

Laboratory and Numerical Modeling Results on Hydrate-Bearing Sediments

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OUTLINE

- THF (Tetrahydrofuran) Hydrate
- JIP Experimental Matrix
- Samples and Laboratory Equipment
- Properties
 - Mechanical Properties: large and small strain
 - Thermal Properties
 - Electrical Properties
- Lensing
- Process Monitoring (Phase Transformation)
- Core Recovery Numerical Modeling

Hydrate Structures

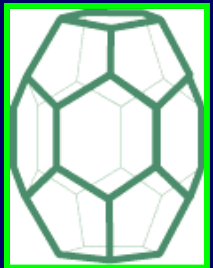
All 3 occur in Gulf of Mexico, with I and II more common



Structure I: 46 water + 8 gas
methane, ethane (up to 5.2 Å)
sites: 2 x 5^{12} and 6 x $5^{12}6^2$



Structure II: 136 water + 24 gas
propane, tetrahydrofuran (5.9-6.9 Å)
sites: 16 x 5^{12} and 8 x $5^{12}6^4$



Structure H: 34 water + 6 gas
iso-pentane (> 6.9 Å)
small, medium, and large sites

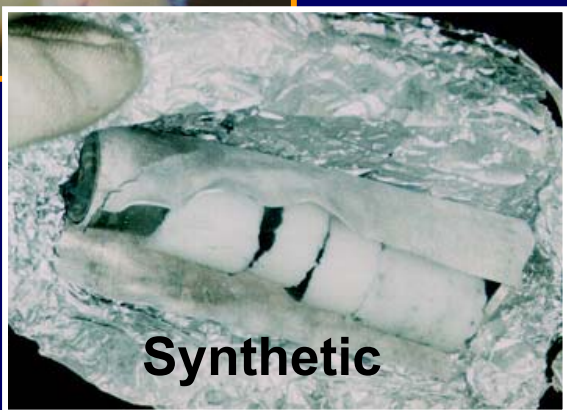
(after Sassen)

