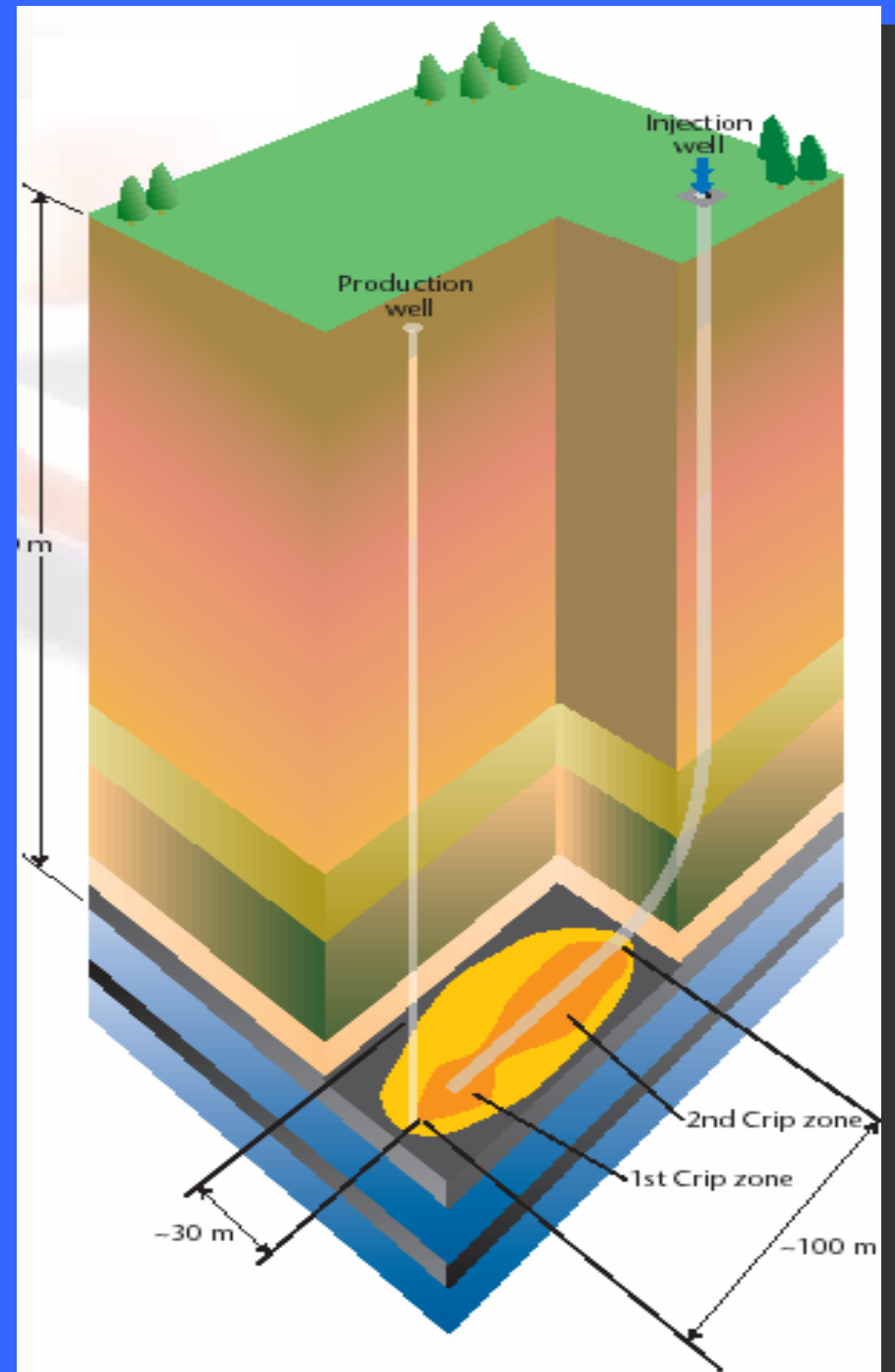


Underground Coal Gasification: Science and Technology Gaps

***Ravi Upadhye, Ph.D., PE
Elizabeth Burton, Ph.D.
Julio Friedmann, Ph.D.***



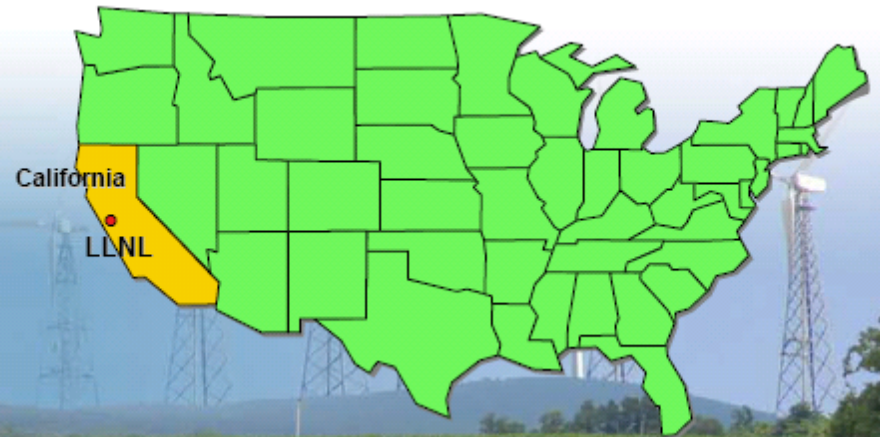
***Lawrence Livermore National
Laboratory***



Lawrence Livermore National Laboratory



*A national research laboratory
operated by the University of
California for the United States
Department of Energy*





DOE/LLNL Activity in UCG

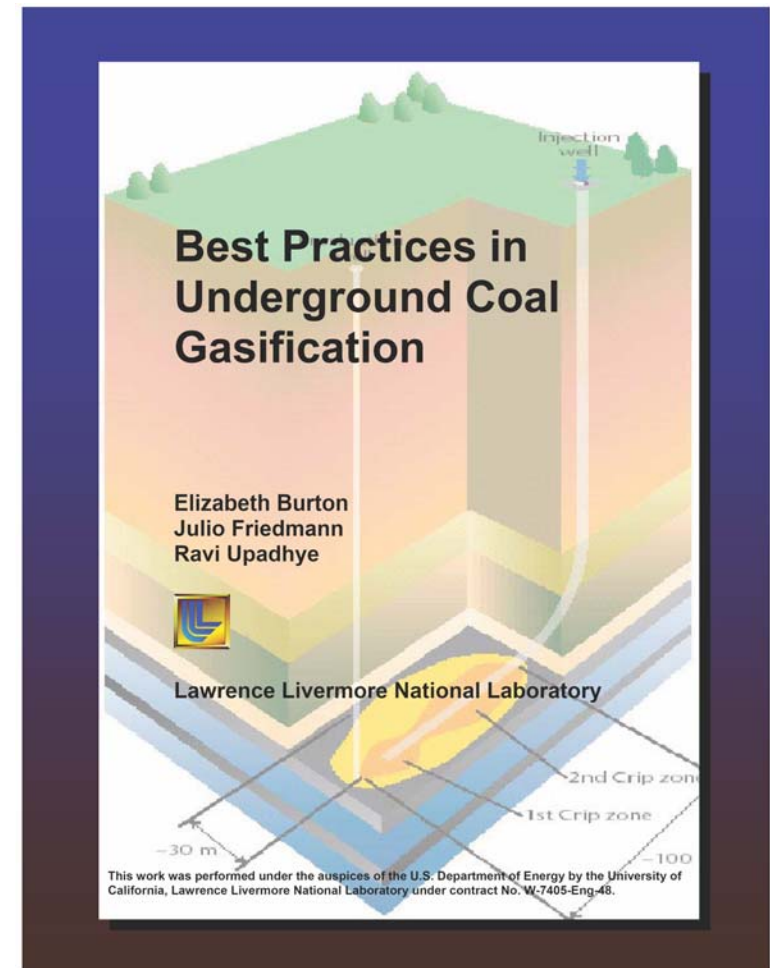
- Invented CRIP (controlled retractable injection point) process (1974-1985)
- Conducted numerous field tests (Hoe Creek, Hanna, Centralia)
- Developed and validated cavity growth models
- Developed methodologies for process control monitoring
- Developed a CFD-based gasification model of the UCG process and integrated it with Aspen Plus
- Developing tools for site assessment
- Developing models for environmental risk assessment
- Applying carbon management and CO₂ sequestration expertise to evaluate UCG-CCS
- Completed draft report on UCG best practices and lessons learned

