#### TWO-PHASE PHOTOBIOLOGICAL ALGAL H<sub>2</sub>-PRODUCTION SYSTEM

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#### Abstract

Continuous production of large volumes of  $H_2$  by algal cells has been achieved by depleting the cells of sulfur (Melis et al., 2000). The operation of this novel algal  $H_2$ -production system occurs in the light with acetate-supplemented medium. Investigations are under way to simplify the system and determine the metabolic pathways involved in the process.

Current year results include the observation that: (i) depletion of nutrients other than sulfur will also inactivate  $O_2$  evolution but at slower rates, (ii) light and acetate are required for rapid inactivation of  $O_2$  evolution, suggesting an energy-dependent process, and (iii) inhibitors of Photosystem II also inhibit H<sub>2</sub> evolution, indicating that residual water-oxidation activity is an important source of reductant for H<sub>2</sub> production.

An automated photobioreactor experimental system was also developed at NREL. We report the design being used and the parameters found important to monitor sulfur deprivation. The automated system will be utilized in the future to measure the effect of a variety of parameters on the  $H_2$ -evolution activity of sulfur-depleted cells.

#### Introduction

Microbial  $H_2$  photoevolution is catalyzed either by nitrogenases or hydrogenases, enzymes that can only function under anaerobic conditions due to their extreme sensitivity to  $O_2$ . Since  $O_2$  is a by-product of photosynthesis, nitrogenase-containing organisms have developed the following spatial and temporal strategies to protect the enzyme from inactivation by  $O_2$ : (a) heterocyst-containing cyanobacteria physically separate  $O_2$  evolution from nitrogenase activity by segregating oxygenic photosynthetic activity in vegetative cells and nitrogenase activity in heterocystis with reduced  $O_2$ -permeability (Fay, 1992) and (b) non-heterocystous cyanobacteria separate  $O_2$ -evolution from nitrogenase activity by performing these functions during, respectively, light and dark periods (Bergman et al., 1997).

Similar strategies cannot be found in phototrophic hydrogenase-containing organisms in nature. In order to sustain  $H_2$  production by green algae in the light, researchers have used a variety of methods to keep the cultures free of  $O_2$ . These include addition of  $O_2$ scavengers such as chromous chloride (Healey, 1970) or dithionite (Randt and Senger, 1985), or purging the cultures with inert gases such as nitrogen (Gfeller and Gibbs, 1984) or helium (Greenbaum et al., 1999). Benemann (1996) has advocated the use of the principle of temporal separation of H<sub>2</sub> and O<sub>2</sub> evolution in green algae ("indirect biophotolysis"), triggered by an unspecified reversible inactivation of Photosystem II (PSII) O<sub>2</sub>-evolution activity. In his proposed model, H<sub>2</sub> photoevolution in the absence of PSII would require the break-down of starch to provide reductants for Photosystem I (PSI) through the chlororespiratory pathway. These reductants would then be used by ferredoxin to reduce protons to  $H_2$ , in a reaction catalyzed by the hydrogenase enzyme. Based on his proposal, we identifed and used sulfur depletion to reversibly inactivate PSII (Melis et al., 2000), achieving apparent temporal separation of O<sub>2</sub> and H<sub>2</sub> evolution in the green alga Chlamydomonas reinhardtii. In the absence of sulfur and in the presence of acetate, the cells shut off most (but not all) of their O<sub>2</sub>-evolving activity, respire all measurable remaining  $O_2$  in a closed environment, and induce the expression of the hydrogenase enzyme. At this point, they can evolve  $H_2$  in the light for up to 4 days. Subsequently, if sulfur is added back to the cultures, they will recover PSII activity and return to a normal growth mode. Cycles of  $O_2$  and  $H_2$  production can be repeated at least 3 times without significant loss of activity. Our results also showed that protein degradation, rather than starch breakdown, correlated with H<sub>2</sub> production by the algal cells. The rates of H<sub>2</sub> evolution by our system were much lower than its potential for electron transport (Melis et al., 2000), suggesting that the system is being limited by some factor other than enzyme activity. Possible limitations include: (a) the rate of substrate degradation, (b) redox control of the rate of electron transport by reducing conditions, (c) limited supply of electrons from residual water oxidation, and (d) competition between the hydrogenase and other physiological pathways.