Methane Hydrate

Structure I



Stern



Synthetic

THF Hydrate

Structure II



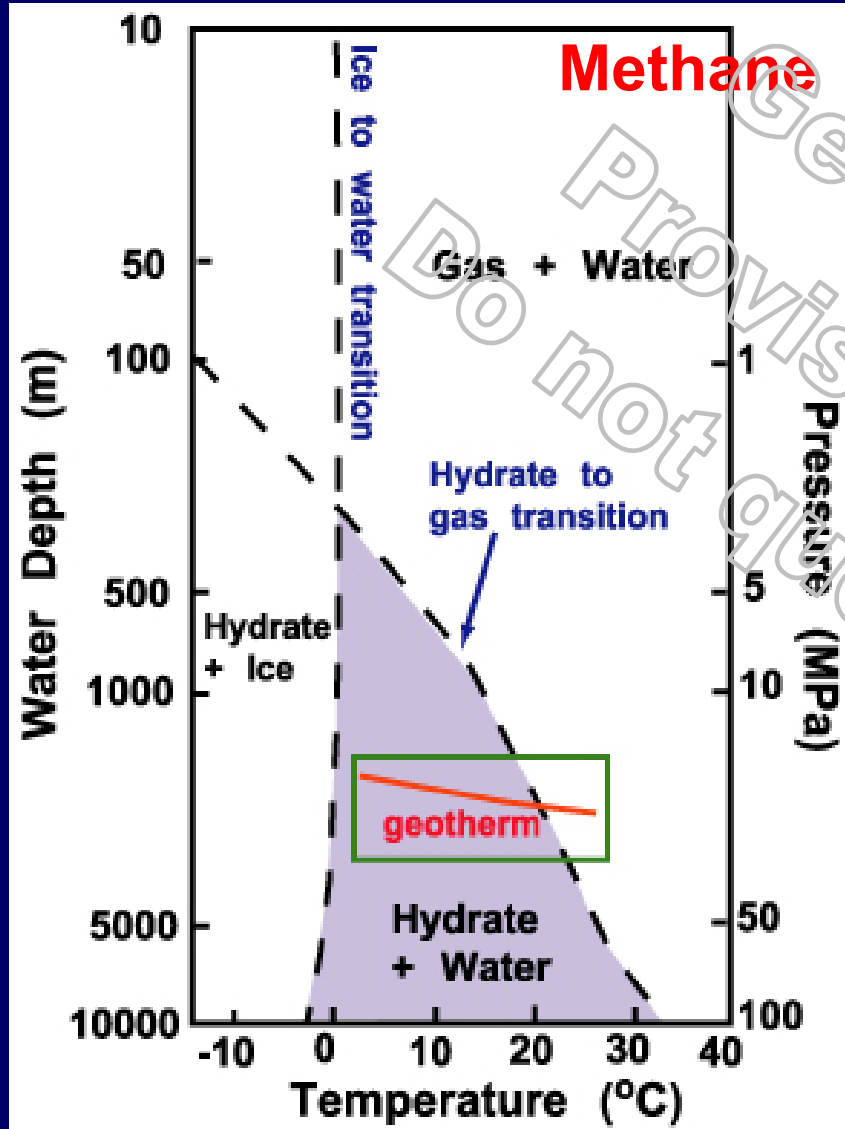
Synthetic



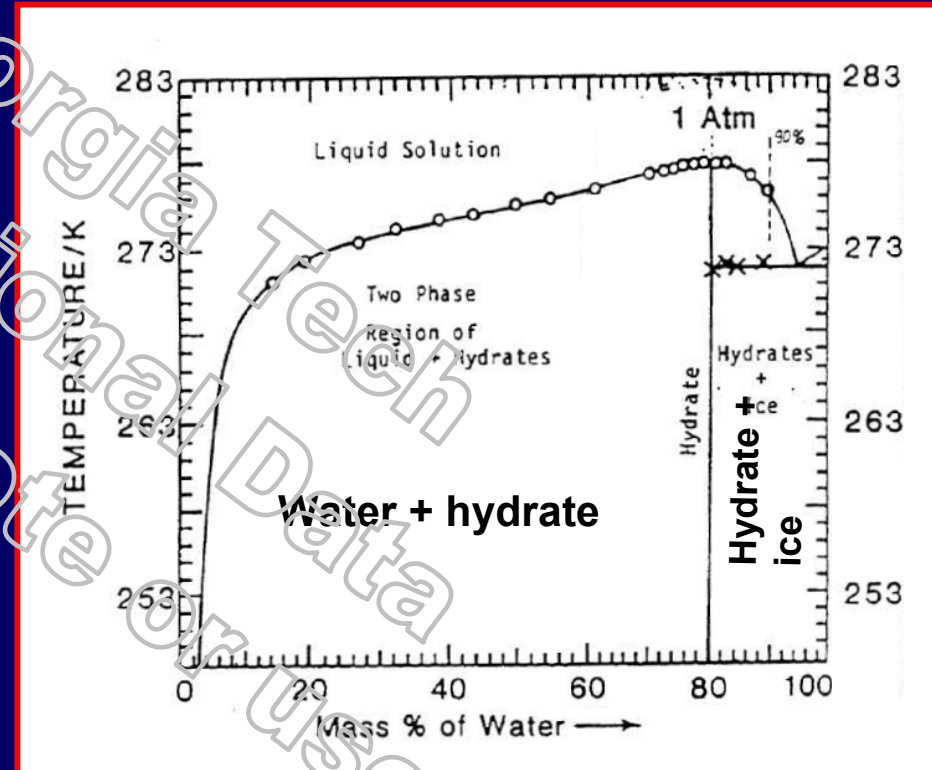
Georgia Tech
Provisional Data
Do not quote or use

Phase Diagrams

METHANE



THF



Methane vs. THF Hydrate

Pressure at phase transformation

Methane 10 -100 MPa for T = 4-30°C

THF 1 atm at T=~-4.4 ° C (optimal stoichiometry)

Molecules per cage

Methane 6

THF 17

Solubility

Methane $(\text{CH}_4 + 3.7 \times 10^5 \text{H}_2\text{O})$

THF fully soluble

Diffusion Time

Methane 22 years for length of 1 meter

THF Advantages for Laboratory Study: Structure II hydrate-former, completely miscible, hydrate formation at close to 1 atm and 0°C
“ We completely understand our system”

Hydrate Properties: Methane vs. THF

Property	Ice	CH ₄ hydrate	THF hydrate
Thermal properties			
Heat capacity (J K ⁻¹ g ⁻¹)	2.097 @ 270 K	2.07 @ 270 K	2.07 @ 270 K
Heat of dissociation (kJ kg ⁻¹)	-333.5 @ 273 K	+338.7 @ 273 K	+262.9 @ 273 K
Thermal conductivity (W m ⁻¹ K ⁻¹)	2.23 @ 263 K	0.5 @ 270 K	0.53 @ 250 K
Thermal diffusivity (m ² s ⁻¹)	15.4x10 ⁻⁷ @ 250 K	3 x 10 ⁻⁷ @ 270 K	
Vol. strain liquid-to-solid	168 x 10 ⁻⁶ @ 280 K	231 x 10 ⁻⁶ @ 200 K	156 x 10 ⁻⁶ @ 200 K
Thermal linear expansivity (K ⁻¹)	56 x 10 ⁻⁶ @ 200 K	77 x 10 ⁻⁶ @ 200 K	52 x 10 ⁻⁶ @ 200 K
Physical properties			
Bulk Compressibility (Pa)	12x10 ⁻¹¹ @ 273 K	~14x10 ⁻¹¹ @ 273 K	~14x10 ⁻¹¹ @ 273 K
Density (kg m ⁻³)	917 @ 273 K	910 @ 273 K	~ 910 @ 273 K
Diff. coef. in water (cm ² s ⁻¹)		1.49 x 10 ⁻⁵	0.56 x 10 ⁻⁵ @ 0.059 M
Strength hydrate + sand (M Pa)	10.5 for ε = 10 ⁻⁶ s ⁻¹		16.0 for ε = 10 ⁻⁶ s ⁻¹
Vp (m s ⁻¹)	~3800 @ 273 K	3369 @ 273 K	3665 @ 273 K
Structural properties			
Cavity size (Å)	n/a	3.95	~ 3.91
Guest size (Å)	n/a	4.36	6.3
Saturation (mole/mole)	n/a	3 x 10 ⁻⁵	0.07
Saturation (mL / L H ₂ O)	n/a	3.5	1 x 10 ⁶
Stoichiometric ratio	n/a	CH ₄ ·5H ₂ O	C ₄ H ₈ O·17H ₂ O