New LLNL-DOE Draft Report



Contents:

- UCG history
- Ignition and gasification process
- Environmental management issues
- Carbon management
- Best practices and lessons-learned
- Technology and R&D gaps; recommendations for future R&D



Copies available soon.

Send requests to: burton14@llnl.gov

Why Consider Underground Coal Gasification?



- **Eliminates conventional coal mining, reducing operating costs, surface damage, eliminating mining safety problems**
- **Accesses otherwise un-mineable coals (deep or thin seams), increasing exploitable reserves**
- **Needs no surface gasification facilities, reducing capital costs**
- **Leaves gasification residuals underground**
- **Eliminates costs, facilities, and environmental issues associated with transport/storage of mined coal or coal gasification residuals (e.g., ash)**
- **Reduces overall greenhouse gas emissions and has advantages for geologic carbon storage**

Timeliness of Underground Coal Gasification R&D Investment



- **Economic pressure to find alternative fuel resources given high cost of oil and natural gas**
- **Increased security risk with continued dependence on imported fuel supplies**
- **Access presently un-mineable coal resources**
 - in U.S., UCG increases exploitable coal resources by ≥ 3 times
- **Potential to optimize UCG as a cost-effective clean-coal technology:**
 - Reduced greenhouse gas emissions per unit coal gasified
 - Potential for combined UCG-CCS
 - Reduced overall environmental impact

Increased recent interest in UCG technology globally creates opportunities to leverage R&D and share lessons-learned and best practices

Limitations and Concerns for UCG

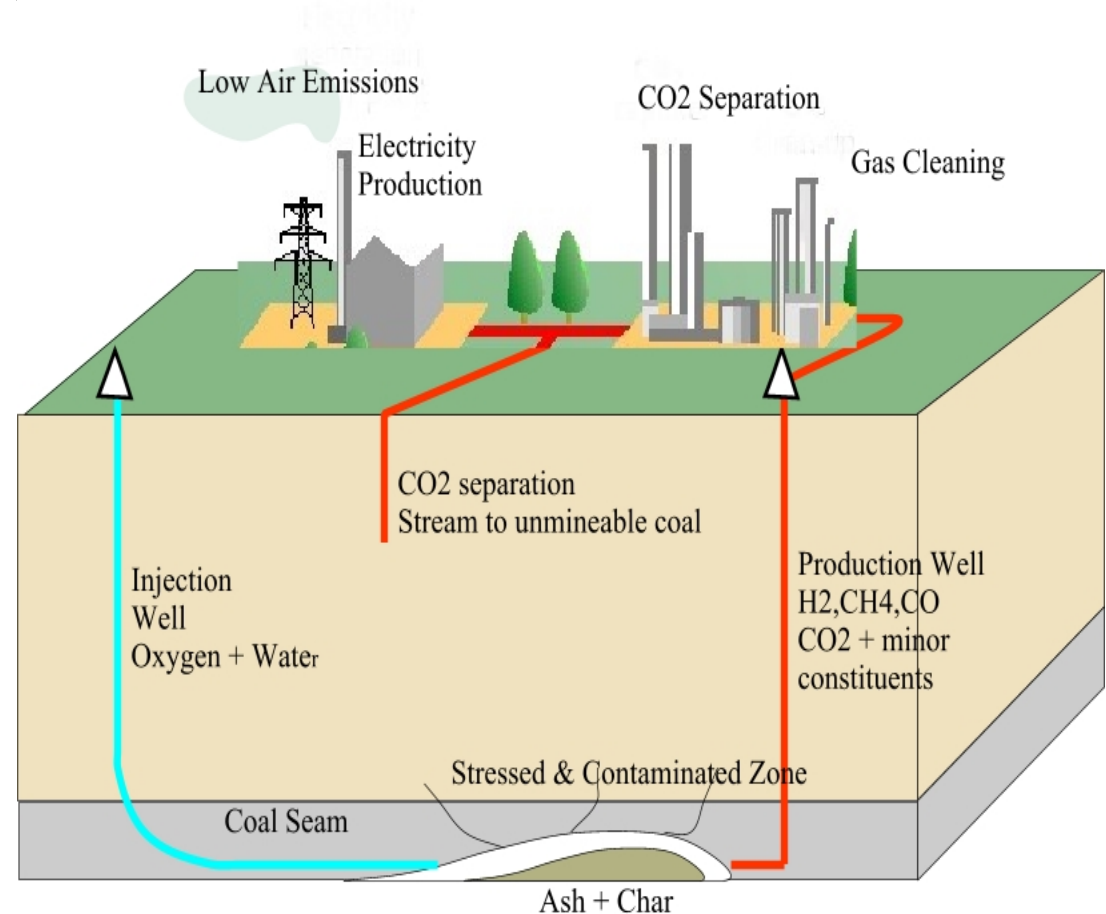


- **UCG has produced significant groundwater contamination and ground subsidence in some previous operations**
- **The increase in exploitable coal possible with UCG may be less when site selection is constrained by geologic and hydrologic criteria to protect the environment**
- **UCG is a non-steady state process-- operations cannot be controlled to the same extent as surface gasification**
 - **Process variables vary as burn progresses and can only be estimated**
 - **Flow rate and composition (heating value) of product gas will vary over time**
- **Business case for UCG will remain difficult until there are enough UCG facilities (power, syngas, chemical feedstocks) to provide economic data**

Science and Technology Gaps



- Environmental and site assessment
- Process and environmental monitoring
- (Computational Fluid Dynamics) CFD gasification model
- Issues with Carbon Capture and Storage (CCS) with UCG



