2 Properties, Production, and Potential for Exposure

2.1 Chemical and Physical Properties

In the crystalline state, one silicon atom and four oxygen atoms are arranged in an ordered, repetitive array of three-dimensional tetrahedrons. The silicon atom is the center of the tetrahedron. Each of the four corners consists of a shared oxygen atom.

Exposure to changes in temperature and pressure, either natural or synthetic, may cause the crystalline structure to change [Iler 1979; Klein and Hurlbut 1993; Navrotsky 1994; Hemley et al. 1994; IARC 1997]. An example of a naturally occurring pressure change is the transformation of alpha quartz to coesite in a rock subjected to the impact of a large meteorite [Iler 1979; Klein and Hurlbut 1993; IARC 1997]. Alpha quartz and beta quartz are the respective designations given to the low- and high-temperature crystal structures. Quartz changes from the alpha to the beta form at 573 °C (1,063 °F) [Ampian and Virta 1992; NIOSH 1983a; Virta 1993; Guthrie and Heaney 1995].

The solubility of quartz in water at room temperature ranges from 6 to 11 micrograms per cubic centimeter (μ g/cm³) (6 to 11 parts per million [ppm]) as SiO₂ [Coyle 1982; Iler 1979]. Quartz is slightly soluble in body fluids, where it forms silicic acid and is excreted by the urinary system [IARC 1987]. The amount of silica dissolved depends on various factors, including particle size, shape, and structure; solution temperature; viscosity; pH; the pro-

portion of dust to liquid; and the presence of trace minerals [King and McGeorge 1938; King 1937; Iler 1979; Wiecek 1988; IARC 1997; Guthrie 1997]. However, the dissolution of quartz does not contribute substantially to its clearance or to changes in its biological activity [IARC 1997; Heppleston 1984; Vigliani and Pernis 1958].

2.2 Number of Workers Potentially Exposed

NIOSH [1991] estimates that at least 1.7 million U.S. workers are potentially exposed to respirable crystalline silica. This estimate is based on information from the National Occupational Exposure Survey (NOES) [NIOSH 1983b] and the *County Business Patterns 1986* [Bureau of the Census 1986]. Table 1 lists the nonmining industries (excluding agriculture) and mining industries with the largest numbers of workers potentially exposed to respirable crystalline silica. In addition, an undetermined portion of the 3.7 million U.S. agricultural workers [Bureau of the Census 1997] may be exposed to dust containing a significant percentage of respirable crystalline silica [Linch et al. 1998].

2.3 Dust-Generating Activities, Uses, and Potential Exposures

Crystalline silica (quartz) is a component of nearly every mineral deposit [Greskevitch et al. 1992]. Thus most crystalline silica exposures are to mixed dust with variable silica content that must be measured by dust collection and

SIC*	Industry	Estimated number of workers potentially exposed (1986) [†]	% total workers exposed (NOES)
Nonmining industries:			
174	Masonry, stonework, tile setting, and plastering	131,986	32.7
734	Services to dwellings and other buildings	65,812	10.3
327	Concrete, gypsum, and plaster products	63,456	33.3
176	Roofing and sheet metal work	51,153	25.3
356	General industrial machinery and equipment	44,991	16.2
807	Medical and dental laboratories	37,063	30.0
493	Combination of gas and electric and other utilities	35,074	21.2
179	Miscellaneous special trade contractors	32,615	7.8
753	Automotive repair shops	30,826	7.1
326	Pottery and related products	29,772	81.7
Mining industries:			
13	Oil and gas extraction	408,175	100‡
12	Bituminous coal and lignite mining	174,131	100
14	Mining and quarrying of nonmetallic minerals, except fuels	100,546	100
10	Metal mining	39,856	100

Table 1. Nonmining and mining industries with the largest numbers of U.S. workers potentially exposed to respirable crystalline silica, 1986

Source: NIOSH [1991].

*Standard industrial classification.

[†]Estimated number of workers potentially exposed to the hazards of flint, quartz, sand, or silica powder; based on data from the *County Business Patterns 1986* [Bureau of the Census 1986] and the National Occupational Exposure Survey (NOES) [NIOSH 1983b]. For SICs in which the estimates differed for individual hazards, the highest percentage was used for that SIC.