Currently, both light and acetate are present during the H<sub>2</sub>-production phase. However, the development of a commercial system for algal H<sub>2</sub> production using sulfur-depleted cells will require the elimination of superfluous nutrients and/or procedures to bring down the cost. Light is required for H<sub>2</sub> evolution. Acetate, however, is not consumed during the time in which H<sub>2</sub> is actually produced, and is thus not necessary for that step (Melis et al., 2000). In the current report, we have examined the requirement for light and acetate during the O<sub>2</sub>-inactivation phase, before H<sub>2</sub> evolution activity (using DCMU) on the H<sub>2</sub>-evolution activity of the sulfur-depleted cells. Finally, we present initial results on a new automated photobioreactor system that can continuously monitor 5 key parameters in algal cultures during sulfur-depletion treatment.

### **Materials and Methods**

## **Cell Growth and Sulfur Depletion**

Wild-type *C. reinhardtii* C137<sup>+</sup> cells were initially grown photoheterotrophically in Trisacetate-phosphate (TAP) medium, pH 7, and bubbled with 3% CO<sub>2</sub> in air at about 25° C. The photobioreactors consisted of flat bottles with stirring capability and placed under continuous cool-white fluorescent illumination at about 100-200  $\mu$ E•m<sup>-2</sup>•s<sup>-1</sup>. The cultures were grown to late log phase, harvested by centrifugation, washed three times in TAP minus sulfur medium, and resuspended in 1.2 l of the same medium to a concentration of 11-18 µg Chl /ml. The sulfur-depleted cell suspension was placed back in the light for up to 150 h.

### **Oxygen and Hydrogen Evolution Measurements**

Oxygen- and hydrogen-evolution activities of the cultures were measured as previously described (Melis et al., 2000) with two different Clark-type electrodes, each poised for the optimal measurement of each gas.

### **Gas Collection Measurements**

The reactor bottles were fitted with a #25 Ace thread and with smaller side-ports for liquid sampling. A threaded glass stopper with capillaries for gas sampling was fitted with a Viton O-ring and used to seal the reactor. Threaded side-arm and gas sampling ports were sealed with rubber-laminated Teflon septa. Figure 1 shows how teflon tubing (HPLC, Aminco), attached to one of the gas ports, was used to conduct gas evolved by the algae in the culture bottles to an upside-down graduated cylinder filled with  $H_2O$ . The gas collection tubing was detached from the culture bottle during liquid and gas sampling to avoid disturbance of gas volume readings in the graduated cylinder.

### Results

# Effect of depletion of nutrients other than sulfur on inactivation of $O_2$ evolution

It is known that, besides sulfur, depletion of nutrients such as phosphorus (Wykoff et al., 1998) or nitrogen (Kumazawa and Mitsui, 1981) from the medium also inactivates the photosynthetic  $O_2$ -evolution activity of algae. However, inactivation by phosphorus depletion is slower than that by sulfur (Wykoff et al., 1998). We investigated the effect of depleting algal cultures of nutrients other than sulfur on the rate of inactivation of PSII activity. Figure 2 (left side) shows the half-life for inactivation of  $O_2$  evolution from algal



# Figure 1. Photobioreactor for algal H<sub>2</sub>-production and system for gas collection

cultures resuspended in media depleted of different nutrients. It is clear that PSII can be inactivated by removing either S, Fe, or Mn from the medium. However, the inactivation is 3 to 4 times faster when sulfur is removed. Figure 2 also shows the effect of combining sulfur depletion with Fe or Mn depletion. The combined depletions do not result in faster inactivation of  $O_2$  evolution, and gave similar rates of subsequent  $H_2$  production (not shown).

### Effect of light and acetate on inactivation of O<sub>2</sub> evolution

We also investigated the need for both acetate and light during inactivation of  $O_2$  evolution. The right side of Figure 2 shows that both acetate and light accelerate the inactivation of PSII. In the presence of acetate but in the dark (D+A), the cultures are inactivated two times slower than in the light; in the absence of acetate but in the light (L-A), inactivation takes four times as long. These results support the notion of an energy-dependent protease contributing to the inactivation of  $O_2$  evolution. Energy-dependent proteases are involved in specific inactivation of various proteins in *E. coli* and other



## Figure 2. Effect of different nutrient and light parameters on the rate of inactivation of $O_2$ evolution.

organisms (Wilson et al., 2000; Wang et al., 1999; Hilliard et al., 1998; Laachouch et al., 1996).