Environmental and Site Assessment Objectives



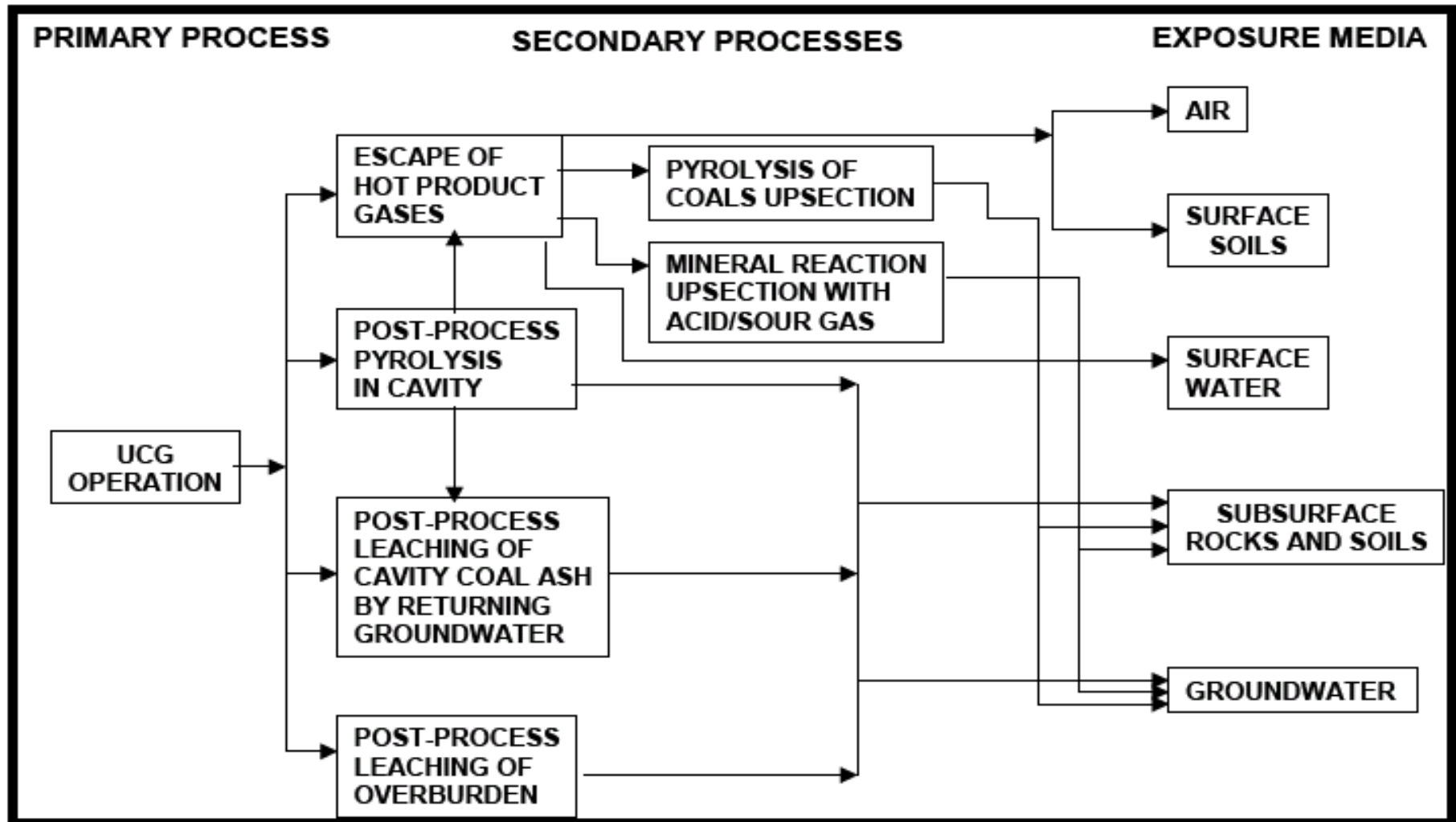
- **Reduce risk of UCG to acceptable levels through assessment tools for:**
 - Site selection and screening
 - Operations and facilities planning based on site-specific risk-indexing
 - Setting operations guidelines based on site parameters
- **Identify UCG operating ranges that limit production of contaminant compounds, and prevent contaminant migration out of the cavity during- and post-UCG**
- **Include evaluation of mitigation and remediation options and economics appropriate to UCG sites, including bioattenuation rates**
- **Include capability to assess combined UCG-CCS**

Environmental and Site Assessment Tools



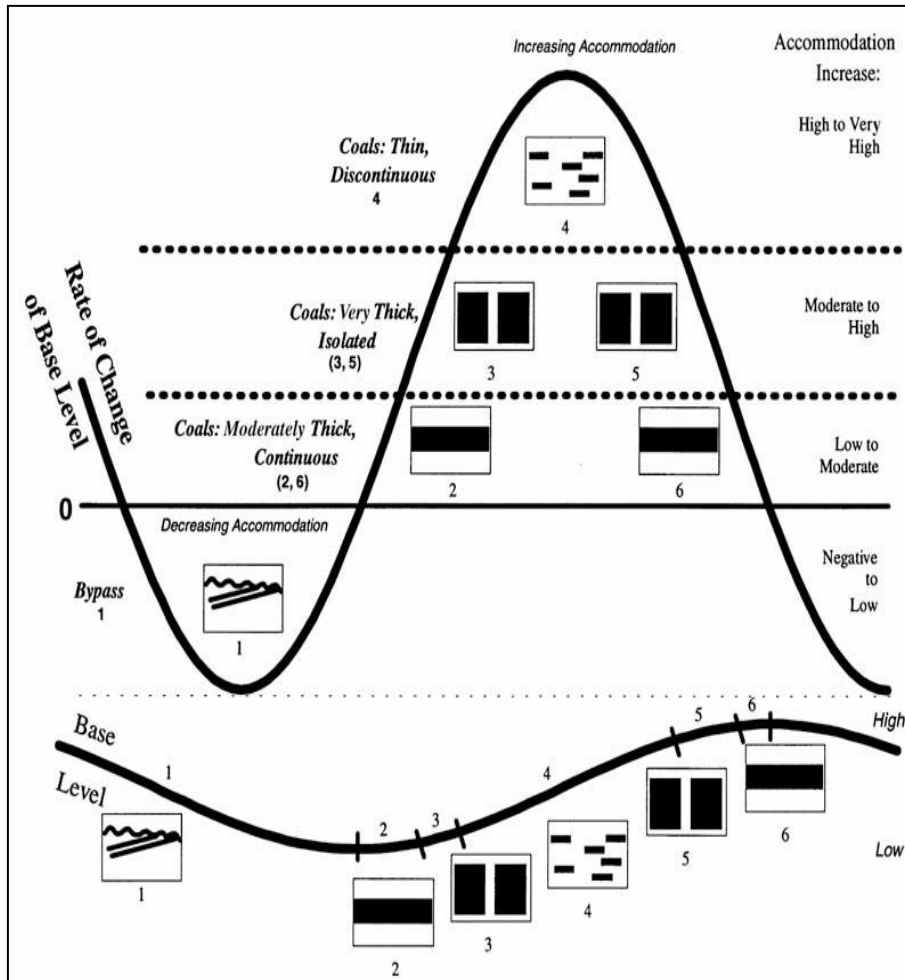
- **Risk-based Decision-making Framework for UCG and UCG-CCS**
- **Site characterizations incorporating geologic and hydrologic systems**
- **Accurate, integrated geomechanical-hydro-chemical simulations of UCG impacts that include:**
 - Effects of thermal changes on density and viscosity, hydrologic gradients, and on contaminant migration
 - Effects of linkage and burn processes, cavity growth, cavity collapse, fracture/fissure propagation, and coal swelling on hydraulic connectivity and gradients
 - Integration and inversion techniques, including stochastic inversion using Monte-Carlo Markov-chain approaches

