



# Emergency Planning and Community Right-To-Know Act Section 313 Reporting Guidance for the Leather Tanning and Finishing Industry

Page 1 of 1

From approved OMB number 2050-0047  
Revised October 14, 2000

## TOXIC CHEMICAL RELEASE INVENTORY REPORTING FORM

FORM R

Emergency Planning and Community Right-To-Know Act of 1986, also known as Title III of the Superfund Authorization Act

APPROPRIATE STATE OFFICE (See instructions & Appendix F)

Enter "X" here if this is a revision  
(For EPA use only)

WHERE TO SEND COMPLETED FORM: See instructions

SECTION 1. FACILITY IDENTIFICATION INFORMATION

REPORTING YEAR: 19

SECTION 2. TRADE SECRET INFORMATION

Are you claiming the toxic chemical identified on page 2 trade secret?  No  Yes (Answer question 2.2; go to Section 3)

Is this copy  Sanitized  Unsanitized (Answer only if "YES" in 2.1)

SECTION 3. CERTIFICATION (Important: Read and sign after completing all form sections.)

I hereby certify that I have reviewed the attached documents and that, to the best of my knowledge and belief, the information submitted using data available to the preparers of this report is true and complete and that the amount of information submitted is of owner/operator or senior management official.

Name and Title: \_\_\_\_\_ Date signed: \_\_\_\_\_

SECTION 4. FACILITY IDENTIFICATION

4.1 Facility or Establishment Name: \_\_\_\_\_

City or Suburb: \_\_\_\_\_ State: \_\_\_\_\_ Zip: \_\_\_\_\_

4.2 This report contains information for: (Important: check a or b, mark c if applicable)

4.3 Technical Contact Name: \_\_\_\_\_ Phone: \_\_\_\_\_

4.3 Public Contact Name: \_\_\_\_\_ Phone: \_\_\_\_\_

4.4 EPA Identification Number(s) (RCRA I.D. No.) (12 characters): \_\_\_\_\_

4.5 Facility NPDES Permit Number(s) (9 characters): \_\_\_\_\_

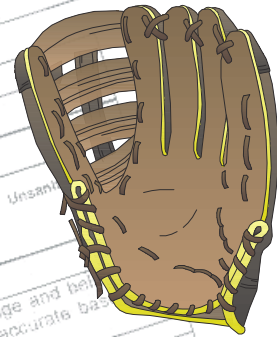
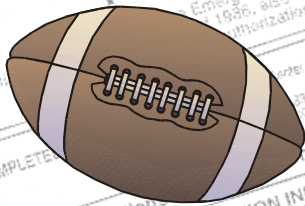
4.6 Underground Injection Well Code (UIC) I.D. Number(s) (12 digits): \_\_\_\_\_

SECTION 5. PARENT COMPANY INFORMATION

5.2 Name of Parent Company: \_\_\_\_\_

Parent Company's DUN & Bradstreet Number: \_\_\_\_\_

EPA Form 600-1 (Rev. 04/97) Previous editions are obsolete.



REPORT

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## **ACKNOWLEDGMENT**

The U.S. EPA wishes to acknowledge the valuable contributions made by the staff and members of Leather Industries of America, Inc. and Garden State Tanning, Inc. Without the insight provided by those in industry with actual experience in fulfilling the reporting requirements of EPCRA Section 313 we would not have been able to produce a document that we believe will be of great assistance to those who must prepare future EPCRA Section 313 reports. Special thanks go to Dr. Nicholas Cory, Technical Director, Leather Industries of America, Inc.; and Mr. Chris Ehret and Mr. Chuck Carbaugh of Garden State Tanning, Inc. for their hard work.

## OVERVIEW

This document supersedes the booklet entitled *Title III Section 313 Release Reporting Guidance, Estimating Chemical Releases from Leather Tanning and Finishing*, dated February 1988. It is intended to assist establishments and facilities performing leather tanning and finishing in complying with the Emergency Planning and Community Right-To-Know Act (EPCRA) Section 313 and Pollution Prevention Act (PPA) Section 6607 reporting requirements, the preparation of Form R or the alternate certification statement, Form A. The EPCRA Section 313 program is commonly referred to as the Toxic Chemical Release Inventory (TRI).

The principal differences in this new document include:

- More detailed examples;
- New EPCRA Section 313 regulations and guidance developed since 1988;
- PPA Section 6607 reporting requirements;
- U.S. Environmental Protection Agency's (U.S. EPA's) interpretive guidance on various issues specific to leather tanning and finishing operations; and
- EPCRA Section 313 issues regarding processes not discussed in the earlier documents.

This document is designed to be a supplement to the annual issue of the *Toxic Chemical Release Inventory Reporting Forms and Instructions (TRI Forms and Instructions)*. It is organized to provide a step-by-step guide to compliance with EPCRA Section 313 and PPA Section 6607, starting with how to determine if your facility must report and ending with guidance for estimating release and other waste management activity quantities.

It is recognized that not all leather tanning and finishing establishments will have all unit operations described in this document. However, each of the unit operations discussed are common operations found in leather tanning and finishing establishments covered by EPCRA Section 313 and PPA Section 6607 reporting requirements. You should select the operation, or combination of operations, that most closely fits the activities at your establishment.

Chapter 1 introduces EPCRA Section 313 and PPA Section 6607 reporting and provides a brief background on Section 313 of EPCRA and Section 6607 of PPA.



Chapter 2 discusses reporting requirements and begins with how to determine whether your facility must report. This determination is based on your answers to a series of four questions:

- Is your facility's primary SIC Code on the EPCRA Section 313 list?
- Does your facility employ ten or more full-time employees or the equivalent?
- Does your facility manufacture, process, or otherwise use any EPCRA Section 313 chemicals or chemical categories?
- Does your facility exceed any of the activity thresholds for an EPCRA Section 313 chemical or chemical category?

If the answer to ANY ONE of the first three questions is "No" you are not required to submit an EPCRA Section 313 report for any chemicals. If you answer "Yes" to the first three questions and "No" to the fourth, you are not required to submit an EPCRA Section 313 report for that chemical or chemical category. If you answer "Yes" to ALL four questions, the next step is to determine what kind of report you must prepare, a Form R or the alternate certification statement, Form A. Chapter 2 provides detailed information on the requirements for each kind of report. Chapter 2 concludes with a discussion on how to address trade secrets and the records that should be kept to support your reporting.

Chapter 3 discusses how to calculate the activity thresholds (manufacture, process, and otherwise use) for the EPCRA Section 313 chemicals or chemical categories. Information is provided on how to determine which EPCRA Section 313 chemicals or chemical categories your facility manufactures, processes, or otherwise uses and how to calculate the quantities of each. Detailed information is also provided on the various exemptions:

- *De minimis* exemption;
- Article exemption;
- Facility-related exemption; and
- Activity-related exemptions.

Chapter 3 concludes with a discussion of how to determine which EPCRA Section 313 chemicals or chemical categories exceed a reporting threshold.

Chapter 4 discusses how to estimate the release and other waste management activity amounts for those EPCRA Section 313 chemicals and chemical categories for which you must prepare a report. The first part of this chapter provides a step-by-step approach designed to minimize the risk of overlooking an activity involving an EPCRA Section 313 chemical or chemical category and any potential sources or types of release and other waste management activities. This procedure consists of:

- Preparation of a detailed **process flow diagram**;
- Identification of EPCRA Section 313 chemicals and chemical categories and potential **sources** of chemical release and other waste management activities;
- Identification of the potential **types** of release and other waste management activities from each source; and
- Determination of the most appropriate methods for **estimating the quantities** of EPCRA Section 313 chemical and chemical category release and other waste management activities.

The second part of Chapter 4 is organized by five major groupings. These subsections describe the typical leather tanning and finishing unit operations where EPCRA Section 313 chemicals and chemical categories are used: beamhouse operations; tanhouse operations; retanning, coloring, and fatliquoring operations; finishing operations; and wastewater treatment. The commonly used EPCRA Section 313 chemicals and chemical categories, process descriptions, release and other waste management activity estimates, example calculations, and common problems are presented.

This document includes examples and common errors applicable to leather tanning and finishing operations. These examples are based on information identified during voluntary site surveys of facilities that have filed EPCRA Section 313 reports in the past, discussions with representatives of the Leather Industries of America, Inc. and Garden State Tanning, Inc., and on questions received by the EPCRA Hotline.

# CHAPTER 1 - INTRODUCTION

## 1.0 PURPOSE

The purpose of this guidance manual is two fold. The primary purpose is to assist leather tanning and finishing facilities in complying with the reporting requirements of Section 313 of the Emergency Planning and Community Right-to-Know Act of 1986 (EPCRA) and of Section 6607 of the Pollution Prevention Act of 1990 (PPA). This manual explains the EPCRA Section 313 reporting requirements and discusses specific release and other waste management activities encountered at many facilities that conduct leather tanning and finishing operations. Since each plant is unique, the recommendations presented may have to be modified for your particular facility. The secondary purpose is to provide information to other interested parties (such as management, legal professionals, inspectors, consultants, teachers, students, and the general public) about the processes and some of the toxic chemicals used in this industry.

This manual is intended solely for guidance and does not alter any statutory or regulatory requirements. The document should be used in conjunction with the appropriate statutes and regulations but does not supersede them. Accordingly, the reader should consult other applicable documents (for example, the statute, the Code of Federal Regulations (CFR), relevant preamble language, and the current *Toxic Chemical Release Inventory Reporting Forms and Instructions (TRI Forms and Instructions)*).

This document supersedes the 1988 document entitled *Title III Section 313 Release Reporting Guidance, Estimating Chemical Releases from Leather Tanning and Finishing*. This new document includes:

- More detailed examples;
- New EPCRA Section 313 regulations and guidance developed since 1988;
- PPA Section 6607 reporting requirements;
- U.S. Environmental Protection Agency's (U.S. EPA's) interpretive guidance on various issues specific to leather tanning and finishing operations; and
- EPCRA Section 313 issues regarding processes not discussed in the earlier document.

It is intended to supplement the *TRI Forms and Instructions* document that is updated and published annually by U.S. EPA. It is essential that you use the current version of the *TRI Forms and Instructions* to determine if (and how) you should report. Changes or modifications to EPCRA Section 313 reporting requirements are reflected in the annual *TRI Forms and Instructions* and should be reviewed before compiling information for the report.

The objectives of this manual are to:

- Reduce the level of effort expended by those facilities that prepare an EPCRA Section 313 report; and
- Increase the accuracy and completeness of the data being reported.

U.S. EPA cannot anticipate every potential issue or question that may apply to your facility. Therefore, this manual attempts to address those issues most prevalent or common for leather tanning and finishing operations. Used in conjunction with the most current *TRI Forms and Instructions* and *Estimating Releases and Waste Treatment Efficiencies for the Toxic Chemical Release Inventory Form (1999 version)*, facilities should be able to provide complete and accurate information for EPCRA Section 313 reporting. Additional discussions on specific issues can be found in U.S. EPA's current edition of *EPCRA Section 313, Questions and Answers* (the 1998 edition is EPA 745-B-98-004), which is available on the U.S. EPA's TRI website (<http://www.epa.gov/tri>) or by contacting the **EPCRA Hotline at 1-800-424-9346**. In the Washington DC metropolitan area, call 703-412-9810.

## **1.1 Background on EPCRA Section 313 and PPA Section 6607**

The following overview of EPCRA Section 313 and Section 6607 of the PPA, will provide you with a basic understanding of the objectives and requirements of this program, and will help you in completing your forms.

One of the primary goals of EPCRA is to increase the public's knowledge of, and access to, information on both the presence of toxic chemicals in their communities and on releases into the environment and other waste management activities of those chemicals.

EPCRA Section 313 requires certain designated businesses (see SIC Code discussion, Chapter 2, Section 2.2) to submit annual reports (commonly referred to as Form R reports and Form A reports) on over 600 EPCRA Section 313 chemicals and chemical categories. Covered facilities report the amounts released or otherwise managed as waste. However, if a facility meets the reporting criteria for listed toxic chemicals, the facility must report even if there are no releases or other waste management quantities associated with these chemicals. Throughout this document, whenever EPCRA Section 313 chemicals are discussed, the discussion includes chemical categories, as appropriate. Chemicals or chemical categories may be added or deleted from the list. Therefore, before completing your annual report, be sure to check the most current list included with the *TRI Forms and Instructions* when evaluating the chemicals and chemical categories present at your facility. Copies of the reporting package can be requested from the EPCRA Hotline, 1-800-424-9346.

All facilities meeting the EPCRA Section 313 reporting criteria must report the annual release and other waste management activity quantities (routine and accidental) of EPCRA Section 313 chemicals and chemical categories to all environmental media. A separate report is required for each EPCRA Section 313 chemical or chemical category that is manufactured (including imported), processed, or otherwise used above the reporting threshold. The reports must be submitted to U.S. EPA and State or Tribal governments, on or before July 1, for activities in the previous calendar year. The owner/operator of the facility on July 1 is primarily responsible for the report, even if the owner/operator did not own the facility during the reporting year. However, property owners with no business interest in the operation of the facility, other than a lessor interest, are exempt from reporting requirements.

EPCRA also mandates U.S. EPA to establish and maintain a publicly available database system consisting of the information reported under Section 313 and under Section 6607 of the PPA. This database, known as the Toxic Chemical Release Inventory (TRI) database, can be accessed through the following sources:

- U.S. EPA Internet site, <http://www.epa.gov/tri>;
- Envirofacts Warehouse Internet site, <http://www.epa.gov/enviro/>; and
- Right-to-Know network, <http://www.rtk.net/trisearch.html>.

However, information qualifying as a trade secret, in accordance with the regulatory requirements, is protected from public release. In addition to being a resource for the public, TRI data are also used in the research and development of regulations related to EPCRA Section 313 chemicals and chemical categories.

To reduce the reporting burden for small businesses, U.S. EPA established an alternate activity threshold of one million pounds manufactured, processed, or otherwise used for facilities with total annual reportable amounts of 500 pounds or less for each EPCRA Section 313 chemical or chemical category. Provided the facility does not exceed either the reportable amount or the alternate threshold, the facility may file a certification form (Form A) rather than file a Form R. By filing the Form A the facility certifies that they do not exceed the reportable amount of 500 pounds or exceed the alternate threshold of one million pounds for the respective chemical or chemical category.

Note that the annual reportable amount includes the quantity of the EPCRA Section 313 chemical or chemical category in all production-related waste management activities, not just releases (see the discussion in Section 2.8 for more detail). Also note that either a Form A or a Form R, but not both, must be submitted for each EPCRA Section 313 chemical or chemical category above any reporting threshold, even if there are zero release and other waste management activity quantities.

Violation of EPCRA Section 313 reporting provisions may result in federal civil penalties of up to \$27,500 per day for each violation (61 FR 69360). State enforcement provisions may also be applicable depending on the state's EPCRA Section 313 reporting regulations.

Members of the Leather Industries of America, Inc. and staff from Garden State Tanning, Inc. provided input on common problems specific to leather tanning and finishing operations encountered by those completing the EPCRA Section 313 reports. U.S. EPA has combined this input with questions forwarded to the EPCRA Hotline and those identified during voluntary site surveys of facilities that have filed EPCRA Section 313 reports in the past.

Selected issues and guidance addressing these common problems are presented throughout this document as applicable.

The *TRI Forms and Instructions* also contain discussions of common problems in completing the EPCRA Section 313 reports. You are encouraged to read this document before filling out the Form R (or Form A) for your facility.

If, after reading this manual, you still have questions about EPCRA Section 313 reporting, please contact the EPCRA Hotline at 1-800-424-9346 or refer to the U.S. EPA's TRI website, <http://www.epa.gov/tri>. Assistance is also available from the designated EPCRA Section 313 Coordinator in the U.S. EPA regional office and the EPCRA contact in your state (see the *TRI Forms and Instructions* for a current list of these contacts). Additional guidance is also available in the resources listed in Appendix A.

## CHAPTER 2 - REPORTING REQUIREMENTS

### 2.0 PURPOSE

The purpose of this chapter is to help you determine if you must prepare an EPCRA Section 313 report(s) and, if so, what kind of a report(s) should be prepared (Form R or the alternate certification statement, the Form A). This chapter presents the EPCRA Section 313 reporting requirements to help you determine if these requirements apply to your facility. It also discusses the reporting of trade secrets and the records that must be kept.

To understand the following discussion you must first understand how EPCRA defines a facility. The term “facility” is defined as, “all buildings, equipment, structures, and other stationary items which are located on a single site or on contiguous or adjacent sites and which are owned or operated by the same person (or by any person who controls, who is controlled by, or who is under common control with such person). A facility may contain more than one “establishment” (40 CFR 372.3). An “establishment” is defined as, “an economic unit, generally at a single physical location, where business is conducted, or where services or industrial operations are performed” (40 CFR 372.3).

U.S. EPA recognizes that for business reasons it may be easier and more appropriate for establishments at one facility to report separately. However, the combined quantities of EPCRA Section 313 chemicals and chemical categories manufactured, processed, or otherwise used in all establishments making up that facility must be considered for threshold determinations. Also, the combined release and other waste management activity quantities reported singly for each establishment must total those for the facility as a whole.

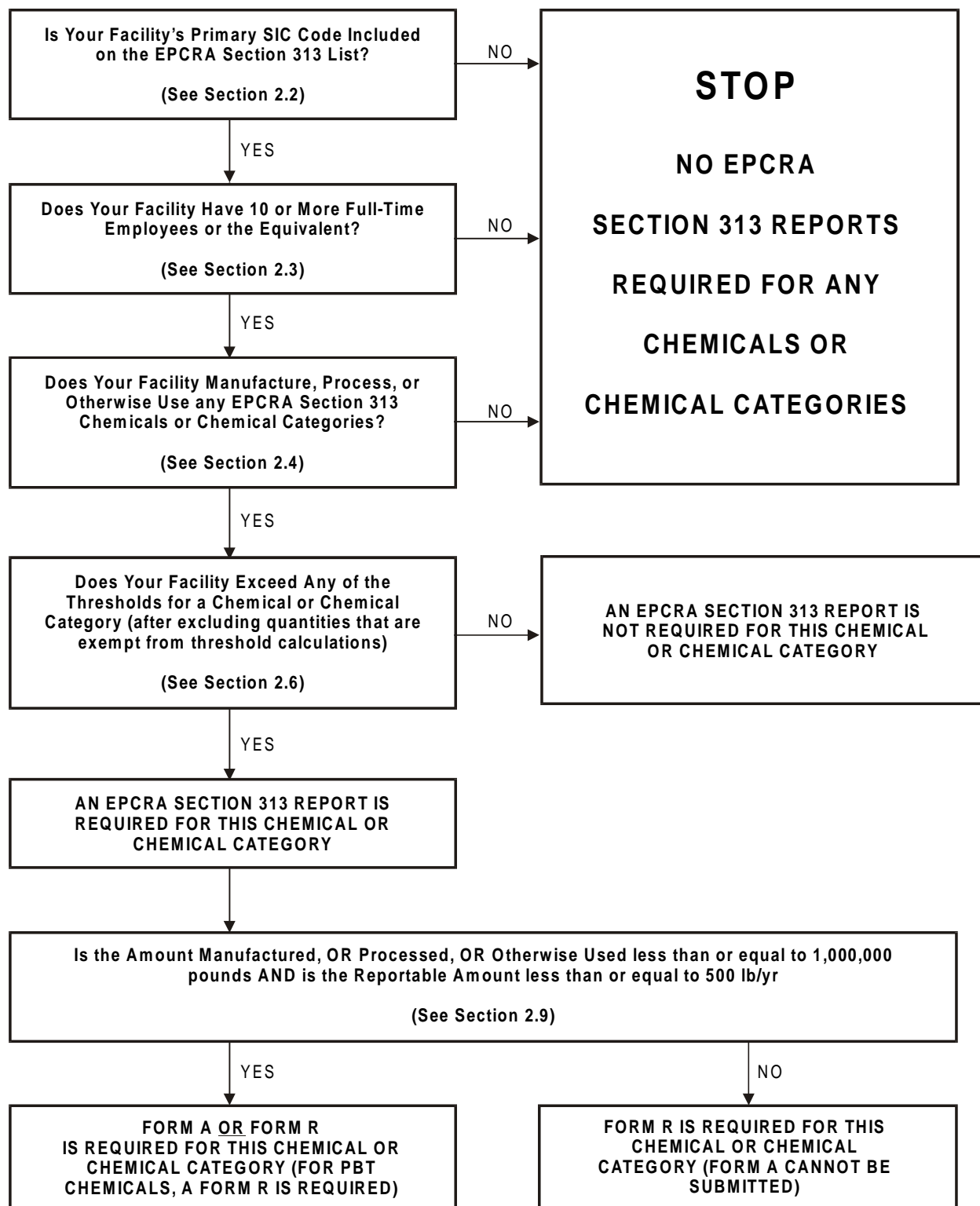
Note that if a facility is comprised of more than one establishment, once an activity threshold is met by the facility, providing the facility meets the SIC Code and employee threshold criteria, release and other waste management activities from all establishments at the facility must be reported.



## 2.1 Must You Report?

How do you determine if your facility must prepare an EPCRA Section 313 report? Your answers to the following four questions will help you decide (illustrated by Figure 2-1):

- 1) Is the primary SIC Code for your facility included in the list covered by EPCRA Section 313 reporting (see Section 2.2)?
- 2) Does your facility have 10 or more full-time employees or the equivalent (see Section 2.3)?
- 3) Does your facility manufacture (which includes importation), process, or otherwise use EPCRA Section 313 chemicals or chemical categories (see Section 2.4)?
- 4) Does your facility exceed any applicable thresholds of EPCRA Section 313 chemicals or chemical categories (for non-PBT chemicals: 25,000 pounds per year for manufacturing; 25,000 pounds per year for processing; or 10,000 pounds per year for otherwise use - see Section 2.5; for PBT chemicals: see Section 2.6 for applicable thresholds)?



**Figure 2-1. EPCRA Section 313 Reporting Decision Diagram**

If you answered “No” to any of the first three questions, you are not required to prepare any EPCRA Section 313 reports. If you answered “Yes” to ALL of the first three questions, you must complete a threshold calculation for each EPCRA Section 313 chemical at the facility, and submit an EPCRA Section 313 report for each chemical and chemical category exceeding the applicable threshold.

**2.2 SIC Code Determination**

Facilities with the SIC Codes presented in Table 2-1 are covered by the EPCRA Section 313 reporting requirements.

**Table 2-1  
SIC Codes Covered by EPCRA Section 313 Reporting**

SIC Codes	Industry	Qualifiers
10	Metal Mining	Except SIC Codes 1011, 1081, and 1094
12	Coal Mining	Except SIC Code 1241
20 through 39	Manufacturing	None
4911, 4931, and 4939	Electric and Other Services and Combination Utilities	Limited to facilities that combust coal and/or oil for the purpose of generating electricity for distribution in commerce
4953	Refuse Systems	Limited to facilities regulated under RCRA Subtitle C
5169	Chemicals and Allied Products	None
5171	Petroleum Bulk Stations and Terminals	None
7389	Business Services	Limited to facilities primarily engaged in solvent recovery services on a contract or fee basis

Leather tanning and finishing facilities are typically classified in SIC Code 3111. You should determine the SIC Code(s) for your facility, based on the activities performed on site. For assistance in determining which SIC Code best suits your facility refer to *Standard Industrial Classification Manual, 1987* published by the Office of Management and Budget.

Note that auxiliary facilities can assume the SIC Code of another covered establishment if the primary function is to support the covered establishment's operations. For the purpose of EPCRA Section 313, auxiliary facilities are defined as those that are primarily engaged in performing support services for another covered establishment or multiple establishments of a covered facility, and are in a different physical location from the primary facility. In addition, auxiliary facilities perform an integral role in the primary facility's activities. In general, the auxiliary facility's basic administrative services (e.g., paperwork, payroll, employment) are performed by the primary facility. If an auxiliary facility's primary function is to support/service a facility with a covered SIC Code, the auxiliary facility assumes the covered SIC Code as its primary SIC Code and must consider the other reporting requirements (40 CFR Section 372.22) to determine if it must comply with the EPCRA Section 313 reporting requirements. However, if the SIC Code for the primary facility is not covered by EPCRA Section 313, then neither the primary nor the auxiliary facility is required to submit a report.

If your facility has more than one SIC Code (i.e., several establishments with different SIC Codes are owned or operated by the same entity and are located at your facility), you are subject to reporting requirements if:

- All the establishments have SIC Codes covered by EPCRA Section 313; OR
- The total value of the products shipped or services provided at establishments with covered SIC Codes is greater than 50% of the value of the entire facility's products and services; OR
- Any one of the establishments with a covered SIC Code ships and/or produces products or provides services whose value exceeds the value of services provided, products produced and/or shipped by every other establishment within the facility.

### **Example - Primary SIC Code**

A facility has two establishments. The first, a general automotive repair service, is in SIC Code 7537. SIC Code 7537 is not a covered SIC Code. However, the second EPCRA Section 313 establishment, a metal products paint shop, is in SIC Code 3479, which is a covered SIC Code. The facility also determines the product is worth \$500/unit as received from the establishment in the non-covered SIC Code and the value of the product is \$1,500/unit after processing by the establishment in the covered SIC Code. The value added by the establishment in the covered SIC Code is more than 50% of the product value; therefore, the primary SIC Code is 3479, a covered SIC Code. Thus, the establishment is covered by EPCRA Section 313 reporting and the entire facility is subject to reporting.

A pilot plant within a covered SIC Code is considered a covered facility and is subject to reporting, provided it meets the employee and activity criteria (note that pilot plants are not eligible for the laboratory exemption, which is discussed in Chapter 3). Warehouses on the same site as facilities in a covered SIC Code are also subject to reporting. Likewise, warehouses that qualify as auxiliary facilities of covered facilities also must report, provided all applicable reporting requirements are met.

While you are currently required to determine your facility's reporting eligibility based on the SIC code system described above, it is important to be aware that the SIC code system will be replaced by a new system in the future. On April 9, 1997 (62 FR 17287), the Office of Management and Budget promulgated the North American Industrial Classification System (NAICS). NAICS is a new economic classification system that replaces the SIC code system as a means of classifying economic activities for economic forecasting and statistical purposes. The transition to the new NAICS may require regulatory actions. As a result, the SIC code system is still required to be used as the mechanism to determine your facility's reporting eligibility. EPA will issue notice in the *Federal Register* to inform you and other EPCRA Section 313 facilities of its plans to adopt the NAICS and how facilities should make their NAICS code determination.

## **2.3 Number of Employees**

If your facility meets SIC Code and activity threshold criteria, you are required to prepare an EPCRA Section 313 report if your facility has 10 or more full-time employees or the equivalent. A full-time employee equivalent is defined as a work year of 2,000 hours. If your

facility's employees hours total 20,000 or more hours in a calendar year, you meet the 10 or more employee threshold criterion.

The following information should be included in your employee calculations:

- Owners;
- Operations/manufacturing staff;
- Clerical staff;
- Temporary employees;
- Sales personnel;
- Truck drivers (employed by the facility);
- Other non-manufacturing or off-site facility employees directly supporting the facility;
- Paid vacation and sick leave; and
- Contractor employees (maintenance, construction, etc. but excluding contracted truck drivers and minor intermittent service vendors (e.g., trash handlers)).

In general, if an individual is employed or hired to work at the facility, all the hours worked by that individual for the facility (including paid leave and overtime) should be counted in determining if the 20,000-hour criterion has been met.

### **Example - Employee Equivalent Calculation**

Your facility has six full-time employees working 2,000 hours/year. You also employ two full-time sales people and a delivery truck driver (employed by the facility) who are assigned to the plant, each working 2,000 hours/year but predominantly on the road or from their homes. The wastewater treatment system (on site and owned by the facility) is operated by a contractor who spends an average of two hours per day and five days per week at the plant. Finally, you built an addition to the plant warehouse during the year, using four contractor personnel who were on site full time for six months (working on average of 1,000 hours each). You would calculate the number of full-time employee equivalents as follows:

- Hours for your nine full-time employees (six plant personnel, two salespeople, and one delivery truck driver) are:  
 $(9 \text{ employees}) \times (2,000 \text{ hours/year}) = 18,000 \text{ hours/year}$
- Hours for the wastewater treatment system operator are:  
 $(2 \text{ hours/day}) \times (5 \text{ days/week}) \times (52 \text{ weeks/year}) = 520 \text{ hours/year};$  and
- Hours for the construction crew are:  
 $(4 \text{ contractors}) \times (1,000 \text{ hours}) = 4,000 \text{ hours/year}.$

Your facility has a total of 22,520 hours for the year, which is above the 20,000 hours/year threshold; therefore, you meet the employee criterion.

## 2.4 Manufacturing, Processing, and Otherwise Use of EPCRA Section 313 Chemicals or Chemical Categories

If you are in a covered SIC Code and have 10 or more full-time employee equivalents, you must determine which EPCRA Section 313 chemicals and chemical categories are manufactured, processed, or otherwise used at your facility. You should prepare a list which includes all chemicals and chemical categories found in mixtures and trade name products at all establishments at the facility. This list should then be compared to the CURRENT list of EPCRA Section 313 chemicals and chemical categories found in the *TRI Forms and Instructions* document for that reporting year (also available from the EPCRA Hotline, 1-800-424-9346). Once you identify the EPCRA Section 313 chemicals and chemical categories at your facility, you must evaluate the activities involving each chemical and chemical category and determine if any activity thresholds have been met.

The original list of chemicals and chemical categories subject to EPCRA Section 313 reporting was a combination of lists from New Jersey and Maryland. Refinements to the list have been made and changes are anticipated to continue. The list can be modified by U.S. EPA initiatives or industry or the public can petition U.S. EPA to modify the list. When evaluating a chemical or chemical category for addition or deletion from the list, U.S. EPA must consider the chemical's potential acute human health effects, chronic human health effects, or its adverse environmental effects. U.S. EPA reviews these petitions and initiates a rulemaking to add or delete the chemical or chemical category from the list, or publishes an explanation why it denied the petition.

Note that chemicals and chemical categories are periodically added, delisted, or modified. Therefore, it is imperative that you refer to the appropriate reporting year's list. You can refer to the U.S. EPA's TRI website, <http://www.epa.gov/tri>, for updated guidance. Also, note that a list of synonyms for EPCRA Section 313 chemicals and chemical categories can be found in the U.S. EPA publication *Common Synonyms for Chemicals Listed Under Section 313 of the Emergency Planning and Community Right-To-Know Act*, (EPA 745-R-95-008). Table 2-2 lists the EPCRA Section 313 chemicals and chemical categories reported for leather tanning

and finishing operations in 1995. This list is not intended to be all inclusive and should only be used as a guide.

**Table 2-2**

**EPCRA Section 313 Chemicals and Chemical Categories Reported by Leather Tanning and Finishing Facilities**

<b>Chemicals</b>	<b>Process</b>
Ammonia (anhydrous and 10% of aqueous)	Wastewater Treatment
Butyl acrylate	Finishing
Butyl alcohol	Finishing
Certain glycol ethers	Finishing
Chlorine	Wastewater Treatment
Chromium	Tanning
Chromium compounds	Tanning
Di (2-ethylhexyl) phthalate	Finishing
Ethylbenzene	Finishing
Formaldehyde	Synthetic Retanning
Formic acid	Coloring
Hydrochloric acid (acid aerosols)	Pickling
Manganese	Wastewater Treatment
Manganese compounds	Wastewater Treatment
Methanol	Finishing
Methyl ethyl ketone	Finishing
Methyl isobutyl ketone	Finishing
Methyl methacrylate	Finishing
N-Methyl-2-pyrrolidone	Finishing
Naphthalene	Synthetic Retanning
Nitrate compounds (only in water and water dissociable)	Wastewater Treatment
Sulfuric acid (acid aerosols)	Pickling
Toluene	Finishing
Triethylamine	Finishing
Xylene (mixed isomers)	Finishing
Zinc compounds	Finishing/Retanning

Ref: 1995 TRI database for SIC Code 3111.



## 2.5 Activity Categories

EPCRA Section 313 defines three activity categories for the listed chemicals and chemical categories: manufacturing (which includes importing), processing, and otherwise use. The activity thresholds are 25,000 pounds per year for manufacturing, 25,000 pounds per year for processing, and 10,000 pounds per year for otherwise use<sup>1</sup>. These thresholds apply to each chemical or chemical category individually. The quantity of chemicals or chemical categories stored on site or purchased is not relevant for threshold determinations. Rather, the determination is based solely on the annual quantity actually manufactured (including imported), processed, or otherwise used. Therefore, EPCRA Section 313 chemicals and chemical categories that are brought on site and stored, but are not incorporated into a product for distribution or are not otherwise used on site during the reporting year, are not considered towards any activity threshold.

Expanded definitions, with examples, of each of the three activities are found in Chapter 3, Tables 3-2, 3-3, and 3-4. The terms are briefly defined in Table 2-3.

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<sup>1</sup>These activity thresholds are for non-PBT chemicals. See Section 2.6 for the activity thresholds applicable to PBT chemicals.

**Table 2-3**

**Activity Categories**

Activity Category	Definition	Threshold <sup>1</sup> (lb/yr)
Manufacture	To produce, prepare, import, or compound an EPCRA Section 313 chemical or chemical category. Manufacture also applies to an EPCRA Section 313 chemical or chemical category that is produced coincidentally during the manufacture, processing, otherwise use, or disposal of another chemical or mixture of chemicals as a byproduct, and an EPCRA Section 313 chemical or chemical category that remains in that other chemical or mixture of chemicals as an impurity during the manufacturing, processing, or otherwise use or disposal of any other chemical substance or mixture. An example would be the production of ammonia or nitrate compounds in a wastewater treatment system.	25,000
Process	To prepare an EPCRA Section 313 chemical or chemical category, or a mixture or trade name product containing an EPCRA Section 313 chemical or chemical category, for distribution in commerce. For example, the addition of EPCRA Section 313 listed pigments to paint should be reported if you exceeded the reporting threshold. Processing includes the preparation for sale to your customers (and transferring between facilities within your company) of a chemical or formulation that you manufacture. For example, if you manufacture an EPCRA Section 313 chemical or chemical category or product, package it, and then distribute it into commerce, this chemical has been manufactured AND processed by your facility.	25,000
Otherwise Use	<p>Generally, use of an EPCRA Section 313 chemical or chemical category that does not fall under the manufacture or process definitions is classified as otherwise use. An EPCRA Section 313 chemical or chemical category that is otherwise used does not function by being incorporated into a product that is distributed in commerce, but may be used instead as a manufacturing or processing aid (e.g., catalyst), in waste processing, or as a fuel (including waste fuel). For example, xylene used as a carrier solvent for paint is classified as otherwise used.</p> <p>On May 1, 1997 U.S. EPA revised the interpretation of otherwise use. The following new otherwise use definition became effective with the 1998 reporting year (62 FR 23834, May 1, 1997):</p> <p>Otherwise use means “any use of a toxic chemical contained in a mixture or other trade name product or waste that is not covered by the terms manufacture or process. Otherwise use of a toxic chemical does not include disposal, stabilization (without subsequent distribution in commerce), or treatment for destruction unless:</p> <ol style="list-style-type: none"> <li>1) The toxic chemical that was disposed, stabilized, or treated for destruction was received from off site for the purposes of further waste management; OR</li> <li>2) The toxic chemical that was disposed, stabilized, or treated for destruction was manufactured as a result of waste management activities on materials received from off site for the purposes of further waste management activities.”</li> </ol>	10,000

<sup>1</sup>These activity thresholds are for non-PBT chemicals. See Section 2.6 for the activity thresholds applicable to PBT chemicals.

### **COMMON ERROR - Coincidental Manufacture**

The coincidental manufacture of an EPCRA Section 313 chemical or chemical category is frequently overlooked. For example, in the treatment of wastewater, ammonia and nitrate compounds may be coincidentally manufactured. Wastewater from leather tanning and finishing facilities typically contains large quantities of proteinaceous materials, which often cause coincidental manufacture of ammonia and nitrate compounds. The amounts of these compounds created in the wastewater treatment process must be estimated and included in the manufacturing threshold determination (see Appendices C and D, respectively, for EPCRA Section 313 reporting guidance for nitrate compounds and ammonia).

Relabeling or redistribution of an EPCRA Section 313 chemical or chemical category where no repackaging occurs does not constitute manufacturing, processing, or otherwise use of that chemical. This type of activity should not be included in threshold determinations.

### **Example - Relabeling**

You buy a mixture in small containers that contains an EPCRA Section 313 chemical or chemical category. When it arrives you put your own label on each container and put the containers in a larger box with several other items you manufacture, and sell the larger box as a kit. The quantity of the EPCRA Section 313 chemical or chemical category in the small containers should not be counted toward the processing threshold (because you did not repackage the chemical) or the otherwise use threshold, nor should it be counted toward the manufacturing activity threshold unless the small containers were imported. However, you must consider other EPCRA Section 313 chemicals and chemical categories that you manufactured in the kit toward manufacturing and processing threshold determinations.

### **Example - Treatment of Wastes from Off Site**

A covered facility receives a waste containing 12,000 pounds of Chemical A, an EPCRA Section 313 chemical, from off site. The facility treats the waste, destroying Chemical A and in the treatment process manufactures 10,500 pounds of Chemical B, another EPCRA Section 313 chemical. Chemical B is disposed on site.

Since the waste was received from off site for the purpose of waste management, the amount of Chemical A must be included in the otherwise use threshold determination for Chemical A. The otherwise use threshold is 10,000 pounds and since the amount of Chemical A exceeds this threshold, all release and other waste management activities for Chemical A must be reported.

Chemical B was manufactured in the treatment of a waste received from off site. The quantity of chemical B should be counted towards the manufacturing threshold. However, the facility disposed of Chemical B on site and waste received from off site for treatment for destruction, disposal, or stabilization is considered to be otherwise used. Therefore, the amount of Chemical B must also be considered in the otherwise use threshold determination. Thus, the reporting threshold for Chemical B has also been exceeded and all release and other waste management activities for Chemical B must be reported.

Also, note that the threshold determinations for the three activity categories (manufacturing, processing, and otherwise use) are mutually exclusive. That is, you must conduct a separate threshold determination for each activity category and if you exceed any threshold, all release and other waste management activities of that EPCRA Section 313 chemical or chemical category at the facility must be considered for reporting.

## **2.6 Persistent, Bioaccumulative, and Toxic (PBT) Chemicals**

U.S. EPA promulgated the final rule for Persistent, Bioaccumulative, and Toxic (PBT) chemicals in the October 29, 1999 Federal Register (64 FR 209). This rule applies for the reporting year beginning January 1, 2000 (for EPCRA Section 313 reports that must be filed by July 1, 2001).

In this rule, U.S. EPA has added seven chemicals and lowered the reporting thresholds for 18 chemicals and chemical categories that meet the EPCRA Section 313 criteria for persistence and bioaccumulation. The PBT chemicals and their thresholds are listed in Table 2-4.