SAND

Grain size	% Hydrate Pressure		Mechanical Large/intermediate strain							Thermal	Mechanical Low strain					Electrical	Distribution			
100	0%	0.01	T	T	T	T	T	T	T	T	O	O	O	O	O	O	O	C		
		0.50	T	T	T	T	T	T	T	T	O	O	O	O	O	O	O	C		
		1.00	T	T	T	T	T	T	T	T	O	O	O	O	O	O	O	C		
		2.00									O	O	O	O	O	O	O	C		
	50%	0.01	T	T	T	T	T	T	T	T	O	O	O	O	O	O	O	C	S	in
		0.50	T	T	T	T	T	T	T	T	O	O	O	O	O	O	O	C	S	in
		1.00	T	T	T	T	T	T	T	T	O	O	O	O	O	O	O	C	S	in
		2.00									O	O	O	O	O	O	O	C	S	in
	100%	0.01	T	T	T	T	T	T	T	T	O	O	O	O	O	O	O	C	S	in
		0.50	T	T	T	T	T	T	T	T	O	O	O	O	O	O	O	C	S	in
		1.00	T	T	T	T	T	T	T	T	O	O	O	O	O	O	O	C	S	in
		2.00									O	O	O	O	O	O	O	C	S	in
20	0%	0.01	T	T	T	T	T	T	T	T	O	O	O	O	O	O	O	C		
		0.50	T	T	T	T	T	T	T	T	O	O	O	O	O	O	O	C		
		1.00	T	T	T	T	T	T	T	T	O	O	O	O	O	O	O	C		
		2.00									O	O	O	O	O	O	O	C		
	50%	0.01	T	T	T	T	T	T	T	T	O	O	O	O	O	O	O	C	S	in
		0.50	T	T	T	T	T	T	T	T	O	O	O	O	O	O	O	C	S	in
		1.00	T	T	T	T	T	T	T	T	O	O	O	O	O	O	O	C	S	in
		2.00									O	O	O	O	O	O	O	C	S	in
	100%	0.01	T	T	T	T	T	T	T	T	O	O	O	O	O	O	O	C	S	in
		0.50	T	T	T	T	T	T	T	T	O	O	O	O	O	O	O	C	S	in
		1.00	T	T	T	T	T	T	T	T	O	O	O	O	O	O	O	C	S	in
		2.00									O	O	O	O	O	O	O	C	S	in
1	0%	0.01	T	T	T	T	T	T	T	T	O	O	O	O	O	O	O	C		
		0.50	T	T	T	T	T	T	T	T	O	O	O	O	O	O	O	C		
		1.00	T	T	T	T	T	T	T	T	O	O	O	O	O	O	O	C		
		2.00									O	O	O	O	O	O	O	C		
	50%	0.01	T	T	T	T	T	T	T	T	O	O	O	O	O	O	O	C	S	in
		0.50	T	T	T	T	T	T	T	T	O	O	O	O	O	O	O	C	S	in
		1.00	T	T	T	T	T	T	T	T	O	O	O	O	O	O	O	C	S	in
		2.00									O	O	O	O	O	O	O	C	S	in
	100%	0.01	T	T	T	T	T	T	T	T	O	O	O	O	O	O	O	C	S	in
		0.50	T	T	T	T	T	T	T	T	O	O	O	O	O	O	O	C	S	in
		1.00	T	T	T	T	T	T	T	T	O	O	O	O	O	O	O	C	S	in
		2.00									O	O	O	O	O	O	O	C	S	in

SILT

CLAY



Georgia Tech JIP Characterization Matrix

Defining Characteristics

Mechanical
(high/intermediate strains)

Thermal

Mechanical
(low strains)

Electrical

Distribution

Grain Size (microns)

Hydrate Concentration (%) - Target values

EFFECTIVE Confining Pressure (MPa)

Longitudinal and lateral stress-strain

Elastic-Plastic Transition

Tensile strength (indirect from Mohr-Coulomb intercept)

Shear strength

Compressive strength

Failure/stability envelopes (Mohr-Coulomb)

Bulk moduli (static)

Triaxial compaction coefficient

Young's modulus

Volume-Pressure compaction curves

Thermal Conductivity

Volume change during phase transformation

P-wave velocities

S-wave velocities

Bulk moduli (dynamic)

Electrical Resistivity

Real permittivity 200 MHz - 1.3 GHz (oedometer at low confinement)

Hydrate distribution (optical/visual--destructive of sample)