Risk-based Decision-making for UCG

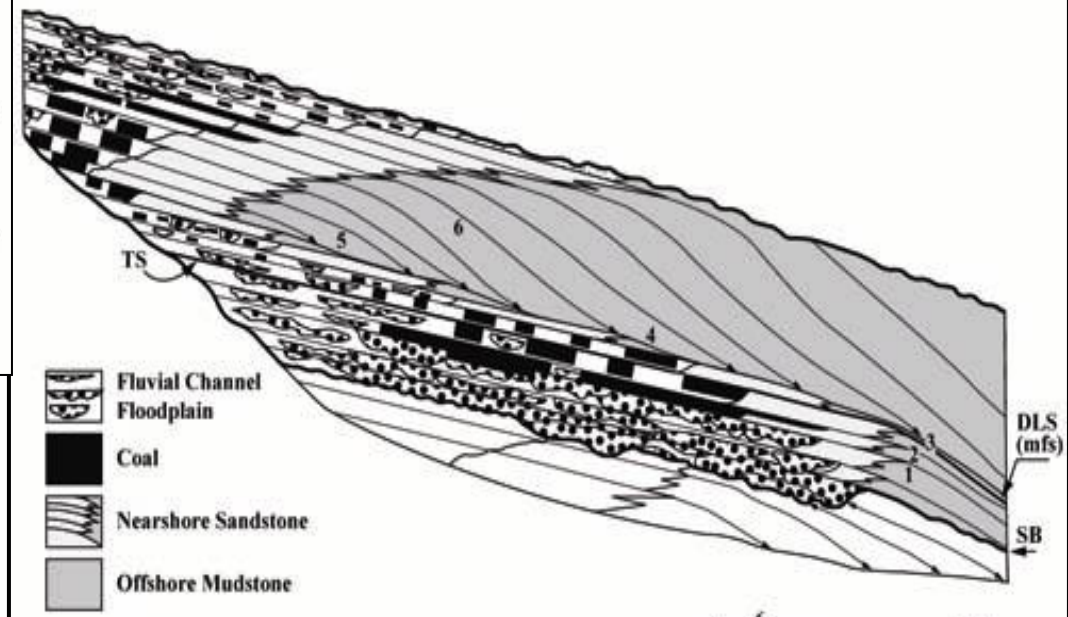


Additional secondary processes apply with CCS

Geologic Characterization for Site Assessment: Stratigraphic Screening



Stratigraphic category	Lateral Isolation	Overlying Unit Character	Relative Risk
1	Low	Sand-prone	High
2	Low	Shale-prone	Moderate
3	High	Shale-prone	Low
4	Moderate	Shale-prone	Moderate
5	Moderate	Sand-prone	High
6	Low	Sand-prone	High



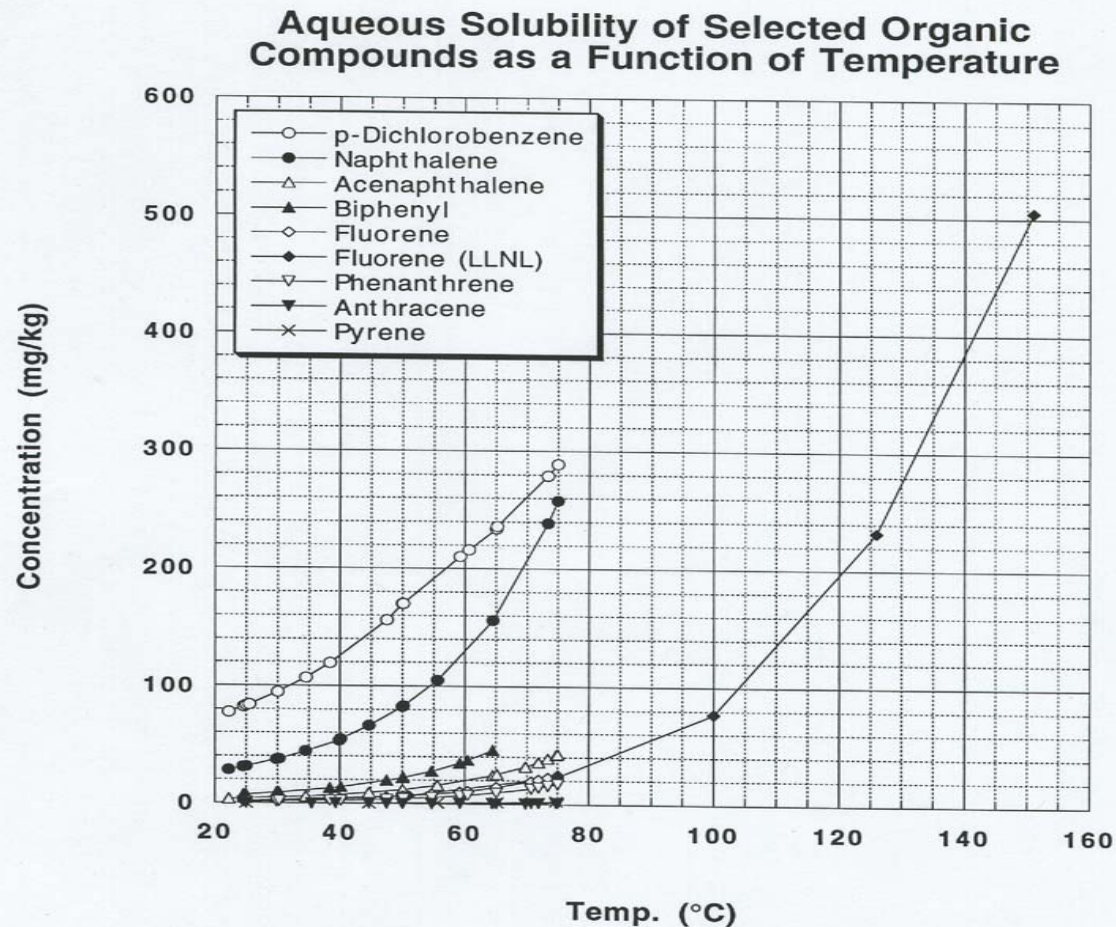
Stratigraphic sequences give first-cut screen for likelihood of good sealing caprock

Simulations of UCG Impacts on Geologic-Hydrologic Systems



- **UCG environmental assessment cannot be done with a standard toolbox of environmental tools**
- **UCG requires use of coupled process, hydrological, geochemical and geomechanical models to capture:**
 - Balancing gasifier operational pressure against hydrologic pressure and other gradients in the field to prevent outward contaminant migration
 - Impact of gasifier operating conditions on creation and behavior of contaminants within the burn chamber
 - Enhanced vertical hydraulic conductivity of the rock matrix above the burn chamber as a result of collapse and fracturing
 - Buoyancy-driven upward flow of groundwater in the vicinity of the burn chamber toward potable water resources at shallower depths