[‡]Exposure is assumed to be 100% in the mining industries.

analysis [Wagner 1995; Donaldson and Borm 1998].

Workers in a large variety of industries and occupations may be exposed to crystalline silica because of its widespread natural occurrence and the wide uses of the materials and products containing it. OSHA compliance officers found respirable quartz in 6,779 personal samples (8-hr TWA) taken in 255 industries that were targeted for inspection (excluding mining and agriculture). In 48% of the industries, average overall exposure exceeded the PEL for respirable quartz [Freeman and Grossman 1995]. Linch et al. [1998] applied an algorithm to OSHA compliance data from the period 1979–1995 and County Business Patterns 1993 data [Bureau of the Census 1993] to estimate the percentage of workers by industry (excluding mining and agriculture) exposed to defined concentrations of respirable crystalline silica (e.g., $\geq 0.05 \text{ mg/m}^3$) in 1993. Area samples and samples involving complaints to OSHA were excluded from the analysis. Although data limitations could have resulted in underestimating or overestimating the number of workers exposed, the authors found 5 three-digit standardized industrial classification (SIC) codes in which an estimated number of workers were exposed to concentrations at least 10 times the NIOSH REL:

C	T	٢	7
S,		L	/

No. workers

174	Masonry and plastering	13,800 (1.8%)
162	Heavy construction	6,300 (1.3%)
172	Painting and paper hanging	3,000 (1.9%)
332	Iron and steel foundries	800 (0.3%)
347	Metal services	400 (0.2%)

Additional three-digit SICs had a number of workers with crystalline silica exposures that were two or five times higher than the NIOSH REL [Linch et al. 1998].

Table 2 lists the main industries around the world in which silica exposure has been reported. Virtually any process that involves movement of earth or disturbance of silica-containing products such as masonry and concrete may expose a worker to silica (see Table 3 for uses of industrial silica sand and gravel). Table 4 presents, from selected States, the most frequently recorded occupations of U.S. residents aged 15 or above whose death certificates list silicosis as an underlying or contributory cause of death [NIOSH 1996a]. In addition, Table 5 lists published case reports of silicosis in workers from other industries and occupations.

2.4 Sampling and Analytical Methods

Historically, several methods have been used to measure worker exposure to airborne crystalline silica (quartz, cristobalite, or tridymite). These methods differ primarily in the analytical technique employed, although they all rely on a collection procedure that uses a cyclone for size-selective sampling. Airborne samples are collected using a cyclone to remove nonrespirable particles and an appropriate filter medium (e.g., polyvinyl chloride) to retain the respirable dust fraction. Preparation of the sample for crystalline silica determination differs depending on the type of analytical technique used. One of three analytical techniques is typically used for the quantitative determination of crystalline silica: X-ray diffraction (XRD) spectrometry, infrared absorption (IR) spectrometry, or colorimetric spectrophotometry. XRD and IR are the most common techniques used for crystalline silica analyses. The quantitative limit of detection for these methods ranges from 5 to 10 μ g per sample; but the accuracy is poor, particularly at the low filter loadings ($\leq 30 \mu g$ per sample) that are typically collected when workplace concentrations of airborne crystalline silica are near the NIOSH REL of 50 μ g/m³ (or 0.05 mg/m³).

Industry or activity	Operations and tasks	Source materials
Agriculture	Plowing, harvesting, using machinery, burning agricultural waste, processing agricultural products	Soil
Mining and related milling operations	Most occupations (underground, surface, mill) and mines (metal and nonmetal, coal), rock drilling, dredging	Ores, associated rock
Quarrying and related milling operations	Crushing stone, sand and gravel processing, stone monument cutting and abrasive blasting, slate work (e.g., pencil manufacturing), diatomite calcination	Sandstone, granite, flint, sand, gravel, slate, diatomaceous earth
Construction	Abrasive blasting of structures and buildings, highway and tunnel construction, excavation and earth moving and digging, masonry, concrete work, demolition, dry sweeping and brushing, pressurized air blowing, jack hammering, laying railroad track, removing rust or paint, sanding and scaling, replacement of asphalt roofing, and hauling, pouring, mixing, or dumping silica-containing materials	Sand, concrete, rock, soil, mortar, plaster, shingles
Glass, including fiberglass	Raw material processing, refractory installation and repair	Sand, crushed quartz, refractory materials
Cement	Raw material processing	Clay, sand, limestone, diatomaceous earth
Abrasives	Silicon carbide production, abrasive products fabrication	Sand, tripoli, sandstone
Ceramics, including bricks, tiles, sanitary ware, porcelain, pottery, refractories, vitreous enamels	Mixing, molding, glaze or enamel spraying, finishing, sculpting, firing	Clay, shale, flint, sand, quartzite, diatomaceous earth
Iron and steel mills	Refractory preparation and furnace repair	Refractory material

