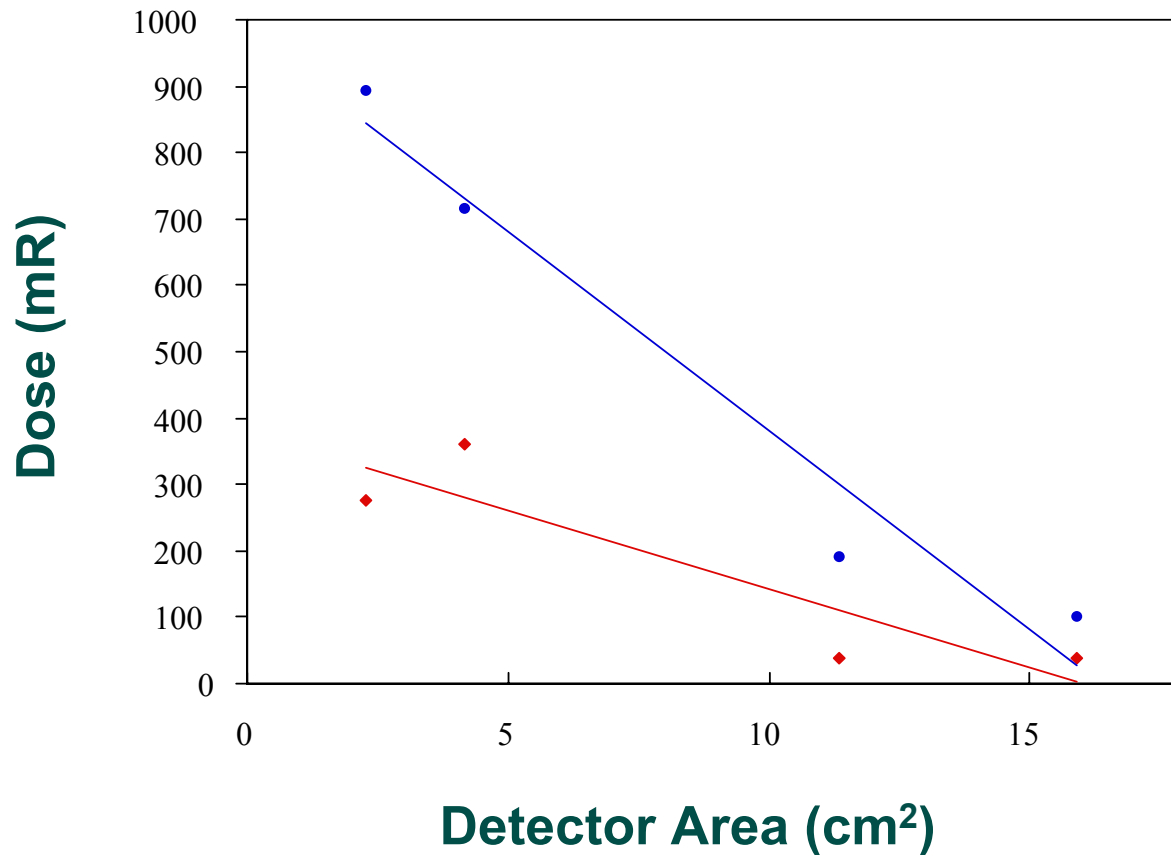
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Sensitivity vs Volume of Indicator Phase



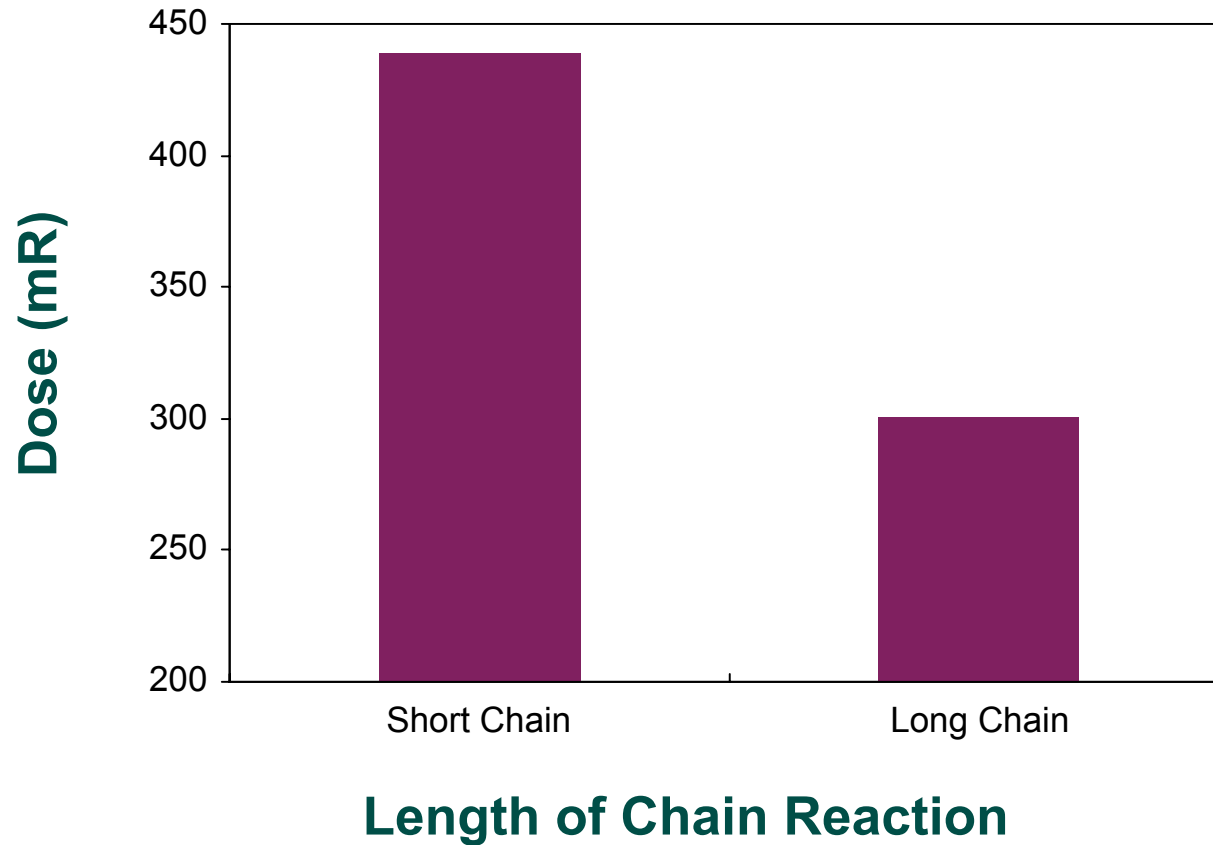
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Sensitivity vs Detector Area



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Sensitivity vs Chain Length



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Features:

Useful Lifetime of Material: Devices function for 8-36 hours

Material Form: Liquid and gel developers

Color Choice: Dark {blue, purple, red} to yellow, or red to blue

Multi-Component System Allows Tuning

Easy to Synthesize

Relatively Stable

Long Shelf Life

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Current Performance:

- **Adjustable Sensitivity - 17 to 2000 mR**
- **Easy to Make and Use**
- **Small - 17 mR Device ~ 2 cm³**
- **Threshold or Continuous Dosimeter**
- **Inexpensive - Chemicals < \$1**
- **Calibrated for ¹³⁷Cs γ -Radiation**
- **Sensitive to β , neutron**

Future Work:

- **Enhance Color Change Range**
- **Mass Produce for Field Test**