**Table 2-4**  
**Reporting Thresholds for EPCRA Section 313 Listed PBT Chemicals**

<b>Chemical Name or Chemical Category</b>	<b>CASRN</b>	<b>Section 313 Reporting Threshold (in pounds unless noted other-wise)</b>
Aldrin	309-00-2	100
Benzo(g,h,i)perylene	191-24-2	10
Chlordane	57-74-9	10
Dioxin and dioxin-like compounds category (manufacturing; and the processing or otherwise use of dioxin and dioxin-like compounds if the dioxin and dioxin-like compounds are present as contaminants in a chemical and if they were created during the manufacturing of that chemical)	NA	0.1 grams
Heptachlor	76-44-8	10
Hexachlorobenzene	118-74-1	10
Isodrin	465-73-6	10
Methoxychlor	72-43-5	100

**Table 2-4 (Continued)**

<b>Chemical Name or Chemical Category</b>	<b>CASRN</b>	<b>Section 313 Reporting Threshold (in pounds unless noted other-wise)</b>
Octachlorostyrene	29082-74-4	10
Pendimethalin	40487-42-1	100
Pentachlorobenzene	608-93-5	10
Polycyclic aromatic compounds category	NA	100
Polychlorinated biphenyl (PCBs)	1336-36-3	10
Tetrabromobisphenol A	79-94-7	100
Toxaphene	8001-35-2	10
Trifluralin	1582-09-8	100
Mercury	7439-97-6	10
Mercury compounds	NA	10

U.S. EPA also added two chemicals to the polycyclic aromatic compounds (PACs) category that is listed above:

- Benzo(j,k)fluorene (fluoranthene)
- 3-methylchloanthrene

These two chemicals are not to be reported individually; rather, they should be included within the PACs compound category.

U.S. EPA finalized two thresholds based on the chemicals' potential to persist and bioaccumulate in the environment. The two levels include setting Section 313 manufacture, process, and otherwise use thresholds to 100 pounds for PBT chemicals and to 10 pounds for that subset of PBT chemicals that are highly persistent and highly bioaccumulative. One exception is the dioxin and dioxin-like compounds category. EPA set the threshold for the dioxin and dioxin-like compound category at 0.1 gram.

U.S. EPA eliminated the *de minimis* exemption for the PBT chemicals. However, this action does not affect the applicability of the *de minimis* exemption to the supplier notification requirements (40 CFR 372.45(d)(1)). U.S. EPA also excluded all PBT chemicals

from eligibility for the alternate threshold of 1 million pounds (see Section 2.9) and eliminated range reporting of PBT chemicals and chemical categories for on-site releases and transfers off-site for further waste management.

Concurrent with the additions and lowered thresholds discussed above, U.S. EPA added “vanadium, except when contained in an alloy” and “vanadium compounds” to the list of toxic chemicals subject to reporting under EPCRA Section 313. The corresponding thresholds for vanadium and vanadium compounds remain 10,000 pounds if otherwise used, 25,000 pounds if processed, and 25,000 pounds if manufactured. Please refer to the discussion on “Qualifiers” in Section 3.1 if vanadium is a concern at your facility.

Note that U.S. EPA is currently developing five guidance documents for chemicals modified in the PBT rule:

- Dioxins and dioxin-like compounds;
- Mercury and mercury compounds;
- Vanadium and vanadium compounds;
- Polycyclic aromatic compounds (PACs) category; and
- Other PBT chemicals.

Please refer to this guidance if applicable to your facility.

## **2.7            How Do You Report?**

You must submit an EPCRA Section 313 report for each EPCRA Section 313 chemical or chemical category that exceeds a threshold for manufacturing, OR processing, OR otherwise use (providing you meet the employee and SIC Code criteria). Provided you do not exceed certain alternate activity thresholds and total annual reportable amounts, you may prepare a Form A (See Section 2.9) rather than a Form R. The *TRI Forms and Instructions* contain detailed directions for the preparation and submittal of EPCRA Section 313 reports for the reporting year. The *TRI Forms and Instructions* are sent to all facilities that submitted EPCRA Section 313 reports the preceding year. However, if you do not receive a courtesy copy, you may request copies of the *TRI Forms and Instructions* from the EPCRA Hotline (1-800-424-9346).

## 2.8 Form R

Form R is the report in which the information required by EPCRA Section 313 is reported. If you are submitting a Form R, it is essential that you use the *TRI Forms and Instructions* for the appropriate reporting year. U.S. EPA encourages the electronic submittal of the Form R, via the Automated Toxic Chemical Release Inventory Reporting Software (ATRS). Use of the ATRS will save preparation time in data entry and photocopying and reduce errors via on-line validation routines and use of pick lists. The ATRS can be found on the Internet at:

- <http://www.epa.gov/atrs>

The ATRS is available in both DOS and Windows versions. More information can be found in the *TRI Forms and Instructions* and by calling the ATRS User Support Hotline at (703) 816-4434.

The Form R consists of two parts:

Part I, Facility Identification Information. This part may be photocopied and re-used for each Form R you submit, except for the signature, which must be original for each submission.

Part II, Chemical Specific Information. You must complete this part separately for each EPCRA Section 313 chemical or chemical category; it cannot be reused year to year even if reporting has not changed.

Submission of incomplete EPCRA Section 313 reports may result in issuance of a Notice of Technical Error (NOTE), Notice of Significant Error (NOSE), or Notice of Non-compliance (NON). See the current *TRI Forms and Instructions* for more detailed information on completing the Form R and submitting the EPCRA Section 313 report.

## 2.9 Alternate Threshold and Form A

U.S. EPA developed the Form A, also referred to as the “Certification Statement,” to reduce the annual reporting burden for facilities with minimal amounts of EPCRA Section 313 chemicals or chemical categories released and otherwise managed as waste (59 FR 61488, November 1994; applicable beginning reporting year 1994 and beyond). On Form A you certify that you are not required to report the release and other waste management information required by EPCRA Section 313 and PPA Section 6607. A facility must meet the following two criteria to use a Form A:

- First, the total annual reportable amount of the EPCRA Section 313 chemical or chemical category cannot exceed 500 pounds per year. The “reportable amount” is defined as the sum of the on-site amounts released (including disposal), treated, combusted for energy recovery, and recycled, combined with the sum of the amounts transferred off site for recycling, energy recovery, treatment, and/or release (including disposal). This total corresponds to the total of data elements 8.1 through 8.7 on the 1999 version of the Form R.
- Second, the amount of the EPCRA Section 313 chemical or chemical category manufactured, processed, OR otherwise used cannot exceed one million pounds. It is important to note that the quantities for each activity are mutually exclusive and must be evaluated independently. If the quantity for any one of the activities exceeds 1,000,000 pounds a Form A cannot be used.

### **Example - Form A Threshold**

If the combined annual reportable amounts from all activities do not exceed 500 pounds, a facility that manufactures 900,000 pounds of an EPCRA Section 313 chemical or chemical category and processes 150,000 pounds of the same chemical or chemical category is eligible to use the Form A because the facility did not exceed the one million pounds for either activity, even though the total usage exceeds one million pounds.

The Form A Certification Statement must be submitted for each eligible EPCRA Section 313 chemical or chemical category. The information on the Form A will be included in the publicly accessible TRI database; however, these data are marked to indicate that they represent certification statements rather than Form Rs. Note that separate establishments at a facility cannot submit separate Form As for the same chemical or chemical category; rather, only



one Form A per EPCRA Section 313 chemical or chemical category can be submitted per facility.

While Form A requests facility identification and chemical identification information, no release and other waste management quantity estimations to any media are required. You must simply certify that the total annual reportable amount did not exceed 500 pounds and that amounts manufactured, processed, or otherwise used did not exceed one million pounds. Once the facility has completed estimates to justify the submission of a Form A, there is a considerable time savings in using the Form A, especially in subsequent years, providing activities involving the chemical or chemical category did not change significantly. It is strongly recommended that you document your initial rationale and refer to it every year, to verify that you have not modified a part of the process that would invalidate the initial rationale supporting submission of Form A.

## **2.10            Trade Secrets**

If you submit trade secret information, you must prepare two versions of the substantiation form as prescribed in 40 CFR Part 350 (see 53 FR 28801, July 29, 1988) as well as two versions of the EPCRA Section 313 report. One set of reports should be “sanitized” (i.e., it should provide a generic name for the EPCRA Section 313 chemical or chemical category identity). This version will be made available to the public. The second version, the “unsanitized” version, should provide the actual identity of the EPCRA Section 313 chemical or chemical category and have the trade secret claim clearly marked in Part I, Section 2.1 of the Form R or Form A. The trade secrets provision only applies to the EPCRA Section 313 chemical or chemical category identity. All other parts of the Form R or Form A must be filled out accordingly.

Individual states may have additional criteria for confidential business information and the submittal of both sanitized and unsanitized reports for EPCRA Section 313 chemicals and chemical categories. Facilities may jeopardize the trade secret status of an EPCRA Section 313 chemical or chemical category by submitting an unsanitized version to a state agency or tribal government that does not require an unsanitized version.

More information on trade secret claims, including contacts for individual state's submission requirements, can be found in the *TRI Forms and Instructions*.

## **2.11**            **Recordkeeping**

Complete and accurate records are absolutely essential to meaningful compliance with EPCRA Section 313 reporting requirements. Compiling and maintaining good records will help you to reduce the effort and cost in preparing future reports, and to document how you arrived at the reported data in the event of U.S. EPA compliance audits. U.S. EPA requires you to maintain records substantiating each EPCRA Section 313 report submission for a minimum of three years. Each facility must keep copies of every EPCRA Section 313 report along with all supporting documents, calculations, work sheets, and other forms that you use to prepare the EPCRA Section 313 report. U.S. EPA may request this supporting documentation during a regulatory audit.

Specifically, U.S. EPA requires the following records be maintained for a period of three years from the date of the submission of a report (summarized from 40 CFR 372.10):

- 1) A copy of each EPCRA Section 313 report that is submitted.
- 2) All supporting materials and documentation used to make the compliance determination that the facility or establishment is a covered facility.
- 3) Documentation supporting the report submitted, including:
  - Claimed allowable exemptions,
  - Threshold determinations,
  - Calculations for each quantity reported as being released, either on or off site, or otherwise managed as waste,
  - Activity determinations, including dates of manufacturing, processing, or use,
  - The basis of all estimates,
  - Receipts or manifests associated with transfers of each EPCRA Section 313 chemical or chemical category in waste to off-site locations, and
  - Waste treatment methods, treatment efficiencies, ranges of influent concentrations to treatment, sequential nature of treatment steps, and operating data to support efficiency claims.

- 4) For facilities submitting a Form A, all supporting materials used to make the compliance determination the facility or establishment is eligible to submit a Form A, including:
- Data supporting the determination the alternate threshold applies,
  - Calculations of the annual reportable amounts,
  - Receipts or manifests associated with the transfer of each EPCRA Section 313 chemical or chemical category in waste to off-site locations, and
  - Waste treatment methods, treatment efficiencies, ranges of influent concentrations to treatment, sequential nature of treatment steps, and operating data to support efficiency claims.

Because EPCRA Section 313 reporting does not require additional testing or monitoring you must determine the best readily available source of information for all estimates. Some facilities may have detailed monitoring data and off-site transfer records that can be used for estimates while others may only have purchase and inventory records. Examples of records that you should keep, if applicable, might include:

- Each EPCRA Section 313 report submitted;
- EPCRA Section 313 Reporting Threshold Worksheets (sample worksheets can be found in Chapter 3 of this document as well as in the *TRI Forms and Instructions*);
- EPCRA Section 313 Reporting Release and Other Waste Management Quantity Estimation Worksheets (sample worksheets can be found in Chapter 4 of this document);
- Engineering calculations and other notes;
- Formulation sheets;
- Purchase records from suppliers;
- Inventory data;
- Material Safety Data Sheets (MSDSs);
- National Pollutant Discharge Elimination System (NPDES)/State Pollutant Discharge Elimination System (SPDES) permits and monitoring reports;
- New Source Performance Standards (NSPS);
- EPCRA Section 312, Tier II reports;
- Monitoring records;
- Air permits;
- Flow measurement data;
- Resource Conservation Recovery Act (RCRA) hazardous waste generator's reports;
- Pretreatment reports filed with local governments;
- Invoices from waste management firms;
- Manufacturer's estimates of treatment efficiencies;

- Comprehensive Environmental Response, Conservation, and Liability Act of 1980 (CERCLA) Reportable Quantity (RQ) reports;
- RCRA manifests; and
- Process flow diagrams (including emissions, releases, and other waste management activities).

## CHAPTER 3 - EPCRA SECTION 313 CHEMICAL OR CHEMICAL CATEGORY ACTIVITY THRESHOLD DETERMINATIONS

### 3.0 PURPOSE

This chapter provides a step-by-step procedure for determining if any EPCRA Section 313 chemicals or chemical categories exceed a reporting threshold. Threshold determinations are essentially a three step process:

- Step 1)* Identify any EPCRA Section 313 chemicals and chemical categories you manufacture/import, process, or otherwise use.
- Step 2)* Identify the activity category and any exempt activities for each EPCRA Section 313 chemical or chemical category.
- Step 3)* Calculate the quantity of each EPCRA Section 313 chemical or chemical category and determine which ones exceed an activity threshold.

### 3.1 Step 1 - Identify Which EPCRA Section 313 Chemicals or Chemical Categories are Manufactured (Including Imported), Processed, or Otherwise Used

Compile lists of all chemicals, chemical categories, compounds, and mixtures at your facility. For facilities with many different chemicals and mixtures it is often helpful to prepare two lists: one with the pure (single ingredient) chemicals (including chemical compounds) and one with the mixtures and trade name products. On the second list, under the name of each mixture/trade name product, write the names of all chemicals in that product. Next, compare the chemicals and chemical categories on both lists to the current EPCRA Section 313 chemicals and chemical categories list found in the *TRI Forms and Instructions* (remember that chemicals and chemical categories may be periodically added and deleted and you should use the current reporting year's instructions). Highlight the EPCRA Section 313 chemicals and chemical categories that are on your lists.

Review the lists to be sure each chemical and chemical category is shown by its correct EPCRA Section 313 name. For example, a common EPCRA Section 313 chemical found in leather tanning and finishing operations is formic acid. Formic acid (Chemical Abstracts Service (CAS) Registry No. 64-18-6) has several synonyms including aminic acid, formic acid, methanoic acid, and hydrogen carboxylic acid. It must be reported on Form R (or Form A), Item 1.2, by its EPCRA Section 313 chemical name, formic acid. Synonyms can be found in the U.S. EPA document *Common Synonyms for Chemicals Listed Under Section 313 of the Emergency Planning and Community Right-to-Know Act* (EPA 745-R-95-008).

While you must consider every chemical and chemical category on the EPCRA Section 313 chemical and chemical category list, you should be aware of the chemicals and chemical categories typically used in leather tanning and finishing operations. As a guide, the EPCRA Section 313 chemicals and chemical categories reported by leather tanning and finishing facilities, and the processes they are typically used in, for reporting year 1995 are listed in Table 2-2.

#### **COMMON ERROR - Glycol Ether Reporting**

You should be sure to check the EPCRA Section 313 definition of glycol ethers. Some leather tanning facilities incorrectly report chemicals in this category.

One of the most commonly reported chemical categories by leather tanning facilities, SIC Code 3111, in 1995 was Certain Glycol Ethers. However, based on information from the Leather Industries of America, many of the commonly used glycol ethers in the leather tanning industry do not fit the EPCRA Section 313 definition of the Certain Glycol Ethers category. This high level of reporting may be the result of assuming that any glycol ether must be reported.

Note that for EPCRA Section 313 purposes, the members of the certain glycol ethers category are glycol ethers derived from ethylene glycol, diethylene glycol, and triethylene glycol. This category does not contain glycol ethers based on propylene glycol, dipropylene glycol, or tripropylene glycol.

Also note that there are two chemicals, 2-Methoxyethanol (CAS Registry No. 109-86-4) and 2-Ethoxyethanol (CAS Registry No. 110-80-5) that are on the individual chemical list and the CAS number list (40 CFR 372.65(a) and (b)). Threshold determinations should be made for each of these chemicals individually and separately from the Certain Glycol Ethers category. See Appendix F for clarification on glycol ethers.

A computerized spreadsheet may be helpful in developing your facility's chemical and chemical category list and performing threshold calculations. The spreadsheet could show the chemical, chemical category or chemical mixture with corresponding component

concentrations; the yearly quantity manufactured, processed, or otherwise used; and the CAS Registry number. The spreadsheet could also be designed to identify the total quantity by activity category (amounts manufactured, processed, and otherwise used) for each EPCRA Section 313 chemical or chemical category in every mixture, compound, and trade name product.

An initial investment of time will be required to develop this spreadsheet; however, the time and effort saved in threshold calculations in subsequent years will be significant. Such a system will also reduce the potential of inadvertently overlooking EPCRA Section 313 chemicals or chemical categories present in mixtures purchased from off-site sources.

To develop the chemical and chemical category list and the associated activity categories you may want to consult the following:

- Material Safety Data Sheets (MSDSs);
- Facility purchasing records;
- New Source Performance Standards (NSPS);
- Inventory records;
- Air and water discharge permits;
- Individual manufacturing/operating functions; and
- Receipts or manifests associated with the transfer of each EPCRA Section 313 chemical and chemical category in waste to off-site locations.

The following is suggested useful information needed to prepare your EPCRA Section 313 reports and should be included for each chemical and chemical category on your spreadsheet:

- The mixture name and associated EPCRA Section 313 chemical and chemical category names;
- The associated Chemical Abstract Service (CAS) Registry numbers;
- The trade name for mixtures and compounds;
- The throughput quantities; and
- Whether the chemical or chemical category is manufactured, processed, or otherwise used at the facility (be sure to include quantities that are coincidentally manufactured and imported, as appropriate).

MSDSs provide important information for the type and composition of chemicals and chemical categories in mixtures, and for determining whether you have purchased raw materials that contain EPCRA Section 313 chemicals and chemical categories. As of 1989, chemical suppliers to facilities in SIC Major Group Codes 20 through 39 are required to notify manufacturing customers of any EPCRA Section 313 chemicals and chemical categories present in mixtures or trade name products distributed to facilities. The notice must be provided to the receiving facility and may be attached or incorporated into that product's MSDS. If no MSDS is required, the notification must be in a letter that accompanies the first shipment of the product to your facility. This letter must contain the chemical name, CAS Registry number, and the weight or volume percent (or a range) of the EPCRA Section 313 chemical or chemical category in mixtures or trade name products.

Carefully review the entire MSDS. Although new MSDSs must list whether EPCRA Section 313 chemicals and chemical categories are present, the language and location of this notification is not currently standardized. Depending on the supplier, this information could be found in different sections of the MSDS. The most likely sections of an MSDS to provide information on EPCRA Section 313 chemicals and chemical categories are:

- Physical properties/chemical composition section;
- Regulatory section;
- Hazardous components section;
- Labeling section; and
- Additional information section.

Also, many EPCRA Section 313 chemicals or chemical categories are present as impurities in mixtures. These quantities must also be considered in threshold determinations unless the concentration is below the *de minimis* value (see Section 3.2.2.1).

#### **COMMON ERROR - Mixture Components**

Facilities often overlook EPCRA Section 313 chemicals and chemical categories that are present in small quantities of bulk solutions. For example, a common chemical used in leather tanning and finishing is xylene. Xylene is often purchased in large quantities for use as a solvent, among other things. Most facilities correctly report for xylene; however, ethylbenzene is typically present at up to 15% in solutions of xylene commercially available. Many facilities have historically overlooked the ethylbenzene in their bulk xylene purchases.



## Qualifiers

Several chemicals on the EPCRA Section 313 chemical and chemical category list include qualifiers related to use or form. Some chemicals are reportable ONLY if manufactured by a specified process or classified in a specified activity category. For example, isopropyl alcohol is only reportable if it is manufactured using the strong acid process and saccharin is reportable only if it is manufactured. Some other chemicals are only reportable if present in certain forms. For example, only yellow or white phosphorus is reportable, while black or red phosphorus is not reportable.

The qualifiers and associated chemicals and chemical categories are presented below. Please make special note of the discussion pertaining to vanadium and vanadium compounds.

- **Aluminum oxide (fibrous)** - Aluminum oxide is only subject to threshold determination and release and other waste management calculations when it is handled in fibrous forms. U.S. EPA has characterized fibrous aluminum oxide for purposes of EPCRA Section 313 reporting as a man-made fiber commonly used in high-temperature insulation applications such as furnace linings, filtration, gaskets, joints, and seals.
- **Ammonia** - (includes anhydrous ammonia and aqueous ammonia from water dissociable ammonium salts and other sources) On June 26, 1995, U.S. EPA qualified the listing for ammonia (CAS Registry No. 7664-41-7) and deleted ammonium sulfate (solution) (CAS Registry No. 7783-20-2) from the EPCRA Section 313 chemical list. Both the qualification and the deletion were effective as of reporting year 1994. The qualifier for ammonia means that anhydrous forms of ammonia are 100% reportable while only 10% of the total aqueous ammonia is reportable. Any evaporation of ammonia from aqueous ammonia solutions is considered anhydrous ammonia. This qualifier applies to both activity threshold determinations and release and other waste management calculations. Note that while ammonium sulfate is no longer an EPCRA Section 313 chemical, 10% of the aqueous ammonia formed from the dissociation of ammonium sulfate (and all other ammonium salts) is reportable, and must be included in both activity threshold determinations and release and other waste management calculations. Additionally, any ammonium nitrate must also be included in the threshold determination and the nitrate portion included in the release and other waste management calculations, for the nitrate compounds category. U.S. EPA has published guidance on

reporting for ammonia and ammonium salts in *Emergency Planning and Community Right-to-Know, EPCRA Section 313, Guidance for Reporting Aqueous Ammonia*, EPA 745-R-95-012 (see Appendix D).

- **Asbestos (friable)** - Asbestos only needs to be considered when it is handled in the friable form. Friable refers to the physical characteristics of being able to crumble, pulverize, or reduce to a powder with hand pressure.
- **Fume or dust** - Two metals (aluminum and zinc) are qualified with “fume or dust.” This definition excludes “wet” forms such as solutions or slurries, but includes powder, particulate, or gaseous forms of these metals. There is no particle size limitation for particulates. For example, use of zinc metal as a paint component is not subject to reporting unless the zinc is in the form of a fume or dust. However, even though elemental zinc is reportable only in the fume or dust form, all forms of zinc compounds are reportable. Note that the entire weight of all zinc compounds should be included in the threshold determination for zinc compounds, while only the metal portion of metal compounds is reported in the release and other waste management amounts. Prior to reporting year 2000, vanadium was also qualified with “fume or dust.” As of reporting year 2000 this qualifier has been removed for vanadium such that all physical forms are now reportable unless the vanadium is contained in an alloy. Please see the discussion on vanadium and vanadium compounds below, if applicable.
- **Hydrochloric acid (acid aerosols)** - On July 25, 1996, U.S. EPA promulgated a final rule delisting non-aerosol forms of hydrochloric acid (CAS Registry No. 7647-01-0) from the EPCRA Section 313 chemical list (effective for the 1995 reporting year). Therefore, threshold determinations and release and other waste management estimates now apply only to the aerosol forms. Under EPCRA Section 313, the term aerosol covers any generation of airborne acid (including mists, vapors, gas, or fog) without any particle size limitation. Therefore, any process that sprays hydrochloric acid “manufactures” hydrochloric acid aerosol and should include this quantity in the manufacturing threshold determination.
- **Manufacturing qualifiers** - Two chemicals, saccharin and isopropyl alcohol, contain qualifiers relating to manufacture. The qualifier for saccharin means that only manufacturers of the chemical are subject to the reporting requirement. The qualifier for isopropyl alcohol means that only facilities that manufacture the chemical by the strong acid process are required to report. Facilities that only process or otherwise use these chemicals are not required to report. Thus, a facility that uses isopropyl alcohol as a solvent should not report for isopropyl alcohol.

- **Nitrate Compounds (water dissociable; reportable only in aqueous solution)** - A nitrate compound is covered by this listing only when in water and if water dissociable. Although the complete weight of the nitrate compound must be used for threshold determinations for the nitrate compounds category, only the nitrate portion of the compound must be considered for release and other waste management calculations. One issue recently raised by industry is how to report nitrate compounds in wastewater and sludge that is applied to farms as a nitrogen source (either on site or off site). Although during such use nitrate compounds may be taken up by plants and cycled back into the ecosystem, U.S. EPA considers that the nitrate compounds in wastewater/sludge to be managed as waste. In this scenario, nitrate compounds should be reported as being disposed to land (either on site or off site as appropriate). U.S. EPA has published guidance for these chemicals in *List of Toxic Chemicals Within the Water Dissociable Nitrate Compounds Category and Guidance for Reporting*, EPA 745-R-96-004 (see Appendix C).
- **Phosphorus (yellow or white)** - Only manufacturing, processing, or otherwise use of phosphorus in the yellow or white chemical forms require reporting. Black and red phosphorus are not subject to EPCRA Section 313 reporting.
- **Sulfuric acid (acid aerosols)** - On June 26, 1995, U.S. EPA promulgated a final rule delisting non-aerosol forms of sulfuric acid (CAS Registry No. 7664-93-9) from the EPCRA Section 313 toxic chemical list (effective for the 1994 reporting year). Therefore, threshold determinations and release and other waste management estimates now apply only to the aerosol forms. Under EPCRA Section 313, the term aerosol covers any generation of airborne acid (including mists, vapors, gas, or fog) without any particle size limitation. Therefore, any process that sprays sulfuric acid “manufactures” sulfuric acid aerosol and should include this quantity in the manufacturing threshold determination. U.S. EPA has published guidance for acid aerosols in *Guidance for Reporting Sulfuric Acid*, EPA 745-R-97-007.
- **Vanadium and vanadium compounds** - Note that prior to reporting year 2000 (effective December 31, 1999 for EPCRA Section 313 reports that must be filed by July 1, 2001), the fume or dust qualifier also applied to vanadium. As of December 31, 1999, U.S. EPA removed this qualifier for vanadium for reporting year 2000 and beyond. Concurrently, U.S. EPA exempted all physical forms of metallic vanadium that are present in alloys. Therefore, vanadium that is present in any physical form of alloys should not be considered for EPCRA Section 313 reporting. However, if vanadium is separated from the alloy, all physical forms of the vanadium are considered to be manufactured and the quantity manufactured should be applied to the 25,000-pound manufacturing threshold. If the vanadium is subsequently processed or otherwise used, the applicable quantity

should also be applied to the processing or otherwise use threshold(s). If a threshold is exceeded, all quantities released or otherwise managed as waste must be reported as appropriate.

Concurrent with this rulemaking, U.S. EPA also added vanadium compounds to the list of toxic chemicals subject to reporting under EPCRA Section 313. U.S. EPA specifically excluded vanadium compounds from the fume or dust qualifier and from the alloy exemption. Therefore, all physical forms of vanadium compounds must be included in threshold determinations and release and other waste management activities estimates.

**3.2 Step 2 - Identify the Activity Category and Any Exempt Activities for Each EPCRA Section 313 Chemical and Chemical Category**

The next step is to identify the activity category (or categories) and any exempt activities for each EPCRA Section 313 chemical and chemical category on your list. Table 3-1 lists the reporting thresholds for each of these activity categories (Tables 3-2 through 3-4 provide detailed definitions of subcategories for each activity category). Each threshold must be individually calculated; they are mutually exclusive and are not additive.

**Table 3-1**

**Reporting Thresholds**

<b>Activity Category</b>	<b>Threshold<sup>1</sup></b>
Manufacture (including import)	25,000 pounds per year
Process	25,000 pounds per year
Otherwise use	10,000 pounds per year

<sup>1</sup>These reporting thresholds are for non-PBT chemicals. See Section 2.6 for the activity thresholds applicable to PBT chemicals.

### **Example - Threshold Determination**

If your facility manufactures 22,000 pounds of an EPCRA Section 313 chemical or chemical category and you also otherwise use 8,000 pounds of the same chemical or chemical category, you have not exceeded either threshold, and an EPCRA Section 313 report for that chemical or chemical category is not required. However, if your facility manufactures 28,000 pounds per year of an EPCRA Section 313 chemical or chemical category and otherwise uses 8,000 pounds of the same chemical or chemical category, you have exceeded the manufacturing threshold and ALL release and other waste management quantities (except those specifically exempted) of that chemical or chemical category must be reported on the Form R, including those from the otherwise use activity.

### **Example - Xylene Isomers**

Leather tanning and finishing operations use the EPCRA Section 313 chemical xylene with the mixed isomers, CAS Registry No. 1330-20-7, being the most frequently reported type. Ortho-, meta-, and para-xylenes are listed on the EPCRA Section 313 chemicals and chemical categories list in addition to xylene (mixed isomers). The mixed isomers classification should be used when a mixture contains any combination of two or three of the isomers. The threshold determination for xylene should be calculated for each isomeric form individually unless the xylenes are manufactured, processed, or otherwise used as a mixture of xylene isomers. For example, a covered facility annually uses 8,000 pounds of para-xylene, 6,000 pounds of ortho-xylene, and 8,000 pounds of mixed isomers as carrier solvents in three separate processing lines. All three activities of xylene are classified as otherwise use as the carrier is intended to evaporate and not remain with the product. There are no other uses of any form of xylene in the facility. The otherwise use activity threshold of 10,000 pounds/year has not been reached for any of the xylenes and an EPCRA Section 313 report need not be prepared for xylene. However, should any two of the streams mix, the facility will exceed the otherwise use threshold for mixed isomers and an EPCRA Section 313 report must be prepared for the mixed isomer form of xylene.

### **COMMON ERROR - Threshold Determination for Recirculation**

Facilities often incorrectly base threshold calculations on the amount of EPCRA Section 313 chemicals or chemical categories in a recirculation system rather than the amount actually used in the reporting year. The amount of the EPCRA Section 313 chemical or chemical category that is actually manufactured (including the quantity imported), processed, or otherwise used, not the amount in storage or in the system, should be the amount applied to the threshold determination. For example, a solvent containing an EPCRA Section 313 chemical or chemical category is used, recirculated on site, and reused as a solvent. The amount of EPCRA Section 313 chemical or chemical category recirculated in the on-site recycling process is not considered in the threshold determination because it is considered to be “direct reuse” and is not reportable. Only the amount of new chemical added to the system should be included in the otherwise used threshold calculation. However, if you send a solvent containing an EPCRA Section 313 chemical or chemical category off site for distillation and subsequent recycling, it should be reported as a transfer to an off-site location for recycling (Part II, Sections 6.2 and 8.5 or the 1999 Form R) because the distillation is considered a waste management activity. The amount of solvent returned to you and subsequently used in the same reporting year must be included in the threshold determination. If the reporting threshold is exceeded, the total quantity recycled should be reported in Section 8.4, i.e., the amount recycled on site must be reported in Section 8.4 each time it is recycled.

Each of the activity categories is divided into subcategories. As discussed in the *TRI Forms and Instructions*, you are required to designate EACH category and subcategory that applies to your facility. Detailed definitions, including descriptions of subcategories for each activity and selected examples, are presented in Tables 3-2, 3-3, and 3-4.

**Table 3-2**

**Definitions and Examples of Manufacturing Subcategories**

<b>Manufacturing Activity Subcategory</b>	<b>Definition</b>	<b>Examples in Leather Tanning and Finishing Operations*</b>
Produced or imported for on-site use/processing	A chemical or chemical category that is produced or imported and then further processed or otherwise used at the same facility.	Sodium dichromate
Produced or imported for sale/distribution	A chemical or chemical category that is produced or imported specifically for sale or distribution outside the manufacturing facility.	
Produced as a byproduct	A chemical or chemical category that is produced coincidentally during the production, processing, or otherwise use of another chemical substance or a mixture and is separated from that substance or mixture. EPCRA Section 313 chemicals or chemical categories produced and released as a result of waste treatment or disposal are also considered byproducts.	Ammonia (From hide deamination), nitrate compounds, chromium shavings and buffings dusts
Produced as an impurity	A chemical or chemical category that is produced coincidentally as a result of the manufacture, processing, or otherwise use of another chemical and remains primarily in the mixture or product with that other chemical.	

\*More complete discussions of the industry-specific examples can be found in Chapter 4 of this guidance manual.

**Table 3-3****Definitions and Examples of Processing Subcategories**

<b>Processing Activity Subcategory</b>	<b>Definition</b>	<b>Examples in Leather Tanning and Finishing Operations*</b>
Reactant	A natural or synthetic chemical or chemical category used in chemical reactions for the manufacture of another chemical substance or product. Examples include feedstocks, raw materials, intermediates, and initiators.	Chromium, copper, and cobalt compounds, formaldehyde, methylene diphenyl diisocyanate, and toluene diisocyanate
Formulation component	A chemical or chemical category that is added to a product or product mixture prior to further distribution of the product and acts as a performance enhancer during use of the product. Examples include additives, dyes, reaction diluents, initiators, solvents, inhibitors, emulsifiers, surfactants, lubricants, flame retardants, and rheological modifiers.	
Article component	A chemical or chemical category that becomes an integral component of an article distributed for industrial, trade, or consumer use.	Chromium, o-phenylphenol, ammonium salts, chromium, copper and cobalt based dyes, dibutyl phthalate, n-methyl-2-pyrrolidone
Repackaging only	A chemical or chemical category that is processed or prepared for distribution in commerce in a different form, state, or quantity. May include, but is not limited to, the transfer of material from a bulk container, such as a tank truck, to smaller containers such as cans or bottles.	

\*More complete discussions of the industry-specific examples can be found in Chapter 4 of this guidance manual.

**Table 3-4****Definitions and Examples of Otherwise Use Subcategories**

Otherwise Use Activity Subcategory	Definition	Examples in Leather Tanning and Finishing Operations*
Chemical processing aid	A chemical or chemical category that is added to a reaction mixture to aid in the manufacture or synthesis of another chemical substance but is not intended to remain in or become part of the product or product mixture. Examples include process solvents, catalysts, inhibitors, initiators, reaction terminators, and solution buffers.	Formic acid, toluene
Manufacturing aid	A chemical or chemical category that aids the manufacturing process but does not become part of the resulting product and is not added to the reaction mixture during the manufacture or synthesis of another chemical substance. Examples include process lubricants, metalworking fluids, coolants, refrigerants, and hydraulic fluids.	
Ancillary or other use	A chemical or chemical category that is used for purposes other than aiding chemical processing or manufacturing. Examples include cleaners, degreasers, lubricants, fuels (including waste fuels), and chemicals used for treating wastes.	Chlorine, chlorine dioxide

\*More complete discussions of the industry-specific examples can be found in Chapter 4 of this guidance manual.

**Example - Chemical Processing Aid**

A spray painting operation uses toluene as the carrier solvent. Ideally all the solvent would evaporate, however, studies have shown 1% of the applied solvent remains on the workpiece. Since the function of the solvent is to improve the application of the paint and is a non-incorporative activity, the entire amount of toluene is considered otherwise used. If the solvent's function was such that it was intended to remain with the workpiece, it would be considered processed, as is the case for pigments, binders, and other paint components intended to remain with the workpiece.

### 3.2.1 Concentration Ranges for Threshold Determination

You should use the best, readily available information or where such data are not available, reasonable estimates for all calculations in EPCRA Section 313 reporting; however,



the exact concentration of an EPCRA Section 313 chemical or chemical category in a mixture or trade name product may not be known. The supplier or MSDS may only list ranges, or upper or lower bound concentrations. U.S. EPA has developed guidance on how to use information in this situation for threshold determinations.

- If the concentration is provided as a lower and upper bound or as a range, you should use the mid-point in your calculations for the threshold determination. For example, the MSDS for the trade name product states methanol is present in a concentration of not less than 20% and not more than 40%, or it may be stated as present at a concentration between 20 to 40%. You should use the mid-point value of 30% methanol in your threshold calculations.
- If only the lower bound concentration of the EPCRA Section 313 chemical or chemical category is specified and the concentration of other components are given, subtract the other component values from 100%. The remainder should be considered the upper bound for the EPCRA Section 313 chemical or chemical category and you should use the given lower bound to calculate the mid-point as discussed above. For example, the MSDS states that a solvent contains at least 50% methyl ethyl ketone (MEK) and 20% non-hazardous surfactants. Subtracting the non-hazardous contents from 100% leaves 80% as the upper bound for MEK. The mid-point between upper (80%) and lower (50%) bounds is 65%, the value you should use in your threshold calculation.
- If only the lower bound is specified and no information on other components is given, you should assume the upper bound is 100% and calculate the mid-point as above.
- If only the upper bound concentration is provided, you should use this value in your threshold calculation.

Special guidance for concentration ranges that straddle the *de minimis* value is presented in Section 3.2.2.1.

### **3.2.2 Evaluation of Exemptions**

When determining thresholds, you can exclude quantities of any EPCRA Section 313 chemicals and chemical categories that are manufactured, processed, or otherwise used in exempt activities. Exemptions are divided into four classes:

1. *De minimis* exemption;
2. Article exemption;
3. Facility-related exemption; and
4. Activity-related exemptions.

#### **COMMON ERROR - Exempt Activities**

If an EPCRA Section 313 chemical or chemical category is used in exempt activities, the quantity used in these activities does not need to be included in your threshold determinations or release and other waste management calculations, even if the chemical or chemical category is used in a reportable activity elsewhere in the facility.

### **3.2.2.1 *De Minimis* Exemption**

If the amount of EPCRA Section 313 chemical(s) or chemical categories present in a mixture or trade name product processed or otherwise used is below its *de minimis* concentration level, that amount is considered to be exempt from threshold determinations and release and other waste management calculations. Note that this exemption does not apply to manufacturing, except for importation or as an impurity as discussed below. Also note that the *de minimis* exemption does not apply to the manufacturing, processing, or otherwise use of the PBT chemicals (refer to Section 2.6). The *de minimis* concentration for EPCRA Section 313 chemicals and chemical categories is 1%, except for Occupational Safety and Health Administration (OSHA)-defined carcinogens, which have a 0.1% *de minimis* concentration. Note that if a mixture contains more than one member of an EPCRA Section 313 chemical category, the weight percent of all members must be summed. If the total meets or exceeds the category's *de minimis* level, the *de minimis* exemption does not apply. U.S. EPA has published several detailed questions and answers and a directive in the current edition of *EPCRA Section 313 Questions and Answers* (1998 edition is EPA 745-B-98-004; see Appendix A, Directive #2) that may be helpful if you have additional concerns about the *de minimis* exemption. The *TRI Forms and Instructions* list each EPCRA Section 313 chemical and chemical category with the associated *de minimis* value.