Table 2. Main industries and activities around the world in which silica exposure has been reported

(Continued)

Sources: IARC [1987; 1997], NIOSH [1979a; 1983a,b; 1996b], DOL, NIOSH [1997], Fulekar and Alam Khan [1995], Jain et al. [1977], Com [1980], Webster [1982], Froines et al. [1986], Davis [1996], Weill et al. [1994], Lucas and Salisbury [1992], Pike [1992], McCunney et al. [1987], Fairfax [1998].

Industry or activity	Operations and tasks	Source materials
Silicon and ferro-silicon foundries (ferrous and nonferrous)	Raw materials handling, casting, molding and shaking out, abrasive blasting, fettling, furnace installation and repair	Sand, refractory material
Metal products, including structural metal, machinery, transportation equipment	Abrasive blasting	Sand
Shipbuilding and repair	Abrasive blasting	Sand
Rubber and plastics	Raw materials handling	Fillers (tripoli, diatomaceous earth)
Paint	Raw materials handling, site preparation	Fillers (tripoli, diatomaceous earth, silica flour)
Soaps and cosmetics	Manufacturing or occupational use of abrasive soaps and scouring powders	Silica flour
Roofing asphalt felt	Filling and granule application	Sand and aggregate, diatomaceous earth
Agricultural chemicals	Raw material crushing, handling, bagging; or dumping products or raw materials	Phosphate ores and rock
Jewelry	Cutting, grinding, polishing, buffing, etching, engraving, casting, chipping, sharpening, sculpting	Semiprecious gems or stones, abrasives, glass
Arts, crafts, sculpture	Pottery firing, ceramics, clay mixing, kiln repairs, abrasive blasting, sand blasting, engraving, cutting, grinding, polishing, buffing, etching, engraving, casting, chipping, sharpening, sculpting	Clays, glazes, bricks, stones, rocks, minerals, sand, silica flour
Dental material	Sand blasting, polishing	Sand, abrasives
Boiler scaling	Coal-fired boilers	Ash and concretions
Automobile repair	Abrasive blasting, sanding, removing paint and rust	Sand, metals, priming putty

Table 2 (Continued). Main industries and activities around the world in which silica exposure has been reported

General use	End use
Sand:	
Glass-making	Containers, flat (plate and window), specialty, fiberglass (unground or ground)
Foundry work	Molding and core, molding and core facing (ground), refractory
Metallurgical work	Silicon carbide, flux for metal smelting
Abrasive work	Blasting, scouring cleansers (ground), sawing and sanding, chemicals (ground and unground)
Fillers	Rubber, paints, putty, whole grain fillers/building products
Ceramics	Pottery, brick, tile
Filtration	Water (municipal, county, local), swimming pool, others
Petroleum manufacturing	Hydraulic fracturing, well packing, and cementing
Recreation	Golf course, baseball, volleyball, play sands, beaches, traction (engine), roofing granules and fillers, other (ground silica or whole grain)
Gravel	Silicon, ferrosilicon, filtration, nonmetallurgical flux, other

Table 3. Industrial silica sand and gravel sold or used by U.S. producers in 1994, by major end use

Sources: IARC [1997]; BOM [1994].