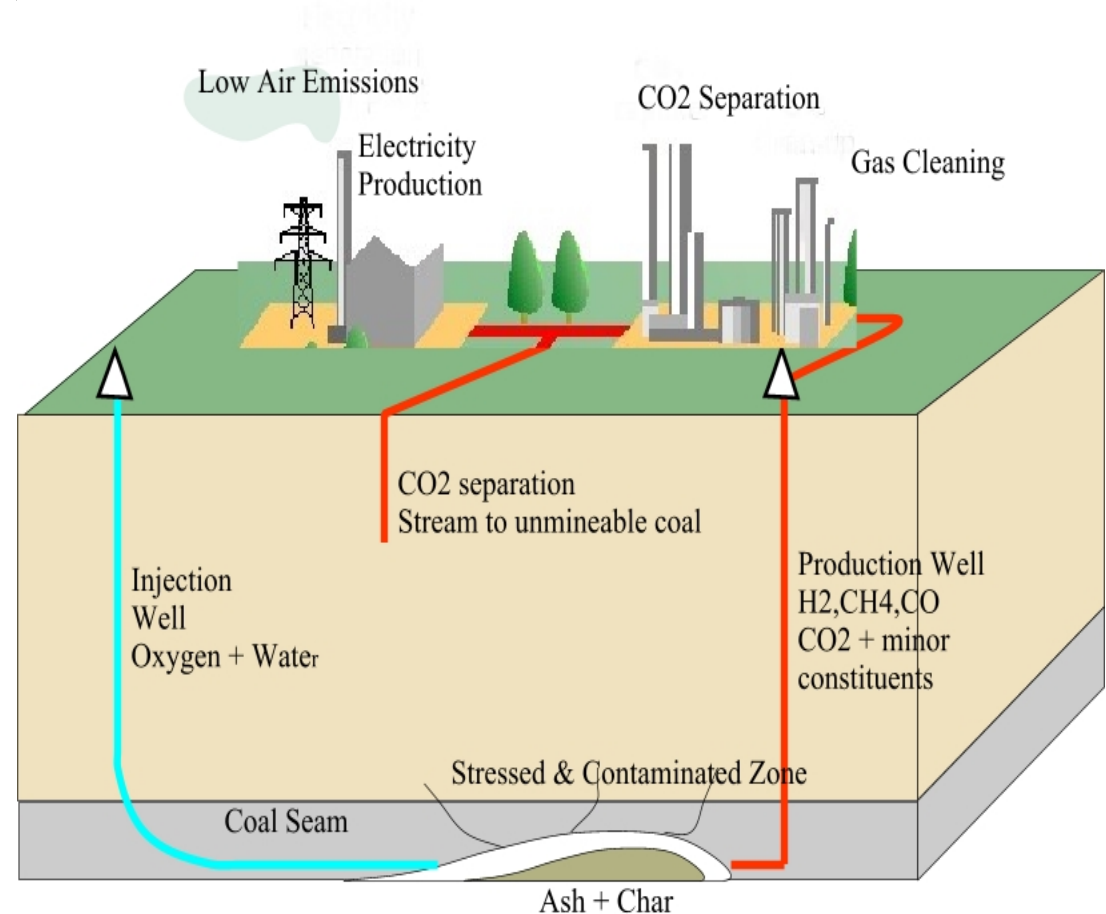
Contaminant Behavior: Increases in Solubility with Temperature



Science and Technology Gaps



- Environmental and site assessment
- Process and environmental monitoring
- (Computational Fluid Dynamics) CFD gasification model
- Issues with Carbon Capture and Storage (CCS) with UCG



Monitoring Objectives



- **Real-time insight into potential hazards (subsidence, groundwater contamination) as the UCG burn progresses through the coal seam**
- **Validate models and improve understanding of the UCG process**
- **Improve understanding of the geomechanical, geochemical, and hydrogeological changes induced by UCG**
- **Allow for the possibility of real-time process control of the UCG “reactor”**

Monitoring Tools

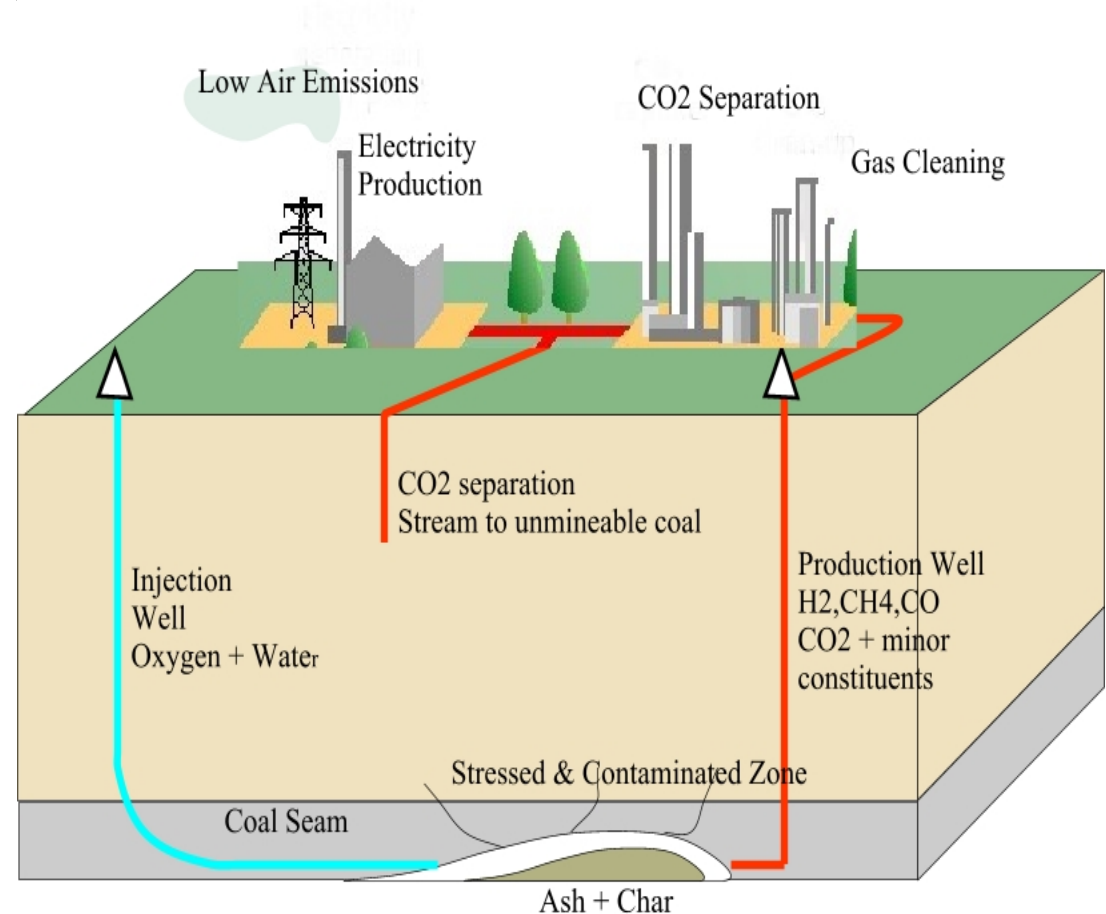


- **Geophysical tools to detect burn front progression, fracturing, cavity collapse, subsidence**
 - Electrical resistance tomography (ERT) and EM induction tomography (EMIT)
 - Seismic monitoring (e.g., 3-d, microseismic)
 - Tilt meters
- **Remote sensing tools to detect surface changes due to UCG**
 - hyperspectral imaging of plant and soil changes from gas leakage, or incipient subsidence
- **Down-hole tools (“smart” borehole casings) for real-time continuous monitoring of temperature, pressure and chemistry**

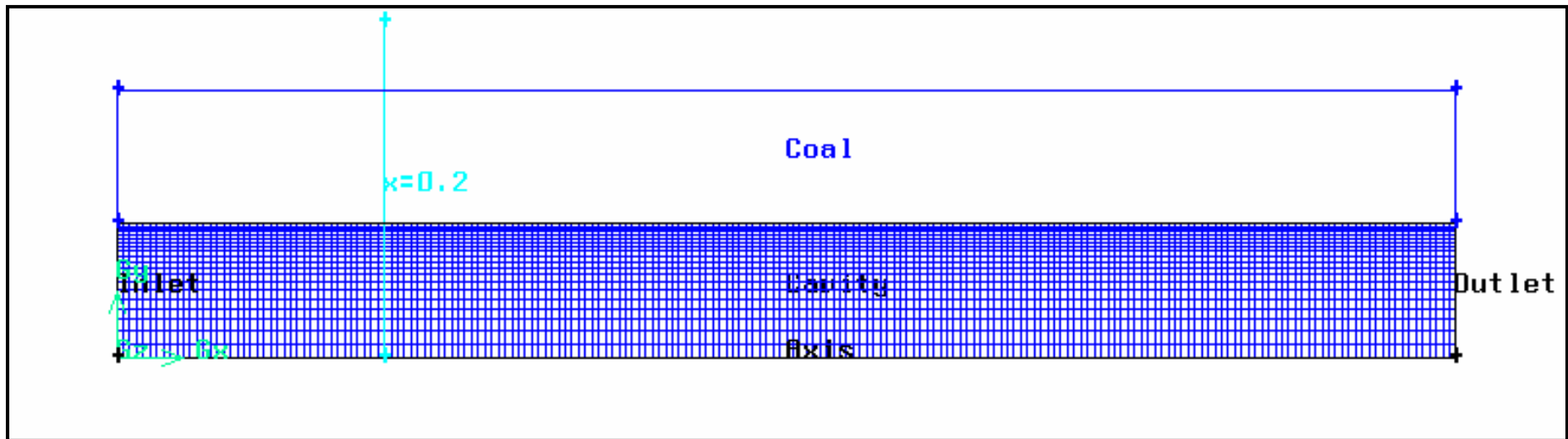
Science and Technology Gaps



- Environmental and site assessment
- Process and environmental monitoring
- (Computational Fluid Dynamics) CFD gasification model
- Issues with Carbon Capture and Storage (CCS) with UCG