Once the *de minimis* level has been equaled or exceeded, the exemption no longer applies to that process stream, even if the EPCRA Section 313 chemical or chemical category later falls below the *de minimis* concentration. All release and other waste management activities

that occur after the *de minimis* concentration has been equaled or exceeded are subject to reporting. The facility does not have to report release and other waste management activities that took place before the *de minimis* concentration was equaled or exceeded.

**Example - De Minimis**

Your facility processes a mixture containing 1.1% nitric acid and 0.6% manganese. The *de minimis* exemption would apply to manganese because the concentration is below 1% which is the *de minimis* level for manganese; however, it would not apply to nitric acid. All of the nitric acid must be included in threshold determinations, and release and other waste management calculations.

The *de minimis* exemption also applies to EPCRA Section 313 chemicals and chemical categories that are coincidentally manufactured below the *de minimis* level only if that chemical is manufactured as an impurity in a mixture that is subsequently distributed in commerce. In addition, the exemption applies to EPCRA Section 313 chemicals and chemical categories below the *de minimis* concentration in an imported mixture or trade name product.

For some mixtures the concentration of EPCRA Section 313 chemicals and chemical categories may be available only as a range. U.S. EPA has developed guidance on how to determine quantities applicable to threshold determinations, and release and other waste management calculations when this range straddles the *de minimis* value. In general, only the quantity of the processed or otherwise used EPCRA Section 313 chemical or chemical category whose concentration exceeds the *de minimis* must be considered. Therefore, U.S. EPA allows facilities to estimate the quantity below the *de minimis* and exclude it from further consideration. The following examples illustrate this point.

### Examples - De Minimis Concentration Ranges

#### Example 1:

A facility processes 8,000,000 pounds of a mixture containing 0.25 to 1.25% manganese. Manganese is subject to a 1% *de minimis* concentration exemption. The amount of mixture subject to reporting is the quantity containing manganese at or above the *de minimis* concentration:

$$8,000,000 \times (0.0125 - 0.0099) \div (0.0125 - 0.0025)$$

The average concentration of manganese that is not exempt (at or above the *de minimis*) is:

$$(0.0125 + 0.01) \div 2$$

Therefore, the amount of manganese that is subject to threshold determination and release and other waste management estimates is:

$$\left[ \frac{8,000,000 \times (0.0125 - 0.0099)}{(0.0125 - 0.0025)} \right] \times \left[ \frac{(0.0125 + 0.01)}{2} \right] = 23,400 \text{ pounds}$$

$$= 23,400 \text{ pounds manganese (which is below the processing threshold)}$$

In this example, because the facility's information pertaining to manganese was available to two decimal places, 0.99 was used to determine the amount below the *de minimis* concentrations. If the information was available to one decimal place, 0.9 should be used, as in Example 2 below.

#### Example 2:

As in Example 1, manganese is present in a mixture, of which 8,000,000 pounds is processed. The MSDS states the mixture contains 0.2% to 1.2% manganese. The amount of mixture subject to reporting (at or above *de minimis*) is:

$$8,000,000 \times (0.012 - 0.009) \div (0.012 - 0.002)$$

The average concentration of manganese that is not exempt (at or above *de minimis*) is:

$$(0.012 + 0.01) \div 2$$

Therefore, the amount of manganese that is subject to threshold determinations and release and other waste management estimates is:

$$\left[ \frac{8,000,000 \times (0.012 - 0.009)}{(0.012 - 0.002)} \right] \times \left[ \frac{(0.012 + 0.01)}{2} \right] = 26,400 \text{ pounds}$$

$$= 26,400 \text{ pounds manganese (which is above the processing threshold)}$$

The exemption does not apply to EPCRA Section 313 chemicals and chemical categories coincidentally manufactured as byproducts and separated from the product, nor does it apply to EPCRA Section 313 chemicals and chemical categories coincidentally manufactured as a result of waste management activities, from either on site or off site. (Under EPCRA Section 313, U.S. EPA does not consider waste to be a mixture.) For example, many facilities treat waste solvents by incinerating them. If coal is used as the primary fuel source to incinerate these waste

solvents, combustion can result in the coincidental manufacture of sulfuric and hydrochloric acid aerosols and metal compounds. Since the *de minimis* exemption does not apply to the coincidental manufacture of EPCRA Section 313 chemicals or chemical categories as a byproduct or in a waste treatment process, the formation of these compounds must be considered for threshold determinations, and release and other waste management calculations.

### **3.2.2.2 Articles Exemption**

An article is defined as a manufactured item that:

- Is formed to a specific shape or design during manufacture;
- Has end-use functions dependent in whole or in part upon its shape or design; and
- Does not release an EPCRA Section 313 chemical or chemical category under normal conditions of processing or otherwise use of the item at the facility.

If you receive a manufactured article from another facility or you produce the article in your facility and process or otherwise use it without changing the shape or design, and your processing or otherwise use does not result in the release of more than 0.5 pound of the EPCRA Section 313 chemical or chemical category in a reporting year from all like articles, then the EPCRA Section 313 chemical or chemical category in that article is exempt from threshold determinations and release and other waste management calculations (U.S. EPA allows a release of 0.5 pound or less to be rounded to zero; the 0.5-pound limit does not apply to each individual article but applies to the sum of all releases from processing or use of all like articles). Section 313 chemicals or chemical categories used to produce an article, however, do not qualify for the article exemption.

The shape and design can be changed somewhat during processing and otherwise use as long as part of the item retains the original dimensions. That is, as a result of processing or otherwise use, if an item retains its initial thickness or diameter, in whole or in part, then it still meets the article definition. If the item's original dimensional characteristics are totally altered during processing or otherwise use, the item would not meet the definition. As an example, items that do not meet the definition would be items that are cold extruded, such as lead

ingots formed into wire or rods. However, cutting a manufactured item into pieces that are recognizable as the article would not change the exemption status as long as the diameter and the thickness of the item remain unchanged. For instance, metal wire may be bent and sheet metal may be cut, punched, stamped, or pressed without losing the article status as long as no change is made in the diameter of the wire or tubing or the thickness of the sheet and no releases above 0.5 pound per year occur for all like articles.

Any processing or otherwise use of an article that results in a release above 0.5 pound per year for each EPCRA Section 313 chemical or chemical category for all like articles negates the exemption. Cutting, grinding, melting, or other processing of a manufactured item could result in a release of an EPCRA Section 313 chemical or chemical category during normal conditions of use and, therefore, could negate the article exemption if the total annual releases from all like articles exceed 0.5 pound in a year. However, if all of the resulting waste is recycled or reused, either on site or off site, so that the release of the EPCRA Section 313 chemical or chemical category does not exceed 0.5 pound for the calendar year, then the article's exemption status is maintained. If the processing or otherwise use of similar manufactured items results in a total release of less than or equal to 0.5 pound of any individual EPCRA Section 313 chemical or chemical category to any environmental media in a calendar year, U.S. EPA will allow this quantity to be rounded to zero and the manufactured items maintain their article status. The 0.5-pound limit does not apply to each individual article, but applies to the sum of all releases from processing or otherwise use of like articles for each EPCRA Section 313 chemical or chemical category. The current edition of *EPCRA Section 313 Questions and Answers* (1998 edition is EPA 745-B-98-004) presents several specific question and answers/discussions pertaining to the articles exemption.

#### **Example - Articles Exemption**

If an article, as part of a coating pre-treatment operation, is subjected to an etching process that removes a portion of an EPCRA Section 313 metal from the surface that is not recycled or reused, this process would constitute a release and negate the article exemption if the total release is greater than 0.5 lb for the reporting year. For example, a copper plate is cleansed by dipping in a sulfuric acid solution. Some of the copper reacts with the acid to form copper sulfate and the used cleaning solution is discharged to a POTW. This process is considered a release of the copper and, if the release from all like articles is greater than 0.5 pound per year, the plate has lost its article exemption.

### 3.2.2.3 Facility-Related Exemption

#### Laboratory Activity Exemption

EPCRA Section 313 chemicals and chemical categories that are manufactured, processed, or otherwise used in laboratories under the supervision of a technically qualified individual are exempted from the threshold determination (and subsequent release and other waste management calculations). This exemption may be applicable in circumstances such as laboratory sampling and analysis, research and development, and quality assurance and quality control activities. It does not include pilot plant scale or specialty chemical production. It also does not include laboratory support activities. For example, chemicals used to maintain laboratory equipment are not eligible for the laboratory exemption.

#### **Example - Laboratory Activity Exemption**

A leather finishing facility has a research laboratory that tests various leather tanning and finishing formulations containing EPCRA Section 313 chemicals and chemical categories by applying these formulations to leather samples being considered for car seat use by an automobile manufacturer. The testing is under the supervision of a “technically qualified individual” in the laboratory. The EPCRA Section 313 chemicals and chemical categories used in this activity would be exempt from EPCRA Section 313 reporting and should not be included in any threshold determination or release and other waste management calculations.

### 3.2.2.4 Activity-Related Exemptions (Otherwise Use Exemptions)

Some exemptions apply to the otherwise use of an EPCRA Section 313 chemical and chemical category. The specific quantities of EPCRA Section 313 chemicals and chemical categories used in these activities do not need to be included in a facility’s threshold determination (nor the associated release and other waste management calculations). The following otherwise use activities are considered exempt:

- **EPCRA Section 313 chemicals and chemical categories used in routine janitorial or facility grounds maintenance.** Examples are bathroom cleaners, fertilizers, and garden pesticides similar in type or concentration to consumer products. Materials used to clean process equipment do not meet this exemption.

- **Personal use of items.** Examples are foods, drugs, cosmetics, and other personal items including those items within the facility such as in a facility operated cafeteria, store, or infirmary. Office supplies such as correction fluid are also exempt.

#### **Example - Personal Use Exemption**

Aqueous ammonia used to clean a cafeteria grill is exempt from threshold determinations and release and other waste management calculations.

- **Structural components of the facility.** Exemptions apply to EPCRA Section 313 chemicals and chemical categories present in materials used to construct, repair, or maintain structural components of a facility. An example common to all facilities would be the solvents and pigments used to paint buildings. Materials used to construct, repair, or maintain process equipment are not exempt.
- **EPCRA Section 313 chemicals and chemical categories used with facility motor vehicles.** This exemption includes the use of EPCRA Section 313 chemicals and chemical categories for the purpose of maintaining motor vehicles operated by the facility. Common examples include gasoline, radiator coolant, windshield wiper fluid, brake and transmission fluid, oils and lubricants, cleaning solutions, and solvents in paint used to touch up the vehicle. Motor vehicles include cars, trucks, forklifts, locomotives, and aircraft. Note that this exemption applies to the OTHERWISE USE of the EPCRA Section 313 chemical and chemical category. The coincidental manufacture of EPCRA Section 313 chemicals and chemical categories resulting from combustion of gasoline is not exempt and should be considered toward the manufacturing threshold.

#### **Example - Motor Vehicle Exemption**

Methanol is purchased for use as a processing aid and as a windshield washer anti-freeze in company vehicles. The amount used for the latter purpose would be subtracted from the facility total **BEFORE** the facility total is compared to the activity threshold. Even if the facility still exceeds the otherwise use threshold, the amount in the anti-freeze is exempt from release and other waste management calculations.

This exemption does NOT apply to stationary equipment. The use of lubricants and fuels for stationary process equipment (e.g., pumps and compressors) and stationary energy sources (e.g., furnaces, boilers, heaters), are NOT exempt.



### **Example - Process Equipment Chemical Use**

Lubricants containing EPCRA Section 313 chemicals and chemical categories used on facility vehicles, or on-site structural maintenance activities that are not integral to the process, are exempt activities. However, lubricants used to maintain pumps and compressors that aid facility process operations are not exempt and the amount of the EPCRA Section 313 chemicals and chemical categories in the lubricant should be applied to the otherwise use threshold.

- **EPCRA Section 313 chemicals and chemical categories in air or water drawn from the environment or municipal sources.** Included are EPCRA Section 313 chemicals and chemical categories present in process water and non-contact cooling water drawn from the environment or a municipal source, or chemicals and chemical categories present in air used either as compressed air or as an oxygen source for combustion.

### **Example - Chemicals in Process Water**

A facility uses river water for one of its processes. This water contains approximately 100 pounds of an EPCRA Section 313 chemical or chemical category. The facility ultimately returns the water that contains the entire 100 pounds of the EPCRA Section 313 chemical or chemical category to the river. The EPCRA Section 313 chemical or chemical category in the water can be considered exempt because the EPCRA Section 313 chemical or chemical category was present as it was drawn from the environment. The facility does not need to consider the EPCRA Section 313 chemical or chemical category drawn with river water for threshold determinations or release and other waste management calculations.

## **3.2.3 Additional Guidance on Threshold Calculations for Certain Activities**

This section covers three specific situations in which the threshold determination may vary from normal facility operations: reuse, remediation, and recycling activities of EPCRA Section 313 chemicals and chemical categories.

### **3.2.3.1 Reuse Activities**

Threshold determinations of EPCRA Section 313 chemicals or chemical categories that are reused at the facility are based only on the amount of the EPCRA Section 313 chemical or chemical category that is added to the system during the year, not the total volume in the system. For example, a facility operates a refrigeration unit that contains 15,000 pounds of anhydrous ammonia at the beginning of the year. The system is charged with 2,000 pounds of

anhydrous ammonia during the year. The facility has therefore otherwise used only 2,000 pounds of the EPCRA Section 313 chemical or chemical category and is not required to report (unless the facility has additional otherwise use activities of ammonia that, when taken together, exceed the reporting threshold). If, however, the whole refrigeration unit was recharged with 15,000 pounds of new or fresh anhydrous ammonia during the year, the facility would exceed the otherwise use threshold, and be required to report.

### **3.2.3.2 Remediation Activities**

EPCRA Section 313 chemicals and chemical categories undergoing remediation (e.g., Superfund remediation) are not being manufactured, processed, or otherwise used. Therefore, they are not included in the activity threshold determinations.

However, if you are conducting remediation of an EPCRA Section 313 chemical or chemical category that is also being manufactured, processed, or otherwise used by the facility above an activity threshold level, you must consider this activity for release and other waste management calculations. You must report any release or other waste management quantities of an EPCRA Section 313 chemical or chemical category due to remediation in Part II, Sections 5 through 8, accordingly, of the 1999 Form R. Those quantities would also be considered as part of the amount for determining Form A eligibility. EPCRA Section 313 chemicals and chemical categories used for remediation should be considered toward threshold determinations. If an EPCRA Section 313 chemical or chemical category exceeds one of the reporting thresholds elsewhere at the facility, all release and other waste management activity quantities of that chemical or chemical category must be reported, including release and other waste management activity quantities resulting from remediation.

Excavation (that is considered part of the remedial action ) of material already landfilled does not constitute a manufacturing, processing, or otherwise use activity. However, routine activities (e.g., dredging a lagoon), even if not performed every year, are not considered to be remedial actions and are always subject to reporting.

### 3.2.3.3 Recycling Activities

For on-site recycling and reuse systems, where the same EPCRA Section 313 chemical or chemical category is recycled and reused multiple times, the quantity recycled or reused should be counted only once (at the time it is introduced into the system) for threshold calculations. (Please note that for reporting on-site waste management activities the quantity of the EPCRA Section 313 chemical or chemical category should be counted every time it exits the recycling unit in Section 8 of Form R.) EPCRA Section 313 chemicals and chemical categories recycled off site and returned to the facility should be treated as newly purchased materials for purposes of EPCRA Section 313 threshold determinations.

### 3.3 Step 3 - Calculate the Quantity of Each EPCRA Section 313 Chemical and Chemical Category and Determine Which Ones Exceed an Activity Threshold

The final step is to determine the quantity and which EPCRA Section 313 chemicals and chemical categories exceed an activity threshold. At this point you should have:

1. Identified each EPCRA Section 313 chemical and chemical category at your facility.
2. Determined the activity category for each EPCRA Section 313 chemical and chemical category (manufactured, processed, or otherwise used).

Now, you must sum the amount for each EPCRA Section 313 chemical and chemical category by activity category, subtract all exempt quantities, and compare the totals to the applicable thresholds. Each EPCRA Section 313 chemical and chemical category exceeding **any one** of the activity thresholds requires the submission of an EPCRA Section 313 report. Provided you meet certain criteria you may prepare a Form A rather than a Form R (see Section 2.8).

### **COMMON ERROR - Assuming a Threshold is Exceeded**

U.S. EPA recently published a report, *The 1994 and 1995 Toxic Release Inventory Data Quality Report*, EPA 745-R-98-002, with the site survey results of over 100 facilities to evaluate EPCRA Section 313 reporting quality. One of the findings of this survey was that facilities that simply assumed that chemical activity thresholds were exceeded were often in error. This resulted in many of these facilities filing EPCRA Section 313 reports when thresholds were actually not exceeded. Unless the facility has strong grounds to support such an assumption, the time spent in explicitly calculating the activity threshold is well spent.

### **COMMON ERROR - Zero Release and Other Waste Management Quantities**

If you meet all reporting criteria and exceed any activity threshold for an EPCRA Section 313 chemical or chemical category, you must file an EPCRA Section 313 report for that chemical or chemical category, even if you have zero release and other waste management activity quantities. Exceeding the chemical activity threshold, not the quantity released or otherwise managed as waste determines whether you must report. Note that if the release and other waste management activity quantity is 500 pounds or less for each chemical or chemical category you may be eligible to use the alternate certification statement, Form A, rather than a Form R (see Section 2.9).

To determine if an EPCRA Section 313 chemical or chemical category exceeds a reporting threshold, you must calculate the annual activity amount of that chemical. Start with the amount of chemical or chemical category at the facility as of January 1, add any amounts brought on site during the year and the amount manufactured (including imported), and subtract the amount left in the inventory on December 31. If necessary, adjust the total to account for exempt activities (see Section 3.2.2 for a discussion of exemptions). You should then compare the result to the appropriate threshold to determine if you are required to submit an EPCRA Section 313 report for that chemical or chemical category. Keep in mind that the threshold calculations are independent for each activity category: manufactured, processed, and otherwise used. If more than one activity category applies, the amount associated with each category is determined separately.

Table 3-5 presents a worksheet that may be helpful when conducting your threshold determinations. Table 3-6 illustrates how the work sheet can be used for the following example:

### **Example - Threshold Worksheet**

Assume your facility purchases, in the applicable reporting year, two mixtures that contain xylene (mixed isomers). You purchased 25,000 pounds of Mixture A (which is 50% xylene, by weight, according to the MSDS) and 110,000 pounds of Mixture B (which contains 20% xylene, by weight). Further, you determine that you process the entire quantity of Mixture A, while you process only half of Mixture B and otherwise use the other half. You do not qualify for any exempt activities.

In this example, you would have processed a total of 23,500 pounds of xylene (12,500 pounds from activities associated with Mixture A and 11,000 pounds from activities associated with Mixture B). You would also have otherwise used a total of 11,000 pounds (all from Mixture B). Therefore, you would not have exceeded the 25,000-pound threshold for processing; however, you would have exceeded the 10,000-pound threshold for otherwise use and would be required to submit an EPCRA Section 313 report that includes releases and other waste management quantities from all activities (including processing).

### Table 3-5. EPCRA Section 313 Reporting Threshold Worksheet

Facility Name: \_\_\_\_\_  
 EPCRA Section 313 Chemical or Chemical Category: \_\_\_\_\_  
 CAS Registry Number: \_\_\_\_\_  
 Reporting Year: \_\_\_\_\_

Date Worksheet Prepared: \_\_\_\_\_  
 Prepared By: \_\_\_\_\_

**Amounts of chemical or chemical category manufactured, processed, or otherwise used.**

Mixture Name or Other Identifier	Information Source	Total Weight (lb)	Percent EPCRA Section 313 Chemical or Chemical Category by Weight	EPCRA Section 313 Chemical or Chemical Category Weight (lb)	Amount of the EPCRA Section 313 Chemical or Chemical Category by Activity (lb):		
					Manufactured	Processed	Otherwise Used
1.							
2.							
3.							
4.							
<b>Subtotal:</b>					(A) _____ lb.	(B) _____ lb.	(C) _____ lb.

**Exempt quantity of chemical or chemical category that should be excluded.**

Mixture Name as Listed Above	Applicable Exemption (de minimis, article, facility, activity)	Fraction or Percent Exempt (if Applicable)	Amount of the EPCRA Section 313 Chemical or Chemical Category Exempt from Above (lb):		
			Manufactured	Processed	Otherwise Used
1.					
2.					
3.					
4.					
<b>Subtotal:</b>			(A <sub>1</sub> ) _____ lb.	(B <sub>1</sub> ) _____ lb.	(C <sub>1</sub> ) _____ lb.

Amount subject to threshold:      (A-A<sub>1</sub>) \_\_\_\_\_ lb.      (B-B<sub>1</sub>) \_\_\_\_\_ lb.      (C-C<sub>1</sub>) \_\_\_\_\_ lb.

Compare to threshold for EPCRA Section 313 reporting.

Activity threshold quantities<sup>1</sup>:      25,000 lb.      25,000 lb.      10,000 lb.

If any one of the thresholds is exceeded, reporting is required for all activities. [Do not submit this worksheet with Form R, retain it for your records.]

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<sup>1</sup>These activity thresholds apply to non-PBT chemicals. See Section 2.6 for activity thresholds applicable to PBT chemicals.

**Table 3-6. Sample EPCRA Section 313 Reporting Threshold Worksheet**

Facility Name: XYZ Leather Tanning Corp.  
 EPCRA Section 313 Chemical or Chemical Category: Xylene (mixed isomers)  
 CAS Registry Number: 1330-20-7  
 Reporting Year: 1998

Date Worksheet Prepared: May 1, 1999  
 Prepared By: A.B. Calloway

**Amounts of chemical or chemical category manufactured, processed, or otherwise used.**

Mixture Name or Other Identifier	Information Source	Total Weight (lb)	Percent EPCRA Section 313 Chemical or Chemical Category by Weight	EPCRA Section 313 Chemical or Chemical Category Weight (lb)	Amount of the EPCRA Section 313 Chemical or Chemical Category by Activity (lb):		
					Manufactured	Processed	Otherwise Used
1. Mixture A	MSDS	25,000	50%	12,500	---	12,500	---
2. Mixture B	MSDS	110,000	20%	22,000	---	11,000	11,000
3.							
4.							
<b>Subtotal:</b>					<b>(A) 0 lb.</b>	<b>(B) 23,500 lb.</b>	<b>(C) 11,000 lb.</b>

**Exempt quantity of chemical or chemical category that should be excluded.**

Mixture Name as Listed Above	Applicable Exemption (de minimis, article, facility, activity)	Fraction or Percent Exempt (if Applicable)	Amount of the EPCRA Section 313 Chemical or Chemical Category Exempt from Above (lb):		
			Manufactured	Processed	Otherwise Used
1. Mixture A	none				
2. Mixture B	none				
3.					
4.					
<b>Subtotal:</b>			<b>(A<sub>1</sub>) 0 lb.</b>	<b>(B<sub>1</sub>) 0 lb.</b>	<b>(C<sub>1</sub>) 0 lb.</b>

Amount subject to threshold: (A-A<sub>1</sub>) 0 lb. (B-B<sub>1</sub>) 23,500 lb. (C-C<sub>1</sub>) 11,000 lb.

Compare to threshold for EPCRA Section 313 reporting. Activity threshold quantities<sup>1</sup>: 25,000 lb. 25,000 lb. 10,000 lb.

If any one of three thresholds is exceeded, reporting is required for all activities. [Do not submit this worksheet with Form R, retain it for your records.]

<sup>1</sup>These activity thresholds apply to non-PBT chemicals. See Section 2.6 for activity thresholds applicable to PBT chemicals.

**Appendix A**

**EPCRA SECTION 313 GUIDANCE RESOURCES**



## CHAPTER 4 - ESTIMATING RELEASE AND OTHER WASTE MANAGEMENT QUANTITIES

### 4.0 PURPOSE

This chapter is intended to guide the user in developing a systematic approach for estimating release and other waste management quantities of EPCRA Section 313 chemicals and chemical categories from leather tanning and finishing operations. Figure 4-1 diagrams a recommended approach for estimating quantities of reportable EPCRA Section 313 chemicals or chemical categories.

This chapter also includes common EPCRA Section 313 reporting and compliance issues as they apply to leather tanning and finishing. The general discussion (Section 4.1) is followed by a presentation of specific examples and issues (Section 4.2).

### 4.1 General Steps for Determining Release and Other Waste Management Activity Quantities

Release and other waste management activity quantities can be determined by completing the following four steps, described in detail in the following sections.

- Step 1)* Prepare a **process flow diagram**.
- Step 2)* Identify EPCRA Section 313 chemicals and chemical categories and potential **sources** of chemical release and other waste management activities.
- Step 3)* Identify release and other waste management activity **types**.
- Step 4)* Determine the most appropriate method(s) and **calculate the estimates** for release and other waste management activity quantities.

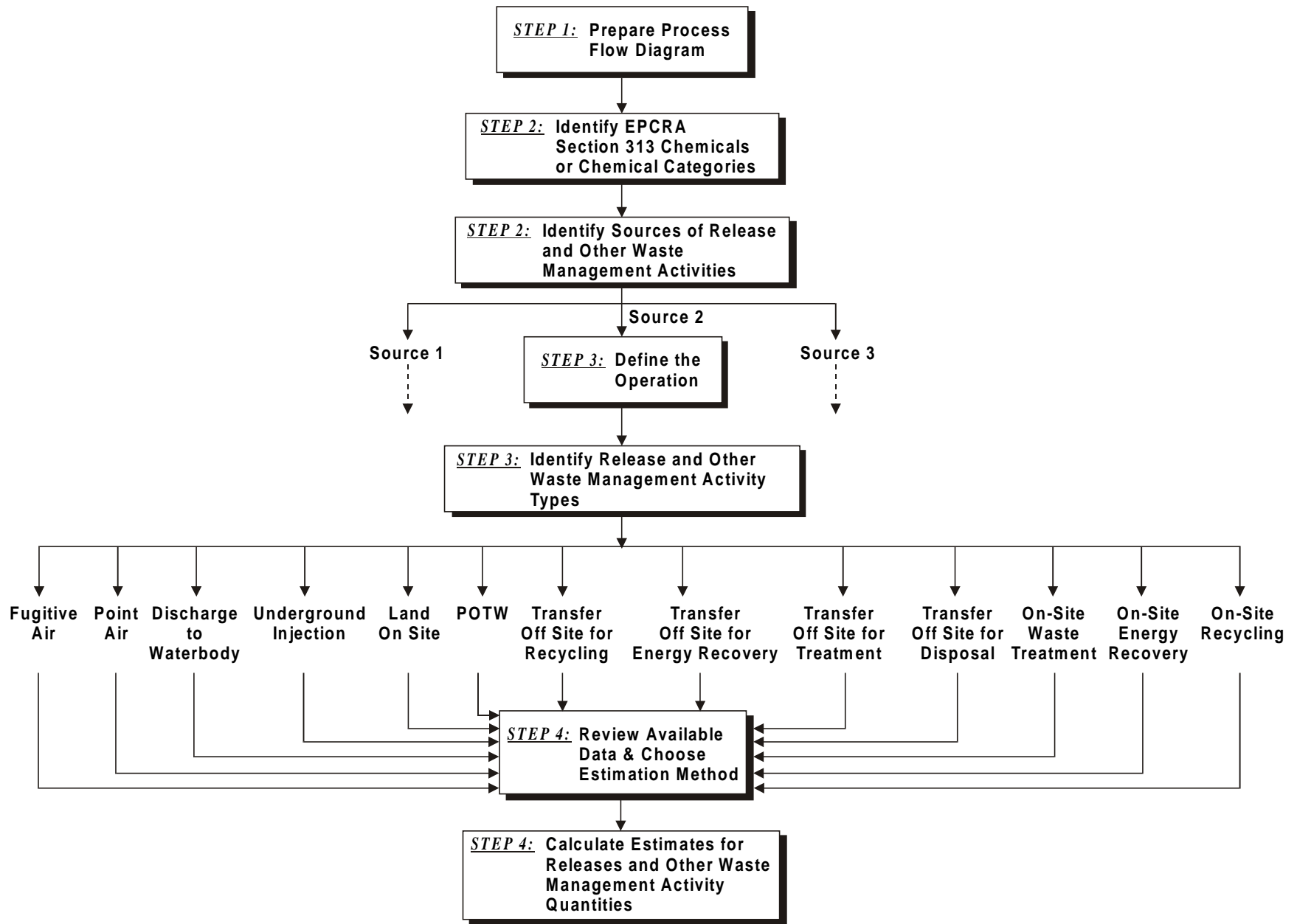


Figure 4-1. Release and Other Waste Management Activity Calculation Approach

For EPCRA Section 313 purposes, “sources” means the streams or units that generate the release and other waste management activity (such as process vents, container residue, or spills) and “types” means the environmental media corresponding to elements in Sections 5 through 8 of the 1999 Form R (for example, releases to fugitive air, releases to stack air, discharges to receiving streams or POTWs, or releases to land).

#### **4.1.1 Step 1: Prepare a Process Flow Diagram**

Preparing a process flow diagram will help you to identify potential sources and types of EPCRA Section 313 chemicals and chemical categories released and otherwise managed as waste at your facility. Depending on the complexity of your facility, you may want to diagram individual processes or operations rather than the entire facility. The diagram should show how materials flow through the processes and identify material input, generation, and output points. Looking at each operation separately, you can determine where EPCRA Section 313 chemicals and chemical categories are used and the medium to which they may be released or otherwise managed as waste.

#### **4.1.2 Step 2: Identify EPCRA Section 313 Chemicals and Chemical Categories and Potential Sources of Chemical Release and Other Waste Management Activities**

Once a process flow diagram has been developed, you must determine the potential sources and the EPCRA Section 313 chemicals and chemical categories that may be released and otherwise managed as waste from each unit operation and process. Remember to include upsets and routine maintenance activities. Potential sources include:

- Accidental spills and releases;
- Air pollution control devices (e.g., baghouses, electrostatic precipitators, and scrubbers);
- Clean up and housekeeping practices;
- Combustion byproducts;
- Container residues;
- Fittings;
- Flanges;
- Process discharge stream;
- Process vents;
- Pumps;
- Recycling and energy recovery byproducts;
- Relief valves;
- Stock pile losses;
- Storage tanks;
- Storm water runoff;
- Tower stacks;
- Transfer operations;

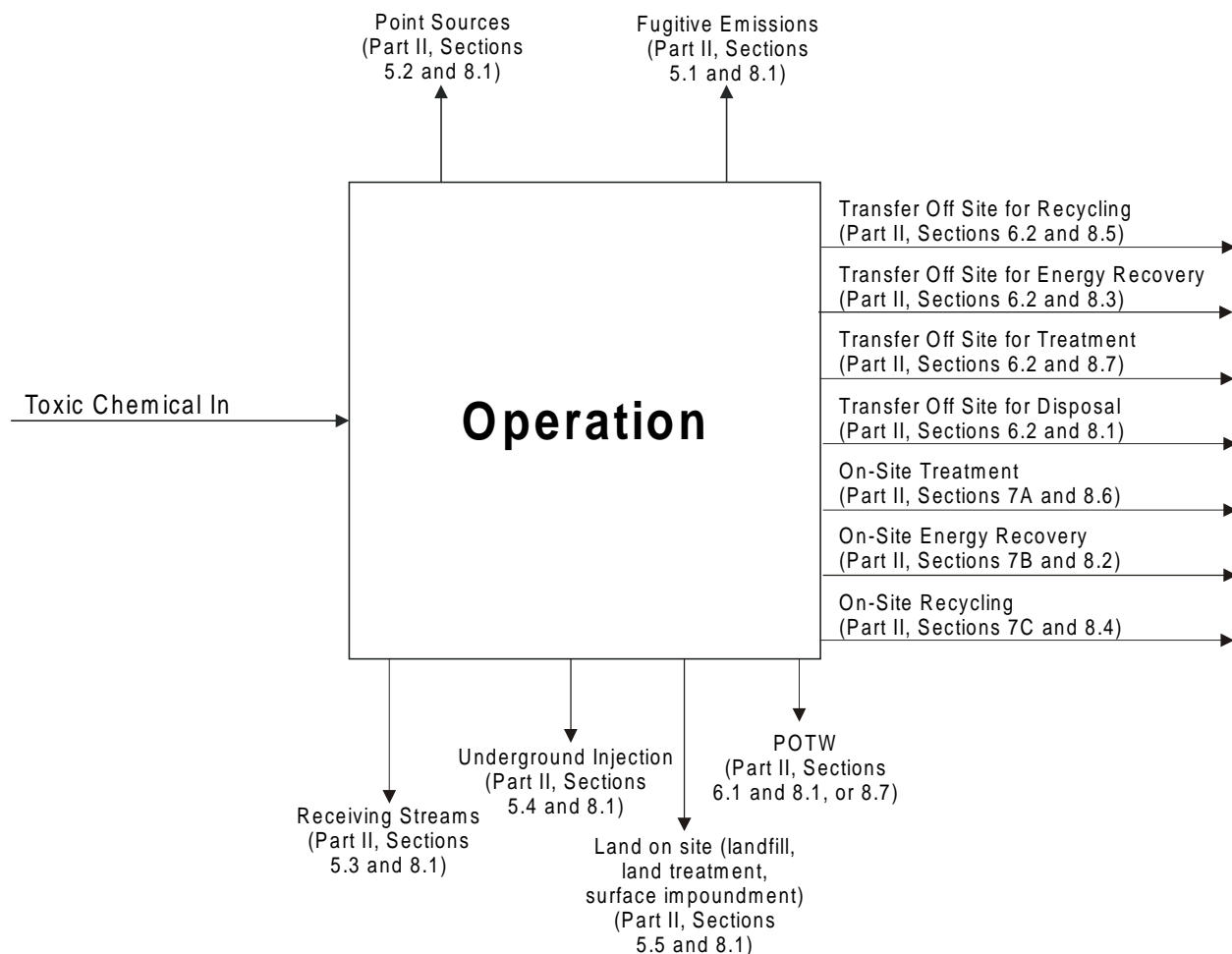
- Treatment sludge;
- Volatilization from process or treatment; and
- Waste treatment discharges.

Next, you must identify the EPCRA Section 313 chemicals and chemical categories that may be released or otherwise managed as waste from each source. A thorough knowledge of the facility operations and processes is required for this determination. You should also consider whether any of the EPCRA Section 313 chemicals or chemical categories are coincidentally manufactured at your facility. Table 2-3 identifies EPCRA Section 313 chemicals and chemical categories typically used in the operations common to leather tanning and finishing. This table can be used as an aid in identifying which chemicals and chemical categories are found in your process. The list may not include all the EPCRA Section 313 chemicals and chemical categories your facility uses, and it may include many chemicals and chemical categories that you do not use.

#### **4.1.3 Step 3: Identify Release and Other Waste Management Activity Types**

For each identified source of an EPCRA Section 313 chemical or chemical category, you should examine all possible release and other waste management activity types. Figure 4-2 schematically represents the possible release and other waste management activity types as they correspond to individual data elements of the Form R. Remember to include both routine operations and accidents when identifying types. This diagram along with the following descriptions can be used as a checklist to make sure all possible types of release and other waste management activities have been considered.

- a. **Fugitive or Non-Point Air Emissions (Part II, Section 5.1 of Form R)** - Includes all emissions to the air that are not released through stacks, vents, ducts, pipes, or any confined air stream. Examples include:
  - Releases from building ventilation systems, such as a roof fan in an open room;
  - Evaporative losses from solvent cleaning tanks, surface impoundments, and spills; and
  - Evaporative losses from solvent cleaning tanks, surface impoundments, and spills; and



**Figure 4-2. Possible Release and Other Waste Management Activity Types<sup>1</sup> for EPCRA Section 313 Chemicals and Chemical Categories**

<sup>1</sup>Sections refer to 1999 Form R. Quantities released to the environment as a result of remedial actions, catastrophic events, or one-time events should be reported in Part II, Section 8 as Subsection 8.8.

- Emissions from any other fugitive or non-point source.
- b. Stack or Point Air Emissions (Part II, Section 5.2 of Form R) -**  
 Includes all emissions to the air that occur through stacks, vents, ducts, pipes, or any confined air stream, including the emissions from storage tanks and air pollution control equipment. Air emissions from paint booths are often channeled through vapor recovery systems and/or air pollution control devices. These are considered stack emissions. Note that emissions released from general room air through a ventilation system are not considered stack or point releases for the purpose of EPCRA Section 313 reporting unless they are channeled through an air pollution control device. Instead, they are considered fugitive releases. However, you should note that for certain state reporting requirements, not

associated with EPCRA Section 313 reporting, some state air quality agencies consider ventilation systems to be a stack or point source.

- c. **Discharges to Receiving Streams or Water Bodies (Part II, Section 5.3 of Form R)** - Includes direct wastewater discharges to a receiving stream or surface water body. Discharges usually occur under a NPDES or SPDES permit.
  
- d. **Underground Injection On-site to Class I Wells (Part II, Section 5.4.1 of Form R) and to Class II through V Wells (Part II, Section 5.4.2 of Form R)** - Includes releases into an underground well at the facility. These wells may be monitored under an Underground Injection Control (UIC) Program permit. RCRA Hazardous Waste Generator Reports may be a good source of information for wastes injected into a Class I well. Injection rate meters may provide information for all the well classes.
  
- e. **Disposal to Land On-site (Part II, Section 5.5 of Form R)** - Includes all releases to land on-site, both planned (i.e., disposal) and unplanned (i.e., accidental release or spill). The four predefined subcategories for reporting quantities released to land within the boundaries of the facility are:
  - (1) **Landfill** - The landfill may be either a RCRA permitted (Part II, Section 5.5.1A) or a non-hazardous waste landfill (Part II, Section 5.5.1B). Both types are included if they are located on site. Leaks from landfills in the years subsequent to the disposal of the EPCRA Section 313 chemicals or chemical categories in the landfill do not need to be reported as a release.
  
  - (2) **Land treatment/application farming** - Land treatment is a disposal method in which a waste containing an EPCRA Section 313 chemical or chemical category is applied to or incorporated into soil. Volatilization of an EPCRA Section 313 chemical or chemical category because of the disposal operation must be included in the total fugitive air releases and should be excluded from land treatment/application farming to avoid double counting.

Sludge and/or aqueous solutions that contain biomass and other organic materials are often collected and applied to farm land. This procedure supplies a nitrogen source for plants and supplies metabolites for microorganisms. U.S. EPA considers this operation to be land treatment/farming if it occurs on site. If a facility sends this material off site for the same purpose, it is considered to be a “transfer to an off-site location, disposal” and should be reported under Sections 6.2 and 8.1 of the Form R.

The ultimate disposition of the chemical or chemical category after application to the land does not change the required reporting. For

example, even if the chemical or chemical category is eventually biodegraded by microorganisms or plants, it is not considered recycled, reused, or treated.