\mathbf{COC}^{\dagger}	Occupation	Number	%
616	Mining machine operator	39	16.0
889	Laborer, except construction	29	11.9
019	Manager or administrator, not elsewhere classified	11	4.5
633	Supervisor or precision production occupations	11	4.5
453	Janitor, cleaner	8	3.3
719	Molding, casting machine operator	8	3.3
243	Supervisor or proprietor of sales occupations	6	2.5
844	Operating engineer	6	2.5
637	Machinist	5	2.1
787	Hand molding, casting, and forming occupations	5	2.1
_	All other occupations	109	44.9
_	Occupation not reported	6	2.5
	TOTAL	243	100.1‡

Table 4. Most frequently recorded occupations of U.S. residents aged 15 or above whose death certificates list silicosis as an underlying or contributory cause of death—selected States, 1991–1992*

Source: NIOSH [1996a].

*Data for 1985–1990 are reported in Table 4–11 of NIOSH [1994d].

[†]COC: 1980 census occupation code. [‡] Column does not add to 100.0 because of rounding.

Industry or occupation	Reference
Agriculture industry or forestry worker	Fennerty et al. [1983]; Dynnik et al. [1981]; Beaumont et al. [1995]
Brewery worker	Nemery et al. [1993]
Confectioner	Canessa et al. [1990]
Crystal cutter	Suskovic et al. [1990]
Drycleaning worker	Seitz et al. [1982]
Filter candle production worker	Vigliani and Mottura [1948]
Grave digger and well digger	al-Kassimi et al. [1991]
Kaolin worker	Rodriguez et al. [1985]
Metal polisher	Malik et al. [1985]
Pit digger	de Barros Hatem and Cavalcanti [1990]
Souvenir casting worker	Carel et al. [1994]
Woodworker	Thoreux et al. [1990]

Table 5. Other occupations^{*} reporting cases of silicosis in workers

*Includes only occupations not listed in Tables 2 or 4.

2.4.1 Sampling Methods

Current sampling methods for crystalline silica involve the use of a cyclone attached to a filter cassette to collect the respirable fraction of the airborne particulate. To minimize measurement bias and variability, these samplers should conform to the criteria of the International Organization for Standardization (ISO), the European Standardization Committee (CEN), and the American Conference of Governmental Industrial Hygienists (ACGIH) for collecting particles of the appropriate size [ISO 1991; CEN 1992; ACGIH 2001]. Also, the cyclone should exhibit sufficient conductivity to minimize the electrostatic effects on particle collection. Cyclones typically used for crystalline silica measurements include the Dorr-Oliver 10-mm nylon cyclone and the Higgins-Dewell conductive cyclone. These cyclones have been evaluated for their compliance with the ISO/CEN/ACGIH respirable aerosol sampling convention. Flow rates of 1.7 L/min for the Dorr-Oliver cyclone and 2.2 L/min for the Higgins-Dewell cyclone provide minimum bias for a wide range of particle size distributions that are likely to occur in the workplace [Bartley et al. 1994]. The Dorr-Oliver 10-mm cyclone is required by MSHA, and the Higgins-Dewell cyclone is used in the United Kingdom. Recently, the GK2.69 cyclone [Kenny and Gussman 1997] has become available with a sampling rate equal to 4.2 L/min. The GK2.69 cyclone is expected to be at least as adequate as the nylon cyclone for conforming to the ISO/ CEN/ACGIH respirable aerosol sampling convention; and it may be preferable for silica sampling since it is conductive, has welldefined dimensional characteristics, and can be used at higher flow rates for better mass sensitivity. Because each type of cyclone exhibits specific particle collection characteristics, the use of a single cyclone type for each application would be advisable until evidence becomes available indicating that bias among cyclone types will not increase laboratoryto-laboratory variability.

Cyclones and filter cassettes should be leak tested to avoid gross failure in the field. The cyclones may be tested using a simple pressure- (or vacuum-) holding test. The filter cassette should also be checked for leakage while attached to the cyclone. Two approaches to testing the cassettes have been used. A micromanometer has been used to measure the pressure drop across a single type of cassette and compare it with the average pressure drop across well-sealed cassettes [Van den Heever 1994]. An alternative approach uses a particle counter to measure the penetration of submicrometer ambient aerosol through the cassette, with the percentage of penetration serving as an indicator of leakage [Baron 2001]. Measurement of cassette leakage by several laboratories indicates that significant leakage can occur in certain situations. Cassettes should be assembled using a press, and they should be routinely checked for leakage.