Pore filling vs. grain boundaries

Do not quote or use



Mechanical Properties

Small Strain

Large Strain

Deformation

at contacts

fabric changes

Stiffness

maximum

decreases

Losses

very low

large - frictional

Volume Change

minimal

potentially large

Diagenetic effects

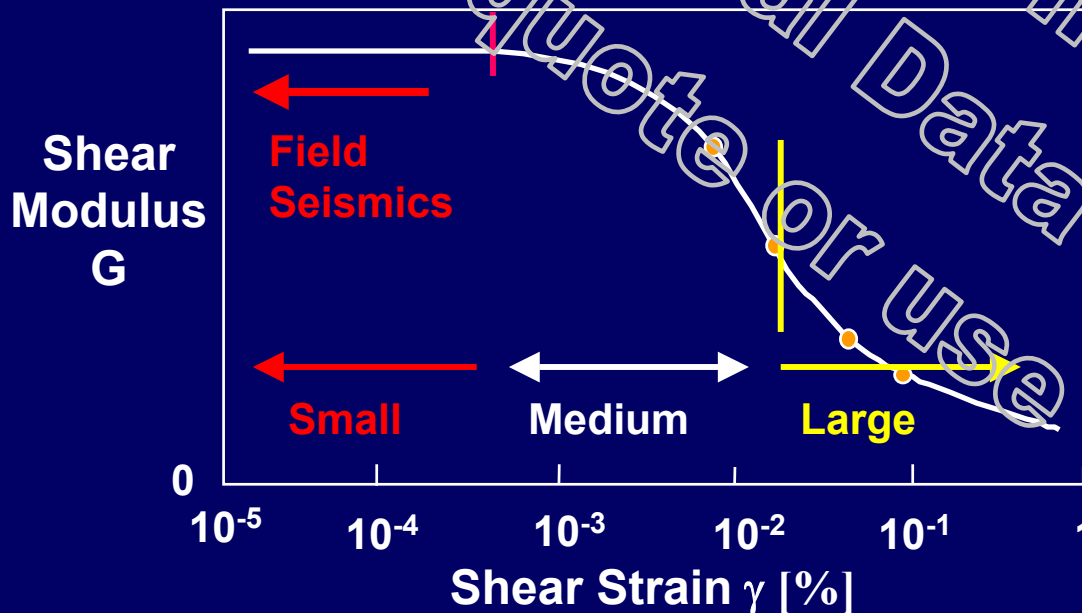
potentially high

small in drained shear

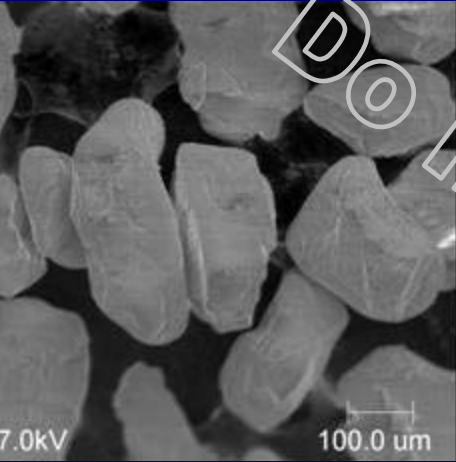
Fabric

constant

changes to critical state

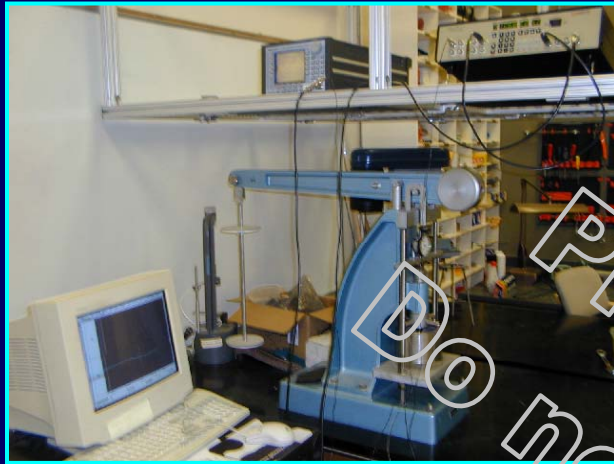


Representative Soils

<p>SAND</p>	<p>SILT (Silica Flour)</p>	<p>CLAY (Kaolinite)</p>
 <p>7.0kV 100.0 um</p>	 <p>15.0kV 500nm</p>	 <p>10.0kV 1.0 um</p>
<p>$D_{50} = 100 \mu\text{m}$ Specific surface = $< 0.1 \text{ m}^2/\text{g}$ Specific gravity = 2.65</p>	<p>$D_{50} = 20 \mu\text{m}$ Specific surface = $160 \text{ m}^2/\text{g}$ Specific gravity = 2.08</p>	<p>$D_{50} = 1.1 \mu\text{m}$ Specific surface = 10 to $20 \text{ m}^2/\text{g}$ Specific gravity = 2.6</p>

Properties: Ordering by both grain size and specific surface

Laboratory Devices



Oedometer ($\varepsilon_h = 0$)