- (3) **Surface impoundment** - A surface impoundment is a natural topographic depression, man-made excavation, or diked area formed primarily of earthen materials that is designed to hold an accumulation of wastes containing free liquids. Examples include: holding, settling, storage, and elevation pits; ponds; and lagoons. Quantities of the toxic chemical released to surface impoundments that are used merely as part of a wastewater treatment process generally must not be reported in this section. However, if the sludge from the surface impoundment contains the EPCRA Section 313 chemical or chemical category, then the EPCRA Section 313 chemicals or chemical categories in the sludge must be estimated in this section unless the sludge is removed and subjected to another waste management activity.
- (4) **Other disposal** - Releases to land that do not fit the categories of landfills, land treatment, or surface impoundment are classified as other disposal. This disposal may include any spills or leaks of the EPCRA Section 313 chemical or chemical category to land.

- f. **Discharges to Publicly Owned Treatment Works (POTW) (Part II, Section 6.1 of Form R)** - Includes the amount of EPCRA Section 313 chemical or chemical category in water transferred to an off-site POTW. Note that metals and metal compounds transferred to a POTW must also be reported in Section 8.1.
- g. **Transfers to Other Off-Site Locations (Part II, Section 6.2 of Form R)** - Includes all off-site transfers containing the EPCRA Section 313 chemical or chemical category for the purposes of disposal, treatment, energy recovery, or recycling. Off-site transfer for disposal includes underground injection, landfill/surface impoundment, other land disposal and transfer to a waste broker for disposal. The amount transferred off site for disposal must also be reported in Section 8.1.

Be sure to consider metals and metal compounds that are present in pigments used in coloring operations. Waste containing these pigments may be present in spent filters or other waste generated from coloring operations.

Also reported in Section 6.2 would be any residual EPCRA Section 313 chemicals or chemical categories in “empty” containers transferred off site. U.S. EPA expects that all containers (bags, totes, drums, tank trucks, etc.) will have a small amount of residual solids and/or liquid. On-site cleaning of containers must be considered for EPCRA Section 313 reporting. If the cleaning occurs with a solvent (organic or aqueous), you

must report the disposition of the waste solvent as appropriate. If the containers are sent off site for disposal or reclamation, you should report the EPCRA Section 313 chemical or chemical category in this section.

### **COMMON ERROR - Shipping Container Residue**

Do not overlook residual chemicals or chemical categories in containers. U.S. EPA recently published *The 1994 and 1995 Toxic Release Inventory Data Quality Report*, EPA 745-R-98-002, presenting the site survey results of over 100 facilities performed to evaluate EPCRA Section 313 reporting quality. This survey found the largest source of overlooked release and other waste management activities was from container residue. So-called “empty” drums may contain an inch or more of liquid after draining and similarly “empty” bags may contain residues of dust and powder. Even though each individual drum or bag may only contain a small amount of an EPCRA Section 313 chemical or chemical category, for facilities that receive hundreds or thousands of drums or bags each year the annual cumulative amount of an EPCRA Section 313 chemical or chemical category can be substantial. The quantities should typically be reported in Section 6.2. (See Table 4-1 for estimates of liquid drum residual and the discussion in the text of this section for estimates of residual from solids.) Please note that unlike RCRA, EPCRA Section 313 does not define what constitutes an “empty” container. EPCRA Section 313 is merely trying to account for all the quantities of toxic chemicals released and otherwise managed as waste.

Actual data and a knowledge of the unloading methods at your facility can be used to estimate the quantity of residual EPCRA Section 313 chemicals or chemical categories in containers. However, U.S. EPA has developed guidance to assist facilities if no site-specific information is available. Table 4-1 provides results from a study of liquid residue quantities left in drums and tanks when emptied. These results are presented as the mass percent of the vessel capacity, and are categorized based on unloading method, vessel material, and bulk fluid material properties such as viscosity and surface tension. No testing was conducted for residual solids in this study. If data or site-specific knowledge is available to estimate the quantity of solid residual in containers, it should be considered. If no data are available, U.S. EPA believes an estimate of 1% residual solid is reasonable.



**Table 4-1**

**Summary of Liquid Residue Quantities From Pilot-Scale  
Experimental Study<sup>a,b</sup>  
(weight percent of drum capacity)**

Unloading Method	Vessel Type	Value	Material			
			Kerosene <sup>c</sup>	Water <sup>d</sup>	Motor Oil <sup>e</sup>	Surfactant Solution <sup>f</sup>
Pumping	Steel drum	Range	1.93 - 3.08	1.84 - 2.61	1.97 - 2.23	3.06
		Mean	2.48	2.29	2.06	3.06
Pumping	Plastic drum	Range	1.69 - 4.08	2.54 - 4.67	1.70 - 3.48	Not Available
		Mean	2.61	3.28	2.30	Available
Pouring	Bung-top steel drum	Range	0.244 - 0.472	0.266 - 0.458	0.677 - 0.787	0.485
		Mean	0.404	0.403	0.737	0.485
Pouring	Open-top steel drum	Range	0.032 - 0.080	0.026 - 0.039	0.328 - 0.368	0.089
		Mean	0.054	0.034	0.350	0.089
Gravity Drain	Slope-bottom steel tank	Range	0.020 - 0.039	0.016 - 0.024	0.100 - 0.121	0.048
		Mean	0.033	0.019	0.111	0.048
Gravity Drain	Dish-bottom steel tank	Range	0.031 - 0.042	0.033 - 0.034	0.133 - 0.191	0.058
		Mean	0.038	0.034	0.161	0.058
Gravity Drain	Dish-bottom glass-lined tank	Range	0.024 - 0.049	0.020 - 0.040	0.112 - 0.134	0.040
		Mean	0.040	0.033	0.127	0.040

<sup>a</sup>From "Releases During Cleaning of Equipment." Prepared by PEI Associates, Inc., for the U.S. Environmental Protection Agency, Office of Pesticides and Toxic Substances, Washington DC Contract No. 68-02-4248. June 30, 1986.

<sup>b</sup>The values listed in this table should only be applied to similar vessel types, unloading methods, and bulk fluid materials. At viscosities greater than 200 centipoise, the residue quantities can rise dramatically and the information on this table is not applicable.

<sup>c</sup>For kerosene, viscosity = 5 centipoise, surface tension = 29.3 dynes/cm<sup>2</sup>

<sup>d</sup>For water, viscosity = 4 centipoise, surface tension = 77.3 dynes/cm<sup>2</sup>

<sup>e</sup>For motor oil, viscosity = 97 centipoise, surface tension = 34.5 dynes/cm<sup>2</sup>

<sup>f</sup>For surfactant solution viscosity = 3 centipoise, surface tension = 31.4 dynes/cm<sup>2</sup>

The following example describes how the information in the table can be used to estimate the quantity of an EPCRA Section 313 chemical or chemical category in water that was used to clean drums on site.

### **Example - Container Residue**

You have determined that a Form R for an EPCRA Section 313 chemical must be submitted. The facility purchases and uses one thousand 55-gallon steel drums that contain a 10% aqueous solution of the chemical. Further, it is assumed that the physical properties of the solution are similar to water. The solution is pumped from the drums directly into a mixing vessel and the “empty” drums are triple-rinsed with water. The rinse water is indirectly discharged to a POTW and the cleaned drums are sent to a drum reclaimer.

From Table 4-1, the average drum residue quantity for this scenario is 2.29%. In this example, it can be assumed that all of the residual solution in the drums was transferred to the rinse water. Therefore, the quantity of the EPCRA Section 313 chemical transferred to the drum reclaimer should be reported as “zero.”

The annual quantity of residual solution that is transferred to the rinse water can be estimated by multiplying the mean weight percent of residual solution remaining in a pumped steel drum by the total annual weight of solution in the drums. If the density is not known, it may be appropriate to use the density of water (8.34 pounds per gallon):

$$(0.0229) \times (55 \text{ gal/drum}) \times (1,000 \text{ drums}) \times (8.34 \text{ lb/gal}) = 10,504 \text{ pounds solution}$$

The concentration of the EPCRA Section 313 chemical in the solution is only 10%.

$$(10,504 \text{ lb solution}) \times (0.1) = 1,050 \text{ pounds of the EPCRA Section 313 chemical}$$

Therefore, 1,050 pounds of the EPCRA Section 313 chemical are transferred to the POTW, and should be reported in Part II, Sections 6.1 and 8.7 of the 1999 Form R. Because they cannot be destroyed, metals cannot be reported as being treated, and metals and metal portions of metal compounds should be reported in Part II, Section 6.1 and 8.1 of the 1999 Form R.

- h. On-Site Waste Treatment (Part II, Section 7A of Form R)** - Includes all on-site waste treatment of EPCRA Section 313 chemicals or chemical categories. The information reported in Section 7A focuses on the treatment of the entire waste stream, not the specific EPCRA Section 313 chemical or chemical category. The information includes type of waste stream (gaseous, aqueous or non-aqueous liquid, or solid); treatment methods or sequence; influent concentrations of the EPCRA Section 313 chemical or chemical category; treatment efficiency (combined removal and destruction) of the entire method or sequence; and whether efficiency data are based on actual operating data. Metals and metal portions of metal compounds treated in a combustion process are not destroyed but should still be reported as going through the treatment process, with a treatment efficiency of zero. Note that only the metal portion of metal compounds should be reported in the Form R. The following example illustrates how Section 7A should be completed for on-site treatment of a wastewater stream containing three EPCRA Section 313 chemicals or chemical categories.

### Example - On-Site Waste Treatment

A process at your facility generates a wastewater stream containing an EPCRA Section 313 chemical (chemical A). A second process generates a wastewater stream containing two EPCRA Section 313 chemicals, a metal (chemical B) and a mineral acid (chemical C). Thresholds for all three chemicals have been exceeded and you are in the process of completing separate Form Rs for each chemical.

The two wastewater streams are combined and sent to an on-site wastewater treatment system before being released to a POTW. This system consists of an oil/water separator that removes 99% of chemical A; a neutralization tank in which the pH is adjusted to 7.5, thereby destroying 100% of the mineral acid (chemical C); and a settling tank where 95% of the metal (chemical B) is removed from the water (and eventually land filled off site).

Section 7A should be completed slightly differently when you file the Form R for each of the chemicals or chemical categories. The table accompanying this example shows how Section 7A should be completed for each chemical or chemical category. First, on each Form R you should identify the type of waste stream in Section 7A.1a as wastewater (aqueous waste, code W). Next, on each Form R you should list the code for each of the treatment steps that is applied to the entire waste stream, regardless of whether the operation affects the chemical or chemical category for which you are completing the Form R (for instance, the first four blocks of Section 7A.1b of all three Form Rs should show: P19 (liquid phase separation), C11 (neutralization), P11 (settling/clarification), and N/A (to signify the end of the treatment system). Note that Section 7A.1b is the only section of the Form R that is not chemical or chemical category specific. It applies to the entire waste stream being treated. Section 7A.1c of each Form R should show the concentration of the specific chemical or chemical category in the influent to the first step of the process (oil/water separation). For this example, assume chemicals or chemical categories A, B, and C are all present at concentrations greater than 1%. Therefore, code "1" should be entered. Section 7A.1d is also chemical specific. It applies to the efficiency of the entire system in destroying and/or removing the chemical or chemical category for which you are preparing the Form R. You should enter 99% when filing for chemical A, 95% for chemical B, and 100% for chemical C. Finally, you should report whether the influent concentration and efficiency estimates are based on operating data for each chemical or chemical category, as appropriate.

Chemical A						
7A.1a	7A.1b	1. <u>P19</u>	2. <u>C11</u>	7A.1c	7A.1d	7A.1e
<u>W</u>	3. <u>P11</u>	4. <u>N/A</u>	5. _____	<u>1</u>	<u>99</u> %	Yes    No
	6. _____	7. _____	8. _____			<u>X</u> _____
Chemical B						
7A.1a	7A.1b	1. <u>P19</u>	2. <u>C11</u>	7A.1c	7A.1d	7A.1e
<u>W</u>	3. <u>P11</u>	4. <u>N/A</u>	5. _____	<u>1</u>	<u>95</u> %	Yes    No
	6. _____	7. _____	8. _____			<u>X</u> _____
Chemical C						
7A.1a	7A.1b	1. <u>P19</u>	2. <u>C11</u>	7A.1c	7A.1d	7A.1e
<u>W</u>	3. <u>P11</u>	4. <u>N/A</u>	5. _____	<u>1</u>	<u>100</u> %	Yes    No
	6. _____	7. _____	8. _____			<u>X</u> _____
<b>[continued on next page]</b>						

Note that the quantity removed and/or destroyed is not reported in Section 7 and that the efficiency reported in Section 7A.1d refers to the amount of EPCRA Section 313 chemical or chemical category destroyed and/or removed from the applicable waste stream. The amount actually destroyed should be reported in Section 8.6 (quantity treated on site). For example, when completing the Form R for chemical B you should report “0” pounds in Section 8.6 because the metal has been removed from the wastewater stream, but not actually destroyed. The quantity of chemical B that is ultimately land filled off site should be reported in Section 6.2 and 8.1. However, when completing the Form R for chemical C you should report the entire quantity in Section 8.6 because raising the pH to 7.5 will completely destroy the mineral acid.

- i. On-Site Energy Recovery (Part II, Section 7B of Form R)** - Includes all on-site energy recovery of reported EPCRA Section 313 chemicals and chemical categories. U.S. EPA’s view is that EPCRA Section 313 chemicals or chemical categories that do not contribute significant heat energy during combustion processes should not be considered for energy recovery. Therefore, only EPCRA Section 313 chemicals or chemical categories with a significant heating value that are combusted in an energy recovery unit, such as an industrial furnace, kiln, or boiler can be reported for energy recovery. If an EPCRA Section 313 chemical or chemical category is incinerated on site but does not significantly contribute energy to the process, (e.g., chlorofluorocarbons (CFCs)) it must be considered on-site waste treatment (see 4.1.3, h. above). Metals and metal portions of metal compounds will never be combusted for energy recovery. Note that only the metal portion of metal compounds should be reported in the Form R.
- j. On-Site Recycling (Part II, Section 7C of Form R)** - Includes all on-site recycling methods used on EPCRA Section 313 chemicals or chemical categories.
- k. Source Reduction and Recycling Activities (Part II, Section 8 of Form R)<sup>1</sup>** - Provide information about source reduction and recycling activities related to the EPCRA Section 313 chemical or chemical category for which release and other waste management activities are being reported. Section 8 uses some data collected to complete Part II, Sections 5 through 7. For this reason, Section 8 should be completed last. The relationship between Sections 5, 6, and 8.8 to Sections 8.1, 8.3, 8.5, and 8.7 are provided in equation forms below.

  - (1) Quantity Released (Part II, Section 8.1 of Form R)** - The quantity reported in Section 8.1 is the quantity reported in all of Section 5 plus the quantity of metals and metal compounds reported as discharged off site to POTWs in Section 6.1 plus the quantity reported as sent off site for disposal in Section 6.2 minus the quantity reported in Section 8.8 that was released on site or sent off site for disposal:

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<sup>1</sup>The Subsection 8.1 through 8.8 designations are for the 1999 Form R. Please refer to the current reporting year *TRI Forms and Instructions* for any changes.

§8.1 = §5 + §6.1 (metals and metal compounds) + §6.2 (disposal only) - §8.8 (on-site release or off-site disposal only)

- (2) **Quantity Used for Energy Recovery On-site (Part II, Section 8.2 of Form R)** - Estimate the quantity of the EPCRA Section 313 chemical or chemical category in wastes combusted for energy recovery on site. This estimate should be the quantity of the chemical or chemical category combusted in the process for which codes were reported in Section 7B. Test data from trial burns or other monitoring data may be used to estimate the quantity of the EPCRA Section 313 chemical or chemical category combusted for energy recovery purposes. If monitoring data are not available, vendor specifications regarding combustion efficiency may be used as they relate to the EPCRA Section 313 chemical or chemical category. A quantity must be reported in Section 8.2 when a method of on-site energy recovery is reported in Section 7B and vice versa.

Two conditions need to be met to report the combustion of an EPCRA Section 313 chemical or chemical category in waste as energy recovery: the chemical or chemical category (1) must have a significant heating value and (2) must be combusted in an energy recovery unit, such as a waste heat boiler, an industrial furnace, or a kiln. If an EPCRA Section 313 chemical or chemical category that does not have a significant heating value (except metals and metal compounds) is combusted for energy recovery on site, it must be considered on-site waste treatment (see 4.1.3.h). Metals and metal compounds in a waste that are combusted on site will never be combusted for energy recovery and are considered to be disposed. Note that “NA” should be reported for EPCRA Section 313 chemicals or chemical categories that do not have a significant heating value. This includes metals, metal portions of metal compounds, halogens, hydrochlorofluorocarbons (HCFCs), and CFCs.

- (3) **Quantity Used for Energy Recovery Off-site (Part II, Section 8.3 of Form R)** - The quantity reported in Section 8.3 is the quantity reported in Section 6.2 for which energy recovery codes are reported. If a quantity is reported in Section 8.8, subtract any associated off-site transfers for energy recovery:

§8.3 = §6.2 (energy recovery) - §8.8 (off-site energy recovery)

Two conditions need to be met to report the combustion of an EPCRA Section 313 chemical or chemical category in waste as energy recovery: the chemical or chemical category (1) must have a significant heating value and (2) must be combusted in an energy recovery unit, such as a waste heat boiler, an industrial furnace, or a kiln. If an EPCRA Section 313 chemical or chemical category

that does not have a significant heating value (except metals and metal compounds) is sent off site for energy recovery, it must be considered off-site waste treatment (see 4.1.3.g). However, this does not apply to metals and metal compounds. Metals and metal compounds sent off site for combustion in energy recovery units must be considered as sent off site for disposal because typically they will ultimately be disposed. Metals and metal portions of metal compounds will never be treated or combusted for energy recovery. Note that only the metal portion of metal compounds should be reported in the Form R. Also note that “NA” should be reported for EPCRA Section 313 chemicals or chemical categories that do not have a significant heating value. This includes metals, metal portions of metal compounds, halogens, HCFCs, and CFCs.

- (4) **Quantity Recycled On-site (Part II, Section 8.4 of Form R) -** Estimate the quantity of the EPCRA Section 313 chemical or chemical category recycled in wastes on site. This estimate should be the quantity of the chemical or chemical category recycled in the process for which codes were reported in Section 7C. A quantity should be reported in Section 8.4 when a method of on-site recycling is reported in Section 7C and vice versa. To estimate this quantity, you should determine if operating data exist that indicate a recovery efficiency and use that efficiency value combined with throughput data to calculate an estimate. If operating data are unavailable, available vendor specifications may be appropriate.
- (5) **Quantity Recycled Off-site (Part II, Section 8.5 of Form R) -** The quantity reported in Section 8.5 must be the same as the quantity reported in Section 6.2 for which recycling codes are reported. If a quantity is reported in Section 8.8, subtract any associated off-site transfers for recycling.

If the facility has knowledge about metals being recovered, this quantity should be reported in Section 8.5.

$$\text{\$8.5} = \text{\$6.2 (recycling)} - \text{\$8.8 (off-site recycling)}$$

**COMMON ERROR - Direct Reuse vs. Recycling**

The direct reuse of an EPCRA Section 313 chemical does not need to be included in the amount reported in Part II, Section 8 of Form R. However, recycling of the chemical should be included.

- (6) **Quantity Treated On-site (Part II, Section 8.6 of Form R) -** Waste treatment in Section 8 is limited to the destruction or chemical conversion of the EPCRA Section 313 chemical or chemical category in wastes. The quantities reported in Section 8.6 will be those that have undergone processes that are a subset of the processes for which codes were reported in Section 7A, where

treatment includes physical removal from a waste stream. To estimate the quantity treated, you should determine if operating data exist that indicate a treatment efficiency (e.g., destruction or chemical conversion of the EPCRA Section 313 chemical or chemical category) and use that efficiency value combined with throughput data to calculate an estimate. Because metals cannot be destroyed or chemically converted into something other than the metal or metal compound, metals cannot be reported as treated in Section 8.6. Note that conversion of a metal from one oxidation state to another (e.g., Cr(VI) to Cr(III)) is not considered treatment for Section 8.6. If operating data are unavailable, available vendor specifications may be appropriate. Section 7A must be completed if a quantity is entered in Section 8.6.

- (7) **Quantity Treated Off-site (Part II, Section 8.7 of Form R)** - The quantity reported in Section 8.7 must be the same as the quantity reported in Section 6.2 for which treatment codes are reported plus quantities sent to a POTW as reported in Section 6.1 except for metals and metal compounds. If a quantity is reported in Section 8.8, subtract any associated off-site transfers for treatment:

$$\text{\$8.7} = \text{\$6.1 (except metals and metal compounds)} + \text{\$6.2 (treatment)} - \text{\$8.8 (off-site treatment)}$$

Because metals cannot be destroyed or chemically converted into something other than the metal or metal compound, metals cannot be reported as treated in Section 8.7. Quantities of metals reported in Section 6.1 and 6.2 should be reported in Section 8.1 (Quantity Released) unless the facility has knowledge that the metal is being recovered.

- (8) **Quantity Released to the Environment as a Result of Remedial Actions, Catastrophic Events, or One-Time Events Not Associated with Production Processes (Part II, Section 8.8 of Form R)** - The purpose of this section is to separate quantities recycled off site, used for energy recovery off site, treated off site, or released (including disposed) that are associated with normal or routine production from those quantities that are not. The quantity reported in Section 8.8 is the quantity of the EPCRA Section 313 chemical or chemical category released directly into the environment or sent off site for recycling, energy recovery, treatment, or disposal during the reporting year because of any of the following events:

- Remedial actions;
- Catastrophic events such as earthquakes, fires, or floods; or
- One-time events not associated with normal or routine production processes.

The quantity reported in Section 8.8 should not be included with quantities reported in Part II, Sections 8.1 through 8.7 of Form R, but should be included in Part II, Sections 5 and 6 of Form R as appropriate.

Spills that occur as a routine part of production operations and could be reduced or eliminated by improved handling, loading, or unloading procedures are included in the quantities reported in Section 8.1 through 8.7 as appropriate. This includes small drippings and spills that often occur during transfer operations and loading/unloading operations associated with leather tanning and finishing processes.

On-site releases and off-site transfers for further waste management from remediation of an EPCRA Section 313 chemical or chemical category or an unpreventable accident unrelated to production (such as a hurricane) are reportable in Section 8.8.

On-site treatment, energy recovery, or recycling of EPCRA Section 313 chemicals or chemical categories in wastes generated as a result of remedial actions, catastrophic events, or one-time events not associated with production processes are not reported in Part II, Section 8.8, nor in Sections 8.1 through 8.7 of Form R.

#### **COMMON ERROR - Double Counting**

Release and other waste management activities should not be “double counted.” A single wastewater discharge should not be listed as both a release to water (on site) and a discharge to POTW (off site). Similarly, a release to land should not be listed as both a release to land (on site) and a transfer to an off-site landfill. Estimates of release and other waste management activities should be prepared for Sections 5 through 7 of the Form R. For the most part, Section 8 relies on the data collected to complete these previous sections. Therefore, Section 8 should be completed last. However, the data elements of Section 8 (8.1 through 8.7) are mutually exclusive and care should be taken to avoid double counting.

#### **4.1.4 Step 4: Determine the Most Appropriate Method(s) and Calculate the Estimates for Release and Other Waste Management Activity Quantities**

After you have identified all of the potential sources for release and other waste management activity types, you must estimate the quantities of each EPCRA Section 313 chemical and chemical category released and otherwise managed as waste. EPA has identified four basic methods that may be used to develop estimates (each method has been assigned a code that must be included when reporting). The methods and corresponding codes are:



- Monitoring Data or Direct Measurement (M);
- Mass Balance (C);
- Emission Factors (E); and,
- Engineering Calculations (O).

Descriptions of these techniques are provided in the U.S. EPA publication, *Estimating Releases and Waste Treatment Efficiencies for the Toxic Chemical Release Inventory Forms* (1999 edition). They are also briefly described below. A more detailed discussion including examples of selected calculation techniques is presented in Appendix B. U.S. EPA does not require you to conduct additional sampling or testing for EPCRA Section 313 reporting; however, you are required to use the best readily available information to determine the method that will result in the most accurate estimate. For example, it may not be appropriate to use emission factors or engineering calculations if more accurate data, such as stack testing results, are available. You are required to identify the primary method used for each estimation.

Many potential sources of data exist for these (and other) methods of developing estimates. Table 4-2 presents potential data sources and the estimation methodology in which they are most likely to be used. Based on site-specific knowledge and potential data sources available, you should be able to determine the best method for calculating each release and other waste management activity quantity.

Once all potential release and other waste management activity sources, types, and estimation methods have been determined, an estimate for each EPCRA Section 313 chemical and chemical category can be developed corresponding to the elements on Form R.

**Table 4-2**

**Potential Data Sources for Release and Other Waste Management Calculations**

<b>DATA SOURCES</b>	
<b><u>Monitoring Data</u></b>	<b><u>Mass Balance</u></b>
<ul style="list-style-type: none"><li>• Air permits</li><li>• Continuous emission monitoring</li><li>• Effluent limitations</li><li>• Hazardous waste analysis</li><li>• Industrial hygiene monitoring data</li><li>• NPDES permits</li><li>• New Source Performance Standards</li><li>• Outfall monitoring data</li><li>• pH for acids and bases</li><li>• POTW pretreatment standards</li><li>• RCRA permit</li><li>• Stack monitoring data</li><li>• Title V permit data</li></ul>	<ul style="list-style-type: none"><li>• Air emissions inventory</li><li>• Hazardous material inventory</li><li>• Hazardous waste manifests</li><li>• MSDSs</li><li>• Pollution prevention reports</li><li>• Spill event records</li><li>• Supply and purchasing records</li></ul>
<b><u>Emission Factors</u></b>	<b><u>Engineering Calculations</u></b>
<ul style="list-style-type: none"><li>• AP-42 chemical specific emission factors</li><li>• Facility or trade association derived <u>chemical-specific</u> emission factors</li></ul>	<ul style="list-style-type: none"><li>• Facility <u>non-chemical specific</u> emission factors.</li><li>• Henry's Law</li><li>• Raoult's Law</li><li>• SOCMI* or trade association non-chemical specific emission factors</li><li>• Solubilities</li><li>• Volatilization rates</li></ul>

\*Synthetic Organic Chemicals Manufacturing Industry.

**4.1.4.1 Monitoring Data or Direct Measurement (code M)**

Using monitoring data or direct measurements is usually the best method for developing chemical release and other waste management activity quantity estimates. Your facility may be required to perform monitoring under provisions of the Clean Air Act (CAA), Clean Water Act (CWA), RCRA, or other regulations. If so, data should be available for developing estimates. Data may have also been collected for your facility through an occupational health and safety assessment. If only a small amount of direct measurement data is available or if you believe the monitoring data are not representative, you must decide if another estimation method would give a more accurate result.

### Example - Monitoring Data

Data from the on-site wastewater treatment facility indicate that the annual average concentration of chromium in the discharge is 8 mg/L. The wastewater treatment facility processed 1.5 million gallons of water in the reporting year. The treated wastewater is discharged to an off-site POTW. The amount of chromium transferred off site to the POTW is estimated as follows:

Amount of chromium transferred

$$= (8 \text{ mg/L}) \times \left( \frac{\text{g}}{1,000 \text{ mg}} \right) \times \left( \frac{\text{lb}}{453.59 \text{ g}} \right) \times \left( \frac{\text{L}}{0.2642 \text{ gal}} \right) \times (1,500,000 \text{ gal/yr})$$
$$= 100 \text{ lb/yr}$$

The 100 pounds per year of chromium transferred to the POTW should be reported in Part II, Sections 6.1 and 8.1 of the 1999 Form R.

### COMMON ERROR - Treatment Efficiencies

Vendor data on treatment efficiencies often represent ideal operating conditions. You should adjust such data to account for downtime and process upsets during the year that would result in lower efficiencies. Remember that efficiencies reported by vendors are often general and may not apply to specific chemicals. For example, an incinerator or flare may be 99.99% efficient in destroying certain organic chemicals, but will have a 0% efficiency in destroying metals.

#### **4.1.4.2 Mass Balance (code C)**

A mass balance involves determining the amount of an EPCRA Section 313 chemical or chemical category entering and leaving an operation. The mass balance is written as follows:

$$\text{Input} + \text{Generation} = \text{Output} + \text{Consumption}$$

where:

- Input refers to the materials (chemicals) entering an operation. For example, chlorine added to process water as a disinfectant would be considered an input to the water treatment operation.
- Generation identifies those chemicals created during an operation (manufactured, including coincidental manufacturing). For example, when nitrogen sources are used in biological wastewater treatment systems, nitrate compounds and additional ammonia may be coincidentally manufactured.

- Output refers to the materials (chemicals) leaving an operation by various avenues. Output (avenues) may include on-site release and other on-site waste management activities; transfers off site for recycling, energy recovery, treatment, storage, or disposal; or the amount of chemical that leaves with the final product. In a leather coloring operation, for example, pigments in the paint may leave the operation as part of the product (the coating), in air emissions from a paint spray booth, and on paint spray booth filters sent for disposal.
- Consumption refers to the amount of chemical converted to another substance during the operation (i.e., reacted).

The mass balance technique may be applied toward manufactured, processed, or otherwise used chemicals and chemical categories. It is typically most useful for otherwise used chemicals or chemical categories that do not become part of the final product, such as catalysts. For large inputs and outputs, a mass balance may not be the best estimation method, because slight uncertainties in mass calculations can yield significant errors in the release and other waste management estimates.

#### **Example - Mass Balance**

A facility otherwise uses a volatile EPCRA Section 313 chemical as a refrigerant and adds 20,000 pounds to the refrigeration system (to make up for system losses). The chemical is released to the air from relief vents, during system filling operations, and from leaks in valves and fittings. During system maintenance, the lines are bled directly into water and the system is vented to the air. Monitoring data of the wastewater, including chemical concentrations and wastewater throughput, indicate that 1,200 pounds of the chemical were discharged to the wastewater. The remaining losses are assumed to be fugitive air releases and are estimated as follows:

Fugitive air releases of the EPCRA Section 313 chemical:

$$\begin{aligned} &= \text{Amount input (lb/yr)} - \text{Amount released to wastewater (lb/yr)} \\ &= 20,000 \text{ lb/yr} - 1,200 \text{ lb/yr} \\ &= 18,800 \text{ lb/yr} \end{aligned}$$

### **COMMON ERROR - Mass Balances for Otherwise Used Chemicals**

Facilities often do not account for the entire quantity of EPCRA Section 313 chemicals or chemical categories that are otherwise used. Many EPCRA Section 313 chemicals and chemical categories in leather tanning and finishing operations are classified as otherwise used. Such chemicals and chemical categories may or may not leave the facility with the product. For those instances where the EPCRA Section 313 chemical or chemical category does not leave the facility in the product, all throughput may be lost during processing through on-site releases to air, water, or land, or it may be shipped off site for further waste management activities. Thus, the entire throughput is often reportable on Form R as releases and other waste management activities to various media. Be sure to consider the entire throughput in these circumstances and partition it as appropriate. A mass balance may be the best starting point to estimate the releases and other waste management quantities. Examples applicable to leather tanning and finishing include triethylamine (CAS Registry No. 121-44-8) and some glycol ethers.

#### **4.1.4.3 Emission Factors (code E)**

An emission factor is a representative value that attempts to relate the quantity of a chemical or chemical category released with an associated activity. These factors are usually expressed as the weight of chemical or chemical category released divided by a unit weight, volume, distance, or duration of the activity releasing the chemical (e.g., pounds of chemical released per pounds of product produced). Emission factors, commonly used to estimate air emissions, have been developed for many different industries and activities. You should carefully evaluate the source of the emission factor and the conditions for its use to determine if it is applicable to the situation at your facility. If there are more than one EPA published emission factors, you should determine which is the most appropriate for your operations and document your rationale.

The most widely known and used source for emission factors is U.S. EPA's publication *Compilation of Air Pollutant Emission Factors (AP-42)*. Volume I of AP-42 contains information on over 200 stationary source categories, including process descriptions and potential sources of air emissions from these processes. Methodologies for estimating the quantity of air pollutant emissions from these sources are presented as Emission Factors. For EPCRA Section 313 purposes only CHEMICAL-SPECIFIC emission factors can be reported as Code "E" - Emission Factor in Part II, Section 5, Column B, Basis for estimate, of the Form R. AP-42 contains emission factors for individual chemicals and for the chemical group Volatile Organic Compounds (VOCs). The VOC emission factors are NOT chemical specific and when used must be reported in Column B as Code "O" - Engineering Calculations. Each chapter in

Volume I covers a major industry or source category. Of special interest to leather tanning and finishing facilities are Chapter 4: Evaporation Loss Sources (in particular Sections 4.2, Surface Coating; and 4.3, Wastewater Collection, Treatment, and Storage), Chapter 6: Organic Chemical Process Industry (in particular Section 6.18, Benzene, Toluene, and Xylenes), Chapter 7: Liquid Storage Tanks, and Section 9.15, Leather Tanning..

AP-42 can be accessed at the following Internet site:

- **<http://www.epa.gov/ttn/chief/ap42.html>**

In an effort to provide current emissions data in an easy-to-access format, U.S. EPA has prepared a CD-ROM entitled Air CHIEF (Air Clearing House for Inventories and Emission Factors). The Air CHIEF CD-ROM is updated annually and is available from the Government Printing Office, and can be ordered from their Web site. In addition to AP-42, the Air CHIEF CD-ROM contains the Factor Information Retrieval (FIRE) data system, a database management system containing U.S. EPA's recommended emission estimation factors for criteria and hazardous air pollutants. The CD-ROM also contains installable copies of software programs for air emission estimation models such as "TANKS" for VOC emission from storage tanks; "WATER8" for air emissions from wastewater systems; and "CHEMDAT8" for VOC emissions from Treatment, Storage, and Disposal Facility (TSDF) processes. Additional information on Air CHIEF and the CD-ROM is available at:

- **<http://www.epa.gov/ttn/chief/airchief.html>**

Your facility may have developed non-chemical-specific emission factors for fugitive or stack emissions based on stack tests for various air permits. Be sure to consider these emission factors if appropriate. However, if such factors are used, they are considered "engineering calculations" for the purposes of EPCRA Section 313 reporting.

### **Example - Emission Factors**

Emission factors have been developed for air releases of fuel constituents and combustion products from boiler operations. AP-42 lists a range of formaldehyde emission factors when No. 6 fuel oil is consumed:

0.024 to 0.061 lb formaldehyde generated/10<sup>3</sup> gal No. 6 fuel oil fired.

Assuming a facility met reporting requirements for formaldehyde, the facility operating a boiler using No. 6 fuel oil could use the above emission factor to determine the amount of formaldehyde generated and subsequently released to the air. If 1,000,000 gallons of No. 6 fuel oil is used during a reporting year, the amount of formaldehyde generated would be between:

$$(0.024 \text{ lb}/10^3 \text{ gal}) \times (1,000,000 \text{ gal}) \text{ and } (0.061 \text{ lb}/10^3 \text{ gal}) \times (1,000,000 \text{ gal})$$
$$= 24 \text{ and } 61 \text{ lb of formaldehyde generated}$$

If there are no engineering controls or air pollution control devices that would destroy or remove the formaldehyde, this quantity should be reported in Part II, Sections 5.2 and 8.1 of the 1999 Form R.

NOTE: No. 6 fuel oil contains other EPCRA Section 313 chemicals and chemical categories and EPCRA Section 313 chemicals and chemical categories may also be coincidentally manufactured during combustion. All should be considered for EPCRA Section 313 reporting.

#### **4.1.4.4 Engineering Calculations (code O)**

Engineering calculations are assumptions and/or judgments used to estimate quantities of EPCRA Section 313 chemicals and chemical categories released or otherwise managed as waste. The quantities are estimated by using physical and chemical properties and relationships (e.g., Ideal Gas law, Raoult's law) or by modifying an emission factor to reflect the chemical properties of the chemical in question. Engineering calculations rely on the process parameters; you must have a thorough knowledge of your facility operations to complete these calculations.

## Examples - Engineering Calculations

### Example 1:

Stack monitoring data are available for xylene but you are required to report for toluene. Toluene is used in the same application as xylene at your facility and the concentrations of the chemicals in the liquid feedstock are approximately the same. You can estimate the emissions of toluene by adjusting the monitoring data of xylene by a ratio of the vapor pressure for xylene to toluene. This example is an engineering calculation based on physical properties and process operation information:

From facility stack monitoring data, you determine that an estimated 200 lb of xylene are released as air emissions during the reporting year. Toluene is also present in the air emissions, but not monitored. The stack operates at approximately 20°C. Based on literature data, the vapor pressure at 20°C for toluene is 22 millimeters of mercury (mmHg) and for xylene is 6 mmHg. Using a ratio of the vapor pressures, the amount of toluene released as air emissions from the stack can be calculated:

$$\begin{aligned} \frac{X \text{ lb/yr toluene}}{200 \text{ lb/yr xylene}} &= \frac{22 \text{ mmHg (vapor pressure of toluene)}}{6 \text{ mmHg (vapor pressure of xylene)}} \\ X \text{ lb/yr toluene} &= \frac{(200 \text{ lb/yr xylene}) (22 \text{ mmHg toluene})}{(6 \text{ mmHg xylene})} \end{aligned}$$

Completing the calculation, you determine that 730 lbs of toluene were released as stack air emissions during the reporting year. The quantity of toluene released should be reported in Section 5.2 of the 1999 Form R.

### Example 2:

A leather coloring process uses 10,000 gallons per year of a paint that is 3% xylene by volume. All of the xylene in the paint is assumed to evaporate during the coating operation. The coloring process is equipped with a fume collection hood that captures 80% of the paint vapors. The remaining 20% of the paint vapors are assumed to be released as fugitive air emissions. The collection hood routes the paint vapors to an incinerator that is vented to the atmosphere and has a destruction efficiency of 99% for xylene. The specific gravity of xylene is 0.86 and the density of water is 8.34 lb/gal. Fugitive air emissions and stack air emissions may be estimated as follows:

1. The total amount of xylene volatilized to air (assumed to be the total amount of xylene in paint)  
$$= (10,000 \text{ gal/yr paint}) \times (0.03, \text{ three percent xylene}) \times (0.86, \text{ xylene specific gravity}) \times (8.34 \text{ lb/gal, density of water})$$
$$= 2,152 \text{ lb/yr xylene evaporated from coloring operations}$$
2. The amount of xylene released as fugitive air emissions  
$$= (2,152 \text{ lb/yr}) \times (0.20; \text{ twenty percent released as fugitive air emissions})$$
$$= 430 \text{ lb/yr}$$

This should be reported in Part II, Section 5.1 and 8.1 of the 1999 Form R.