2.4.2 Analytical Methods

2.4.2.1 XRD Spectrometry

XRD methods used for crystalline silica determination include NIOSH Method 7500 [NIOSH 1998], OSHA Method ID-142 [OSHA 1996], MSHA Method P-2 [MSHA 1999], and the Health and Safety Executive (HSE) Method for the Determination of Hazardous Substances (MDHS) 51/2 [HSE 1988]. Details of these methods are presented in Table 6. XRD is capable of distinguishing the three prevalent polymorphs of crystalline silica (quartz, cristobalite, and tridymite) and can simultaneously analyze for each polymorph while correcting for interferences that may be present on the sample [Madsen et al. 1995]. Although most samples collected in industrial workplaces are relatively free of mineral interferences, an XRD scan of some samples should be performed to ensure the absence of interferences through confirmation of the correct peak ratios for the three largest peaks.

Item	NIOSH Method 7500	OSHA Method ID-142	MSHA Method P-2	MDHS 51/2
Silica polymorph	Quartz, cristobalite, tridymite	Quartz, cristobalite	Quartz, cristobalite	Quartz
Sampler	10-mm nylon cyclone, 1.7 L/min; Higgins-Dewell cyclone, 2.2 L/min	10-mm nylon Dorr-Oliver cyclone, 1.7 L/min	10-mm nylon Dorr-Oliver cyclone, 1.7 L/min	Higgins-Dewell cyclone, 1.9 L/min
Filter	37-mm, 5-μm PVC membrane	37-mm, 5-μm PVC membrane	37-mm, 5-μm PVC membrane	25-mm, 5-μm PVC membrane
Volume	400–1,000 L; total dust < 2 mg	408–816 L; total dust < 3 mg	400–1,000 L; total dust < 3 mg	≥456 L; total dust < 2 mg
Filter preparation	RF plasma asher, muffle furnace, or filter dissolution in THF	Dissolve filter in THF	RF plasma asher	None
Redeposition	On 0.45-µm silver membrane filter	On 0.45-µm silver membrane filter	On 0.45-µm silver membrane filter	None
Drift correction	Silver internal standard	Silver internal standard	Silver internal standard	External standard (e.g., aluminum plate)
X-ray source	Cu K _α ; 40 kV, 35 mA	Cu K _α ; 40 kV, 40 mA	Cu K _α ; 55 kV, 40 mA	Cu K _α ; 45 kV, 45 mA
Calibration	Suspensions of SiO ₂ in 2-propanol (deposited on silver membrane filter)	Suspensions of SiO ₂ in 2-propanol (deposited on silver membrane filter)	Suspensions of SiO_2 in 2-propanol (deposited on silver membrane filter)	Sampling from a generated atmosphere of standard quartz dust
Proficiency testing	PAT	PAT	PAT	WASP
Range (µg quartz)	20-2000	50–160 (validation range)	20-500	50-2000
LOD (µg quartz)	5 (estimated)	10	5	3
Precision	$\overline{\text{RSD}} = 0.08$ 50-200 µg	$\overline{CV} = 0.106 @$ 50-160 µg	CV = 10 % @ 20-500 μg	CV = 5 % @ 50 µg

Table 6. XRD	sampling and	analytical	methods for	crystalline silica
		•/		•/

*Abbreviations: Cu = copper; CV = coefficient of variation (equivalent to RSD); \overline{cv} = pooled coefficient of variation; K_{α} = electron ionization energy; kV = kilovolt(s); LOD = limit of detection; mA = milliampere(s); MDHS = Methods for the Determination of Hazardous Substances (Health and Safety Executive, United Kingdom); MSHA = Mine Safety and Health Administration; NIOSH = National Institute for Occupational Safety and Health; OSHA = Occupational Safety and Health Administration; PAT = proficiency analytical testing; PVC = polyvinyl chloride; RF = radio frequency; RSD = relative standard deviation; \overline{RSD} = pooled relative standard deviation (equivalent to \overline{cv}); THF = tetrahydrofuran; WASP = Workplace Analysis Scheme for Proficiency; XRD = X-ray diffraction.