LLNL's CFD Model: Approach



- Cylindrically symmetric cavity
- Considers influx of water and coal pyrolysis
- 1-cm thermal wave ahead of surface reactions
- Coal = $\text{CH}_{0.08}$
- WG shift reaction and coal gasification reactions are considered to be volumetric, but known kinetics are used
- Radiation effects are ignored

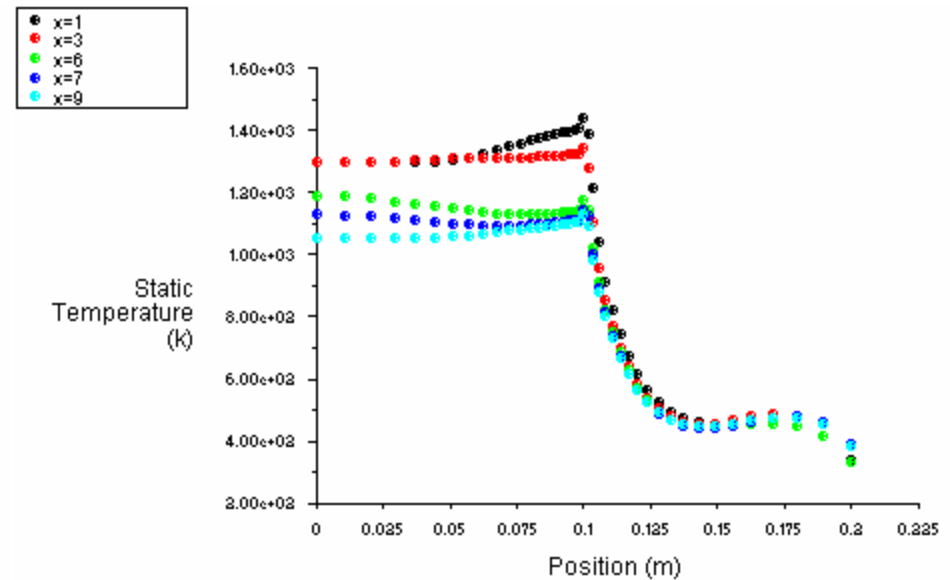
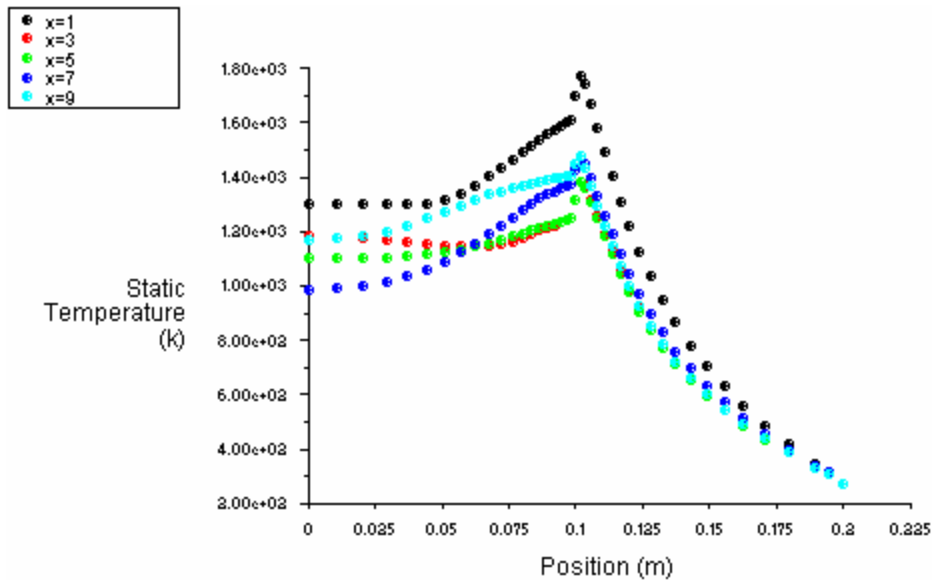
LLNL's CFD Model: Predicted gas Compositions



Typical UCG gas compositions adjusted to 33 mol% water content.

Component	UCG Model	Field Measurement ⁽¹⁾
H ₂	27.2	27.3
CO	13.0	6.4
CO ₂	19.4	27.2
CH ₄	7.4	6.4
H ₂ O	33.0	33.0

Temperature Profiles from the CFD Model



Static Temperature

Jul 21, 2004
FLUENT 6.1 (2d, segregated, spe6, ske)

Static Temperature

Jul 21, 2004
FLUENT 6.1 (2d, segregated, spe7, ske)

Temp vs radial position, without and with water influx (0.3 kg/kg coal)