3. The amount of xylene released as stack air emissions  
$$= (2,152 \text{ lb/yr}) \times (0.80, \text{ eighty percent capture efficiency}) \times (1.0 - 0.99, \text{ percent not incinerated})$$
$$= 17 \text{ lb/yr}$$

This should be reported in Part II, Section 5.2 and 8.1 of the 1999 Form R.



Engineering calculations can also include computer models. Several computer models are available for estimating emissions from landfills, wastewater treatment, water treatment, and other processes.

Non-chemical-specific emission factors, Synthetic Organic Chemicals Manufacturing Industry (SOCMI) emission factors, industry-determined emission factors for processes or equipment, and site-specific emission factors also can be used, but must be classified as “Engineering Calculations” for EPCRA Section 313 reporting.

#### **4.1.4.5 Estimating Release and Other Waste Management Quantities**

Once all sources, types, and appropriate estimation methodologies have been identified, you can estimate the release and other waste management activity quantities of EPCRA Section 313 chemicals or chemical categories for each element of the Form R. The recommended approach is that you estimate amounts from all sources at your facility to each type as identified by the elements of Form R. Table 4-3 presents a worksheet that may be helpful in compiling this information.

If you prepare a Form R, you must also enter on-site treatment information in Section 7A, including the code for each treatment method used, the destruction and removal efficiency for the EPCRA Section 313 chemical or chemical category in the treated waste stream, and the concentration of the EPCRA Section 313 chemical or chemical category in the influent to treatment. You should report treatment methods that do not actually destroy or remove the chemical or chemical category by entering “zero (0)” for removal efficiency. Similarly, on-site energy recovery methods and on-site recycling methods must be reported in Sections 7B and 7C, respectively.

**Table 4-3**

**Release and Other  
Waste Management Quantity Estimation Worksheet**

Facility Name: \_\_\_\_\_

Date Worksheet Prepared: \_\_\_\_\_

EPCRA Section 313 Chemical or Chemical Category: \_\_\_\_\_

Prepared by: \_\_\_\_\_

CAS Registry Number: \_\_\_\_\_

Reporting Year: \_\_\_\_\_

<b>ON SITE</b>			
<b>Release or Other Waste Management Activity Type</b>	<b>Amount (lb)</b>	<b>Basis of Estimate</b>	<b>Form R Element* (1999 version)</b>
<b>FUGITIVE AIR</b>			
Equipment Leaks			5.1 and 8.1 or 8.8
Process Areas			5.1 and 8.1 or 8.8
Evaporative Losses, Spills, Surface Impoundments			5.1 and 8.1 or 8.8
Total =			5.1 and 8.1 or 8.8
<b>STACK AIR</b>			
Process Vents			5.2 and 8.1 or 8.8
Storage Tanks			5.2 and 8.1 or 8.8
Control Device Stacks			5.2 and 8.1 or 8.8
Other			5.2 and 8.1 or 8.8
Total =			5.2 and 8.1 or 8.8
<b>RECEIVING STREAM/WATER BODY DISCHARGE</b>			
Stormwater Discharge			5.3 and 8.1 or 8.8
On-Site Treatment Plant Discharge			5.3 and 8.1 or 8.8
Total =			5.3 and 8.1 or 8.8
<b>ON-SITE UNDERGROUND INJECTION</b>			
Underground Injection to Class I Wells			5.4 and 8.1 or 8.8
Underground Injection to Class II - V Wells			5.4 and 8.1 or 8.8
Total =			5.4 and 8.1 or 8.8

\*Entries for Section 8.8 only if release is result of remedial action, catastrophic event, or one-time event not associated with production process.

**Table 4-3 (Continued)**

<b>ON SITE</b>			
<b>Release or Other Waste Management Activity Type</b>	<b>Amount (lb)</b>	<b>Basis of Estimate</b>	<b>Form R Element* (1999 version)</b>
<b>ON-SITE LAND</b>			
RCRA Subtitle C Landfill			5.5 and 8.1 or 8.8
Other Landfill			5.5 and 8.1 or 8.8
Land Treatment/Application Farming			5.5 and 8.1, 8.6, or 8.8
Surface Impoundment			5.5 and 8.1 or 8.8
Other Disposal			5.5 and 8.1 or 8.8
Total =			5.5 and 8.1 or 8.8
<b>ON-SITE ENERGY RECOVERY</b>			
Industrial Kiln			8.2
Industrial Furnace			8.2
Industrial Boiler			8.2
Other Energy Recovery Methods			8.2
Total =			8.2
<b>ON-SITE RECYCLING</b>			
Solvents/Organics Recovery			8.4
Metals Recovery			8.4
Acid Regeneration			8.4
Other Reuse or Recovery			8.4
Total =			8.4
<b>ON-SITE TREATMENT</b>			
Air Emissions Treatment			8.6
Biological Treatment			8.6
Chemical Treatment			8.6
Incineration/Thermal Treatment			8.6
Physical Treatment			8.6
Solidification/Stabilization			8.6
Total =			8.6

\*Entries for Section 8.8 only if release is result of remedial action, catastrophic event, or one-time event not associated with production process.

**Table 4-3 (Continued)**

<b>OFF SITE</b>				
<b>Release or Other Waste Management Activity Type</b>	<b>Amount (lb)</b>	<b>Basis of Estimate</b>	<b>Form R Element* (1999 version)</b>	<b>Off-Site Location (name)</b>
<b>OFF-SITE DISPOSAL</b>				
Solidification/Stabilization (metals and metal compounds only)			6.2 and 8.1 or 8.8	
Amount of metal and metal compounds to POTW			6.1 and 8.1 or 8.8	
Wastewater Treatment (excluding POTWs) metals and metal compounds only			6.2 and 8.1 or 8.8	
Underground Injection			6.2 and 8.1 or 8.8	
Landfill/Surface Impoundment			6.2 and 8.1 or 8.8	
Land Treatment			6.2 and 8.1 or 8.8	
Other Land Disposal			6.2 and 8.1 or 8.8	
Other Off-Site Management			6.2 and 8.1 or 8.8	
Total =			6.2 and 8.1 or 8.8	
<b>OTHER AMOUNTS SENT OFF SITE</b>				
Amounts sent for storage			6.2 and 8.1 or 8.8	
Amounts sent for unknown waste management practice			6.2 and 8.1 or 8.8	
Total =			6.2 and 8.1 or 8.8	
<b>OFF-SITE TREATMENT</b>				
Solidification/Stabilization			6.2 and 8.7 or 8.8	
Incineration/Thermal Treatment			6.2 and 8.7 or 8.8	
Incineration/Insignificant Fuel Value			6.2 and 8.7 or 8.8	
Wastewater Treatment (to POTW excluding metals and metal compounds)			6.1 and 8.7 or 8.8	
Wastewater Treatment (excluding POTW and metal and metal compounds)			6.2 and 8.7 or 8.8	
Sent to Waste Treatment Broker			6.2 and 8.7 or 8.8	
Total =			6.2 and 8.7 or 8.8	

\*Entries for Section 8.8 only if release is result of remedial action, catastrophic event, or one-time event not associated with production process.

**Table 4-3 (Continued)**

<b>OFF SITE</b>				
<b>Release or Other Waste Management Activity Type</b>	<b>Amount (lb)</b>	<b>Basis of Estimate</b>	<b>Form R Element* (1999 version)</b>	<b>Off-Site Location (name)</b>
<b>OFF-SITE ENERGY RECOVERY</b>				
Off-Site Energy Recovery			6.2 and 8.3 or 8.8	
Sent to Energy Recovery Broker			6.2 and 8.3 or 8.8	
Total =			6.2 and 8.3 or 8.8	
<b>OFF-SITE RECYCLING</b>				
Solvents/Organics Recovery			6.2 and 8.5 or 8.8	
Metals Recovery			6.2 and 8.5 or 8.8	
Other Reuse or Recovery			6.2 and 8.5 or 8.8	
Acid Regeneration			6.2 and 8.5 or 8.8	
Sent to Recycling Waste Broker			6.2 and 8.5 or 8.8	
Total =			6.2 and 8.5 or 8.8	

\*Entries for Section 8.8 only if release is result of remedial action, catastrophic event, or one-time event not associated with production process.

## 4.2 Determination of Release and Other Waste Management Quantities from Leather Tanning and Finishing Operations

Leather tanning and finishing converts raw hides and skins into leather that has thermal stability, is soft and flexible, and non-putrescible. Current leather tanning and finishing operations involve many chemical and mechanical operations. The processes involved in leather tanning and finishing will be discussed in five major groupings; detailed process flow diagrams are presented in the subsequent discussion on each set of unit operations.

- Beamhouse operations – receiving and storage of hides, soaking, unhairing, reliming, delimiting, and bating;
- Tanyard operations – pickling, tanning, wringing, sorting, trimming, siding, splitting, and shaving;
- Retanning, coloring and fatliquoring operations;
- Finishing operations – setting out, drying, conditioning, staking, dry milling, buffing, finishing, plating, grading, and measuring; and
- Wastewater treatment – pre-treatment and treatment.

For discussion purposes, some of the individual process steps have been combined. Not all leather tanning facilities will have all operations and facilities may vary in the sequence of operations. You should analyze the process flow in your facility and prepare a site-specific process flow diagram showing the individual operations present in your facility.

In 1995 the five EPCRA Section 313 chemicals and chemical categories most commonly reported by facilities in SIC Code 3111 were chromium compounds, formic acid, ammonia, certain glycol ethers, and chromium. Table 2-2 provides a complete list of EPCRA Section 313 chemicals and chemical categories commonly reported by SIC Code 3111 facilities in 1995.

Typical sources of release and other waste management activities of EPCRA Section 313 chemicals and chemical categories in leather tanning facilities include wastewater from several of the wet operations such as unhairing, delimiting, tanning, wringing and sorting, retanning, coloring, and fatliquoring. Various dry operations such as buffing and finishing release air pollutants in dust form, while processes involving solvents may release volatile organic chemicals (VOCs) and other chemicals to the air through evaporation. Scraps and shavings from trimming, siding, splitting, shaving, buffing and finishing activities may contribute

to solid wastes, along with shipping container residues. On-site wastewater treatment systems that are typically observed at these facilities generate sludge that may contain EPCRA Section 313 chemicals or chemical categories.

Release and other waste management activity types for EPCRA Section 313 chemicals and chemical categories include stack and fugitive air, wastewater discharge either direct to a receiving stream or off site to a POTW, and land or off-site disposal of solid wastes.

#### **4.2.1 Beamhouse Operations**

In a typical tannery, beamhouse operations begin when the raw hides are received (generally fresh (green) hides are transported to the tannery in a refrigerated truck). They are normally placed in a large, cool, well-ventilated storage facility. The hides usually have been cured with a concentrated salt (sodium chloride) solution to prevent putrefaction. The hides are sorted, weighed and assembled in packs for further processing. When the hides are ready for processing they are soaked to remove the salt and restore the moisture. Wetting agents and bactericides are typically added in the soaking step. After soaking, hair, epidermis, and soluble proteins are removed from the hides by chemical and sometimes mechanical means. The final step in the beamhouse operations is to remove residual unhairing chemicals and any remaining non-leather making substances from the hides in the bating process. (Ref: *Leather Facts*, Third Edition. New England Tanners Club, Peabody, MA. 1994.)

Anhydrous ammonia is a prevalent EPCRA Section 313 chemical in this industry. There are two sources of aqueous ammonia in beamhouse operations. The first source is the proteins removed in the soaking and unhairing steps which can be converted to ammonia. The second source is the addition of ammonia salts, ammonium chloride and ammonium sulfate, in the delimiting step. In both cases the ammonia that is generated is considered to be the result of the coincidental manufacturing of ammonia. Both operations take place in water and the resulting ammonia would thus be considered aqueous ammonia as described in the U.S. EPA publication, EPA 745-R-95-012, *Emergency Planning and Community Right-to-Know, EPCRA Section 313, Guidance for Reporting Aqueous Ammonia*. (see Appendix D)

Total aqueous ammonia is the sum of the two forms of ammonia, anhydrous ammonia ( $\text{NH}_3$ ) and ammonium ion ( $\text{NH}_4^+$ ). The amount of each form of ammonia found in a solution of ammonia is a function of the pH and temperature of that solution.

### **The First Source of Ammonia – Unhairing and Liming**

The high pH typically found in the unhairing and liming water solutions, pH ~12.0-12.6, will result in the predominant manufacture of aqueous ammonia in this step. A small portion of aqueous ammonia involved in the process is released to the atmosphere as anhydrous ammonia. The following information is quoted from the above ammonia guidance publication.

“If the source of aqueous ammonia is anhydrous ammonia in water, total aqueous ammonia (calculated in terms of  $\text{NH}_3$  equivalents) is equal to the quantity of anhydrous ammonia manufactured, processed, or otherwise used...

If a facility manufactures, processes or otherwise uses aqueous ammonia, the quantity applied toward threshold determinations for the ammonia listing is 10 percent of the total quantity of the aqueous ammonia manufactured processed or otherwise used. The quantity reported when calculating the amount of ammonia that is released, transferred or otherwise managed is 10 percent of the total quantity of the aqueous ammonia released or transferred.”

### **The Second Source of Ammonia – Addition of Ammonia Salts**

The following information is quoted from the above ammonia guidance publication.

“Water dissociable ammonium salts are not reportable in their entirety under the ammonia listing; these salts are reportable to the extent that they dissociate in water, and only 10 percent of the total aqueous ammonia that results when these salts dissociate is reportable...

If a facility dissolves a water dissociable salt in water that facility has manufactured aqueous ammonia and 10 percent of the total aqueous ammonia manufactured

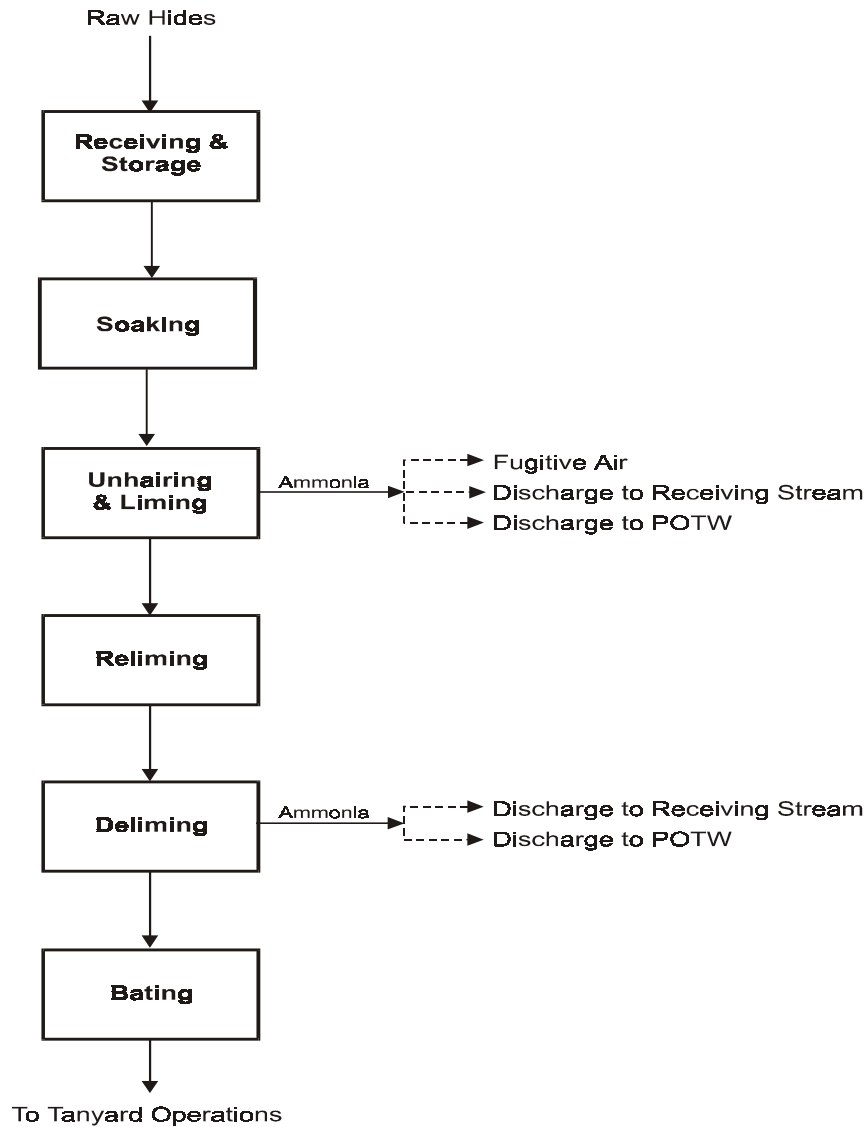


from these salts is to be included in the manufacturing threshold determinations under the ammonia listing.”

For additional guidance please see the copy of the complete ammonia guidance publication included as Appendix D to this document and the example in this section.

### **Step 1: Prepare a Process Flow Diagram**

A site-specific process flow diagram should be prepared to help identify all potential sources and types of chemical and chemical category release and other waste management activities. A typical flow diagram is presented in Figure 4-3.



**Figure 4-3. Process Flow – Beamhouse Operations**

**Step 2: Identify EPCRA Section 313 Chemicals and Chemical Categories and Potential Sources of Chemical Release and Other Waste Management Activities**

The leather tanning industry typically uses only four bactericides: o-phenylphenol; 2-(thiocyanomethylthio)benzothiazole; diiodomethyl-p-tolyl sulfone; and 1,2-benzisothiazolin-3-one. Of those, the only EPCRA Section 313 chemical is o-phenylphenol, a synonym for 2-phenylphenol, CAS Registry No. 90-43-7. Anhydrous ammonia may be generated by the deamination of hide substances during unhairing, and ammonium sulfate and ammonium chloride salts are used in deliming (see Appendix D for guidance on reporting for

ammonia). Careful control of pH in beamhouse operations ensures that hydrogen sulfide is not generated.

Typical sources of EPCRA Section 313 chemicals and chemical categories are process wastewaters, evaporation of volatile compounds such as ammonia, and residues in “empty” shipping containers.

### **Step 3: Identify Release and Other Waste Management Activity Types**

Types of release and other waste management activities include wastewater discharges, either direct to a receiving stream or indirect to a POTW, which may contain bactericides and aqueous ammonia; fugitive emissions from volatilization (typically, only anhydrous ammonia is expected); and on-site or off-site management of container residues, which may result in the release or management of wastes that contain EPCRA Section 313 chemicals or chemical categories to on-site or off-site disposal, treatment, energy recovery, or recycling, as appropriate. Note that if your facility has tannery-specific information that suggests some of the ammonia remains in the finished product, this amount is not reportable as part of a release or otherwise waste managed quantity; however, this information will be useful for the mass balance approach to determine this quantity.

### **Step 4: Determine the Most Appropriate Method(s) and Calculate the Estimates for Release and Other Waste Management Activity Quantities**

Monitoring data for on-site wastewater treatment plant permits and NPDES permit requirements can generally provide wastewater concentrations of EPCRA Section 313 chemicals and chemical categories that are directly or indirectly discharged in your facility’s wastewater. Pre-treatment permit compliance monitoring may provide information on wastewater concentrations discharged to POTWs. Appendix B provides an example using NPDES data for estimating the quantity of an EPCRA Section 313 chemical or chemical category directly discharged in wastewater (the concepts of this example also apply to indirect discharge to a POTW). The ammonia will partition itself between the wastewater and the sludge during the treatment process. Amounts in each will be determined by analysis.

The releases and waste management activities associated with anhydrous ammonia are of particular concern for leather tanning and finishing operations, specifically during hide deamination where anhydrous ammonia will be generated (and the subsequent treatment of process water). Site-specific emission factors, or other standardized emission factors for the industry, may be combined with estimates of the anhydrous ammonia that is generated from the process, and the amount of anhydrous ammonia purchased to estimate the fugitive air release and the discharges in wastewater. Appendix G presents emissions factors for the hide deamination process from Leather Industry of America.

If EPCRA Section 313 chemicals or chemical categories are present in any materials brought on site in containers, the liquid residues in empty shipping containers can be estimated using the factors in Table 4-1. For dry materials a residue factor of 1% can be used if actual data are not available (see Section 4.1.3.g for a complete discussion on container residue).

#### **Example - Fugitive Anhydrous Ammonia Release Calculation**

In this example the amount of anhydrous ammonia that is coincidentally manufactured and the amount released as fugitive air emissions and discharged to a POTW in a leather tanning facility is estimated. Per engineering judgement, the amount of ammonia generated is estimated to be 0.15% of the raw hide weight. Also, studies indicated that approximately 3% of the ammonia generated at this facility will be volatilized and released to air. Finally, it is known that 7% of the total ammonia generated and purchased remains with the final product.

A leather tanning facility receives an average of 15,000 brine-cured hides per week. Each brine-cured, pre-fleshed hide weighs approximately 65 pounds. The facility operates 50 weeks per year. In addition, the facility purchased and used 150,000 pounds of ammonia for deliming, neutralization, and dyeing. Using the ammonia generation value of 0.15% of the raw hide weight, the total annual generation of ammonia is calculated to be:

Total ammonia generated:

$$\begin{aligned} &= (15,000 \text{ hides/wk}) \times (50 \text{ wk/yr}) \times (65 \text{ lb/hide}) \times (0.0015, 0.15\%) \\ &= 73,125 \text{ pounds ammonia per year} \end{aligned}$$

The total amount of ammonia dissolved in water is the amount generated plus the amount purchased:

$$\begin{aligned} &= 73,125 \text{ pounds} + 150,000 \text{ pounds} \\ &= 223,125 \text{ pounds per year} \end{aligned}$$

**Amount Released to Air:**

Per the assumed emission factor, 3% of the ammonia generated and purchased will volatilize as an air emission. You must determine whether all, or some of this quantity should be partitioned between fugitive and stack air releases. For this example, it is assumed that 100% of the air release is fugitive:

$$\begin{aligned} &= (223,125 \text{ pounds}) \times (0.03, 3\% \text{ factor}) \\ &= 6,694 \text{ pounds per year} \end{aligned}$$

The fugitive air releases of ammonia will occur primarily in the beamhouse because of the high pH, a range of 12.2 to 12.5, at which the unhairing operations are conducted. The addition of ammonium salts in the delimiting step rapidly drops the pH to around 8.5 and the release of ammonia to the atmosphere decreases to essentially zero. Also, because of the pH at which typical wastewater treatment plants operate, pH near 7, there are expected to be minimal if any fugitive air release of ammonia if an on-site wastewater treatment plant was used. The fugitive air releases should be reported in Part II, Section 5.1 and included in Section 8.1 of the 1999 Form R.

**Amount Discharged to POTW:**

7% of the ammonia generated and purchased remains with the final product. Using a mass balance approach, this amount combined with the quantity released to air (calculated above) should be subtracted from the total ammonia with the remainder being the quantity sent to the POTW.

Quantity remaining with product:

$$\begin{aligned} &= (223,125 \text{ pounds}) \times (0.07, 7\% \text{ factor}) \\ &= 15,619 \text{ pounds per year} \end{aligned}$$

Quantity sent to POTW:

$$\begin{aligned} &= (\text{total ammonia generated and purchased}) - (\text{air emissions}) - (\text{ammonia in product}) \\ &= (223,125 \text{ pounds}) - (6,694 \text{ pounds}) - (15,619 \text{ pounds}) \\ &= 200,813 \text{ pounds per year} \end{aligned}$$

This is the quantity of ammonia in solution that will be discharged to the POTW. As detailed in U.S. EPA's guidance for reporting ammonia for EPCRA Section 313 (Appendix D), 10% of the aqueous ammonia is considered to be reportable. Therefore, the quantity that should be reported in Part II, Section 6.1 (quantity discharged to a POTW) and included in Section 8.7 (quantity treated off site) is:

$$\begin{aligned} &= (200,813 \text{ pounds}) \times (0.01, 10\%) \\ &= 2,008 \text{ pounds per year} \end{aligned}$$

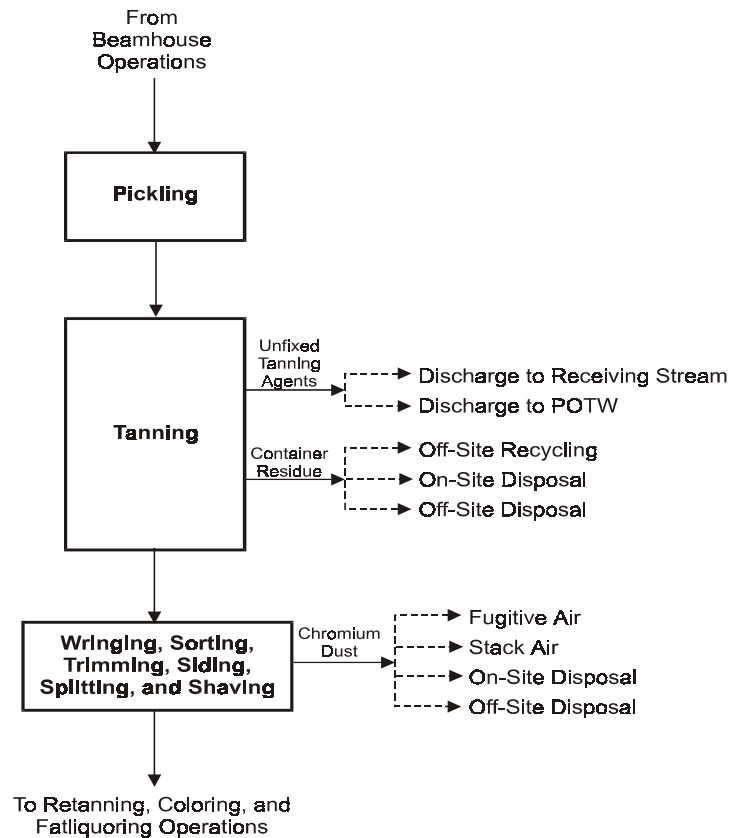
## 4.2.2 Tanyard Operations

The actual tanning of leather takes place in the tanyard operations. Salt and acid are added in the pickling step to provide the low pH environment that will prepare the hides to accept the tanning chemicals in the next step (note that some facilities consider delimiting and salting to be included as the first step, prior to pickling, of tanyard operations). Tanning converts the hides into a stable, non-putrescible material. The industry commonly uses the chrome tanning method in which the chrome tanning agent is introduced into a revolving drum

containing the hides floating in brine. Following the application of the tanning chemicals the excess moisture is removed, perimeter areas are trimmed to remove the less desirable material, and the thickness is adjusted to uniform dimension suitable for the desired end use of the finished leather (Ref: *Leather Facts*, Third Edition. New England Tanners Club, Peabody, MA. 1994).

### Step 1: Prepare a Process Flow Diagram

A site-specific process flow diagram should be prepared to help identify all potential sources and types of chemical and chemical category release and other waste management activities. A typical flow diagram is presented in Figure 4-4.



**Figure 4-4. Process Flow – Tanyard Operations**

## **Step 2: Identify EPCRA Section 313 Chemicals and Chemical Categories and Potential Sources of Chemical Release and Other Waste Management Activities**

The predominant EPCRA Section 313 chemicals and chemical categories involved in tanyard operations are chromium and chromium compounds used in the tanning step and formic acid used in the pickling step. Solutions of sulfuric acid and hydrochloric acid are also used in pickling; however, only the acid aerosol forms of sulfuric and hydrochloric acids are reportable (see Section 3.1). Other EPCRA Section 313 chemicals that may be processed or otherwise used include tetrachloroethylene, to degrease oil-tanned skins, and formaldehyde, occasionally used as a pretannage. The latter has been largely replaced with glutaraldehyde. There is also the possibility of coincidentally manufacturing anhydrous ammonia during pickling and tanning operations.

Typical sources of EPCRA Section 313 chemicals and chemical categories are process wastewaters, evaporation of volatile compounds, dust particles from trimming and shaving operations, and residues in “empty” shipping containers.

## **Step 3: Identify Release and Other Waste Management Activity Types**

Types of releases and other waste management activities include direct and indirect wastewater discharges, which may contain aqueous ammonia, chromium, aluminum, or formaldehyde; fugitive emissions from evaporation of any volatile EPCRA Section 313 chemicals or chemical categories; fugitive and stack emissions of dust containing chromium; and on-site or off-site management of container residues and any dust collected in air pollution control devices, which may result in the release or management of wastes that contain EPCRA Section 313 chemicals or chemical categories to on-site or off-site disposal, treatment, energy recovery, or recycling, as appropriate.

Note that any EPCRA Section 313 chemical or chemical category sent through a pollution control device is considered to have been treated for destruction if it is converted to another chemical or it is HCl or H<sub>2</sub>SO<sub>4</sub> acid aerosols. The treatment efficiency of the unit should be reported in Section 7A and the quantity treated for destruction should be reported in Section 8.6. Also, note that any EPCRA Section 313 chemical or chemical category sent through an air

pollution control device is considered to have been captured for further waste management activities if it is not converted to another chemical or it is not HCl or H<sub>2</sub>SO<sub>4</sub> acid aerosols. The capture efficiency of the unit should be reported in Section 7A and the quantity captured should be reported in Sections 6 and/or 8 depending on the final disposition of the chemical or chemical category.

#### **Step 4: Determine the Most Appropriate Method(s) and Calculate the Estimates for Release and Other Waste Management Activity Quantities**

Monitoring data for on-site wastewater treatment plant permits and NPDES permit requirements can generally provide wastewater concentrations of EPCRA Section 313 chemicals and chemical categories that are directly or indirectly discharged in your facility's wastewater. Example 6 in Appendix B provides an example using NPDES data for estimating the quantity of an EPCRA Section 313 chemical directly discharged in wastewater (the concepts of this example also apply to indirect discharge to a POTW).

Volatile EPCRA Section 313 chemicals and chemical categories, not intended to remain with the product, e.g., tetrachloroethylene used as a degreasing agent, can be assumed to be 100% released as either fugitive or stack emissions, as appropriate, after subtracting any potential container residue. Liquid residues in the shipping containers can be estimated using the factors in Table 4-1. For dry materials a residue factor of 1% can be used if actual data are not available (see Section 4.1.3.g for a complete discussion on container residue).

See Section 4.2.1 and Appendix D for a more detailed discussion of ammonia, if applicable.

#### **Example - Chromium Release and Other Waste Management Activities**

In this example mass balance and engineering calculation methods are used to estimate the release and other waste management quantities of chromium.

On January 1 of the reporting year your facility had in its inventory three hundred and fifty, 55-gallon steel drums containing a 15% (on the basis of Cr<sub>2</sub>O<sub>3</sub>) solution of chromium sulfate. During the year your facility purchased 2,250 drums and on 31 December 600 drums remained in the inventory. The solution is poured out of the open-top drums, and the "empty" drums returned to the supplier to be refilled and sold again to your facility. The solution has physical properties, including density, similar to water.

*[continued on next page]*



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### Threshold Determination

The first step is to determine if your facility has exceeded the processing threshold of 25,000 pounds per year for EPCRA Section 313 reporting. To do so, calculate the amount of chromium compound, in the form of the 15% solution of chromium sulfate, processed by your facility during the reporting year. If other facility operations use chromium compounds such as certain pre-metallized dyes, you must include that amount in any threshold determinations and release and other waste management quantity estimates. Keep in mind that it is the weight of the metal **compound** that is used for the threshold determination, while only the weight of the **parent metal** is estimated in the release and other waste management calculations.

The amount of chromium compound used at the facility during the reporting year is:

$$(350 + 2,250 - 600 \text{ drums}) \times (55 \text{ gal/drum}) \times (8.34 \text{ lb/gal, density of water}) \times (0.15, 15\% \text{ Cr}_2\text{O}_3)$$
$$= 137,610 \text{ pounds chromium compound (as Cr}_2\text{O}_3\text{)}.$$

This exceeds the 25,000 pounds per year threshold for reporting the processing of an EPCRA Section 313 chemical or chemical category. Thus, you must now determine the quantities of chromium released and or otherwise waste managed. The atomic weight of chromium is 52, the atomic weight of oxygen is 16, and the molecular weight of  $\text{Cr}_2\text{O}_3$  is 152.

The amount of chromium processed at the facility is:

$$(137,610 \text{ pounds Cr compound}) \times (104/152, \text{ weight of chromium/weight of Cr}_2\text{O}_3)$$
$$= 94,154 \text{ pounds chromium.}$$

Now you must determine how to calculate the estimates for the release and other waste management activity quantities to be reported. Consider the ways in which chromium leaves your facility:

- As a component of the final product, including the product itself and the leather dusts, shavings, and other scraps generated in the manufacture of the final product;
- In the residue in the “empty” shipping containers returned to the supplier; and
- In the liquid effluent and sludge solids from the wastewater treatment plant.

### Leather product

Based on process knowledge, you estimate that 80% of the chromium is taken up into the leather:

$$(0.80) \times (94,154 \text{ lb/yr total Cr}) = 75,323 \text{ pounds of chromium.}$$

You also estimate that 5% by weight of the leather is lost in the form of dust and scraps in finishing the leather. The dust and scraps are sold to a fertilizer manufacturer for direct reuse. Thus, the entire amount of chromium taken up in the leather, 75,323 pounds, is sold in commerce. You do not have to report it on the Form R.

### “Empty” shipping containers

According to Table 4-1, an average of 0.034% of a liquid with water characteristics remains as the residue in an open-top steel drum, unloaded by pouring. The amount of chromium returned to the supplier in the “empty” drums can be estimated as:

$$(0.00034) \times (94,154 \text{ lb/yr total Cr}) = 32 \text{ pounds of chromium returned to the supplier.}$$

[continued on next page]

*[continued from previous page]*

If you know that the drums are refilled and returned to you without being cleaned or otherwise managed, your facility should not report this amount as a release or other waste management activity on the Form R. However, if they are cleaned, this quantity should be reported as being sent off site for disposal, treatment, energy recovery, or recycling, as appropriate.

The liquid effluent from your facility wastewater treatment plant discharges directly to a receiving stream while the sludge from the plant is disposed on site in a RCRA Subtitle C landfill. The SPDES permit for the effluent requires monitoring for, among other items, flow and chromium concentration. The flow was 150,000,000 gallons for the reporting year and the chromium content averaged 0.75 µg/L.

### **Wastewater treatment plant**

The amount of chromium in the liquid effluent is:

$$(150,000,000 \text{ gal/yr}) \times (0.75 \text{ } \mu\text{g/L Cr concentration}) \times (10\text{E-}9 \text{ kg/}\mu\text{g}) \times (2.2 \text{ lb/kg}) \times (3.78 \text{ L/gal}) \\ = 0.9 \text{ pounds chromium in the liquid effluent.}$$

This quantity should be reported in Part II, Sections 5.3 and 8.1 of the 1999 Form R.

You can now use a mass balance to determine the amount of chromium in the wastewater treatment plant sludge:

$$(94,154 \text{ lb total Cr}) - (75,323 \text{ lb Cr in leather}) - (32 \text{ lb Cr in residue}) - (0.9 \text{ lb Cr in effluent}) \\ = 18,798.1 \text{ pounds chromium in sludge.}$$

This should be reported in Part II, Sections 5.5.1A and 8.1 of the 1999 Form R.

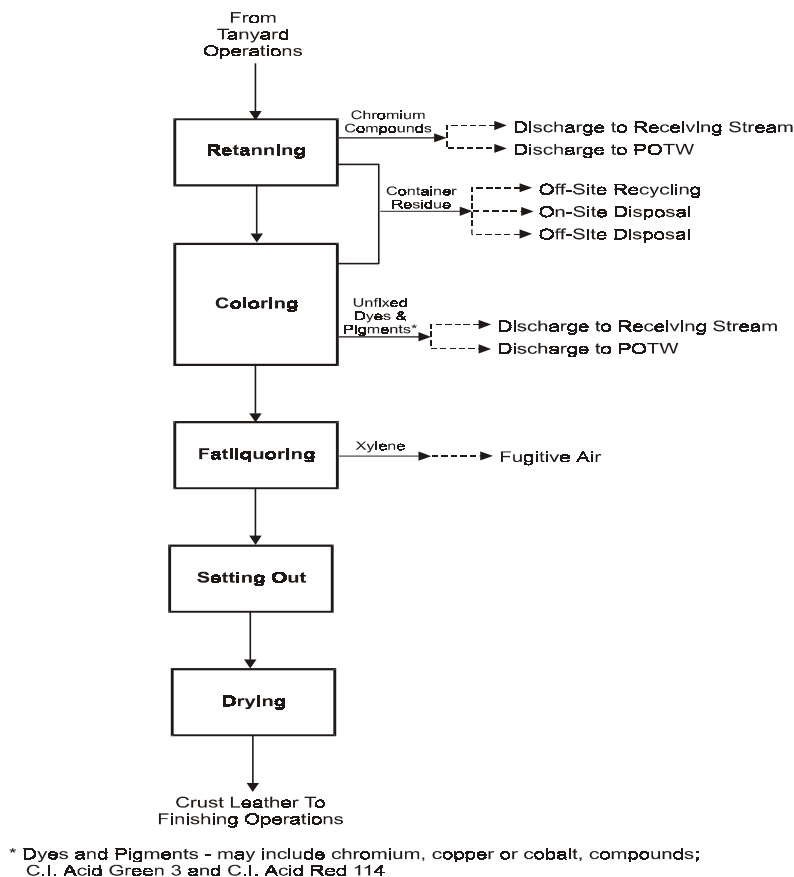
Note that this quantity is also the amount that was removed from the wastewater stream in your on-site wastewater treatment system. Therefore, Part II, Sections 7 and 8.6 should also be completed. The quantity reported in Section 8.6 should be the amount actually destroyed and for the purposes of EPCRA Section 313 reporting metals cannot be destroyed. Therefore, you must complete Section 7 (with the appropriate removal efficiency of 100% in this case) and enter zero in Section 8.6 as the quantity destroyed (see the example for On-Site Treatment in Section 4.1.3 for further discussion).

## **4.2.3 Retanning, Coloring and Fatliquoring**

Retanning, coloring, and fatliquoring make the physical properties of the leather match the desired end use of the finished product. In the retanning process several tanning agents may be combined to achieve a certain leather quality. Dyes may be added to achieve the desired color. The fatliquoring process involves the use of oils and waxes to lubricate the collagen fibers to produce the flexibility and softness needed for the final leather product (Ref: *Leather Facts*, Third Edition. New England Tanners Club, Peabody, MA. 1994).

## Step 1: Prepare a Process Flow Diagram

A site-specific process flow diagram should be prepared to help identify all potential sources and types of chemical and chemical category release and other waste management activities. A typical flow diagram is presented in Figure 4-5.



**Figure 4-5. Process Flow - Retanning, Coloring, and Fatliquoring Operations**

## Step 2: Identify EPCRA Section 313 Chemicals and Chemical Categories and Potential Sources of Chemical Release and Other Waste Management Activities

Tanyard operations may use the following EPCRA Section 313 chemicals and chemical categories: formaldehyde, formic acid, sulfuric acid (only an EPCRA Section 313 chemical in acid aerosol form), chromium compounds, copper compounds, cobalt compounds, 2-phenylphenol, C.I. Acid Green 3, C.I. Acid Red 114, and xylene.