2.4.2.2 IR Spectrometry

IR methods used for crystalline silica determination include NIOSH Methods 7602 and 7603 [NIOSH 1994a,c], MSHA Method P-7 [MSHA 1994], and MDHS 37 and 38 [HSE 1987, 1984]. Details of these methods are presented in Table 7. Although IR is less specific than XRD (IR methods cannot readily distinguish crystalline silica polymorphs), the technique is less expensive and can be optimized for measuring quartz in well-defined sample matrices [Madsen et al. 1995; Smith 1997; Hurst et al. 1997]. Samples that contain other silicates (such as kaolinite) and amorphous silica can present interferences in the analyses. Also, a potential for bias exists when correcting for matrix absorption effects, with an increasing risk of bias at lower quartz concentrations.

2.4.2.3 Colorimetric Spectrophotometry

The NIOSH colorimetric method for crystalline silica (NIOSH Method 7601) [NIOSH 1994b] is significantly less precise than IR or XRD methods. The colorimetric analytical method exhibits a nonlinear dependence on the mass of crystalline silica present [Eller et al. 1999a]. The linear range of the method is limited, and the blank values for samples can be high (20 µg silica or higher) [Talvitie 1951, 1964; Talvitie and Hyslop 1958]. High intralaboratory variability of the method (up to twice that of IR or XRD) has been noted in studies conducted in the Proficiency Analytical Testing Program (PAT) [Shulman et al. 1992]. The colorimetric method cannot distinguish between silica and silicates, since it is based on the measurement of silicon.

2.4.2.4 Factors Affecting the Sensitivity and Accuracy of Analytical Techniques

Samples prepared for XRD analyses are measured directly (MDHS 51/2) or are redeposited onto 25-mm silver membrane filters (NIOSH Method 7500 and OSHA Method ID-142). IR samples can be measured directly (MDHS 37), redeposited on an acrylic copolymer membrane filter (NIOSH Method 7603 and MSHA Method P-7), or incorporated into a potassium bromide (KBr) pellet (NIOSH Method 7602 and MDHS 38). Techniques used for redepositing the sample (both IR and XRD) are difficult to perform at low sample loadings and require the laboratory analyst to demonstrate good intralaboratory reproducibility. However, these techniques can be optimized by preparing multiple working standards from multiple suspensions of calibration standards and by ensuring that the sample is redeposited evenly as a thin layer on the filter. No statistically significant difference has been observed between ashing the filter (muffle furnace and low-temperature asher) and dissolving the filter by tetrahydrofuran before redepositing the sample [Eller et al. 1999a].

The instrument response of all three analytical techniques is influenced by the size of the particles in the sample. With XRD, the diffraction intensity (as measured by peak height) can vary considerably with particle size, with smaller particles showing lower intensities [Bhaskar et al. 1994]. The sensitivity of IR analyses decreases with increasing particle size. The colorimetric method requires the use of a precisely timed heating step with phosphoric acid to digest amorphous silica and silicates during sample preparation, causing a possible loss of some small crystalline silica particles [Eller et al. 1999a]. Since particle size affects the sensitivity of all three analytical techniques, the particle size distribution of the calibration standard should closely match the size of the particles retained on the collected sample.

For all analytical techniques, strict adherence to standardized procedures is necessary to produce accurate results. Specifically, appropriate

Item	NIOSH Method 7602	NIOSH Method 7603	MSHA P-7	MDHS 37	MDHS 38
Matrix		Coal mine dust	Coal mine dust		
Sampler	10-mm nylon cyclone, 1.7 L/min; Higgins-Dewell cyclone, 2.2 L/min	10-mm nylon cyclone, 1.7 L/min; Higgins-Dewell cyclone, 2.2 L/min	10-mm nylon Dorr-Oliver cyclone, 2.0 L/min	Higgins-Dewell cyclone, 1.9 L/min	Higgins- Dewell cyclone, 1.9 L/min
Filter	37-mm filter; 5-μm PVC or MCE membrane	37-mm filter; 5-μm PVC membrane	37-mm filter; 5-µm PVC membrane, preweighed	37-mm filter; 5-μm PVC membrane	37-mm filter; 5-µm PVC membrane
Volume	400–800 L; total dust <2 mg	300-1,000 L; total dust <2 mg	Not stated	≥456 L; total dust <1 mg	≥456 L; total dust <0.7mg
Filter preparation	RF plasma asher or muffle furnace	RF plasma asher or muffle furnace	RF plasma asher	None	Muffle furnace
Analytical sample preparation	Mix residue with KBr, press 13-mm pellet	Redeposit on 0.45-µm acrylic copolymer filter	Redeposit on 0.45-µm acrylic copolymer filter	None	Mix residue with KBr, press 13-mm pellet
Standard	Polystyrene film	Polystyrene film	Polystyrene film	Polystyrene film	Polystyrene film
Calibration	Quartz diluted in KBr	Standard suspension of quartz in 2-propanol	Standard suspension of quartz in 2-propanol	Sampling from a generated atmosphere of standard quartz dust	Sampling from a generated atmosphere of standard quartz dust
Proficiency testing	PAT	PAT	PAT	WASP	WASP
Range (µg quartz)	10-160	30-250	25-250	10-1,000	5-700
LOD (µg quartz)	5 (estimated)	10 (estimated)	10	Varies with particle size	Varies with particle size