Thermal properties



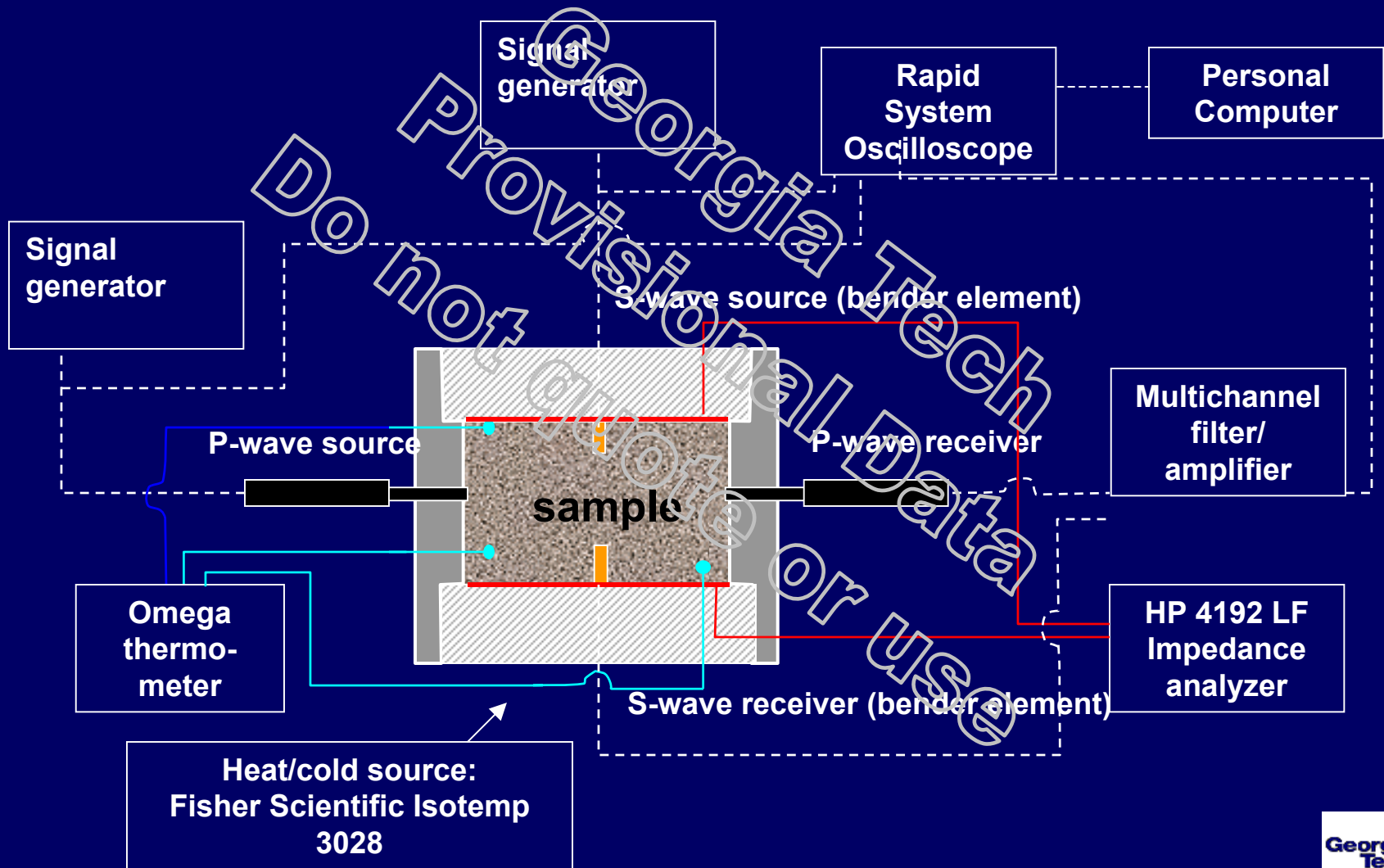
Triaxial test



High-pressure cell



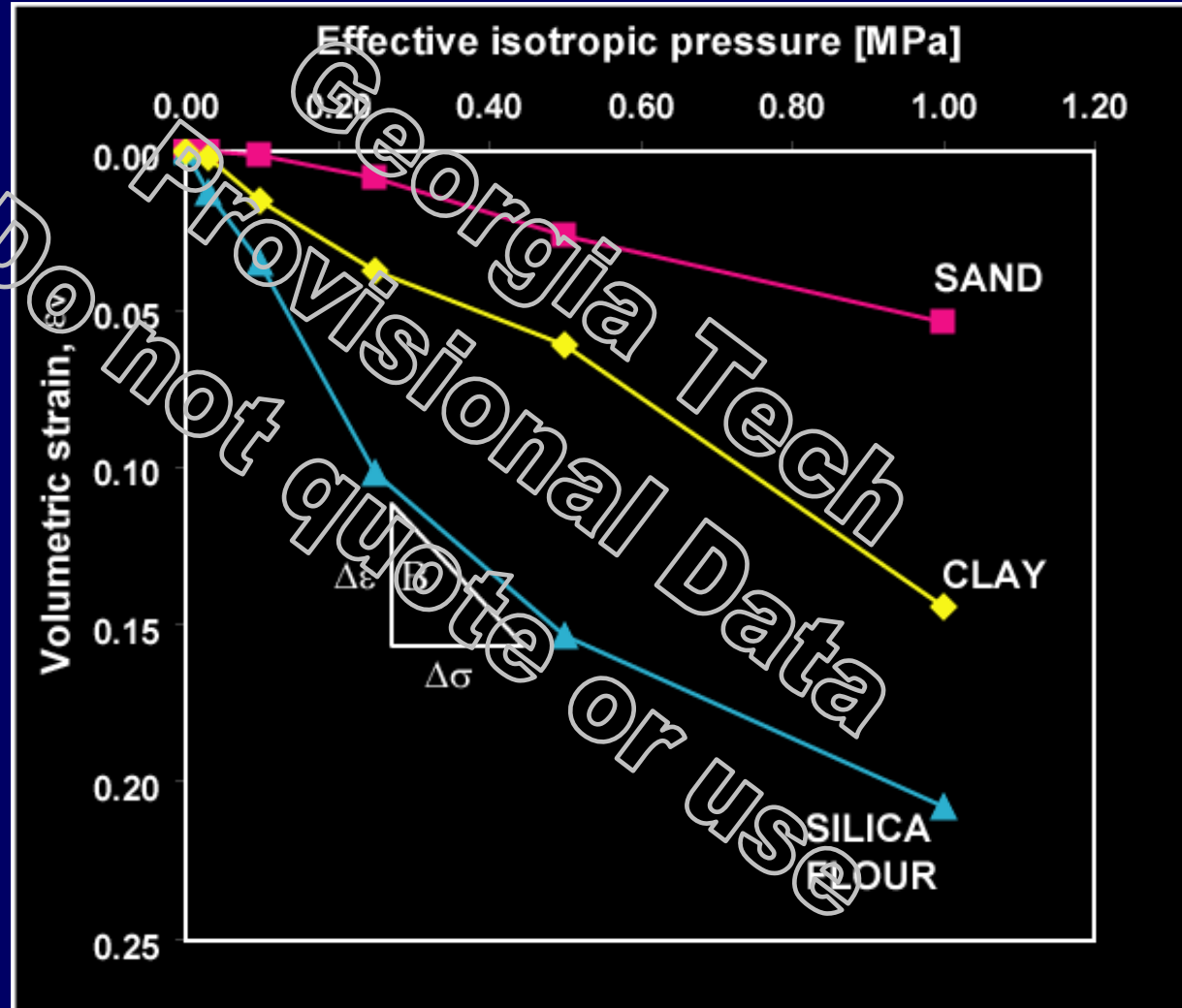
Oedometer cell and peripheral equipment



Mechanical Properties

Isotropic Loading

(bulk modulus B)