Resins such as melamine, used to retan leather, may contain formaldehyde. Synthetic tanning agents may also contain formaldehyde used to polymerize sulfonated phenols to synthesize the syntans. While a residual amount of free formaldehyde is sometimes present in these products, the formaldehyde usually combines with the leather, rather than being released. A minority of tanners may still occasionally use formaldehyde as a retanning agent (e.g. chamois producers), but the industry has phased out its use and replaced it with alternatives, such as glutaraldehyde.

Chromium compounds can also be found in the mineral tanning agents and the pre-metallized dyes. The pre-metallized dyes may sometimes also be based on copper or cobalt. Most of the dye is absorbed into the leather. Other EPCRA Section 313 chemicals include 2-phenylphenol which functions as a fungicide in these operations; the two dyes, C.I. Acid Green 3 and C.I. Acid Red 114, which are not commonly used in leather manufacturing facilities but may be used by some facilities for coloring processes; and xylene, which may be used as a solvent for the application of silicone and fluorocarbon polymers in the fatliquoring operation.

Typical sources of EPCRA Section 313 chemicals and chemical categories are process wastewaters, fugitive air emissions from evaporation of volatile compounds, and residues in “empty” shipping containers.

### **Step 3: Identify Release and Other Waste Management Activity Types**

Types of release and other waste management activities are similar to those presented in beamhouse and tanyard operations. They include direct or indirect discharges to a receiving stream or to a POTW; fugitive air emissions from evaporation of any volatile materials such as xylene used as a solvent; and on-site or off-site management of container residues, which may result in the release and other waste management of EPCRA Section 313 chemicals or chemical categories to on-site or off-site disposal, treatment, energy recovery, or recycling, as appropriate.

#### **Step 4: Determine the Most Appropriate Method(s) and Calculate the Estimates for Release and Other Waste Management Activity Quantities**

After identifying the sources and types of EPCRA Section 313 chemicals and chemical categories, the method(s) used to estimate the quantities released or otherwise managed as waste from retanning, coloring, and finishing operations are identical to those previously discussed.

Monitoring data gathered for on-site wastewater treatment plant operating permits and/or NPDES permit requirements can generally provide flow data and wastewater concentrations that can be used to estimate quantities of EPCRA Section 313 chemicals and chemical categories directly discharged to receiving streams or indirectly discharged to POTWs.

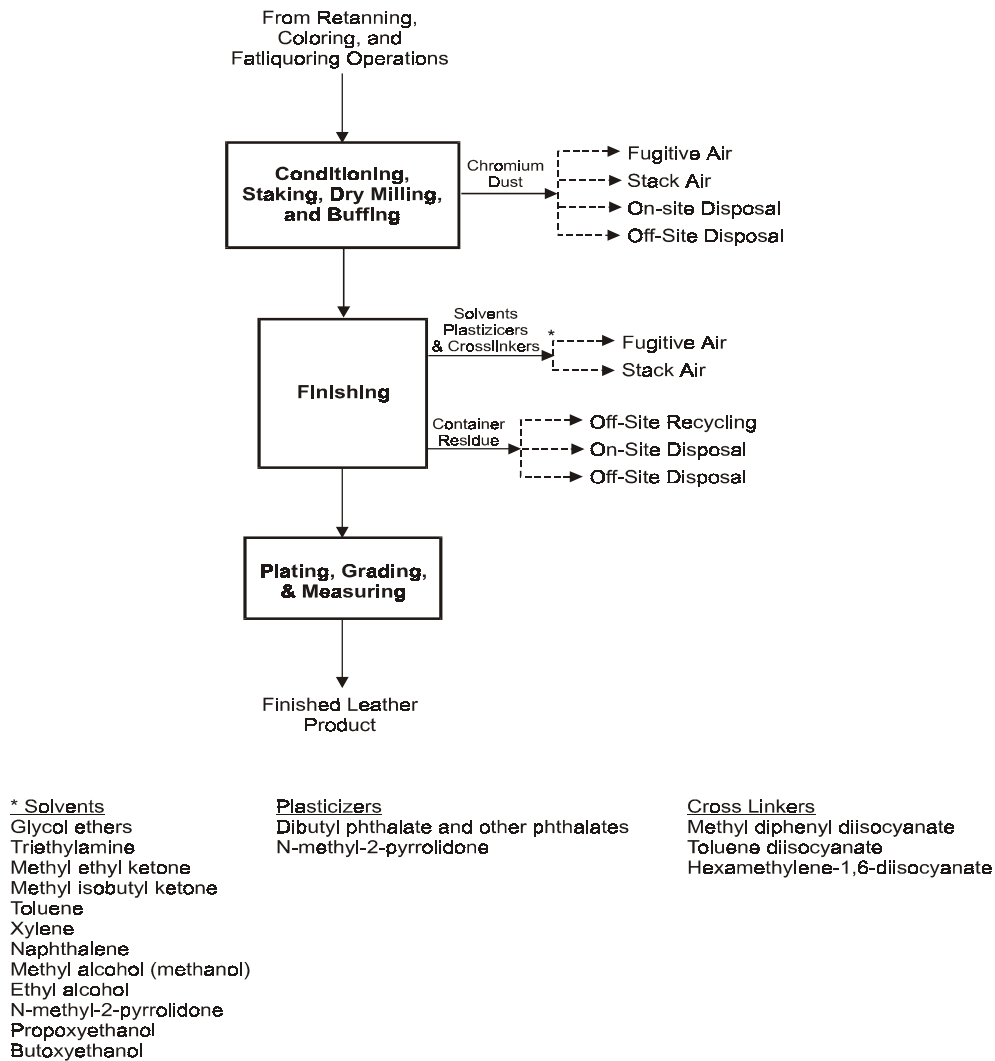
Volatile EPCRA Section 313 chemicals and chemical categories, not intended to remain with the product, e.g., xylene used as a solvent, can be assumed to be 100% released as either fugitive or stack emissions, as appropriate after subtracting any potential container residue. 2-butoxyethanol, propoxyethanol, and diethylene glycol monomethyl ether are examples of certain glycol ethers that may also be applicable. Liquid residues in shipping containers can be estimated using the factors in Table 4-1. For dry materials a residue factor of 1% can be used if actual data are not available (see Section 4.1.3.g for a complete discussion on container residue).

#### **4.2.4 Finishing Operations**

Finishing operations may be comprised of several steps designed to achieve the desired end use physical properties and dimensions of the final leather. The steps may include achieving the proper level of residual moisture, mechanical softening, smoothing/buffing the surface and imparting the desired grain pattern, grading, and measuring and cutting patterns to fit customer specifications (Ref: *Leather Facts*, Third Edition. New England Tanners Club, Peabody, MA. 1994).

## Step 1: Prepare a Process Flow Diagram

A site-specific process flow diagram should be prepared to help identify all potential sources and types of chemical and chemical category release and other waste management activities. A typical flow diagram is presented in Figure 4-6.



**Figure 4-6. Process Flow – Finishing Operations**

## **Step 2: Identify EPCRA Section 313 Chemicals and Chemical Categories and Potential Sources of Chemical Release and Other Waste Management Activities**

Leather dust containing chromium may be generated in some of the unit operations during finishing. Formaldehyde may be used to crosslink casein-based finishing formulations in the production of glazed leather. The other EPCRA Section 313 chemicals and chemical categories that may be used in finishing operations serve as solvents, plasticizers, or crosslinkers, and are listed in the process flow diagram. According to the Leather Industries of America, a range of leather finishing formulations use propylene glycol and other glycol ethers based on propylene glycol, which are not included in the Certain Glycol Ether category as defined in EPCRA Section 313; however, other glycol ethers used by the leather industry are included in the Certain Glycol Ether category. You should review the EPCRA Section 313 guidance presented in Appendix F if any glycol ethers are used at your facility.

Typical sources of EPCRA Section 313 chemicals and chemical categories include the generation of dusts containing chromium or chromium compounds; evaporation of volatile compounds, particularly solvents and plasticizers used in the final finishing steps; and residues in “empty” shipping containers.

## **Step 3: Identify Release and Other Waste Management Activity Types**

Typically, fewer release and other waste management types are expected during finishing operations, compared to other processes. The expected types include fugitive and stack air releases of dusts and any volatile materials, and on-site or off-site management of container residues and any dusts collected in air pollution control devices, which may result in the release and other waste management of EPCRA Section 313 chemicals or chemical categories to on-site or off-site disposal, treatment, energy recovery, or recycling, as appropriate. Note that any EPCRA Section 313 chemical or chemical category sent through a pollution control device is considered to have been treated for destruction if it is converted to another chemical or it is HCl or H<sub>2</sub>SO<sub>4</sub> acid aerosols. The treatment efficiency of the unit should be reported in Section 7A and the quantity treated for destruction should be reported in Section 8.6. Also, note that any EPCRA Section 313 chemical or chemical category sent through an air pollution control device is considered to have been captured for further waste management activities if it is not converted

to another chemical or it is not HCl or H<sub>2</sub>SO<sub>4</sub> acid aerosols. The capture efficiency of the unit should be reported in Section 7A and the quantity captured should be reported in Sections 6 and/or 8 depending on the final disposition of the chemical or chemical category. Wastewater is not typically generated in finishing operations.

#### **Step 4: Determine the Most Appropriate Method(s) and Calculate the Estimates for Release and Other Waste Management Activity Quantities**

Many of the solvents and other EPCRA Section 313 chemicals and chemical categories processed and otherwise used in finishing operations are very volatile. One can assume that all of the applied amount will evaporate and be released to the air, either as a fugitive emission, or as a stack emission if vapors are isolated or channeled through an air pollution control device. Do not forget to account for any potential container residue before estimating the quantity that may be volatilized during processing or otherwise use activities. Potential container residue quantities can be estimated as discussed in Section 4.1.3(g). The total amount released can be estimated using a mass balance approach based on purchasing records and beginning and end-of-year facility inventory amounts.

If dust containing EPCRA Section 313 chemicals or chemical categories is generated (typically chromium) it may be collected in a fabric filter (baghouse). If so, quantities of chromium passing through the baghouse should be estimated as stack emissions, and the quantity collected should be reported based on the method used to manage the waste dust. Also, use of the baghouse (an air pollution control device) is considered on-site treatment. Therefore, Part II, Section 7A and 8.6 of the 1999 Form R should be completed as appropriate. Section 4.1.3(h) presents an example for on-site treatment.

Dusts and trim scraps are typically sold and weighed and the chromium concentration is expected to be the same as in the finished product. If sold, this quantity does not have to be reported, but should be considered in mass balance calculations when determining releases and other waste management quantities. If disposed or otherwise managed as waste, the quantity of EPCRA Section 313 chemical or chemical compound should be reported or approximated with estimates based on the mass disposed multiplied by the concentration.

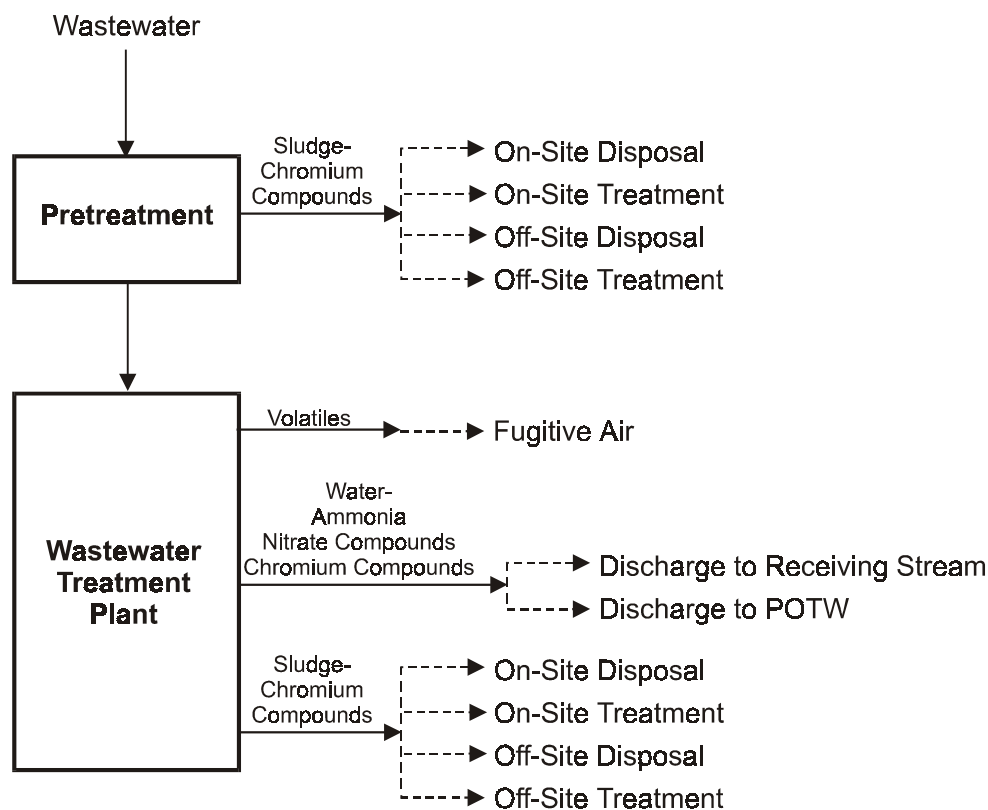


#### **4.2.5 Wastewater Treatment**

Many leather tanning facilities operate their own wastewater treatment plants, some with direct discharge to receiving streams and some with indirect discharge to a POTW. The incoming wastewater is characterized by high biochemical oxygen demand (BOD) and total suspended solids (TSS) content. In many cases the chromium and other heavy metals are removed by pre-treatment using high pH liquors from beamhouse operations to precipitate the metals. Direct discharge permits (NPDES/SPDES) or POTW influent specifications typically establish limits for chromium, sulfides, pH, nitrates, BOD, and TSS.

##### **Step 1: Prepare a Process Flow Diagram**

A site-specific process flow diagram should be prepared to help identify all potential sources and types of chemical and chemical category release and other waste management activities. A typical flow diagram is presented in Figure 4-7.



**Figure 4-7. Process Flow – Wastewater Treatment**

**Step 2: Identify EPCRA Section 313 Chemicals and Chemical Categories and Potential Sources of Chemical Release and Other Waste Management Activities**

Manganese sulfate, phosphoric acid, ammonia, chlorine, chlorine dioxide, chromium, and chromium compounds are some of the EPCRA Section 313 chemicals and chemical categories found in leather tanning and finishing wastewater treatment operations. The wastewater treatment system may use manganese sulfate as a catalyst for sulfide oxidation, while phosphoric acid may be used as a nutrient source in the biological treatment process. Both would be classified as otherwise used under EPCRA Section 313. Chromium and chromium compounds from the facility process operations may pass through the system and are found in the wastewater effluent and sludges. Ammonia and nitrate compounds can be coincidentally manufactured in biological treatment systems when proteins break down. Ammonia may also be added as a metabolic nitrogen source for the microbes. Chlorine and chlorine dioxide are used for effluent disinfection.

Typical sources of EPCRA Section 313 chemicals and chemical categories are treatment system effluents, evaporation of volatile compounds, and residues in “empty” shipping containers.

### **Step 3: Identify Release and Other Waste Management Activity Types**

Types of release and other waste management activities include direct discharge to receiving streams, indirect discharges to POTWs, fugitive air emissions from evaporation of any volatile materials, and on-site or off-site management of container residues (typically phosphoric acid and treatment system nutrients) and sludges, which may result in the release or management of wastes that contain EPCRA Section 313 chemicals or chemical categories to on-site or off-site disposal, treatment, energy recovery, or recycling, as appropriate.

### **Step 4: Determine the Most Appropriate Method(s) and Calculate the Estimates for Release and Other Waste Management Activity Quantities**

As discussed in Step 4 of previous sections (and in example 6 of Appendix B), monitoring data gathered for direct discharge permits and POTW influent specifications can provide flow data and concentrations of EPCRA Section 313 chemicals and chemical categories in the wastewater entering and leaving the wastewater treatment system. Also, liquid residues in shipping containers can be estimated using the factors in Table 4-1. For dry materials a residue factor of 1% can be used if actual data are not available (see Section 4.1.3.g for a complete discussion on container residue).

The quantity of anhydrous ammonia that is coincidentally manufactured should be applied to both the 25,000-pound manufacturing threshold as well as the 10,000-pound otherwise use threshold (assuming it is consumed as a metabolic nitrogen source). However, beginning with reporting year 1994, U.S. EPA revised the ammonia listing as follows: “ammonia (includes anhydrous ammonia and aqueous ammonia from water dissociable salts and other sources; 10% of total aqueous ammonia is reportable under this listing).” Appendix D contains the U.S. EPA published guidance for reporting ammonia and ammonium salts.

### **COMMON ERROR - Reporting**

A common error in wastewater treatment is the failure to report the ammonia that may be coincidentally manufactured as a byproduct during the treatment process. Other reportable chemicals also may be coincidentally manufactured during such treatments depending upon the chemicals present.

U.S. EPA has also provided guidance on reporting for the nitrate compounds category. This guidance is presented in Appendix C.

The quantity of phosphoric acid or any other EPCRA Section 313 chemical that is added as a nutrient should be applied to the 10,000-pound otherwise use activity threshold. If any threshold is exceeded, an EPCRA Section 313 report should be completed. However, if the pH of the effluent from the wastewater treatment system is between 6 and 9, U.S. EPA considers the mineral acid to be neutralized and zero discharge to the POTW or receiving stream should be reported. Please note that even if all release and other waste management quantities are zero, an EPCRA Section 313 report must still be completed. In such instances, you may be eligible to complete an alternate certification statement (Form A).

Finally, emission factors for volatile EPCRA Section 313 chemicals from wastewater treatment plants are available in Chapter 4.3 of AP-42.

## Appendix A

### TRI GUIDANCE RESOURCES

#### A.1 EPCRA Section 313 RELATED REFERENCES

40 CFR 372, Toxic Chemical Release Reporting; Community Right-to-Know; Final Rule  
See 53 FR 4500, February 16, 1988.

Toxic Chemical Release Inventory Reporting Forms and Instructions for the Current Reporting Year - See also Automated Toxic Chemical Release Inventory Reporting Software (ATRS) under Section A.2, Internet Sites.

U.S. EPA publishes this document each year to provide current guidance for preparing the Form R and Form A reports. This document contains the most up-to-date list of chemicals for which reports are required. It includes a blank Form R and Form A and provides step-by-step instructions for completing each report. It also has a list of U.S. EPA regional and state contacts for EPCRA Section 313 reporting. The current version of this document should always be consulted in preparing the EPCRA Section 313 report.

Common Synonyms for Chemicals Listed Under EPCRA Section 313 of the Emergency Planning and Community Right-to-Know Act (EPA 745-R-95-008)

This glossary contains chemical names and their synonyms for substances covered by the reporting requirements of EPCRA Section 313. The glossary was developed to aid in determining whether a facility manufactures, processes, or uses a chemical subject to EPCRA Section 313 reporting.

Consolidated List of Chemicals Subject to the Emergency Planning and Community Right-to-Know Act (EPCRA) and Section 112(r) of the Clean Air Act (as amended) (EPA 740-R-95-001)

List of chemicals covered by EPCRA Sections 302 and 313, CERCLA Hazardous Substances, and CAA 112(r). The list contains the chemical name, CAS Registry Number, and reporting requirement(s) to which the chemical is subject.

Data Quality Checks to Prevent Common Reporting Errors on Form R/A (EPA 745-R-98-012).

The Emergency Planning and Community Right-to-Know Act: EPCRA Section 313 Release Reporting Requirements, August, 1995 (EPA 745/K-95-052)

This brochure alerts businesses to their reporting obligations under EPCRA Section 313 and assists in determining whether their facility is required to report. The brochure contains U.S. EPA Regional contacts, the list of EPCRA Section 313 toxic chemicals and a description of the Standard Industrial Classification (SIC) codes subject to EPCRA Section 313.

EPCRA Section 313 Questions and Answers: 1998 Version, (EPA 745-B-98-004).

Executive Order 12856 - Federal Compliance with Right-to-Know Laws and Pollution Prevention Requirements: Questions and Answers (EPA 745-R-95-011)

This document assists federal facilities in complying with Executive Order 12856. This information has been compiled by U.S. EPA from questions received from federal facilities. This document is intended for the exclusive use of federal facilities in complying with Sections 302, 303, 304, 311, 312, and 313 of the Emergency Planning and Community Right-to-Know Act (EPCRA) of 1986 and the Pollution Prevention Act of 1990, as directed by the Executive Order.

Supplier Notification Requirements (EPA 560/4-91-006)

This pamphlet assists chemical suppliers who may be subject to the supplier notification requirements under EPCRA Section 313. The pamphlet explains the supplier notification requirements, gives examples of situations which require notification, describes the trade secret provision, and contains a sample notification.

Toxic Chemical Release Inventory - Data Quality Checks to Prevent Common Reporting Errors on Form R/Form A (EPA 745-R-98-012)

This is a compilation of Notices of Data Change, Significant Error, Noncompliance, or Technical Error. It provides a listing of common errors found on the Form R reports submitted to U.S. EPA. It also provides a discussion of the types of errors which result in each of the above Notices as well as a list of Notice of Technical Error codes and descriptions.

Trade Secrets Rule and Form

See 53 FR 28772, July 29, 1988. This rule implements the trade secrets provision of the EPCRA (Section 322) and includes a copy of the trade secret substantiation form.

**A.2            INFORMATION SOURCES**

Most of the materials included as reference in this manual are available from the following sources:

National Center for Environmental Publications and Information (NCEPI)  
P.O. Box 42419  
Cincinnati, OH 45242-2419  
(800) 490-9198  
Fax: (513)489-8695  
Internet: <http://www.epa.gov/ncepihom/index.html>

Emergency Planning and Community Right-to-Know (EPCRA) Information Hotline  
U.S. Environmental Protection Agency  
(800) 424-9346 or (703) 412-9810 (for the Washington, D.C. metropolitan area)  
TDD: (800) 553-7672

**Internet Sites**

- TRI homepage: <http://www.epa.gov/tri>  
This site contains information on the Toxic Release Inventory and provides links to a variety of data and documents related to the TRI program.

- Automated Toxic Chemical Release Inventory Reporting Software (ATRS):  
<http://www.epa.gov/atrs>  
 This site provides access to the automated EPCRA Section 313 reporting forms for electronic submittal of required data to U.S. EPA.
- Air CHIEF CD-ROM  
<http://www.epa.gov/ttn/chief/airchief.html>  
 This site provides information on the Air CHIEF CD-ROM, contents, ordering information, system requirements, and sources for additional information.
- Clearinghouse for Inventories and Emission Factors (CHIEF):  
<http://www.epa.gov/ttn/chief/>  
 This site provides access to the latest information and tools for estimating emissions of air pollutants and performing emission inventories.
- Code of Federal Regulations, 40 CFR: <http://www.epa.gov/epacfr40>  
 This site was created by U.S. EPA to expand access to Title 40 - Environmental Protections of the Code of Federal Regulations.
- Compilation of Air Pollutant Emission Factors (AP-42):  
<http://www.epa.gov/ttn/chief/ap42etc.html>  
 This site provides access to files containing guidance for estimating emissions from specific sources and emission factors.
- Federal Register Notice: <http://www.epa.gov/EPA-TRI>  
 This site provides access to all Federal Register notices related to the TRI program from 1994 to current.
- Material Safety Data Sheets (MSDSs):  
<http://msds.pdc.cornell.edu/issearch/msdssrch.htm>  
 A key word searchable database of 325,000 MSDSs.
- TANKS: <http://www.epa.gov/ttn/chief/tanks.html>  
 This site contains information on TANKS, a DOS-based computer software program that computes estimates of VOC emissions from fixed and floating-roof storage tanks.
- WATER8/CHEMDAT8: <http://www.epa.gov/ttn/chief/software.html#water8>  
 WATER8 is an analytical model for estimating compound-specific air emissions from wastewater collection and treatment systems. CHEMDAT8 is a Lotus 1-2-3 spreadsheet for estimating VOC emissions from TSDf processes.

### **A.3            INDUSTRY-SPECIFIC TECHNICAL GUIDANCE DOCUMENTS**

In 1988 and 1990, U.S. EPA developed a group of individual guidance documents for industries or activities in industries who primarily manufacture, process, or otherwise use EPCRA Section 313 chemicals. See list of industries/activities below. U.S. EPA is currently revising some of these documents and preparing additional documents. The newer versions will be available beginning in the Fall of 1998.

Chemical Distribution Facilities, January 1999 (EPA 745-B-99-005)

Coal Mining Facilities, January 1999 (EPA 745-B-99-002)

Coincidental Manufacture/By-products (EPA 745-B-00-014)

Electricity Generating Facilities, January 1999 (EPA 745-B-99-003)

Estimating Releases and Waste Treatment Efficiencies

Food Processors, September 1998 (EPA 745-R-98-011)

Formulation of Aqueous Solutions

Foundry Operations (EPA 745-B-00-016)

Leather Tanning and Finishing Industry, April 2000 (EPA 745-B-00-012)

Metal Working and Electroplating Operations, April 2000 (EPA 745-B-00-015)

Metal Mining Facilities, January 1999 (EPA 745-B-99-001)

Monofilament Fiber Manufacture (EPA 745-B-00-013)

Pulp, Paper, and Paperboard Production (EPA 745-B-00-010)

Petroleum Terminals and Bulk Storage Facilities, January 1999 (EPA 745-B-99-006)

Presswood & Laminated Wood Products Manufacturing

Printing Industry, April 2000 (EPA 745-B-00-005)

RCRA Subtitle C TSD Facilities and Solvent Recovery Facilities, January 1999 (EPA 745-B-99-004)

Rubber and Plastics Manufacturing, April 2000 (EPA 745-B-00-017)

Semiconductor Manufacture, July 1999 (EPA 745-R-99-007)



Smelting Operations (EPA 745-B-00-009)

Spray Application and Electrodeposition of Organic Coatings, December 1998 (EPA 745-B-99-014)

Textile Processing Industry, April 2000 (EPA 745-B-00-008)

Welding Operations (EPA 745-B-00-011)

Wood Preserving Operations (EPA 745-B-00-007)

U.S. EPA, Office of Compliance, published a series of documents in 1995 called Sector Notebooks. These documents provide information of general interest regarding environmental issues associated with specific industrial sectors. The Document Control Numbers (DCN) range from EPA/310-R-95-001 through EPA/310-R-95-018.

#### **A.4            CHEMICAL-SPECIFIC GUIDANCE DOCUMENTS**

U.S. EPA has also developed a group of guidance documents specific to individual chemicals and chemical categories. These are presented below.

Emergency Planning and Community Right-to-Know EPCRA Section 313: Guidance for Reporting Aqueous Ammonia, July 1995 (EPA 745-R-95-012)

Emergency Planning and Community Right-to-Know EPCRA Section 313: List of Toxic Chemicals within the Chlorophenols Category, November 1995 (EPA 745-B-95-004)

Estimating Releases for Mineral Acid Discharges Using pH Measurements, U.S. Environmental Protection Agency, June 1991.

Guidance for Reporting Sulfuric Acid (acid aerosols including mists, vapors, gas, fog, and other airborne forms of any particle size), November 1997 (EPA-745-R-97-007)

Toxic Release Inventory List of Toxic Chemicals within the Glycol Ethers Category and Guidance for Reporting, May 1995 (EPA 745-R-95-006)

Toxic Release Inventory List of Toxic Chemicals within the Nicotine and Salts Category and Guidance for Reporting, February 1995 (EPA 745-R-95-004)

Toxic Release Inventory List of Toxic Chemicals within the Polychlorinated Alkanes Category and Guidance for Reporting, February 1995 (EPA 745-R-95-001)

Toxic Release Inventory List of Toxic of Chemicals within the Polycyclic Aromatic Compounds Category, February 1995 (EPA 745-R-95-003)

Toxic Release Inventory List of Toxic Chemicals within the Strychnine and Salts Category and Guidance for Reporting, February 1995 (EPA 745-R-95-005)

Toxic Release Inventory List of Toxic of Chemicals within the Water Dissociable Nitrate Compounds Category and Guidance for Reporting, May, 1996 (EPA 745-R-96-004)

Toxics Release Inventory - List of Toxic Chemicals Within Ethylenebisdithiocarbamic Acid Category, November 1994, EPA 745-B-94-003.

Toxics Release Inventory - Copper Phthalocyanine Compounds Excluded for the Reporting Requirements Under the Copper Compounds Category on the EPCRA Section 313 List, April 1995, EPA 745-R-95-007.

Toxics Release Inventory - List of Toxic Chemicals Within Warfarin Category, November 1994, EPA 745-B-94-004.

## **A.5            OTHER USEFUL REFERENCES**

Burgess, W.A. Recognition of Health Hazards in Industry. Harvard School of Public Health. Boston, Massachusetts, John-Wiley & Sons.

CRC Handbook of Chemistry and Physics. Latest Edition, Robert C. Weast, Editor, CRC Press, Inc., Florida.

Kirk Othmer - Encyclopedia of Chemical Technology. Latest Edition, John Wiley & Sons, New York.

Locating and Estimating Air Emissions from Various Sources. Available from: National Technical Information Services (NTIS), (703) 487-4650.

The Merck Index. Latest Edition, Merck & Co., Inc., New Jersey.

Perry, R.H. and C.H. Chilton, Chemical Engineer's Handbook. Latest Edition, McGraw-Hill Book Company, New York.

Sax, N.I. and R.J. Lewis, Sr., Hawley's Condensed Chemical Dictionary. Latest Edition, Van Nostrand Reinhold Company, New York.

**Appendix B**

**BASIC CALCULATION TECHNIQUES**

## Appendix B

### BASIC CALCULATION TECHNIQUES

This section will provide the basic techniques needed to use specific types of data or engineering calculations. Examples are provided for:

- (1) Stack monitoring data;
- (2) Industrial hygiene data;
- (3) Raoult's Law;
- (4) Air emission factors;
- (5) RCRA hazardous waste analysis data;
- (6) NPDES monitoring data.

#### (1) Stack Monitoring Data

The following is an example of a release calculation using monitoring data.

**Example:** Stack monitoring data are available for a paint booth. The measured average concentration of toluene is 0.1 ppmv (dry gas basis). The moisture content in the stack is typically 10%, and stack conditions are maintained at 80°C and atmospheric pressure. The stack gas velocity is 8 m/s. The diameter of the stack is 0.3 m. Calculate the point air release of toluene.

Step 1. Calculate volumetric flow of stack gas stream.

$$\text{Volumetric flow} = (\text{gas velocity}) \times [(\pi) \times (\text{internal stack diameter})^2/4]$$

$$\text{Volumetric flow} = (8.0 \text{ m/s}) \times [(\pi) \times (0.3 \text{ m})^2/4] = 0.6 \text{ m}^3/\text{s}$$

Step 2. Correct for moisture content in stack gas stream.

Stack exhausts may contain large amounts of water vapor. The concentration of the chemical in the exhaust is often presented on a dry basis. For an accurate release rate, correct the vent gas flow rate for the moisture content by multiplying by the term (1 - fraction water vapor). The dry gas rate can then be multiplied by the chemical concentration.

(Note: If the toluene concentration is on a wet gas basis, no correction is necessary for moisture content.)

$$\text{Dry volumetric flow} = (\text{Volumetric flow}) \times (1 - \text{fraction water vapor})$$

$$\text{Dry volumetric flow} = (0.6 \text{ m}^3/\text{s}) \times (1 - 0.10) = 0.5 \text{ m}^3/\text{s}$$

Step 3. Convert ppmv to mg/m<sup>3</sup>.

- ppmv is defined as one part of a chemical in 10<sup>6</sup> parts of gas (1.0 m<sup>3</sup>/10<sup>6</sup> m<sup>3</sup>).
- Use the molar volume of a gas, corrected for stack temperature and pressure conditions, calculated by the ideal gas law (PV = nRT). Note that the molar volume of an ideal gas at 273 K and 1 atm is 22.4 L/mole.
- Molecular weight of toluene (MW) = 92.14 g/mole.
- R = the Ideal Gas Constant (0.082057 L - atm per mole-Kelvin)

To calculate the molar volume of stack gas, use the ideal gas equation.

$$\text{Molar volume} = \frac{V}{n} = \frac{RT}{P}$$

For the example, the stack conditions are 80° C (353 K) and atmospheric pressure (1 atm).

$$\text{Molar volume} = \left( 0.082057 \frac{\text{L-atm}}{\text{mole-K}} \right) \times (353 \text{ K}) / (1 \text{ atm})$$

$$= 29.0 \text{ L/mole}$$

The conversion of ppmv to mg/m<sup>3</sup> can now be calculated.

$$\left( \frac{\text{mg}}{\text{m}^3} \right) = (\text{concentration of chemical, ppmv}) \times \left( \frac{1}{\text{molar volume of gas}} \right) \times (\text{MW})$$

Using the example, the concentration of toluene is calculated as follows:

$$\left( \frac{0.1 \text{ m}^3}{10^6 \text{ m}^3} \right) \times \left( \frac{\text{mole}}{29.0 \text{ L}} \right) \times \left( \frac{92.14 \text{ g}}{\text{mole}} \right) \times \left( \frac{\text{L}}{10^{-3} \text{ m}^3} \right) \times \left( \frac{1,000 \text{ mg}}{1 \text{ g}} \right) = 0.3 \text{ mg/m}^3$$

Step 4. Calculate air releases.

Air releases are calculated as follows:

$\text{Air Release} = (\text{volumetric flow, m}^3/\text{s}) \times (\text{concentration, mg/m}^3) \times (\text{operating time, s/yr})$
--

The paint booth is used 8 hours per day, 5 days per week, 52 weeks per year.

$$\text{Operating time} = \left( 8 \frac{\text{hr}}{\text{day}} \right) \times \left( 5 \frac{\text{day}}{\text{week}} \right) \times \left( 52 \frac{\text{week}}{\text{yr}} \right) = 2,080 \text{ hr/yr}$$

$$\begin{aligned} \text{Air Release} &= (0.5 \text{ m}^3/\text{s}) \times (0.3 \text{ mg/m}^3) \times \left( \frac{3,600 \text{ s}}{\text{hr}} \right) \times \left( \frac{2,080 \text{ hr}}{\text{yr}} \right) \times \left( \frac{\text{lb}}{454 \text{ g}} \right) \times \left( \frac{\text{g}}{1,000 \text{ mg}} \right) \\ &= 2.5 \text{ lb/yr of toluene} \end{aligned}$$

It is important to note that this calculation assumes the measured emissions are representative of actual emissions at all times; however, this is not always the case. Ideally, a continuous emissions monitor provides the most representative data.

Also note that monitoring and stack data may have units that are different than those used in the example. Modify conversion factors and constants to reflect your data when calculating air releases.

## (2) **Industrial Hygiene Data**

The following is an example of a release calculation using industrial hygiene data.

**Example:** Occupational industrial hygiene data shows that workers are exposed to an average of 0.1 ppmv benzene (wet gas basis). The density of benzene vapor is 0.2 lb/ft<sup>3</sup>. The ventilation system exhausts 20,000 acfm of room air at 70°F. The plant operates 24 hours per day, 330 days per year.

The benzene concentration is on a wet gas basis, therefore a moisture correction of the ventilation flow rate is not necessary. The industrial hygiene data is collected at the same ambient conditions as the ventilation system, therefore no

adjustment for temperature or pressure needs to be performed. A conservative estimation of benzene fugitive releases could be calculated as follows:

$$\text{Air Release} = (\text{ventilation flow rate, ft}^3/\text{min}) \times (\text{operating time, min/yr}) \times (\text{concentration of chemical, ppmv}) \times (\text{vapor density of chemical, lb/ft}^3)$$

Benzene releases per year would be calculated as follows:

$$\left( \frac{20,000 \text{ ft}^3}{\text{min}} \right) \times \left( \frac{60 \text{ min}}{\text{hr}} \right) \times \left( \frac{24 \text{ hr}}{\text{day}} \right) \times \left( \frac{330 \text{ day}}{\text{yr}} \right) \times \left( \frac{0.1 \text{ ft}^3 \text{ benzene}}{10^6 \text{ ft}^3 \text{ air}} \right) \times \left( \frac{0.2 \text{ lb}}{\text{ft}^3} \right)$$

$$= 190 \text{ lb/yr of benzene}$$

### (3) Raoult's Law

The following is an example of a release calculation using Raoult's Law. Raoult's Law states that the partial pressure of a compound in the vapor phase over a solution may be estimated by multiplying its mole fraction in the liquid solution by the vapor pressure of the pure chemical.

$$P_A = X_{A,L}P^\circ = X_{A,G}P_T$$

where:

$P^\circ$	=	Vapor pressure of pure liquid chemical A;
$X_{A,L}$	=	Mole fraction of chemical A in solution;
$X_{A,G}$	=	Mole fraction of chemical A in the gas phase;
$P_A$	=	Partial pressure of chemical A in the gas phase; and
$P_T$	=	Total pressure.

**Example:** A wash tank holds a solution containing 10% by weight of o-xylene (A) and 90% by weight of toluene (B). The tank is vented to the atmosphere; the process vent flow rate is estimated as 100 acfm (2.83m<sup>3</sup>/min) based on a minimum fresh air ventilation rate. The molecular weight of o-xylene is 106.17 g/mole and toluene is 92.14 g/mole. The vapor pressure of o-xylene is 10 mm of Hg (0.19 psia). The total pressure of the system is 14.7 psia (atmospheric conditions). The process tank is in service 250 days/yr. Calculate the air release of o-xylene.

Step 1: Calculate the mole fraction of o-xylene in the liquid solution.

$$X_{A,L} = \frac{\frac{\text{wt fraction A}}{MW_A}}{\frac{\text{wt fraction A}}{MW_A} + \frac{\text{wt fraction B}}{MW_B}}$$

Where:

$X_{A,L}$  = Mole fraction of chemical A in liquid solution;  
 MW = Molecular weight of chemical, g/mole; and  
 wt fraction = Weight fraction of chemical in material.

$$X_{A,L} = \frac{\left[ \frac{0.1}{106.17} \right]}{\left[ \frac{0.1}{106.17} + \frac{0.9}{92.14} \right]}$$

$$X_{A,L} = 0.09$$

Step 2: Calculate the mole fraction of o-xylene in the gas phase.

$$X_{A,G} = \frac{X_{A,L} P^\circ}{P_T}$$

where:

$X_{A,G}$  = Mole fraction of chemical A in gas phase;  
 $X_{A,L}$  = Mole fraction of chemical A in liquid solution;  
 $P^\circ$  = Vapor pressure of pure liquid chemical A, psia; and  
 $P_T$  = Total pressure of system, psia.

$$X_{A,G} = [0.09] \times \left[ \frac{0.19 \text{ psia}}{14.7 \text{ psia}} \right] = 0.001$$

Step 3: Calculate releases using Raoult's Law.



$$\text{Emissions} = (X_{A,G}) \times (\text{AFR}) \times (t) \times (\text{MW}_A) \times \left( \frac{1}{\text{MV}} \right)$$

where:

Emissions	=	Air release of pollutant A, g-A/yr;
$X_{A,G}$	=	Mole fraction of chemical A in gas phase;
AFR	=	Air flow rate of room, m <sup>3</sup> /min;
t	=	Operating time of wash tank, min/yr;
MW	=	Molecular weight of chemical, g/g-mole; and
MV	=	Gas molar volume (22.4 L/mole at standard temperature and pressure).