Table 7. \mathbf{IR}^* sampling and analytical methods for crystalline silica

See footnote at end of table.

(Continued)

Item	NIOSH Method 7602	NIOSH Method 7603	MSHA P–7	MDHS 37	MDHS 38
Precision	RSD <0.15	$\overline{\text{RSD}} = 0.098$	CV = 5-10 %	CV = 5 %	CV = 5 %
	@ 30μg	@ 100-500µg	@ 100-500 μg	@ 50 μg	@ 50 μg

Table 7 (Continued). IR* sampling and analytical methods for crystalline silica

*Abbreviations: CV = coefficient of variation (equivalent to RSD, relative standard deviation); IR = infrared absorption; KBr = potassium bromide; MCE = methyl cellulose ester; MDHS = Methods for the Determination of Hazardous Substances (Health and Safety Executive, United Kingdom); MSHA = Mine Safety and Health Administration; NIOSH = National Institute for Occupational Safety and Health; LOD = limit of detection; PAT = proficiency analytical testing; PVC = polyvinyl chloride; RF = radio frequency; RSD = pooled relative standard deviation (equivalent to CV, pooled coefficient of variation); WASP = Workplace Analysis Scheme for Proficiency.

calibration of the technique has been shown to be critical in the accurate measurement of crystalline silica [Eller et al. 1999b]. Also, it is essential that only standard reference materials from the National Institute of Standards and Technology (NIST) (for which particle size and phase purity has been established) be used to prepare calibration curves for quartz (1878a) and cristobalite (1879a) [Eller et al. 1999a]. No standard reference material for tridymite is available, since this silica polymorph rarely exists in the workplace. However, a wellcharacterized sample of tridymite of the appropriate particle size is available from the U.S. Geological Survey^{*} and can be used as a reference standard.

Direct-on-filter techniques are used by the United Kingdom, the European Union, and Australia [Madsen et al. 1995]. These techniques require less time and labor than others and are amenable to both XRD and IR analyses [Lorberau et al. 1990]. However, direct-on-filter techniques are affected by the manner in which the particles are deposited on the filter sample (particle deposition may be nonuniform). Thus care must be taken when choosing the area of the filter to measure so that results can be compared with other methods. Sample overloading is possible for a sample collected over a full work shift.

2.4.3 Feasibility of Measuring Crystalline Silica at Various Concentrations

The efficacy of sampling and analytical methods for measuring concentrations of hazardous materials may be established using the NIOSH accuracy criterion [NIOSH 1995b], which requires better than 25% accuracy at concentrations of expected method application. Accuracy, as a percentage of true concentration values, is defined in terms of an interval expected to contain 95% of (future) measurements. To account for uncertainty in method evaluations, the upper 95% confidence limit on the accuracy is measured and used in the criterion. Generally, the accuracy of a method is measured over a range of concentrations bracketing the OSHA PEL. Use of a range of measurements means that accuracy is assuredboth at concentrations below the PEL (for possible use in action level determinations) and, more significantly, at the PEL (where method results must be legally defensible).