Improvements Planned for CFD Model

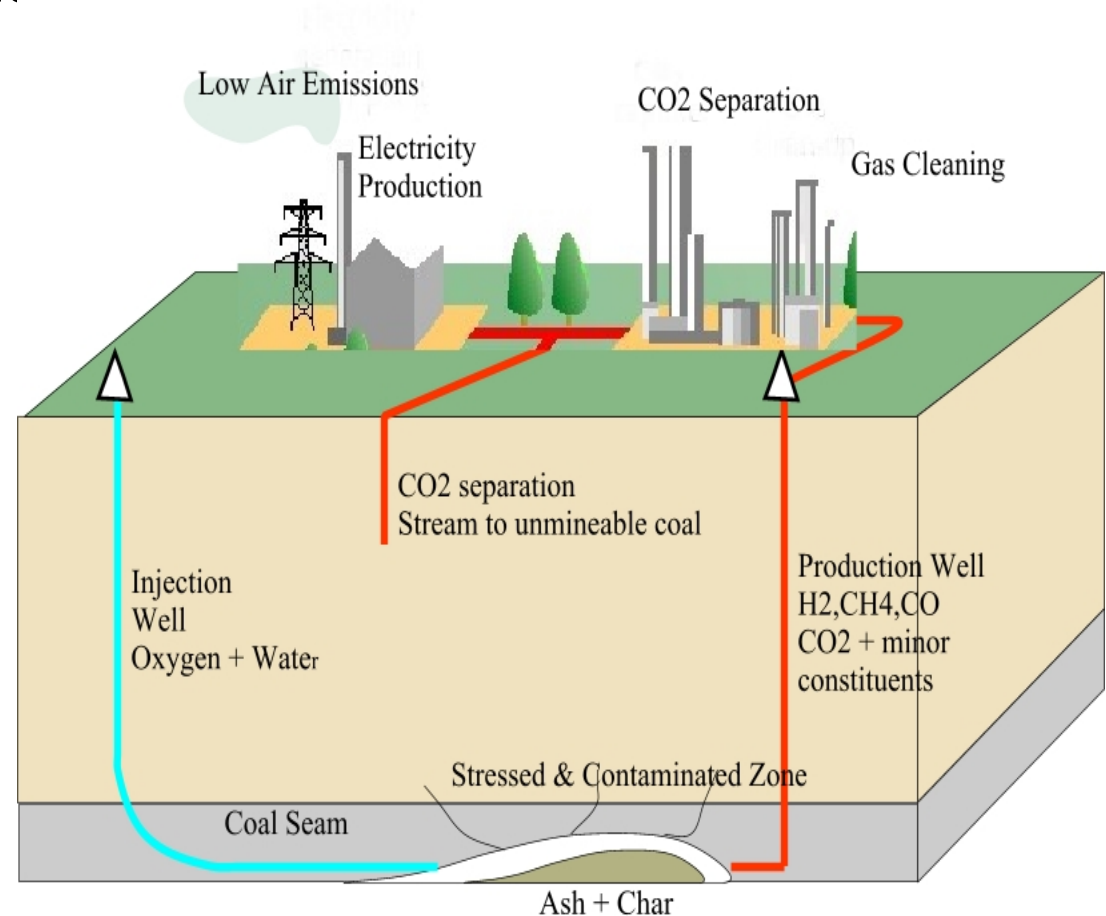


- **Steady-state to dynamic**
- **2D-3D model, no assumption of axial symmetry**
- **Include radiative heat transfer**
- **Include coal-methane kinetics**
- **Accommodate surface reactions**
- **Include separate kinetics for combustion and reforming**
- **Improvements in treatment of the porous zone**
- **Calculation of local recession rate for coal**
- **Integration with environmental/hydrogeologic models**

Science and Technology Gaps



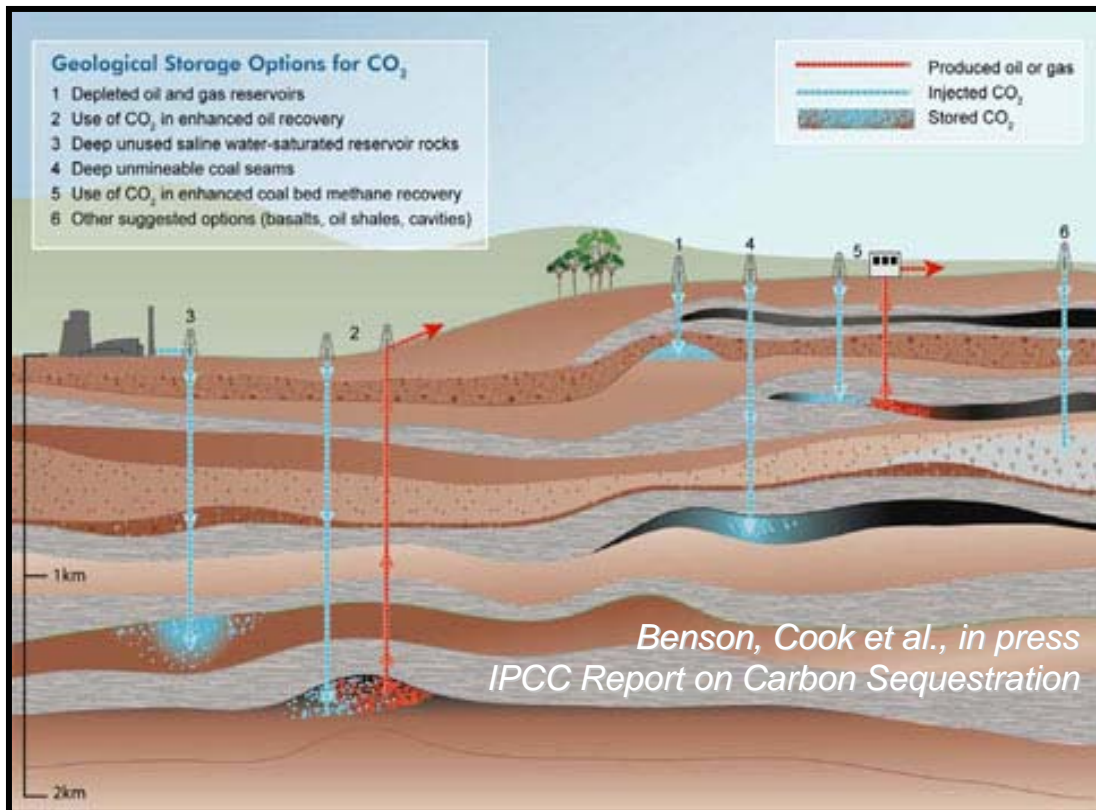
- Environmental and site assessment
- Process and environmental monitoring
- (Computational Fluid Dynamics) CFD gasification model
- Issues with Carbon Capture and Storage (CCS) with UCG



UCG with Carbon Capture and Storage (UCG-CCS)



Carbon capture and storage (CCS) aims to reduce net greenhouse gas emissions, chiefly CO₂, through storage of gas underground .



Storage Targets

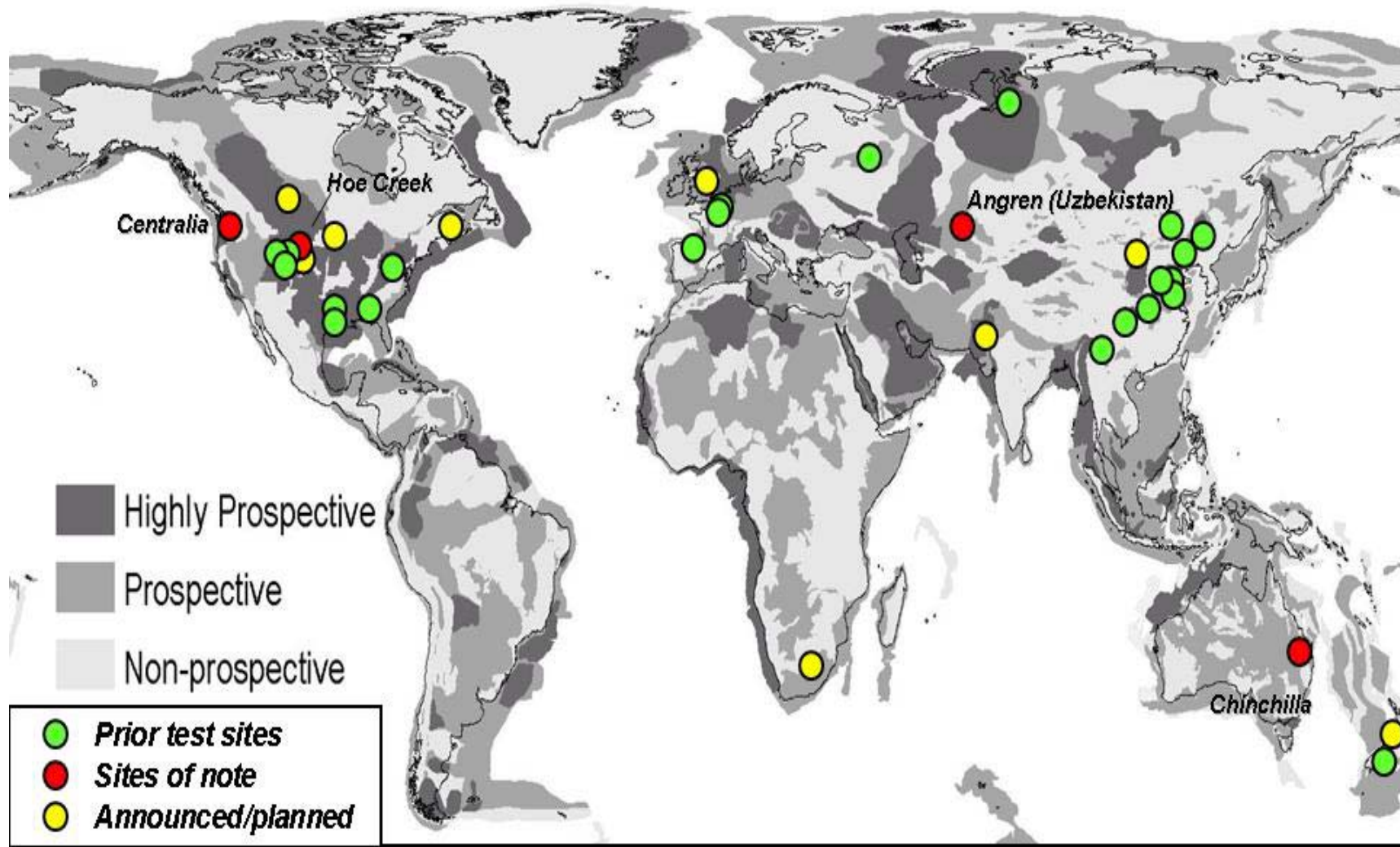
- Saline Aquifers
- Depleted Oil & Gas fields (w/ or w/o EOR and EGR)
- Unmineable Coal Seams (w/ or w/o ECBM)