If conditions vary from standard temperature and pressure the gas molar volume can be calculated using the ideal gas law and tank conditions as presented in Example 1.

Emissions = (0.001) ×

$$\left( \frac{2.83 \text{ m}^3}{\text{min}} \right) \times \left( \frac{250 \text{ day}}{\text{yr}} \right) \times \left( \frac{24 \text{ hr}}{\text{day}} \right) \times \left( \frac{60 \text{ min}}{\text{hr}} \right) \times \left( \frac{\text{mole}}{22.4 \text{ L}} \right) \times \left( \frac{106.17 \text{ g}}{\text{mole}} \right) \times \left( \frac{\text{L}}{10^{-3} \text{ m}^3} \right)$$

$$= 4.8 \times 10^6 \text{ g/yr}$$

The emission of o-xylene is calculated as shown below.

$$\text{Emissions} = (4.8 \times 10^6 \text{ g/yr}) \times \left( \frac{\text{lb}}{454 \text{ g}} \right) = 10,570 \text{ lb/yr of o-xylene}$$

Air releases for toluene can be calculated in a similar manner.

#### (4) **Air Emission Factor**

The following is an example of a release calculation using air emission factors.

**Example:** An industrial boiler uses 300 gallons per hour of No. 2 fuel oil. The boiler operates 2,000 hours per year. Calculate emissions of formaldehyde using the AP-42 emission factors.

$$\text{AE} = (\text{EF}) \times (\text{AU}) \times (\text{OT})$$

where:

AE = Annual emissions of pollutant, lb/yr  
EF = Emission factor of pollutant, lb/10<sup>3</sup> gallon of fuel. EF for formaldehyde for an industrial boiler burning No. 2 fuel oil is 0.035 to 0.061 lb/10<sup>3</sup> gallons.  
AU = Quantity of fuel used, gal/yr.  
OT = Operating time, hr/yr.

Using an emission factor of 0.061 pounds of formaldehyde per gallon of fuel, the air releases are calculated as follows:

$$AE = \left( \frac{0.061 \text{ lb}}{10^3 \text{ gal}} \right) \times \left( \frac{300 \text{ gal}}{\text{hr}} \right) \times \left( \frac{2,000 \text{ hr}}{\text{yr}} \right) = 36.6 \text{ lb/yr of formaldehyde}$$

(5) **RCRA Waste Analysis**

The following is an example of a calculation using RCRA waste analysis data.

**Example:** Spent paint wastes were disposed at an off-site waste treatment facility. The quantity of paint waste shipped was five 55-gallon drums per year. Analysis of the waste showed 5% cadmium by weight. Estimating the density of the paint waste to be 9.5 lb/gallon, the amount of cadmium to off-site disposal is calculated as follows:

$\text{Amount of cadmium} = (\text{amount of paint waste disposed, gal/yr}) \times (\text{concentration of cadmium, lb/lb}) \times (\text{density of paint waste, lb/gal})$
---

$$\left( \frac{5 \text{ drums}}{\text{yr}} \right) \times \left( \frac{55 \text{ gal}}{\text{drum}} \right) \times \left( \frac{9.5 \text{ lb}}{\text{gal}} \right) \times \left( \frac{5 \text{ lb Cd}}{100 \text{ lb waste}} \right) = 131 \text{ lb/yr of cadmium}$$

(6) **NPDES Data**

The following is an example of a calculation using NPDES data.

NPDES permits require periodic monitoring of the effluent stream. In this example, quarterly samples were taken to be analyzed for silver content. Each sample was an hourly, flowrate-based composite taken for one day to be representative of the discharge for that day. The total effluent volume for that day was also recorded. The following data were collected on each sample day.

<u>Yearly Quarter Sample Number</u>	<u>Discharge Flow Rate (10<sup>6</sup> gal/day)</u>	<u>Total Silver (µg/L)</u>
1	0.5	10
2	0.6	10
3	0.4	6
4	0.2	<3

To calculate the amount of silver in pounds discharged on each sample day, the concentration of silver in the discharge is multiplied by the discharge flow rate for that day, as shown below for the first quarter sample.

Amount of silver = (daily flow rate) × (silver concentration)
---

$$\text{First Quarter: } \left( \frac{10 \mu\text{g}}{\text{L}} \right) \times \left( \frac{1 \text{g}}{10^6 \mu\text{g}} \right) \times \left( \frac{1 \text{lb}}{454 \text{g}} \right) \times \left( \frac{3.785 \text{L}}{\text{gal}} \right) \left( \frac{0.5 \times 10^6 \text{gal}}{\text{day}} \right)$$

$$= 0.04 \text{ lb/day of silver}$$

The amount of silver discharged during each of the other three monitoring events was similarly determined to be:

0.05 lb/day; 0.02 lb/day, and 0.005 lb/day.

For the last data point the concentration of silver was reported by the laboratory to be less than the detection limit of 3 µg/L. For this calculation the detection limit was used to calculate the daily discharge, a conservative assumption.

The average daily discharge was calculated to be:

$$\left( \frac{0.04 + 0.05 + 0.02 + 0.005}{4} \right) \text{ lb/day} = 0.03 \text{ lb/day}$$

The plant operates 350 days/year (plant shuts down for two weeks in July).

The estimated annual discharge of silver is calculated as follows:

$$\text{Annual discharge} = (350 \text{ day/yr}) (0.03 \text{ lb/day}) = 10.5 \text{ lb of silver/yr}$$

**Appendix C**

**LIST OF TOXIC CHEMICALS WITHIN THE WATER DISSOCIABLE  
NITRATE COMPOUNDS CATEGORY AND GUIDANCE FOR REPORTING**



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# TOXICS RELEASE INVENTORY

## List of Toxic Chemicals Within the Water Dissociable Nitrate Compounds Category and Guidance for Reporting

EPCRA Section 313 of the Emergency Planning and Community Right-to-Know Act of 1986 (EPCRA) requires certain facilities manufacturing, processing, or otherwise using listed toxic chemicals to report their environmental releases of such chemicals annually. Beginning with the 1991 reporting year, such facilities also must report pollution prevention and recycling data for such chemicals, pursuant to section 6607 of the Pollution Prevention Act, 42 U.S.C. 13106. When enacted, EPCRA Section 313 established an initial list of toxic chemicals that was comprised of more than 300 chemicals and 20 chemical categories. EPCRA Section 313(d) authorizes EPA to add chemicals to or delete chemicals from the list, and sets forth criteria for these actions.

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## Section 1. Introduction

On November 30, 1994 EPA added 286 chemicals and chemical categories, which include 39 chemicals as part of two delineated categories, to the list of toxic chemicals subject to reporting under EPCRA Section 313 of the Emergency Planning and Community Right-to-Know Act of 1986 (EPCRA), 42 U.S.C. 11001. These additions are described at 59 FR 61432, and are effective January 1, 1995 for reports due July 1, 1996. Six chemical categories (nicotine and salts, strychnine and salts, polycyclic aromatic compounds, water dissociable nitrate compounds, diisocyanates, and polychlorinated alkanes) are included in these additions. At the time of the addition, EPA indicated that the Agency would develop, as appropriate, interpretations and guidance that the Agency determines are necessary to facilitate accurate reporting for these categories. This document constitutes such guidance for the water dissociable nitrate compounds category.

### Section 1.1 Who Must Report

A plant, factory, or other facility is subject to the provisions of EPCRA Section 313, if it meets all three of the following criteria:

- It conducts manufacturing operations (is included in Standard Industrial Classification (SIC) codes 20 through 39); and
- It has 10 or more full-time employees (or the equivalent 20,000 hours per year); and
- It manufactures, imports, processes, or otherwise uses any of the toxic chemicals listed on the EPCRA Section 313 list in amounts greater than the “threshold” quantities specified below.

### Section 1.2 Thresholds

Thresholds are specified amounts of toxic chemicals used during the calendar year that trigger reporting requirements.

If a facility *manufactures* or *imports* any of the listed toxic chemicals, the thresholds quantity will be:

- 25,000 pounds per toxic chemical or category over the calendar year.

If a facility *processes* any of the listed toxic chemicals, the threshold quantity will be:

- 25,000 pounds per toxic chemical or category over the calendar year.

If a facility *otherwise uses* any of the listed toxic chemicals (without incorporating it into any product or producing it at the facility), the threshold quantity is:

- 10,000 pounds per toxic chemical or category over the calendar year.

EPCRA Section 313 requires threshold determinations for chemical categories to be based on the total of all chemicals in the category manufactured, processed or otherwise used. For example, a facility that manufactures three members of a chemical category would count the total amount of all three chemicals manufactured towards the manufacturing threshold for that category. When filing reports for chemical categories, the releases are determined in the same manner as the thresholds. One report is filed for the category and all releases are reported on this form.

### **Section 1.3 Chemicals Within the Water Dissociable Nitrate Compounds Category**

EPA is providing a list of CAS numbers and chemical names to aid the regulated community in determining whether they need to report for the water dissociable nitrate compounds category. The list includes individual chemicals within the water dissociable nitrate compounds category. If a facility is manufacturing, processing, or otherwise using a chemical which is on this list, they must report this chemical. However, this list is not exhaustive. If a facility is manufacturing, processing, or otherwise using a water dissociable nitrate compound, they must report the chemical, even if it does not appear on the list.

### **Section 1.4 *De Minimis* Concentrations**

The water dissociable nitrate compounds category is subject to the one percent *de minimis* concentration. Thus, mixtures that contain members of this category in excess of the *de minimis* should be factored into threshold and release determinations.

## Section 2. Guidance for Reporting Chemicals within the Water Dissociable Nitrate Compounds Category

Note: for the purposes of reporting under the nitrate compounds category, water dissociable means that the nitrate ion dissociates from its counterion when in solution.

### Section 2.1 Chemicals within the Water Dissociable Nitrate Compounds Category

Chemicals within the nitrate compounds category are only reportable when in aqueous solution. All water dissociable nitrate compounds are included in the nitrate compounds category, including ammonium nitrate. Specifically listed EPCRA Section 313 chemicals *are not* included in threshold determinations for chemical categories such as the water dissociable nitrate compounds category. Specifically listed toxic chemicals are subject to their own individual threshold determinations. As of December 1, 1994, ammonium nitrate (solution) is not an individually listed chemical on the EPCRA Section 313 list. However, ammonium nitrate is still subject to reporting under the nitrate compounds category. In addition, the aqueous ammonia from the dissociation of ammonium nitrate when in aqueous solution is subject to reporting under the ammonia listing.

### Section 2.2 Determining Threshold and Release Quantities for Nitrate Compounds

The total nitrate compound, including both the nitrate ion portion and the counterion, is included in the nitrate compounds category. When determining threshold amounts, the total weight of the nitrate compound is to be included in all calculations. However, only the nitrate ion portion is to be included when determining the amount of the chemicals within the nitrate compounds category that is released, transferred, or otherwise managed in wastes.

**Example 1:** In a calendar year, a facility processes 100,000 pounds of ammonium nitrate ( $\text{NH}_4\text{NO}_3$ ), *in aqueous solution*, which is released to wastewater streams then transferred to a POTW. The quantity applied towards threshold calculations for the nitrate compounds category is the total quantity of the nitrate compound or 100,000 pounds. Since this quantity exceeds the 25,000 pound processing threshold, the facility is required to report for the nitrate compounds category. Under the nitrate compounds category, only the weight of the nitrate ion portion of ammonium nitrate is included in release transfer calculations. The molecular weight of the ammonium nitrate is 80.04 and the weight of the nitrate ion portion is 62.01 or 77.47 percent of the molecular weight of ammonium nitrate. Therefore, the amount of nitrate ion reported as transferred to the POTW is 77.47 percent of 100,000 pounds or 77,470 pounds (reported as 77,000 pounds). The aqueous ammonia from ammonium nitrate is reportable under the EPCRA Section 313 listing for ammonia. For determining thresholds and calculating releases under the ammonia listing, see the separate directive, *Guidance for Reporting Aqueous Ammonia* (EPA document #745-R-95-0003, July 1995).



**Example 2:** In a calendar year, a facility manufactures as by-products 20,000 pounds of sodium nitrate ( $\text{NaNO}_3$ ) and 10,000 pounds of calcium nitrate ( $\text{Ca}(\text{NO}_3)_2$ ), both in aqueous solutions, and releases these solutions to wastewater streams. The total quantity of nitrate compounds manufactured by the facility is the sum of the two chemicals, or 30,000 pounds, which exceeds the manufacturing threshold quantity of 25,000 pounds. The facility therefore is required to report for the nitrate compounds category. By weight, the nitrate ion portion is 72.96 percent of sodium nitrate and is 75.57 percent of calcium nitrate. Of the 20,000 pounds of the sodium nitrate in solution, 72.96 percent or 14,592 pounds is nitrate ion, and similarly, of the 10,000 pounds of the calcium nitrate in solution, 75.57 percent or 7,557 pounds is nitrate ion. The total nitrate ion in aqueous solution released by the facility is the sum of the nitrate ion in the two solutions or 22,149 pounds (reported as 22,000 pounds).

### **Section 2.3 Reporting Nitrate Compounds Generated from the Partial or Complete Neutralization of Nitric Acid**

Nitric acid is an individually listed chemical on the original EPCRA Section 313 list and is reported as a separate chemical if the manufacture, process, or otherwise use thresholds are exceeded. The partial or complete neutralization of nitric acid results in the formation of nitrate compounds which are reported as chemicals within the nitrate compounds category if their manufacture, process, or otherwise use thresholds are exceeded.

Mineral acids such as nitric acid may be present in aqueous waste streams that are sent to on-site neutralization or are discharged to a publicly owned treatment works (POTW) or other off-site treatment facility. As stated in the *Toxic Chemical Release Inventory Reporting Form R and Instructions* document (revised 1993 version, EPA 745-K-94-001), on-site acid neutralization and its efficiency must be reported in Part II, section 7A of Form R (waste treatment methods and efficiency section). For purposes of reporting on Form R, EPA considers a waste mineral acid at a pH 6 or higher to be 100 percent neutralized (water discharges to receiving streams or POTWs are reported as zero). The nitrate compounds produced from the complete neutralization (pH 6.0 or above) of nitric acid are reportable under the nitrate compounds category and should be included in all threshold and release calculations. Two Form R reports would be required if the manufacture, process or otherwise use thresholds are exceeded for nitric acid and for the nitrate compounds category.

If the nitric acid treatment efficiency is not equal to 100 percent (pH is less than 6), the amount of the acid remaining in the waste stream which is released to the environment on-site or off-site must be reported in Part II of Form R. The nitrate compounds produced from the partial neutralization of nitric acid are reportable under the nitrate compounds category and should be included in all threshold and release calculations. Two reports would again be required if the manufacture, process or otherwise use thresholds are exceeded for nitric acid and for the nitrate compounds category.

### Section 2.3.1 Estimating Nitric Acid Releases

The pH of the waste stream can be used to calculate the amount of nitric acid in the stream and the efficiency of neutralization. The pH is a measure of the acidity or alkalinity of a waste stream and can be obtained readily using a pH meter or pH sensitive paper. The pH scale itself varies from 0 to 14.

The total nitric acid concentration (ionized and un-ionized) in pounds per gallon can be calculated by using the pH value of the solution, the molecular weight and ionization constant of the acid, and appropriate conversion factors. The total acid concentration for nitric acid for different pH values is listed in Table 1. The calculation of mineral acid concentrations and the derivation of Table 1 are discussed in a separate directive, *Estimating Releases for Mineral Acid Discharges Using pH Measurements*, and an addendum to this directive.

The procedure outlined in this guidance document for calculating the quantity of nitrate compounds formed from the complete or partial neutralization of nitric acid can be used if nitric acid is the only mineral acid in a solution. In addition, the calculation of nitric acid releases using only pH measurements is a rough estimate. The subsequent calculation of nitrate compound releases is therefore also only a rough estimate. The estimates can be made for a waste stream with a steady pH below 6 or for one whose pH temporarily drops to below pH 6. Facilities should use their best engineering judgement and knowledge of the solution to evaluate how reasonable the estimates are.

**Example 3:** In a calendar year, a facility transfers 1.0 million gallons of a solution containing nitric acid ( $\text{HNO}_3$ ), at pH 4, to a POTW. Using Table 1 (next page), a pH of 4 corresponds to a concentration of 0.0000520 lbs  $\text{HNO}_3$ /gallon of solution. The weight of  $\text{HNO}_3$  transferred can be estimated using the equation:

$$\text{Transfer of } \text{HNO}_3 = (\text{Concentration of } \text{HNO}_3) \times (\text{effluent flow rate})$$

Substituting the example values into the above equation yields:

$$\text{Transfer of } \text{HNO}_3 = 0.0000520 \text{ lbs/gal } \text{HNO}_3 \times 1,000,000 \text{ gal solution/year} = 52 \text{ lbs/year}$$

**Example 4:** A facility had an episodic release of nitric acid ( $\text{HNO}_3$ ) in which the waste stream was temporarily below pH 6. During the episode, the wastewater (pH 2.0) was discharged to a river for 20 minutes at a rate of 100 gallons per minute. Using Table 1, a pH of 2.0 for  $\text{HNO}_3$  represents a concentration of 0.0052000 lbs  $\text{HNO}_3$ /gallon of solution. The amount of the  $\text{HNO}_3$  released can be estimated using the following equation:

$$\text{Release of } \text{HNO}_3 = (\text{concentration of } \text{HNO}_3) \times (\text{effluent flow rate})$$

Substituting the example values in the above equation:

$$\begin{aligned} \text{Release of } \text{HNO}_3 &= 0.0052000 \text{ lbs/gal} \times 100 \text{ gal/min} \times 20 \text{ min} \\ &= 10 \text{ lbs} \end{aligned}$$

**Table C-1  
Nitric Acid Concentration Versus pH**

<b>pH</b>	<b>Nitric Acid Concentration (lbs/gallon)</b>	<b>pH</b>	<b>Nitric Acid Concentration (lbs/gallon)</b>
0.0	0.5200000	3.0	0.0005200
0.2	0.3300000	3.2	0.0003300
0.4	0.2100000	3.4	0.0002100
0.6	0.1300000	3.6	0.0001300
0.8	0.0830000	3.8	0.0000830
1.0	0.0520000	4.0	0.0000520
1.2	0.0330000	4.2	0.0000330
1.4	0.0210000	4.4	0.0000210
1.6	0.0130000	4.6	0.0000130
1.8	0.0083000	4.8	0.0000083
2.0	0.0052000	5.0	0.0000052
2.2	0.0033000	5.2	0.0000033
2.4	0.0021000	5.4	0.0000021
2.6	0.0013000	5.6	0.0000013
2.8	0.0008300	5.8	0.0000008
		6.0	0.0000005

**Section 2.3.2 Estimating Treatment Efficiencies for Nitric Acid Neutralization**

Nitric acid solutions that are neutralized to a pH of 6 or above have a treatment efficiency of 100 percent. If nitric acid is neutralized to a pH less than 6, then the reportable treatment efficiency is somewhere between 0 and 100 percent. It is possible to estimate the neutralization treatment efficiency using nitric acid concentration values directly from Table 1 in the equation given below. The concentrations correspond to the pH values before and after treatment.

$$\text{Treatment efficiency} = \frac{(I - E)}{I} \times 100$$

where:

- I = Acid concentration before treatment; and
- E = Acid concentration after treatment.

**Example 5:** A nitric acid (HNO<sub>3</sub>) waste stream of pH 2.4 is neutralized to pH 4.6. Using Table 1, the initial nitric acid concentration is 0.0021000 mol/liter and the final concentration is 0.0000130 mol/liter. Substituting these values into the equation for treatment efficiency:

$$\begin{aligned}\text{Treatment Efficiency} &= \frac{(0.0021000 - 0.0000130)}{0.0021000} \times 100 \\ &= 99.4 \text{ percent}\end{aligned}$$

For strong acids only (including nitric acid), the net difference in pH before and after treatment can be used to estimate the treatment efficiency since pH is directly proportional to the acid concentration. For example, a pH change of one unit results in a treatment efficiency of 90 percent, whether the pH change is from pH 1 to pH 2 or from pH 4 to pH 5. Table 2 summarizes treatment efficiencies for various pH changes (the pH change is the difference between the initial pH and the pH after neutralization). In the table, some pH changes result in the same treatment efficiency values due to rounding to one decimal place.

**Table C-2**  
**Nitric Acid Treatment Efficiencies for Various pH Changes**

pH Unit Change	Treatment Efficiency (%)	pH Unit Change	Treatment Efficiency (%)
1.0	90.0	2.0	99.0
1.1	92.1	2.1	99.2
1.2	93.7	2.2	99.4
1.3	95.0	2.3	99.5
1.4	96.0	2.4	99.6
1.5	96.8	2.5	99.7
1.6	97.5	2.6	99.8
1.7	98.0	2.7	99.8
1.8	98.4	2.8	99.8
1.9	98.7	2.9	99.9
		3.0	99.9

**Example 6:** If a nitric acid (HNO<sub>3</sub>) waste stream of pH 2 is treated to pH 4, the pH change is 2 units. Using Table 2 above, the treatment efficiency is given as 99.0 percent.

### **Section 2.3.3 Estimating Releases of Nitrate Compounds Generated from the Neutralization of Nitric Acid**

The nitrate compounds produced from the complete neutralization (pH 6.0 or above) or partial neutralization (pH less than 6) or nitric acid are reportable under the nitrate compounds category if the appropriate threshold is met and should be included in all threshold and release calculations. In order to determine the quantity of a nitrate compound generated and released, the quantity of nitric acid released must be known (or calculated from the equations used in Examples 3 and 4 above) as well as the nitric acid treatment efficiency (calculated from the equations used in Examples 5 and 6 above).

The neutralization of nitric acid will most likely result in the generation of monovalent nitrate compounds (such as sodium nitrate and potassium nitrate). The quantity of these compounds formed in kilomoles will be equal to the quantity of the nitric acid neutralized in kilomoles. If divalent nitrate compounds are formed (such as calcium nitrate), the quantity of these compounds formed in kilomoles will be equal to one-half the quantity of the nitric acid neutralized in kilomoles. Similarly, if trivalent nitrate compounds are formed (such as iron (III) nitrate), the quantity formed of these compounds in kilomoles will be equal to one-third the quantity of the nitric acid neutralized in kilomoles. Note: to calculate the releases of nitrate compounds generated from the neutralization of nitric acid, the molecular weight of the nitrate compound formed must be used. Molecular weights of some of the individual chemicals within the water dissociable nitrate compounds category are given in Table 3.

**Example 7:** In a calendar year, a facility transfers 50,000 pounds of nitric acid (HNO<sub>3</sub>) to an on-site treatment facility. The nitric acid treatment efficiency is 95 percent, and the nitrate compound formed as a result of the treatment is sodium nitrate (NaNO<sub>3</sub>). The quantity of nitric acid transferred that is neutralized (generating sodium nitrate) is 95 percent of 50,000 pounds or 47,500 pounds. The molecular weight of nitric acid is 63.01 kg/kmol, and the molecular weight of sodium nitrate is 84.99 kg/kmol. The quantity of nitric acid neutralized is converted first to kilograms then to kilomoles using the following equations:

$$\begin{aligned}\text{Kilograms HNO}_3 \text{ neutralized} &= (\text{lbs HNO}_3 \text{ neutralized}) \times (0.4536 \text{ kg/lb}) \\ \text{Kilomoles HNO}_3 \text{ neutralized} &= (\text{kg HNO}_3) \div (\text{MW of HNO}_3 \text{ in kg/kmol})\end{aligned}$$

Substituting the example values into the above equation yields:

$$\text{Kilograms HNO}_3 \text{ neutralized} = 47,500 \text{ lbs} \times 0.4536 \text{ kg/lb} = 21,546 \text{ kg}$$

$$\text{Kilomoles HNO}_3 \text{ neutralized} = 21,546 \text{ kg} \div 63.01 \text{ kg/kmol} = 341.9 \text{ kmol}$$

The quantity of sodium nitrate generated in kilomoles is equal to the quantity of nitric acid neutralized (341.9 kmol). The quantity of sodium nitrate generated in kilomoles is converted first to kilograms then to pounds using the following equations:

$$\begin{aligned}\text{Kilograms NaNO}_3 \text{ generated} &= (\text{kmol NaNO}_3) \times (\text{MW of NaNO}_3 \text{ in kg/kmol}) \\ \text{Pounds NaNO}_3 \text{ generated} &= (\text{kg NaNO}_3) \times (2.205 \text{ lbs/kg})\end{aligned}$$

Substituting the values into the above equation yields:

$$\begin{aligned}\text{Kilograms NaNO}_3 \text{ generated} &= 341.9 \text{ kmol} \times 84.99 \text{ kg/kmol} = 29,058 \text{ kg} \\ \text{Pounds NaNO}_3 \text{ generated} &= 29,058 \text{ kg} \times 2.205 \text{ lbs/kg} = 64,073 \text{ pounds} \\ &(\text{reported as } 64,000 \text{ pounds})\end{aligned}$$

The 64,000 pounds of sodium nitrate generated is the quantity used to determine whether thresholds have been met or exceeded. The quantity of nitrate ion released is calculated as in Example 1 above.

## **Section 2.4 Generation of Nitrate Compounds from Biological Wastewater Treatment**

If a facility treats wastewater on-site biologically, using the activated sludge process, for example, the facility may be generating nitrate compounds as by-products of this biological process. The nitrate ion generated from this process will be associated with various counterions (e.g., sodium ion, potassium ion). In the absence of information on the identity of the counterion, a facility should assume for the purposes of EPCRA Section 313 threshold determinations that the counterion is sodium ion.

### Section 3. CAS Number List of Some of the Individual Chemicals within the Water Dissociable Nitrate Compounds Category

EPA is providing the following table of CAS numbers and chemical names to aid the regulated community in determining whether they need to report for the water dissociable nitrate compounds category. If a facility is manufacturing, processing, or otherwise using a chemical which is listed below, they must report this chemical. However, this list is not exhaustive. If a facility is manufacturing, processing, or otherwise using a water dissociable nitrate compound, they must report this chemical, even if it does not appear on the following list.

**Table C-3**  
**Listing by CAS Number and Molecular Weight of Some of the Individual Chemicals within the Water Dissociable Nitrate Compounds Category**

Chemical Name	Molecular Weight*	CAS Number
Aluminum nitrate, nonahydrate	213.00	7784-27-2
Ammonium nitrate	80.04	6484-52-2
Cerium (III) ammonium nitrate, tetrahydrate	486.22	13083-04-0
Cerium (IV) ammonium nitrate	548.23	10139-51-2
Barium nitrate	261.34	10022-31-8
Beryllium nitrate, trihydrate	133.02	7787-55-5
Cadmium nitrate	236.42	10325-94-7
Cadmium nitrate, tetrahydrate	236.42	10022-68-1
Calcium nitrate	164.09	10124-37-5
Calcium nitrate, tetrahydrate	164.09	13477-34-4
Cerium (III) nitrate, hexahydrate	326.13	10294-41-4
Cesium nitrate	194.91	7789-18-6
Chromium (III) nitrate, nonahydrate	238.01	7789-02-8
Cobalt (II) nitrate, hexahydrate	182.94	10026-22-9
Copper (II) nitrate, trihydrate	187.56	10031-43-3
Copper (II) nitrate, hexahydrate	187.56	13478-38-1
Dysprosium (III) nitrate, pentahydrate	348.51	10031-49-9
Erbium (III) nitrate, pentahydrate	353.27	10031-51-3
Gadolinium (III) nitrate, hexahydrate	343.26	19598-90-4
Gallium nitrate, hydrate	255.73	69365-72-6
Iron (III) nitrate, hexahydrate	241.86	13476-08-9
Iron (III) nitrate, nonahydrate	241.86	7782-61-8

**Table C-3 (Continued)**

<b>Chemical Name</b>	<b>Molecular Weight*</b>	<b>CAS Number</b>
Lanthanum (III) nitrate, hexahydrate	324.92	10277-43-7
Lead (II) nitrate	331.21	10099-74-8
Lithium nitrate	68.95	7790-69-4
Lithium nitrate, trihydrate	68.95	13453-76-4
Magnesium nitrate, dihydrate	148.31	15750-45-5
Magnesium nitrate, hexahydrate	148.31	13446-18-9
Manganese (II) nitrate, tetrahydrate	178.95	20694-39-7
Neodymium (III) nitrate, hexahydrate	330.25	16454-60-7
Nickel (II) nitrate, hexahydrate	182.70	13478-00-7
Potassium nitrate	101.10	7757-79-1
Rhodium (III) nitrate, dihydrate	288.92	13465-43-5
Rubidium nitrate	147.47	13126-12-0
Samarium (III) nitrate, hexahydrate	336.37	13759-83-6
Scandium (III) nitrate	230.97	13465-60-6
Scandium (III) nitrate, tetrahydrate	230.97	16999-44-3
Silver nitrate	169.87	7761-88-8
Sodium nitrate	84.99	7631-99-4
Strontium nitrate	211.63	10042-76-9
Strontium nitrate, tetrahydrate	211.63	13470-05-8
Terbium (III) nitrate, hexahydrate	344.94	13451-19-9
Thorium (IV) nitrate	480.06	13823-29-5
Thorium (IV) nitrate, tetrahydrate	480.06	13470-07-0
Yttrium (III) nitrate, hexahydrate	274.92	13494-98-9
Yttrium (III) nitrate, tetrahydrate	274.92	13773-69-8
Zinc nitrate, trihydrate	189.39	131446-84-9
Zinc nitrate, hexahydrate	189.39	10196-18-6
Zirconium (IV) nitrate, pentahydrate	339.24	13986-27-1

\*For hydrated compounds, e.g., aluminum nitrate, nonahydrate, the molecular weight excludes the weight of the hydrate portion. For example, the same molecular weight is provided for aluminum nitrate, nonahydrate and aluminum nitrate.



**Appendix D**

**GUIDANCE FOR REPORTING AQUEOUS AMMONIA**

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# EMERGENCY PLANNING AND COMMUNITY RIGHT-TO-KNOW EPCRA Section 313 Guidance for Reporting Aqueous Ammonia

EPCRA Section 313 of the Emergency Planning and Community Right-to-Know Act of 1986 (EPCRA) requires certain facilities manufacturing, processing, or otherwise using listed toxic chemicals to report their environmental releases of such chemicals annually. Beginning with the 1991 reporting year, such facilities also must report pollution prevention and recycling data for such chemicals, pursuant to section 6607 of the Pollution Prevention Act, 42 U.S.C. 13106. When enacted, EPCRA Section 313 established an initial list of toxic chemicals that was comprised of more than 300 chemicals and 20 chemical categories. EPCRA Section 313(d) authorizes EPA to add chemicals to or delete chemicals from the list, and sets forth criteria for these actions.

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## Section 1. Introduction

On June 30, 1995 EPA finalized four actions in response to a petition received in 1989 to delete ammonium sulfate (solution) from the list of toxic chemicals subject to reporting under EPCRA Section 313 of the Emergency Planning and Community Right-to-Know Act of 1986 (EPCRA), 42 U.S.C. 11001. The four actions taken are summarized as follows: (1) deleted ammonium sulfate (solution) from the EPCRA Section 313 list of toxic chemicals, (2) required that threshold and release determinations for aqueous ammonia be based on 10 percent of the total aqueous ammonia present in aqueous solutions of ammonia, (3) modified the ammonia listing by adding the following qualifier: ammonia (includes anhydrous ammonia and aqueous ammonia from water dissociable ammonium salts and other sources; 10 percent of total aqueous ammonia is reportable under this listing), and (4) deleted ammonium nitrate (solution) as a separately listed chemical on the EPCRA Section 313 list of toxic chemicals. All actions are effective for the 1994 reporting year for reports due July 1, 1995, with the exception of the deletion of ammonium nitrate (solution) as a separately listed chemical, which is effective for the 1995 reporting year for reports due July 1, 1996. At the time that these actions were finalized, EPA indicated that the Agency would develop, as appropriate, interpretations and guidance that the Agency determines are necessary to facilitate accurate reporting for aqueous ammonia. This document constitutes such guidance for reporting under the ammonia listing.

### Section 1.1 Who Must Report

A plant, factory, or other facility is subject to the provisions of EPCRA Section 313, if it meets all three of the following criteria:

- It conducts manufacturing operations (is included in Standard Industrial Classification (SIC) codes 20 through 39); and
- It has 10 or more full-time employees (or the equivalent 20,000 hours per year); and
- It manufactures, imports, processes, or otherwise uses any of the toxic chemicals listed on the EPCRA Section 313 list in amounts greater than the “threshold” quantities specified below.

### Section 1.2 Thresholds

Thresholds are specified amounts of toxic chemicals used during the calendar year that trigger reporting requirements.

If a facility *manufactures* or *imports* any of the listed toxic chemicals, the threshold quantity will be:

- 25,000 pounds per toxic chemical or category over the calendar year.

If a facility *processes* any of the listed toxic chemicals, the threshold quantity will be:

- 25,000 pounds per toxic chemical or category over the calendar year.

If a facility *otherwise uses* any of the listed toxic chemicals (without incorporating it into any product or producing it at the facility), the threshold quantity is:

- 10,000 pounds per toxic chemical or category over the calendar year.

### **Section 1.3 Chemical Sources of Aqueous Ammonia**

If a facility manufactures, processes, or otherwise uses anhydrous ammonia or aqueous ammonia, they must report under the ammonia listing. EPA is providing a table of Chemical Abstract Service (CAS) numbers and chemical names to aid the regulated community in determining whether they need to report under the ammonia listing for aqueous ammonia. This table includes a list of water dissociable ammonium salts which, when placed in water, are a source of aqueous ammonia. The table contains only commonly used ammonium salts and therefore is not exhaustive. If a facility manufactures, processes, or otherwise uses aqueous ammonia, regardless of its source, it must report under the ammonia listing, even if the source of the aqueous ammonia is not listed in the table provided in this document.

### **Section 1.4 *De Minimis* Concentrations**

The ammonia listing is subject to the one percent *de minimis* concentration. Thus, solutions containing aqueous ammonia at a concentration in excess of *one percent of the 10 percent reportable under this listing* should be factored into threshold and release determinations.

## Section 2. Guidance for Reporting Aqueous Ammonia

Note: for the purposes of reporting under the ammonia listing for aqueous ammonia, water dissociable ammonium salts means that the ammonium ion dissociates from its counterion when in solution.

### Section 2.1 Determining Threshold and Release Quantities for Ammonia

If a facility manufactures, processes, or otherwise uses *anhydrous ammonia*, the quantity applied towards threshold determinations for the ammonia listing is the total quantity of the anhydrous ammonia manufactured, processed, or otherwise used. The quantity reported when calculating the amount of ammonia that is released, transferred, or otherwise managed is the total quantity of *anhydrous ammonia* released or transferred.

If the facility manufactures, processes, or otherwise uses *anhydrous ammonia* in quantities that exceed the appropriate threshold and subsequently dissolves some or all of the *anhydrous ammonia* in water, then the following applies: 1) threshold determinations are based on 100 percent of the *anhydrous ammonia* (simply 10 percent of *aqueous ammonia*); 2) release, transfer, and other waste management quantities for the *aqueous ammonia* are calculated as 10 percent of total ammonia; 3) release, transfer, and other waste management quantities for the *anhydrous ammonia* are calculated as 100 percent of the *anhydrous ammonia*.

If a facility manufactures, processes, or otherwise uses *aqueous ammonia*, the quantity applied toward threshold determinations for the ammonia listing is 10 percent of the total quantity of the *aqueous ammonia* manufactured, processed, or otherwise used. The quantity reported when calculating the amount of ammonia that is released, transferred, or otherwise managed is 10 percent of the total quantity of *aqueous ammonia* released or transferred.

If a facility dissolves a water dissociable ammonium salt in water that facility has manufactured *aqueous ammonia* and 10 percent of the total *aqueous ammonia* manufactured from these salts is to be included in manufacturing threshold determinations under the ammonia listing.

If *aqueous ammonia* from water dissociable ammonium salts is processed or otherwise used, then 10 percent of the total *aqueous ammonia* is to be included in all processing and otherwise use threshold determinations under the ammonia listing.

**Example 1:** In a calendar year, a facility places 25,000 lbs of anhydrous ammonia in water for processing and processes 25,000 lbs of aqueous ammonia from an ammonium salt. The facility must include all of the 25,000 lbs of anhydrous ammonia in the determination of the processing threshold, but only 10 percent (or 2,500 lbs) of the aqueous ammonia from the ammonium salt in the processing threshold determination.

Total aqueous ammonia is the sum of the two forms of ammonia (un-ionized,  $\text{NH}_3$ , and ionized,  $\text{NH}_4^+$ ) present in aqueous solutions. A precise calculation of the weight of total aqueous ammonia would require determining the ratio of the two forms of ammonia present using the pH and temperature of the solution. The weight of total aqueous ammonia can be more

easily calculated by assuming that aqueous ammonia is comprised entirely of the  $\text{NH}_4^+$  form or the  $\text{NH}_3$  form. For the purpose of determining threshold and release quantities under EPCRA Section 313, EPA recommends that total aqueous ammonia be calculated in terms of  $\text{NH}_3$  equivalents (i.e., for determining weights, assume total ammonia is comprised entirely of the  $\text{NH}_3$  form). This method is simpler than using pH and temperature data to determine the ratio of the two forms present and is consistent with the presentation of total ammonia toxicity in a separate EPA document, *Ambient Water Quality Criteria for Ammonia* (EPA document #440/5-85-001, January 1985).

## **Section 2.2 Chemical Sources of Aqueous Ammonia**

Aqueous ammonia may be generated in solution from a variety of sources that include the release of anhydrous ammonia to water and the dissociation of ammonium salts in water. Water dissociable ammonium salts are not reportable in their entirety under the ammonia listing; these salts are reportable to the extent that they dissociate in water, and only 10 percent of the total aqueous ammonia that results when these salts dissociate is reportable. If these salts are not placed in water, they are not reportable.

If these salts are purchased neat or as solids by a facility, then placed in water by that facility, the facility is *manufacturing* aqueous ammonia.