NIOSH has evaluated both the XRD silica method (NIOSH Method P&CAM 259, the forerunner to NIOSH Method 7500) [NIOSH 1979b] and an IR silica method (MSHA Method P-7, equivalent to NIOSH Method 7603) in a collaborative test among several laboratories [NIOSH, BOM 1983]. One result of the test was that the accuracy of the methods was estimated by evaluating the intralaboratory variability at various filter loadings. The concentrations to which these filter loadings correspond depend on the flow rate of the presampler used. Experimental conditions and results relevant to the derivation of these estimates are summarized in Tables 8 and 9. The results of the collaborative tests indicate that both the XRD and IR methods tested meet the NIOSH accuracy criterion [NIOSH 1995b] over the range of filter loadings measured. Currently, OSHA uses the 10-mm nylon cyclone at a sampling rate of 1.7 L/min for sampling crystalline silica. The concentrations relevant to the collaborative test conditions are listed in Tables 10 and 11 and assume an 8-hr sampling period. As indicated in Tables 10 and 11, the traditional nylon cyclone meets the

^{*}Tridymite reference material may be obtained from Dr. Stephen A. Wilson, U.S. Geological Survey, Box 25046, MS 973, Denver, CO 80225 (telephone: 303–236–2454; FAX: 303–236–3200; e-mail: swilson @usgs.gov; Web site: http://minerals.cr.usgs.gov/ geochem).

	Filter loading		
Item	69.4 μg	98.4 μg	204 µg
Degrees of freedom	12	11	12
RSD for sampling and analytical methods $\left(\%\right)^{*,\dagger}$	8.8	6.3	8.1

Table 8. Intralaboratory results for evaluation of XRD silica method

Source: NIOSH, BOM [1983].

^{*}RSD = relative standard deviation. RSD for sampling and analytical methods represents the RSD in mass estimates, accounting for intersampler and analytical variability.

[†]Implications for XRD: Pooled filter levels and pump error (assumed to be <5%) indicate that the overall imprecision is as follows: Total RSD for sampling and analytical methods is 9.3%. Therefore, the upper 95% confidence limit on the accuracy (35 degrees of freedom) is 21%.

	Filter loading		
Item	67.2 μg	99.7µg	161 µg
Degrees of freedom	10	12	11
RSD for sampling and analytical methods $\left(\%\right)^{*,\dagger}$	5.8	7.8	7.4

Table 9. Intralaboratory results for evaluation of IR silica method

Source: NIOSH, BOM [1983].

*RSD = relative standard deviation. RSD for sampling and analytical methods represents the RSD in mass estimates, accounting for intersampler and analytical variability.

[†]Implications for IR: Pooled filter levels and pump error (assumed to be <5%) indicate that the overall imprecision is as follows: Total RSD for sampling and analytical methods is 7.1%. Therefore, the upper 95% confidence limit on the accuracy (33 degrees of freedom) is 17%.

		Filter loading		
Cyclone and sampling rate	69.4 μg	98.4 μg	204 μg	Applicable exposure limit
Nylon cyclone, 1.7 L/min	85	121	251	100
GK2.69 cyclone, 4.2 L/min	34	49	102	50

Table 10. XRD method evaluation: concentration ranges bracketing applicable exposure limits for which the NIOSH accuracy criterion is met* (µg/m³)

*Eight-hour sampled masses are combined with results of NIOSH, BOM [1983].

Table 11. IR method: concentration ranges bracketing applicable exposure limits for which the NIOSH accuracy criterion is met* (ug/m³)

(µg/m)					
Cyclone and sampling rate	6.72 μg	99.7 μg	161 μg	Applicable exposure limit	
Nylon cyclone, 1.7 L/min	83	123	198	100	
GK2.69 cyclone, 4.2 L/min	34	50	80	50	

*Eight-hour sampled masses are combined with results of NIOSH, BOM [1983].

accuracy criterion over a range of concentrations bracketing $100 \ \mu g/m^3$.

Since the GK2.69 cyclone is expected to conform to the ISO/CEN/ACGIH respirable aerosol sampling convention, the NIOSH intralaboratory collaborative tests can be used to establish confidence limits on its accuracy. The results of the collaborative tests indicate that the GK2.69 cyclone meets the accuracy criterion over a range of concentrations bracketing $50 \text{ }\mu\text{g/m}^3$, as illustrated in Tables 10 and 11.