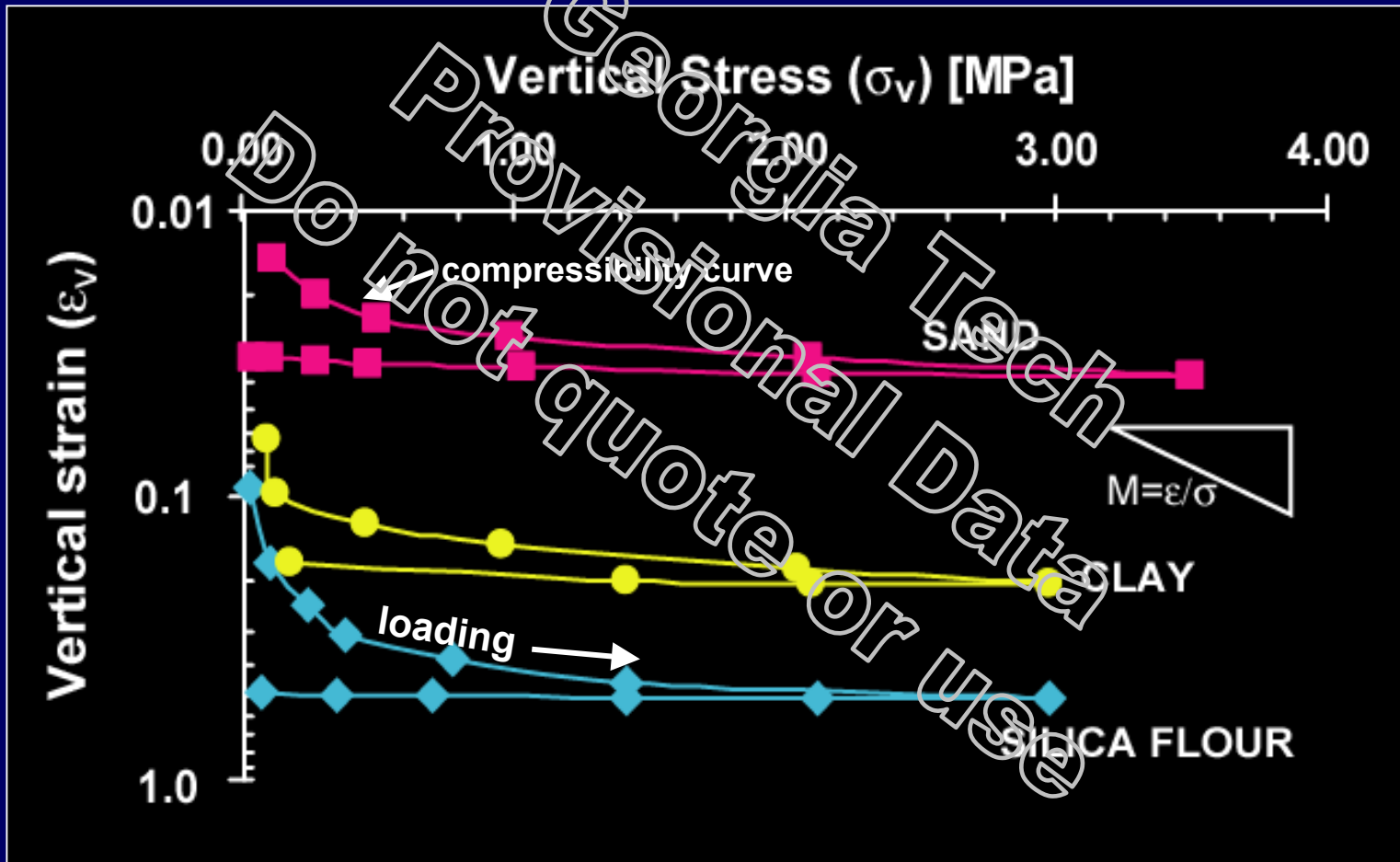


Constrains bulk moduli of soils after hydrate has formed & dissociated (100%)

Mechanical Properties

Zero-Lateral Strain Loading

(constrained modulus M)

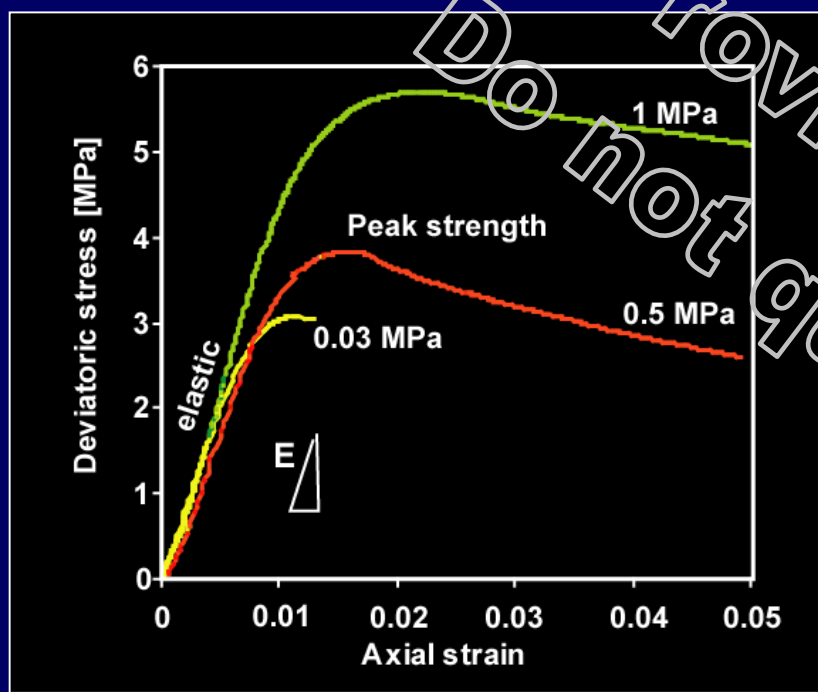


Differences in non-recoverable strain after formation/dissociation of hydrate

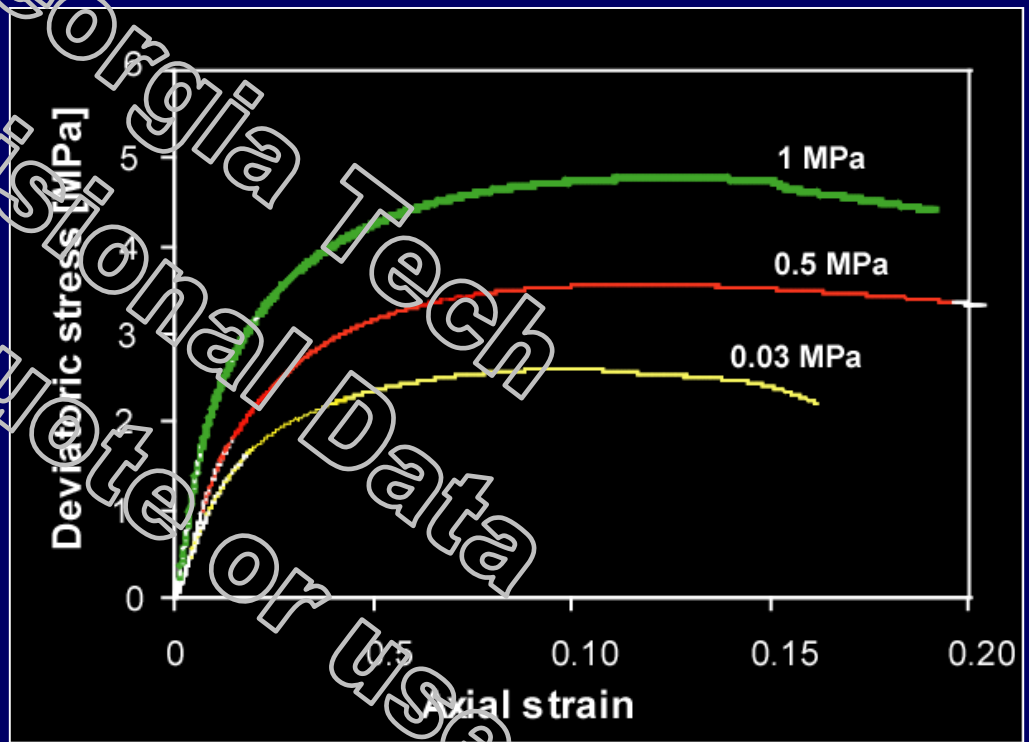
Mechanical Properties

Stress-Strain Path

(Young's modulus E)