### **Section 2.2.1 Reporting Aqueous Ammonia Generated from Anhydrous Ammonia in Water**

If the source of aqueous ammonia is anhydrous ammonia in water, total aqueous ammonia (calculated in terms of  $\text{NH}_3$  equivalents) is equal to the quantity of anhydrous ammonia manufactured, processed, or otherwise used. A hypothetical scenario demonstrating the calculations involved in reporting aqueous ammonia generated from anhydrous ammonia in water is given in Example 2.

**Example 2:** In a calendar year, a facility uses 30,000 pounds of anhydrous ammonia to neutralize acids in a wastewater stream. The neutralized waste stream (containing aqueous ammonia from dissociated ammonium salts) is then transferred to a POTW. The quantity to be applied toward threshold determinations is the total quantity of anhydrous ammonia used in the waste stream neutralization, or 30,000 pounds. The quantity of ammonia reported as transferred is 10 percent of the total quantity of aqueous ammonia transferred, or 3,000 pounds.

### **Section 2.2.2 Reporting Aqueous Ammonia Generated from the Dissociation of Ammonium Salts (Other Than Ammonium Nitrate)**

If the source of aqueous ammonia is the dissociation of ammonium salts in water, total aqueous ammonia (calculated in terms of  $\text{NH}_3$  equivalents) is calculated from the weight percent (wt%) of the  $\text{NH}_3$  equivalents of the ammonium salt. The  $\text{NH}_3$  equivalent wt% of an ammonium salt is calculated using the following equation:

$$\text{NH}_3 \text{ equivalent wt\%} = (\text{NH}_3 \text{ equivalent weight}) / (\text{MW ammonium salt}) \times 100$$

If the source of aqueous ammonia is a monovalent compound (such as ammonium chloride,  $\text{NH}_4\text{Cl}$ , ammonium nitrate,  $\text{NH}_4\text{NO}_3$ , or ammonium bicarbonate ( $\text{NH}_4\text{HCO}_3$ ), the  $\text{NH}_3$  equivalent weight is equal to the MW of  $\text{NH}_3$  (17.03 kg/kmol). If divalent compounds are involved (such as ammonium carbonate,  $(\text{NH}_4)_2\text{CO}_3$ ), then the  $\text{NH}_3$  equivalent weight is equal to the MW of  $\text{NH}_3$  multiplied by two. Similarly, if trivalent compound are involved, then the  $\text{NH}_3$  equivalent weight is equal to the MW of  $\text{NH}_3$  multiplied by three.

Example 3:

The  $\text{NH}_3$  equivalent wt% of ammonium chloride is calculated as follows:

$$\text{NH}_3 \text{ equivalent wt\%} = (\text{NH}_3 \text{ equivalent weight})/(\text{MW ammonium chloride}) \times 100$$

$$\text{NH}_3 \text{ equivalent wt\%} = (17.03)/(53.49) \times 100$$

$$\text{NH}_3 \text{ equivalent wt\%} = 31.84\%$$

The  $\text{NH}_3$  equivalent wt% of ammonium carbonate is calculated as follows:

$$\text{NH}_3 \text{ equivalent wt\%} = 2 \times (\text{NH}_3 \text{ equivalent weight})/(\text{MW ammonium chloride}) \times 100$$

$$\text{NH}_3 \text{ equivalent wt\%} = 2 \times (17.03)/(96.09) \times 100$$

$$\text{NH}_3 \text{ equivalent wt\%} = 35.45\%$$

To aid the regulated community in reporting under the ammonia listing for aqueous ammonia, the table of chemical sources of aqueous ammonium provided in Section 3 of this document includes, in addition to CAS number, chemical name, and molecular weight, the  $\text{NH}_3$  equivalent wt% of the commonly used, water dissociable ammonium salts listed in this table.

Example 4: In a calendar year, a facility uses 100,000 pounds of ammonium chloride,  $\text{NH}_4\text{Cl}$ , *in aqueous solution* which is released to wastewater streams, then transferred to a POTW. The  $\text{NH}_3$  equivalent wt% of ammonium chloride is 31.84% (taken from Table 1 in Section 3 below or calculated as in Example 3 above). The total quantity of aqueous ammonia present in solution is 31.84% of the 100,000 pounds of ammonia chloride used, or 31,840 pounds. The quantity applied towards threshold determinations for the ammonia listing is 10 percent of the total quantity of aqueous ammonia present in solution, or 3,184 pounds. The quantity of ammonia reported as released or transferred is 10 percent of the total quantity of aqueous ammonia released or transferred, or 3,184 pounds.

**Example 5:** In a calendar year, a facility uses 500,000 pounds of ammonium carbonate,  $(\text{NH}_4)_2\text{CO}_3$ , and 400,000 pounds of ammonium bicarbonate,  $\text{NH}_4\text{HCO}_3$ , in aqueous solution which is released to wastewater streams, then transferred to a POTW. The  $\text{NH}_3$  equivalent wt% of ammonium carbonate is 35.45%, and the  $\text{NH}_3$  equivalent wt% of ammonium bicarbonate is 21.54% (taken from Table 1 in Section 3 below or calculated as in Example 3 above). The quantity of aqueous ammonia present in solution from ammonium carbonate is 35.45% of the 500,000 pounds of ammonia carbonate used, or 177,250 pounds. The quantity of aqueous ammonia present in solution from ammonium bicarbonate is 21.54% of the 400,000 pounds of ammonia bicarbonate used or 86,160 pounds. The total quantity of aqueous ammonia present in solution is 263,410 pounds. The quantity applied towards threshold determinations for the ammonia listing is 10 percent of the total quantity of aqueous ammonia present in solution, or 26,341 pounds. The quantity of ammonia reported as released or transferred is 10 percent of the total quantity of aqueous ammonia released or transferred, or 26,341 pounds.

### **Section 2.2.3 Reporting Aqueous Ammonia Generated from the Dissociation of Ammonium Nitrate**

Some sources of aqueous ammonia may be reportable under other EPCRA Section 313 category listings. Ammonium nitrate (solution) is relevant to reporting under the ammonia listing to the extent that 10 percent of the total aqueous ammonia that results when ammonium nitrate dissociates is reported when determining thresholds and calculating releases. However, under the nitrate compound category listing, ammonium nitrate (and other mixed salts containing ammonium and nitrate) must be reported in its entirety. When reporting ammonium nitrate under this category listing, the total nitrate compound, including both the nitrate ion portion and the ammonium counterion, is included when determining threshold quantities. However, only the nitrate ion portion is included when determining the amount of ammonium nitrate that is released, transferred, or otherwise managed in wastes. The calculations involved in determining threshold and release quantities for reporting under the nitrate compound category listing are described in a separate directive, *List of Toxic Chemicals within the Water Dissociable Nitrate Compounds Category and Guidance for Reporting* (EPA document #745-R-95-002, February 1995). Note: reporting ammonium nitrate under the ammonia listing and nitrate compounds category listing is effective for the 1995 reporting year for reports due July 1, 1996.



Example 6: In a calendar year, a facility uses 1,250,000 pounds of ammonium nitrate,  $\text{NH}_4\text{NO}_3$ , *in aqueous solution* which is released to wastewater streams, then transferred to a POTW. The  $\text{NH}_3$  equivalent wt% of ammonium nitrate is 21.28% (taken from Table 1 in Section 3 below or calculated as in Example 3 above). The total quantity of aqueous ammonia present in solution is 21.28% of the 1,250,000 pounds of ammonia chloride used, or 266,000 pounds. The quantity applied towards threshold determinations for the ammonia listing is 10 percent of the total quantity of aqueous ammonia present in solution, or 26,600 pounds. The quantity of ammonia reported as released or transferred is 10 percent of the total quantity of aqueous ammonia released or transferred, or 26,600 pounds. For determining thresholds and calculating releases under the nitrate compound category listing, see the separate directive, *List of Toxic Chemicals within the Water Dissociable Nitrate Compounds Category and Guidance for Reporting* (EPA document #745-R-95-002, February, 1995).

Example 7: In a calendar year, a facility transfers 100,000 pounds of nitric acid (HNO<sub>3</sub>) to an on-site treatment facility. The nitric acid is neutralized with anhydrous ammonia, and treatment efficiency is 95 percent (the nitrate compound formed as a result of the treatment is ammonium nitrate, NH<sub>4</sub>NO<sub>3</sub>). The neutralized waste stream (containing aqueous ammonia from dissociated ammonium nitrate) is then transferred to a POTW. The quantity of nitric acid neutralized is 95 percent of 100,000 pounds or 95,000 pounds. The quantity of nitric acid neutralized is converted first to kilograms then to kilomoles using the following equations:

$$\begin{aligned}\text{Kilograms HNO}_3 \text{ neutralized} &= (\text{lbs HNO}_3 \text{ neutralized}) \times (0.4536 \text{ kg/lb}) \\ \text{Kilomoles HNO}_3 \text{ neutralized} &= (\text{kg HNO}_3) \div (\text{MW of HNO}_3 \text{ in kg/kmol})\end{aligned}$$

Substituting the appropriate values into the above equations yields:

$$\begin{aligned}\text{Kilograms HNO}_3 \text{ neutralized} &= 95,000 \text{ lbs} \times 0.4536 \text{ kg/lb} = 43,092 \text{ kg} \\ \text{Kilomoles HNO}_3 \text{ neutralized} &= 43,092 \text{ kg} \div 63.01 \text{ kg/kmol} = 683.9 \text{ kmol}\end{aligned}$$

The quantity of anhydrous ammonia used in kilomoles in the acid neutralization and the quantity of ammonium nitrate generated in kilomoles from the neutralization are equal to the quantity of nitric acid neutralized (683.9 kmol). The quantity of anhydrous ammonia used in kilograms and pounds in the acid neutralization is calculated as follows:

$$\begin{aligned}\text{Kilograms NH}_3 \text{ used} &= (\text{kmol NH}_3) \times (\text{MW of NH}_3 \text{ in kg/kmol}) \\ \text{Pounds NH}_3 \text{ used} &= (\text{kg NH}_3) \times (2.205 \text{ lbs/kg})\end{aligned}$$

Substituting the appropriate values into the above equation yields:

$$\begin{aligned}\text{Kilograms NH}_3 \text{ used} &= (683.9 \text{ kmol}) \times (17.03 \text{ kg/kmol}) = 11,647 \text{ kmol} \\ \text{Pounds NH}_3 \text{ used} &= (11,647 \text{ NH}_3) \times (2,205 \text{ lbs/kg}) = 25,682 \text{ pounds}\end{aligned}$$

The quantity reported applied towards threshold determinations for the ammonia listing is the total quantity of anhydrous ammonia used in the acid neutralization, or 25,682 pounds. The quantity of ammonia reported as released or transferred is 10 percent of the total quantity of aqueous ammonia released or transferred, or 2,568 pounds. For determining thresholds and calculating releases under the nitrate compound category listing, see the separate directive, *List of Toxic Chemicals within the Water Dissociable Nitrate Compounds Category and Guidance for Reporting* (EPA document #745-R-95-002, February 1995).

### Section 3. CAS Number and List of Some Chemical Sources of Aqueous Ammonia

EPA is providing the following table of CAS numbers and chemical names to aid the regulated community in determining whether they need to report under the ammonia listing for aqueous ammonia. If a facility manufactures, processes, or otherwise uses, *in aqueous solution*, a chemical which is listed below, they must report 10 percent of the total aqueous ammonia that is the result of the dissociation of this chemical. However, this list is not exhaustive. If a facility manufactures, processes, or otherwise uses, *in aqueous solution*, a water dissociable ammonium compound, they must report 10 percent of the total aqueous ammonia that is the result of the dissociation of the compound, even if the compound does not appear in the following table.

**Table D-1**  
**Listing by CAS Number and Molecular Weight of**  
**Some Chemical Sources of Aqueous Ammonia**

Chemical Name	Molecular Weight*	NH <sub>3</sub> Equivalent Wt%	CAS Number
Ammonium acetate	77.08	22.09	631-61-8
Ammonium aluminum sulfate (Ammonium aluminum disulfate)	237.14	7.181	7784-25-0
Ammonium antimony fluoride (Diammonium pentafluoroantimonate)	252.82	13.47	32516-50-0
Ammonium arsenate (Ammonium arsenate, hydrogen) (Ammonium arsenate, dihydrogen)	158.97	10.71	13462-93-6
Ammonium arsenate (Diammonium arsenate) (Diammonium arsenate, hydrogen) (Diammonium arsenate, monohydrogen)	176.00	19.35	7784-44-3
Ammonium arsenite	124.96	13.63	13462-94-7
Ammonium azide	60.06	28.35	12164-94-2
Ammonium benzenesulfonate	175.20	9.720	19402-64-3
Ammonium benzoate	139.15	12.24	1863-63-4
Ammonium bromate	145.94	11.67	13843-59-9
Ammonium bromide	97.94	17.39	12124-97-9
Ammonium cadmium chloride (Ammonium cadmium trichloride)	236.81	7.191	18532-52-0
Ammonium carbamate	78.07	21.81	1111-78-0
Ammonium carbonate carbamate	157.13	21.68	8000-73-5
Ammonium carbonate (Diammonium carbonate)	96.09	35.45	506-87-3

**Table D-1 (Continued)**

Chemical Name	Molecular Weight*	NH <sub>3</sub> Equivalent Wt%	CAS Number
Ammonium carbonate, hydrogen (Ammonium bicarbonate)	79.06	21.54	1066-33-7
Ammonium cerium nitrate (Ammonium hexanitratocerate) (Ammonium hexanitratocerate (IV)) (Diammonium cerium hexanitrate)	548.23	6.213	16774-21-3
Ammonium cerous nitrate (Ammonium cerous nitrate, tetrahydrate)	486.22	7.005	13083-04-0
Ammonium chlorate	101.49	16.78	10192-29-7
Ammonium perchlorate	117.49	14.49	7790-98-9
Ammonium chloride	53.49	31.84	12125-02-9
Ammonium chromate (Ammonium chromate (VI)) (Diammonium chromate)	152.07	22.40	7788-98-9
Ammonium chromate (Ammonium dichromate) (Ammonium dichromate (VI)) (Ammonium bichromate) (Diammonium dichromate)	252.06	13.51	7789-09-5
Ammonium chromium sulfate (Ammonium chromic sulfate)	265.17	6.422	13548-43-1
Ammonium citrate (Ammonium citrate, monohydrogen) (Ammonium citrate, dibasic) (Diammonium citrate) (Diammonium citrate, hydrogen)	226.19	15.06	3012-65-5
Ammonium citrate (Ammonium citrate, tribasic) (Triammonium citrate)	243.22	21.01	3458-72-8
Ammonium cobalt sulfate (Ammonium cobaltous sulfate)	289.14	11.78	13596-46-8
Ammonium cupric chloride (Ammonium chlorocuprate (II)) (Diammonium copper tetrachloride) (Diammonium tetrachlorocuprate)	241.43	14.11	15610-76-1
Ammonium cyanate (Ammonium isocyanate)	60.06	28.35	22981-32-4
Ammonium cyanide	44.06	38.65	12211-52-8
Ammonium cyanoaurate, monohydrate (Ammonium tetracyanoaurate, monohydrate)	319.07	5.337	14323-26-3

**Table D-1 (Continued)**

Chemical Name	Molecular Weight*	NH <sub>3</sub> Equivalent Wt%	CAS Number
Ammonium cyanoaurate (Ammonium dicyanoaurate)	267.04	6.377	31096-40-9
Ammonium ferricyanide (Ammonium hexacyanoferrate (III)) (Triammonium hexacyanoferrate)	266.07	19.20	14221-48-8
Ammonium ferrocyanide (Ammonium hexacyanoferrate (II)) (Tetraammonium ferrocyanide) (Tetraammonium hexacyanoferrate)	284.11	23.98	14481-29-9
Ammonium fluoride	37.04	45.98	12125-01-8
Ammonium fluoride (Ammonium difluoride) (Ammonium bifluoride) (Ammonium fluoride, hydrogen) (Ammonium difluoride, hydrogen) (Ammonium bifluoride, hydrogen)	57.04	29.86	1341-49-7
Ammonium fluoroborate (Ammonium tetrafluoroborate)	104.84	16.24	13826-83-0
Ammonium fluorogermanate (IV) (Ammonium hexafluorogermanate (IV)) (Diammonium hexafluorogermanate)	222.66	15.30	16962-47-3
Ammonium fluorophosphate (Ammonium hexafluorophosphate)	163.00	10.45	16941-11-0
Ammonium fluorosulfate (Ammonium fluorosulfonate)	117.10	14.54	13446-08-7
Ammonium formate	63.06	27.01	540-69-2
Ammonium gallium sulfate	282.90	6.020	15335-98-5
Ammonium hydroxide	35.05	48.59	1336-21-6
Ammonium iodide	144.94	11.75	12027-06-4
Ammonium iridium chloride (Ammonium chloroiridate (III)) (Ammonium hexachloroiridate) (Triammonium hexachloroiridate)	459.05	11.13	15752-05-3
Ammonium iron sulfate (Ammonium ferric sulfate) (Ammonium iron disulfate)	269.02	6.330	10138-04-2
Ammonium iron sulfate (Ammonium ferrous sulfate) (Diammonium iron disulfate) (Diammonium ferrous disulfate)	286.05	11.91	10045-89-3

**Table D-1 (Continued)**

Chemical Name	Molecular Weight*	NH <sub>3</sub> Equivalent Wt%	CAS Number
Ammonium lactate (Ammonium 2-hydroxypropionate)	107.11	15.90	515-98-0
Ammonium laurate (Ammonium dodecanoate)	217.35	7.835	2437-23-2
Ammonium magnesium sulfate	252.50	13.49	14727-95-8
Ammonium malate	168.15	20.26	6283-27-8
Ammonium malate, hydrogen (Ammonium bimalate)	151.12	11.27	5972-71-4
Ammonium molybdate (Diammonium molybdate)	196.01	17.38	13106-76-8
Ammonium molybdate (Ammonium heptamolybdate) (Ammonium molybdate, hydrate) (Ammonium molybdate, tetrahydrate) (Ammonium <i>paramolybdate</i> , tetrahydrate)	1,163.8	8.780	12054-85-2
Ammonium nickel chloride, hexahydrate	183.09	9.301	16122-03-5
Ammonium nickel sulfate (Ammonium nickel sulfate, hexahydrate) (Ammonium nickel disulfate, hexahydrate) (Diammonium nickel disulfate, hexahydrate)	286.88	11.87	7785-20-8
Ammonium nitrate	80.04	21.28	6484-52-2
Ammonium nitrate sulfate	212.18	24.08	12436-94-1
Ammonium nitrite	64.04	26.59	13446-48-5
Ammonium oleate	299.50	5.686	544-60-5
Ammonium oxalate	124.10	27.45	1113-38-8
Ammonium palladium chloride (Ammonium chloropalladate (II)) (Ammonium tetrachloropalladate (II)) (Diammonium tetrachloropalladate)	284.31	11.98	13820-40-1
Ammonium phosphate (Ammonium orthophosphate)	149.09	34.27	10124-31-9
Ammonium phosphate (Ammonium biphosphate) (Ammonium phosphate, hydrogen) (Ammonium phosphate, dihydrogen) (Ammonium orthophosphate, dihydrogen) (Ammonium phosphate, monobasic)	115.03	14.80	7722-76-1

**Table D-1 (Continued)**

Chemical Name	Molecular Weight*	NH <sub>3</sub> Equivalent Wt%	CAS Number
Ammonium phosphate (Ammonium phosphate, hydrogen) (Ammonium orthophosphate, monohydrogen) (Ammonium phosphate, dibasic) (Ammonium orthophosphate, dibasic) (Diammonium phosphate) (Diammonium orthophosphate) (Diammonium phosphate, hydrogen) (Diammonium phosphate, monohydrogen) (Diammonium orthophosphate, hydrogen)	132.06	25.79	7783-28-0
Ammonium phosphinate (Ammonium hypophosphite)	83.03	20.51	7803-65-8
Ammonium phosphite (Ammonium biphosphite) (Ammonium phosphite, dihydrogen)	99.03	17.20	13446-12-3
Ammonium picramate	216.15	7.879	1134-85-6
Ammonium propionate	91.11	18.69	17496-08-1
Ammonium rhodium chloride (Ammonium chlororhodate (III)) (Ammonium hexachlororhodate (III)) (Triammonium rhodium hexachloride) (Triammonium hexachlororhodate)	369.74	13.82	15336-18-2
Ammonium salicylate (Ammonium 2-hydroxybenzoate)	155.15	10.98	528-94-9
Ammonium selenide	115.04	29.61	66455-76-3
Ammonium silicon fluoride (Ammonium fluorosilicate) (Ammonium hexafluorosilicate) (Diammonium silicon hexafluoride) (Diammonium fluorosilicate) (Diammonium hexafluorosilicate)	178.15	19.12	16919-19-0
Ammonium stearate (Ammonium octadecanoate)	301.51	5.648	1002-89-7
Ammonium succinate (Diammonium succinate)	152.15	22.39	2226-88-2
Ammonium sulfamate (Ammonium amidosulfate) (Ammonium amidosulfonate)	114.12	14.92	7773-06-0
Ammonium sulfate (Diammonium sulfate)	132.13	25.78	7783-20-2

**Table D-1 (Continued)**

Chemical Name	Molecular Weight*	NH <sub>3</sub> Equivalent Wt%	CAS Number
Ammonium sulfate (Ammonium bisulfate) (Ammonium sulfate, hydrogen) (Ammonium sulfate, monohydrogen)	115.10	14.80	7803-63-6
Ammonium <i>persulfate</i> (Ammonium peroxy sulfate) (Ammonium peroxydisulfate) (Diammonium persulfate) (Diammonium peroxydifulsite)	228.19	14.93	7727-54-0
Ammonium sulfide (Ammonium bisulfide) (Ammonium sulfide, hydrogen)	51.11	33.32	12124-99-1
Ammonium sulfide (Ammonium monosulfide) (Diammonium sulfide)	68.14	49.99	12135-76-1
Ammonium sulfide (Diammonium pentasulfide)	196.39	17.34	12135-77-2
Ammonium sulfite, monohydrate (Diammonium sulfite, monohydrate)	116.13	29.33	7783-11-1
Ammonium sulfite (Ammonium bisulfite) (Ammonium sulfite, hydrogen)	99.10	17.18	10192-30-0
Ammonium tetrachloroaurate (III), hydrate	356.82	4.772	13874-04-9
Ammonium thiocarbamate	94.13	18.09	16687-42-6
Ammonium thiocarbonate (Diammonium trithiocarbonate)	144.27	23.61	13453-08-2
Ammonium thiocyanate (Ammonium isothiocyanate) (Ammonium sulfocyanate) (Ammonium rhodanate) (Rhodanid)	76.12	22.37	1762-95-4
Ammonium dithionate	196.19	17.36	60816-52-6
Ammonium thiosulfate (Ammonium hyposulfite) (Diammonium thiosulfate)	148.20	22.98	7783-18-8
Ammonium tin bromide (Ammonium bromostannate (IV)) (Ammonium hexabromostannate (IV)) (Diammonium hexabromostannate)	634.19	5.371	16925-34-1



Chemical Name	Molecular Weight*	NH <sub>3</sub> Equivalent Wt%	CAS Number
Ammonium tin chloride (Ammonium chlorostannate (IV)) (Ammonium hexachlorostannate (IV)) (Diammonium tin hexachloride) (Diammonium hexachlorostannate)	367.48	9.269	16960-53-5
Ammonium titanium fluoride (Ammonium fluorotitanate (IV)) (Ammonium hexafluorotitanate (IV)) (Diammonium titanium hexafluoride) (Diammonium hexafluorotitanate)	197.95	17.21	16962-40-6
Ammonium titanium oxalate, monohydrate (Diammonium dioxalatooxotitanate, monohydrate)	276.00	12.34	10580-03-7
Ammonium tungstate (Ammonium tungstate (VI)) (Ammonium <i>paratungstate</i> ) (Hexaammonium tungstate)	1,779.2	5.743	12028-06-7
Ammonium tungstate (Ammonium tungstate (VI)) (Ammonium <i>paratungstate</i> ) (Decaammonium tungstate)	3,058.6	5.568	11120-25-5
Ammonium valerate (Ammonium pentoate)	119.16	14.29	42739-38-8
Ammonium zinc chloride (Ammonium chlorozincate) (Ammonium tetrachlorozincate) (Diammonium tetrachlorozincate)	243.27	14.00	14639-97-5

\*For hydrated compounds, e.g., ammonium sulfite, monohydrate, the molecular weight excludes the weight of the hydrate portion.

**Appendix E**

**UNIT CONVERSION FACTORS**

**(From U.S. Coast Guard Commandant Instruction M.16465.12A)**

## CONVERSION FACTORS

To Convert	To	Multiply By
<b>Length</b>		
inches	millimeters	25.4
inches	feet	0.0833
feet	inches	12
feet	meters	0.3048
feet	yards	0.3333
feet	miles (U.S. statute)	0.0001894
yards	feet	3
yards	miles (U.S. statute)	0.0005682
miles (U.S. statute)	feet	5280
miles (U.S. statute)	yards	1760
miles (U.S. statute)	meters	1609
miles (U.S. statute)	nautical miles	0.868
meters	feet	3.271
meters	yards	1.094
meters	miles (U.S. statute)	0.0006214
nautical miles	miles (U.S. statute)	1.152
<b>Area</b>		
square inches	square centimeters	6.452
square inches	square feet	0.006944
square feet	square inches	144
square feet	square meters	0.09290
square meters	square feet	10.76
square miles	square yards	3,097,600
square yards	square feet	9
<b>Volume</b>		
cubic inches	cubic centimeters	16.39
cubic inches	cubic feet	0.0005787
cubic feet	cubic inches	1728
cubic feet	cubic meters	0.02832
cubic feet	U.S. gallons	7.481
cubic meters	cubic feet	35.31
liters	quarts (U.S. liquid)	1.057
quarts (U.S. liquid)	liters	0.9463
U.S. gallons	barrels (petroleum)	0.02381
U.S. gallons	cubic feet	0.1337
U.S. gallons	Imperial gallons	0.8327
barrels (petroleum)	U.S. gallons	42
Imperial gallons	U.S. gallons	1.201
milliliters	cubic centimeters	1

## CONVERSION FACTORS (Continued)

To Convert	To	Multiply By
<b>Time</b>		
seconds	minutes	0.01667
seconds	hours	0.0002778
seconds	days	0.00001157
minutes	seconds	60
minutes	hours	0.01667
minutes	days	0.0006944
hours	seconds	3600
hours	minutes	60
hours	days	0.04167
<b>Mass or Weight</b>		
pounds	kilograms	0.4536
pounds	short tons	0.0005
pounds	long tons	0.000464
pounds	metric tons	0.0004536
tons (short)	pounds	2000
tons (metric)	pounds	2205
tons (long)	pounds	2240
kilograms	pounds	2.205
tonnes (metric tons)	kilograms	1000
<b>Energy</b>		
calories	Btu	0.003968
calories	joules	4.187
Btu (British thermal units)	calories	252.0
Btu	joules	1055
joules	calories	0.2388
joules	Btu	0.0009479
<b>Velocity</b>		
feet per second	meters per second	0.3048
feet per second	miles per hour	0.6818
feet per second	knots	0.5921
meters per second	feet per second	3.281
meters per second	miles per hour	2.237
miles per hour	meters per second	0.4470
miles per hour	feet per second	1.467
knots	meters per second	0.5148
knots	miles per hour	1.151
knots	feet per second	1.689
pounds per cubic foot	grams per cubic centimeter	0.01602
grams per cubic centimeter	pounds per cubic foot	62.42
grams per cubic centimeter	kilograms per cubic meter	1000
kilograms per cubic meter	grams per cubic centimeter	0.001

## CONVERSION FACTORS (Continued)

To Convert	To	Multiply By
<b>Pressure</b>		
ponds per square inch (absolute) (psia)	kilonewtons per square meter (kN/m <sup>2</sup> )	6.895
psia	atmospheres	0.0680
psia	inches of water	27.67
psia	millimeters of mercury (torr)	51.72
pounds per square inch (gauge) (psig)	psia	add 14.70
millimeters of mercury (torr)	psia	0.01934
millimeters of mercury (torr)	kN/m <sup>2</sup>	0.1333
inches of water	psia	0.03614
kilograms per square centimeter	millimeters of mercury (torr)	735.6
inches of water	kN/m <sup>2</sup>	0.2491
kilograms per square centimeter	atmospheres	0.9678
atmospheres	kN/m <sup>2</sup>	101.3
kilograms per square centimeter	psia	14.22
atmospheres	psia	14.70
bars	kN/m <sup>2</sup>	100
kilonewtons per square meter (kN/m <sup>2</sup> )	psia	0.1450
bars	atmospheres	0.9869
kilonewtons per square meter (kN/m <sup>2</sup> )	atmospheres	0.009869
bars	kilograms per square centimeter	1.020
<b>Viscosity</b>		
centipoises	pounds per foot per second	0.0006720
pounds per foot per second	centipoises	1488
centipoises	poises	0.01
centipoises	Newton seconds per square meter	0.001
poises	grams per centimeter per second	1
grams per centimeter per second	poises	1
Newton seconds per square meter	centipoises	1000
<b>Thermal Conductivity</b>		
Btu per hour per foot per °F	watts per meter-kelvin	1.731
Btu per hour per foot per °F	kilocalories per hour per meter per °C	1.488
watts per meter-kelvin	Btu per hour per foot per °F	0.5778
kilocalories per hour per meter per °C	watts per meter-kelvin	1.163
kilocalories per hour per meter per °C	Btu per hour per foot per °F	0.6720
<b>Heat Capacity</b>		
Btu per pound per °F	calories per gram per °C	1
Btu per pound per °F	joules per kilogram-kelvin	4187
joules per kilogram-kelvin	Btu per pound per °F	0.0002388
calories per gram per °C	Btu per pound per °F	1
<b>Concentration (in water solution)</b>		
parts per million (ppm)	milligrams per liter	1
milligrams per liter	ppm	1
milligrams per cubic meter	grams per cubic centimeter	1×10 <sup>-9</sup>
grams per cubic centimeter	milligrams per cubic meter	1×10 <sup>9</sup>
grams per cubic centimeter	pounds per cubic foot	62.42

## CONVERSION FACTORS (Continued)

To Convert	To	Multiply By
<b>Temperature</b>		
degrees Kelvin ( $^{\circ}$ K)	degrees Rankine ( $^{\circ}$ R)	1.8
degrees Rankine ( $^{\circ}$ R)	degrees Kelvin ( $^{\circ}$ K)	0.5556
degrees centigrade ( $^{\circ}$ C)	degrees Fahrenheit ( $^{\circ}$ F)	first multiply by 1.8, then add 32
degrees Fahrenheit ( $^{\circ}$ F)	degrees centigrade ( $^{\circ}$ C)	first subtract 32, then multiply by 0.5556
degrees centigrade ( $^{\circ}$ C)	degrees Kelvin ( $^{\circ}$ K)	add 273.2
degrees Fahrenheit ( $^{\circ}$ F)	degrees Kelvin ( $^{\circ}$ K)	add 459.7
<b>Flow</b>		
cubic feet per second	U.S. gallons per minute	448.9
U.S. gallons per minute	cubic feet per second	0.002228
<b>Universal Gas Constant (R)</b>		
8.314 joules per gram mole-kelvin		
1.987 calories per gram mole-kelvin		
1.987 Btu per pound mole per $^{\circ}$ F		
10.73 psia-cubic feet per pound mole per $^{\circ}$ F		
82.057 atm-cubic centimeters per gram mole-kelvin		
62.361 millimeters mercury liter per gram mole-kelvin		

**Appendix F**

**EXCERPT FROM: LIST OF CHEMICALS WITHIN THE  
GLYCOL ETHERS CATEGORY**



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# TOXICS RELEASE INVENTORY

## List of Toxic Chemicals within the Glycol Ethers Category

Section 313 of the Emergency Planning and Community Right-to-Know Act (EPCRA) requires certain facilities manufacturing, processing, or otherwise using listed toxic chemicals to report their environmental releases of such chemicals annually. Beginning with the 1991 reporting year, such facilities also must report pollution prevention and recycling data for such chemicals, pursuant to section 6607 of the Pollution Prevention Act, 42 U.S.C. 13106. When enacted, EPCRA Section 313 established an initial list of toxic chemicals that was comprised of more than 300 chemicals and 20 chemical categories. EPCRA Section 313(d) authorizes EPA to add chemicals to or delete chemicals from the list, and sets forth criteria for these actions.

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## Section 1. Introduction

On June 28, 1994, EPA promulgated a final rule (published in the Federal Register July 5, 1994) modifying the definition of the glycol ethers category on the list of toxic chemicals under Section 313 of the Emergency Planning and Community Right-to-Know Act of 1986 (EPCRA), 42 U.S.C. 11001 *et seq.* The effect of this modification, which is described at 59 FR 34386, is that many high molecular weight glycol ethers were excluded from the category. As described in the final rule, this modification was made retroactive for the 1993 reporting year for reports due July 1, 1994. This modification is also effective for all subsequent reporting years. As part of this modification and as the result of public comment, EPA changed the category name at 40 CFR 372.65(c) from 'glycol ethers' to 'certain glycol ethers.' However, this document will refer to the constituents of this category simply as glycol ethers. At the time of the modification, EPA indicated that the Agency would work with the public and the regulated community to develop, as appropriate, interpretations and guidance that the Agency determines are necessary to facilitate accurate reporting for the modified glycol ethers category. This document constitutes such guidance.

### Who Must Report

A plant, factory, or other facility is subject to the provisions of EPCRA section 313, if it meets all three of the following criteria:

- It conducts manufacturing operations (is included in Standard Industrial Classification (SIC) codes 20 through 39); and
- It has 10 or more full-time employees (or the equivalent 20,000 hours per year); and
- It manufactures, imports, processes, or otherwise uses any of the toxic chemicals listed on the EPCRA section 313 list in amounts greater than the "threshold" quantities specified below.

### Thresholds

Thresholds are specified amounts of toxic chemicals *manufactured, processed, or otherwise used* during the calendar year that trigger reporting requirements.

If a facility *manufactures or imports* any of the listed toxic chemicals, the threshold quantity will be:

- 25,000 pounds per toxic chemical or category over the calendar year.

If a facility *processes* any of the listed toxic chemicals, the threshold quantity will be:

- 25,000 pounds per toxic chemical or category over the calendar year.

If a facility *otherwise uses* any of the listed toxic chemicals, the threshold quantity is:

- 10,000 pounds per toxic chemical or category over the calendar year.

EPCRA section 313 requires threshold determinations for chemical categories to be based on the total of all chemicals in the category manufactured, processed, or otherwise used. For example, a facility that manufactures three members of a chemical category would count the total amount of all three chemicals manufactured towards the manufacturing threshold for that category. When filing reports for chemical categories, the releases are determined in the same manner as the thresholds. One report is filed for the category and all releases are reported on this form.

### **Glycol Ethers Category Definition**

The glycol ethers category is defined by the following formula:



where:

- n = 1, 2, or 3;
- R = Alkyl C7 or less, or phenyl or alkyl substituted phenyl;
- R' = H or alkyl C7 or less, or  
OR' consisting of a carboxylic acid ester, sulfate, phosphate, nitrate, or sulfonate.

Chemicals that meet this category definition are reportable.

EPA is providing three lists of CAS numbers and chemical names to aid the regulated community in determining whether they need to report for the glycol ethers category. Section 2 (pages 5 to 159) lists individual chemicals that meet the definition of the EPCRA section 313 'certain glycol ethers' category. This list consists only of chemicals that have been assigned CAS numbers and, thus, is not exhaustive. If a facility manufactures, processes, or otherwise uses, in greater than threshold quantities, a glycol ether that meets the category definition, whether or not that chemical is on the list, they must report the chemical.

Section 3 (pages 160 to 161) lists chemical mixtures which contain glycol ethers that meet the definition of the EPCRA section 313 'certain glycol ethers' category. EPA has tried to make this list as complete as possible; however, not all mixtures that contain reportable glycol ethers will necessarily appear on this list. If a facility manufactures, processes, or otherwise uses a mixture which includes components, in greater than threshold quantities, that meet the definition of the EPCRA section 313 'certain glycol ethers' category, they must report the glycol ether component(s) that meet the category definition.

Section 4 (pages 162 to 180) lists oligomeric and polymeric chemicals (for the category formulae,  $R - (OCH_2CH_2)_n - OR'$ , n is unspecified) that might contain a glycol ether component (n = 1, 2, or 3). EPA has tried to make this list as complete as possible; however, not all oligomeric or polymeric chemicals that might contain glycol ethers will necessarily appear on this list. If a facility manufactures, processes, or otherwise uses an oligomeric or polymeric chemical that contains components, in greater than threshold quantities, that meet the definition

of the EPCRA section 313 'certain glycol ethers' category, they must report the glycol ether component(s) that meet the category definition.

### **Ethylene Glycol Ethers versus Propylene Glycol Ethers**

The members of this category are glycol ethers derived from ethylene glycol, diethylene glycol, and triethylene glycol. This category does not contain glycol ethers based on propylene glycol, dipropylene glycol, or tripropylene glycol.

### **Individually Listed Glycol Ethers**

There are two chemicals, 2-methoxyethanol (CAS number 109-86-4) and 2-ethoxyethanol (CAS number 110-80-5) that are on the individual chemical list and CAS number lists (40 CFR 37265(a) and (b)). Threshold determinations should be made for each of these chemicals individually and separately from the glycol ethers category.

### ***De Minimis* Concentrations**

The glycol ethers category is subject to the one percent *de minimis* concentration. Thus, mixtures that contain members of this category at concentrations in excess of one percent should be factored into threshold and release determinations.

**Appendix G**

**EMISSION FACTORS FOR ANHYDROUS AMMONIA FROM THE  
LEATHER INDUSTRIES OF AMERICA**

The use of actual measurements is encouraged; however, the Leather Industries of America (LIA) have provided some typical emission factors and related data that may be reasonable in lieu of monitoring data. This information is presented in the table below.

**G-1**

**Emission Factors and Typical Throughput Parameters for Anhydrous Ammonia Estimates**

<b>Parameter</b>	<b>Factor</b>
Total anhydrous ammonia generated during hide deamination	0.15% of the weight of brine-cured, pre-fleshed hide
	0.10% of the weight of fresh (“green”) hide
Fugitive air release of anhydrous ammonia from hide deamination	3% of total ammonia generated and purchased
Ammonia discharged to water	90% of total ammonia generated and purchased (of which 10% is reportable as anhydrous ammonia)
Ammonia remaining in commercial product (as the ammonium ion, not reportable)	7% of total ammonia generated and purchased

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The pages listed in bold text in the index correspond to the primary uses or definitions of the associated term. Additionally, this index includes a list of primary purposes for examples and common errors that are presented throughout the document.

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