Utilization of Interfacial Engineering to Improve Food Emulsion Properties







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Traditional (Single-Step) Method of Emulsion Formation



Homogenization

Limitations of Single-Step Method

- Some emulsifiers are efficient at forming small droplets during homogenization, but not at providing good stability.
- Some emulsifiers provide good stability, but are inefficient at forming small droplets during homogenization.
- Existing emulsifiers provide limited scope for developing encapsulation & delivery systems.

Hypothesis: Can we develop a new emulsifier strategy that combines the fast adsorption of small molecules and the good stability of large polymers to create emulsions with improved performance and functionality?



Multi-Step Emulsion Formation Primary

Secondary

Separate Oil and Water Phases

-Potential (mV)



Control of Interfacial Characteristics

- Control of Interfacial Properties
 - Charge Sign and Density
 - Thickness
 - Packing & Permeability
 - Rheology
 - Responsiveness

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• Control of Emulsion Properties – Stability, Rheology, Release



Potential Advantages of Multilayer Emulsions

- Improvement in Stability of Emulsions to Environmental Stresses
 - pH, Ionic Strength
 - Heating, Chilling, Freezing
 - Dehydration



- Encapsulation and Delivery of Functional Components
 - Protection of Labile Ingredients
 - Controlled or Triggered Release



Potential Disadvantages of Multilayer Emulsions

• More Difficult to Prepare – Prone to flocculation



More Expensive to Prepare

 Additional ingredients required
 Additional processing required



Problem: Research is needed to establish optimum preparation conditions (ingredients and processes) for preparation of multilayer emulsions with specific performance and functionality.

Overall Project Objectives Optimization & Application of Multilayer Technology



Understanding Formation of Multilayer Emulsions: Theoretical Model



*C*_{SAT} = Characteristic Saturation Concentration
The minimum amount of polymer required to completely cover all of the droplet surfaces.



$C_{ADS} = Characteristic Adsorption Concentration$

• The minimum polymer concentration required to ensure adsorption occurs faster than collisions.

C_{DEP} = Characteristic Depletion Concentration

• The polymer concentration where depletion flocculation is first observed.





Theoretical Stability Map Effect of Droplet & Polymer Concentration



For stable system: C_{Ads} < C < C_{Dep}

Multilayer Emulsion Formation: Bridging & Depletion (pH 3.5)

Pectin Concentration



Multilayer Emulsion Formation: Establishment of "Creaming Stability" Map

| | | Ű | | | |] | рH | 3.5 | | | | | | | рH | [7. 0 |
|-----|---------------------------|------|---------------------------------|---|---|---|----|-----|-------------------------------|------|---------------------------------|---|---|---|----|---------------|
| | pH < pI | | Droplet Concentration (wt %) | | | | | | pH > pI | | Droplet Concentration (wt %) | | | | | |
| | | | 0.5 | 1 | 3 | 5 | 8 | 10 | | | 0.5 | 1 | 3 | 5 | 8 | 10 |
| Bri | Pectin (wt %) dging | 0 | S | S | S | S | S | S | Pectin (wt %) Depletion | 0 | S | S | S | S | S | S |
| | | 0.01 | S | С | C | С | S | S | | 0.01 | S | S | S | S | S | S |
| | | 0.02 | S | S | C | С | S | S | | 0.02 | S | S | S | S | S | S |
| | | 0.04 | S | S | C | С | С | S | | 0.04 | S | S | S | S | S | S |
| | | 0.06 | S | S | S | С | С | С | | 0.06 | S | S | S | S | S | S |
| | | 0.08 | S | S | S | С | С | С | | 0.08 | S | S | S | S | S | S |
| | | 0.10 | S | S | S | С | С | С | | 0.10 | С | С | С | С | С | С |
| | | 0.30 | S | S | S | S | S | S | | 0.30 | С | C | С | С | С | С |
| | | | | | | | | | | | | | | | | |

S

N

S

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0.50

0.50

Understanding Formation of Multilayer Emulsions: Experiments



Measure:

• Adsorption (ζ-potential); Microstructure (Microscopy); Stability (Laser Diffraction, Creaming); Rheology (Shear rheometry)

Multilayer Emulsion Formation: Establishment of Adsorption Range



ζ-Potential Measurements Show Where Adsorption Occurs

Interfacial Characteristics: Determination of Interfacial Properties



ζ-Potential Dynamic Light Scattering

Interfacial Properties:

- Polymer Concentration
- Thickness
- Packing
- Charge



Protein-Pectin Multilayers: $\Gamma_{Sat} = 1.6 \text{ mg m}^{-2}$; $\delta = 45 \text{ nm}$; $\phi = 2\%$

Application of Technology: Improving Emulsion Stability

Motivation

• Multilayer technology could be used to improve the stability of many food emulsions to environmental stresses

Environmental Stresses

- pH, Ionic Strength
- Thermal processing, Chilling, Freezing
- Dehydration



Multilayer Emulsion Properties: Extension of pH Range



3.0 3.5 4.0 4.5 5.0 5.5 6.0 6.5 7.0

3.0 3.5 4.0 4.5 5.0 5.5 6.0 6.5 7.0

Multilayer Emulsion Properties: Extension of NaCl Stability Range







Origin of Stability:

- Reduction in VDW
- Increase in ES
- Increase in Steric





Multilayer Emulsion Properties: Improvement of Thermal Stability





1° <mark>β-Lg</mark> 2° Pectin

pH 4

Multilayer Emulsion Properties: Improvement of Freeze-Thaw Stability



1° β-Lg 2° Carrageenan 3 ° Gelatin

0% Sucrose

Multilayer Emulsion Properties: Improvement of Dehydration Stability

Primary





Secondary







Multilayer Emulsion Properties: Improvement of Oxidative Stability







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Current Status of Multilayer Emulsions

- Emulsions containing lipid droplets coated by nanolaminated layers can be prepared by a simple cost effective method using food ingredients
- These emulsions have improved stability to environmental stresses, such as heating, freezing, drying, pH extremes, and high mineral contents
- Future studies are needed to determine their suitability for use in real foods (encapsulation, controlled release, triggered release)



Future Work

- Investigate use of mixed biopolymers to form the outer interfacial layers (so can control charge and thickness).
- Investigate methods of cross-linking adsorbed layers so that they retain functionality over wider range of conditions.
- Investigate digestibility of multilayer emulsions.
- Investigate applications to real food systems, *e.g.*, beverage emulsions.



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