### Source of reductant for H<sub>2</sub> evolution

Sulfur depletion for 24 h inhibits more than 90% of the O<sub>2</sub>-evolution activity in algal cells (Melis et al., 2000). Our previous observation that protein consumption alone could potentially provide the reductant needed for the amount of H<sub>2</sub> evolved by sulfur-depleted algal cells seemed to indicate that at least some of the reductant for H<sub>2</sub> evolution originated from protein degradation. However, given that the subsequent rate of H<sub>2</sub> evolution is only 10% of the capacity of electron transport chain (based on the O<sub>2</sub> evolution rate measured at the beginning of the sulfur-deprivation experiment), it is also possible that the electrons used to reduce protons to H<sub>2</sub> come from residual PSII activity. In order to test this idea, we added DCMU, a specific inhibitor of PSII activity, to the algal cultures following onset of H<sub>2</sub> evolution. This treatment resulted in inhibition of H<sub>2</sub> production by the algal hydrogenase originated from residual water-oxidation activity.

## Automated Photobioreactor Experimental System

We have developed an automated system that allows us to continuously monitor a series of physical and electrochemical parameters in our sulfur-depleted, algal  $H_2$ -production system, as shown in Figure 3. Preliminary experiments showed that pH, eH (redox

potential),  $pO_2$  (dissolved  $O_2$ ), temperature, and quantity of evolved gas are important parameters to record.



## Figure 3. Schematic of an automated system to monitor algal H<sub>2</sub>-production

In order to test this design, photobioreactors with additional ports for each of the sensors were fabricated. Each photobioreactor has 3 ports for the sensors, one for culture sampling or chemical injection, and one for gas outlet, as shown in Figure 4.

The method adopted for gas-to-liquid volume conversion is described in Figure 5. It consists of an intermediate bottle (bottle 1 in Fig. 3) full of water that collects the gas evolved by the photobioreactor cultures. The collected gas displaces the liquid in the intermediate bottle, which in turn is syphoned to a second bottle (bottle 2 in Fig. 3). Bottle 2 is located on an electronic balance. The changes in the weight of bottle 2 are a measure of the rate of gas evolution by the algal cultures. We found that the inside diameter of the connecting tubes had to be no less than 0.5 mm, otherwise the time response of the system was excessively high. This type of gas-to-liquid conversion and measurement system is temperature and pressure sensitive but under properly controlled conditions gives an error of less than 3%.

Finally, an integrated microprocessor system that simultaneously monitors four separate algal culture vessels was assembled, according to the schematic shown in Figure 6.

Preliminary experiments have been conducted in order to confirm consistent changes in all the monitored parameters in all four vessels (not shown).



Figure 4. Design of a photobioreactor for the automated algal  $H_2$ -production system.



Figure 5. Vessel for gas-to-liquid conversion.



Figure 6. Schematic of an integrated system for simultaneously monitoring five parameters from four photobioreactors in an algal  $H_2$ -production system.

#### **Summary and Conclusions**

Since algal H<sub>2</sub> photoproduction from water was discovered almost 60 years ago (Gaffron and Rubin, 1942), H<sub>2</sub> could be collected only if O<sub>2</sub>, produced by photosynthesis, was removed continuously by chemical or mechanical means. Earlier this year, we described a physiological method (sulfur depletion) to reversibly inactivate O<sub>2</sub>-evolution activity in an algal culture; this inactivation leads to the induction of hydrogenase activity and subsequent production of large quantities of H<sub>2</sub> for 3-4 days (Melis et al., 2000). We now report that, while depletion of a number of other nutrients besides sulfur can also inactivate O2-evolution, their rate of inactivation is slower (Fig. 2). We also provide evidence that this inactivation must involve energy-dependent proteases (Fig. 2), since it occurs much more slowly in the dark and/or in the absence of acetate. Finally, we present preliminary evidence suggesting that residual water-oxidation activity may be the source of most of the reductant for the H<sub>2</sub> evolution process (Fig. 3). Clearly, much more detailed work will be required to clarify the metabolic pathways involved in the process, and the inter-relationship between H<sub>2</sub> evolution and protein degradation.