Sand + 50% hydrate



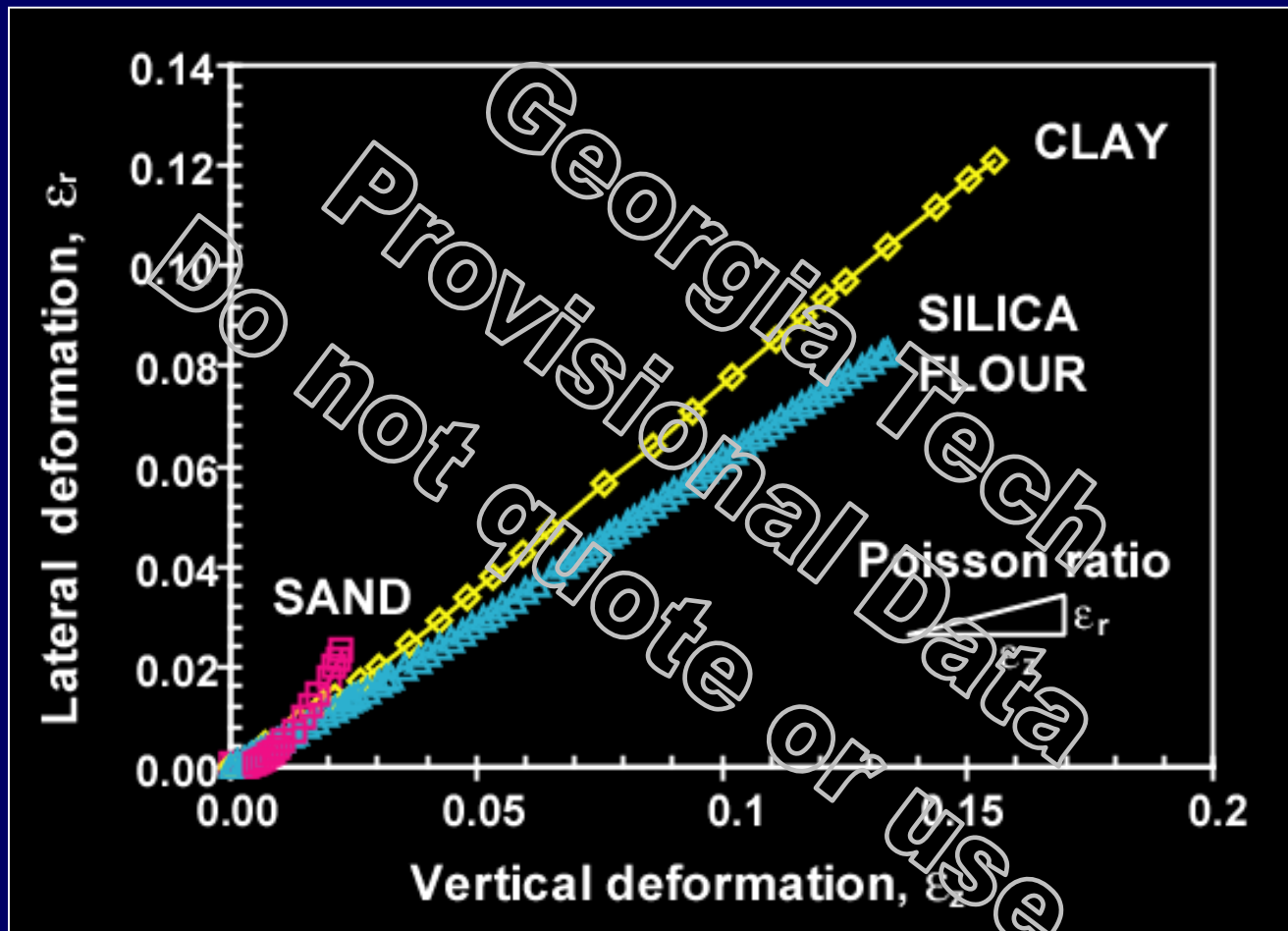
Silica flour + 50% hydrate

Undrained, elastoplastic deformation



Mechanical Properties

Poisson Ratio

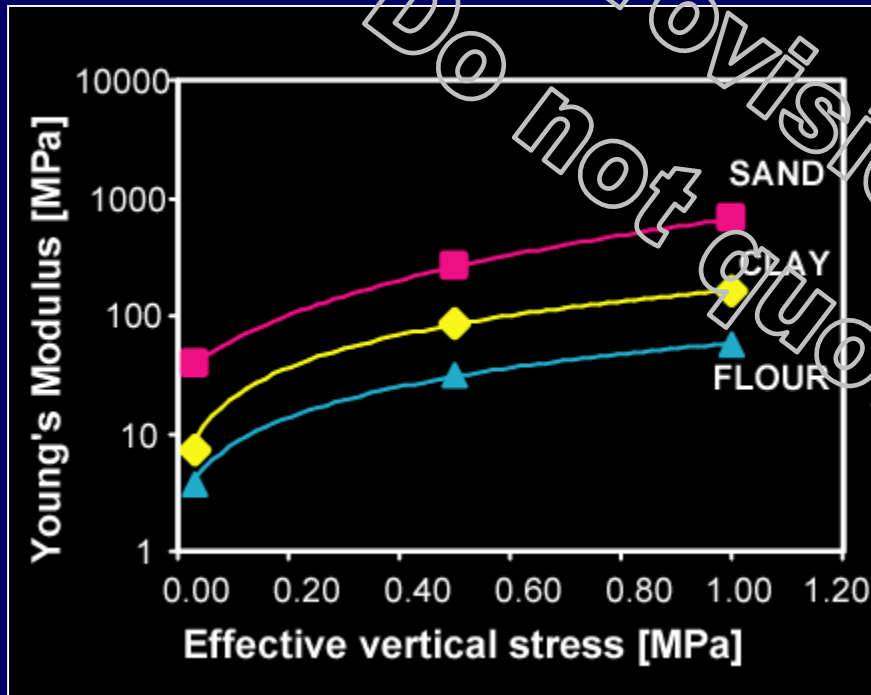


Poisson ratio greater than 0.5, as is common in dilatant soil