CCS with UCG: Advantages



- **Good coincidence between CCS and UCG sites**
- **Substantial increase in capacity:**
 - The UCG burn enhances permeability and creates cavity to hold CO₂
 - Rubbleized coal, ash and char residuals have significantly enhanced surface area for CO₂ sorption.
- **Facilities in place:**
 - Gas processing plant and wells already in place
 - Eliminates substantial cost for CCS project (e.g., well drilling)
- **Response of coal likely to enhance sequestration capacity and lower leakage risk:**
 - Sorption of CO₂ onto organic matrix will immobilize and attenuate potential CO₂ migration and leakage
 - Coals swell and plasticize in the presence of CO₂, which could close fractures and porosity

Global Prospects for Carbon Storage and UCG Sites



From Bradshaw and Dance, 2005

CO₂ storage within UCG zones: Caveats



UCG processes cause thermal, geomechanical, and geochemical changes to the reservoir:

- Heating/quenching effects on fractures and rock properties
- Enhanced permeability from acid leaching of ash, tars, char, coal, rock minerals
- Changes in fluid density from temperature and TDS
- Increased solubility of organic contaminants in CO₂
- Increased solubility of metals in acid groundwaters

Objective: Evaluate benefits against drawbacks to quantify risk of UCG/CCS operations

Integrated UCG-CCS Tools

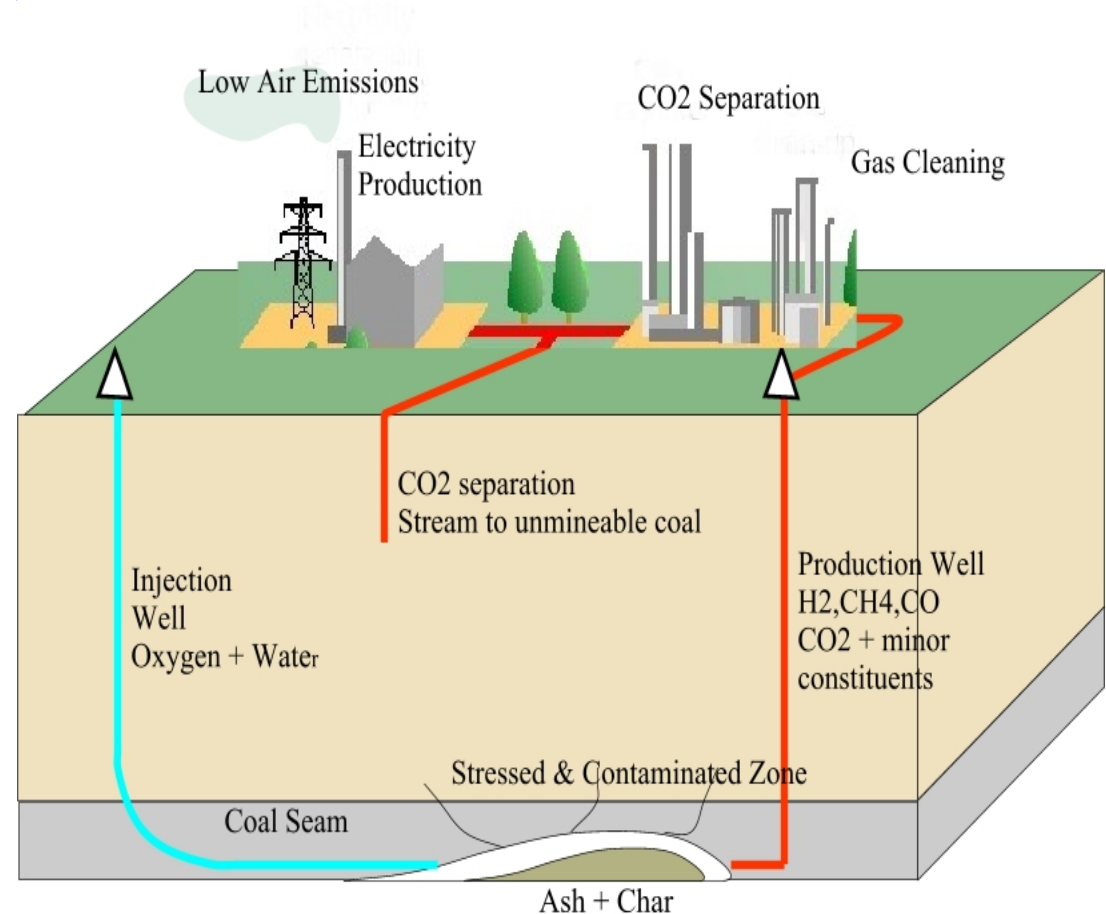


- **Expansion of environmental and site assessment capabilities to include aspects pertinent to CCS**
 - Include CCS aspects in RBDM framework
 - Enhanced solubility of organic compounds in CO₂
 - Leakage risk for CO₂
- **Expansion of monitoring capabilities to include detection of CO₂ migration underground and to detect leakage to surface**
- **Consider engineering and economic aspects of CCS during UCG planning**

Science and Technology Gaps



- Environmental and site assessment
- Process and environmental monitoring
- (Computational Fluid Dynamics) CFD gasification model
- Issues with Carbon Capture and Storage (CCS) with UCG



Disclaimer and Auspices Statements



This document was prepared as an account of work sponsored by an agency of the United States Government. Neither the United States Government nor the University of California nor any of their employees, makes any warranty, express or implied, or assumes any legal liability or responsibility for the accuracy, completeness, or usefulness of any information, apparatus, product, or process disclosed, or represents that its use would not infringe privately owned rights. Reference herein to any specific commercial product, process, or service by trade name, trademark, manufacturer, or otherwise, does not necessarily constitute or imply its endorsement, recommendation, or favoring by the United States Government or the University of California. The views and opinions of authors expressed herein do not necessarily state or reflect those of the United States Government or the University of California, and shall not be used for advertising or product endorsement purposes.

This work was performed under the auspices of the U.S. Department of Energy by University of California, Lawrence Livermore National Laboratory under Contract W-7405-Eng-48.