Even without a clear knowledge of all the pathways involved in the transport of reductant to hydrogenase in sulfur-depleted cells, it is evident that  $H_2$  production can be sustained for up to 4 days (Melis et al., 2000). Moreover, the system can be recycled back and forth between photosynthetic growth and  $H_2$  production (data not shown). This algal  $H_2$ production system does not seem to be a pure "indirect biophotolysis" system, as proposed by Benemann (1996). Indeed, it is similar in concept to systems in which photosynthetically-produced  $O_2$  is removed by addition of  $O_2$  scavengers or by purging with neutral gases. The significant difference is that sulfur-depleted cells operate with only 10% of their normal oxygenic PSII activity, while previous systems (listed in Table I), presumably operate with 100% functional PSII. Nevertheless, this does not seem to make a large difference in terms of the actual rates of steady-state  $H_2$  production, as shown in Table I. Indeed, sulfur-depleted cells produce  $H_2$  at rates comparable to systems in which 100% of the PSII are operational, with the additional advantage that  $O_2$ is removed by physiological means, not by the introduction of extraneous chemical reductants or inert gases.

Up until now, only nitrogenase-based systems were capable of sustained  $H_2$  photoproduction without expensive O<sub>2</sub>-removal systems (Benemann, 1996). Indeed, cyanobacteria, in the absence of fixed nitrogen sources can produce  $H_2$  for months, particularly if their uptake hydrogenase activity is concomitantly inactivated (Markov et al., 1996). Table II shows a comparison of the rates of  $H_2$  production by a variety of nitrogenase-containing cyanobacteria, and our sulfur-depleted algal system. Given the different pigment composition of cyanobacteria and green algae, we show the data on a per mg dry weight basis. The sulfur-depleted algal cells evolve  $H_2$  at rates higher than most optimized cyanobacterial systems. This is not surprising, given that nitrogenases are known to be sluggish enzymes that, besides reductants, require ATP.

## Table I. Comparison of Rates and Volumes of H<sub>2</sub> Collected from Different Algal Systems and Methods Used to Sustain High H<sub>2</sub> Evolution Rates

Organism	Initial rate ( $\mu$ moles H <sub>2</sub> • mg Chl <sup>-1</sup> • h <sup>-1</sup> )	Steady-state rate ( $\mu$ moles H <sub>2</sub> • mg Chl <sup>-1</sup> • h <sup>-1</sup> )	Total volume H <sub>2</sub> collected	Culture volume and Chl content
Chlamydomonas moewsuii (Healey, 1970)	-	5 [chromous chloride]*	-	-
<i>Chlamydomonas</i> <i>reinhardtii</i> F60 (Gfeller & Gibbs, 1984)	-	5.7 $[N_2 purging]$	0.25 ml in 3.5 h	0.3 mg Chl in 3 ml
Scenedesmus obliquus (Randt & Senger, 1985)	54	13 [Na dithionite]	-	-
Chlamydomonas reinhardtii (Greenbaum, personal communication)	50	≅ 10 [He purging]		
<i>Chlamydomonas</i> <i>reinhardtii</i> (this work)	90	$\cong 10$ [Sulfur depletion]	325 ml in 95 h	18 mg Chl in 1 liter

\* Values in brackets indicate the means by which O<sub>2</sub> was removed from the cultures.

## Table II. Comparison of the Rates of H<sub>2</sub> Evolution among Different Nitrogenase-Based Systems and our Sulfur-Depleted Green Algal System.

Organism	Rate of H <sub>2</sub> evolution (ml•mg dry weight <sup>-1</sup> •d <sup>-1</sup> )	Reference	
Anabaena cylindrica	0.09-0.03	Miyamoto et al., 1979	
Oscillatoria sp. Miami BG7	0.14	Kumazawa and Mitsui, 1981	
Anabaena variabilis (no uptake hydrogenase)	0.22	Markov et al., 1996	
Sulfur-depleted Chlamydomonas reinhardtii	0.34	This work	

In conclusion, we have shown that green algae can produce significant amounts of bulk  $H_2$  gas at rates comparable to other oxygenic photosynthetic organisms, when their  $O_2$ evolving capability is reduced by physiological means. Hydrogen production depends on the depletion of sulfur from the medium, is reversible, and results in the generation of pure  $H_2$  (co-evolved  $CO_2$  stays in solution). We are currently investigating in more detail the metabolic pathways involved in the evolution of  $H_2$  under sulfur-depletion conditions.

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