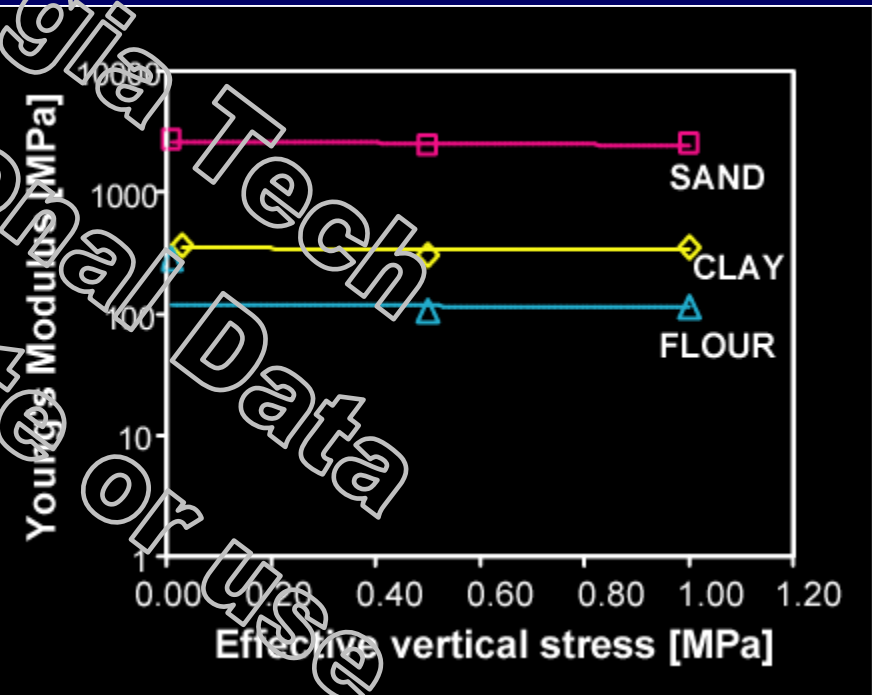
Mechanical Properties

Young's Modulus (E)

Sediment and Water



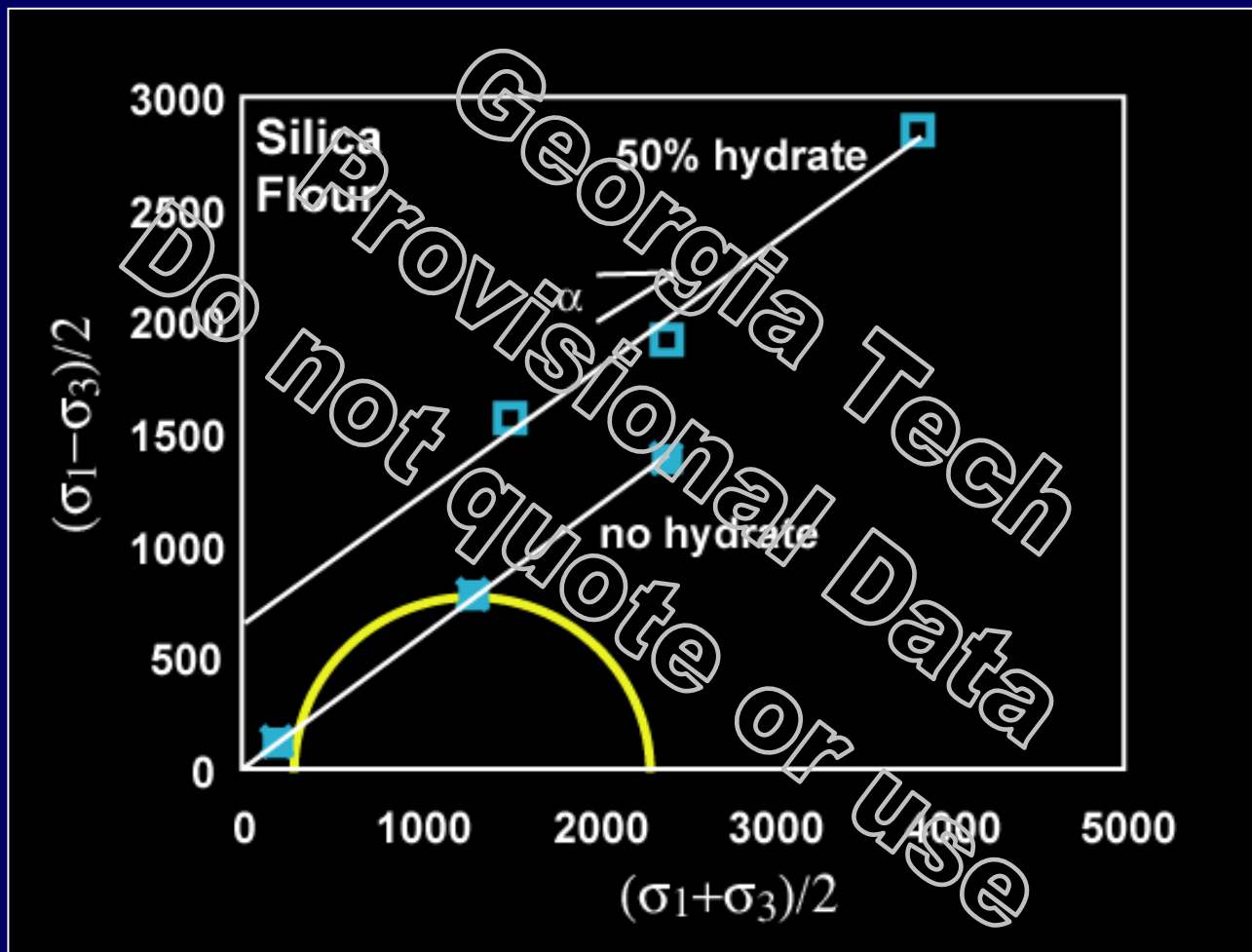
Sediment and 50% Hydrate



Once hydrate has formed, high-strain elastic properties remain constant

Mechanical Properties

Mohr-Coulomb failure criterion

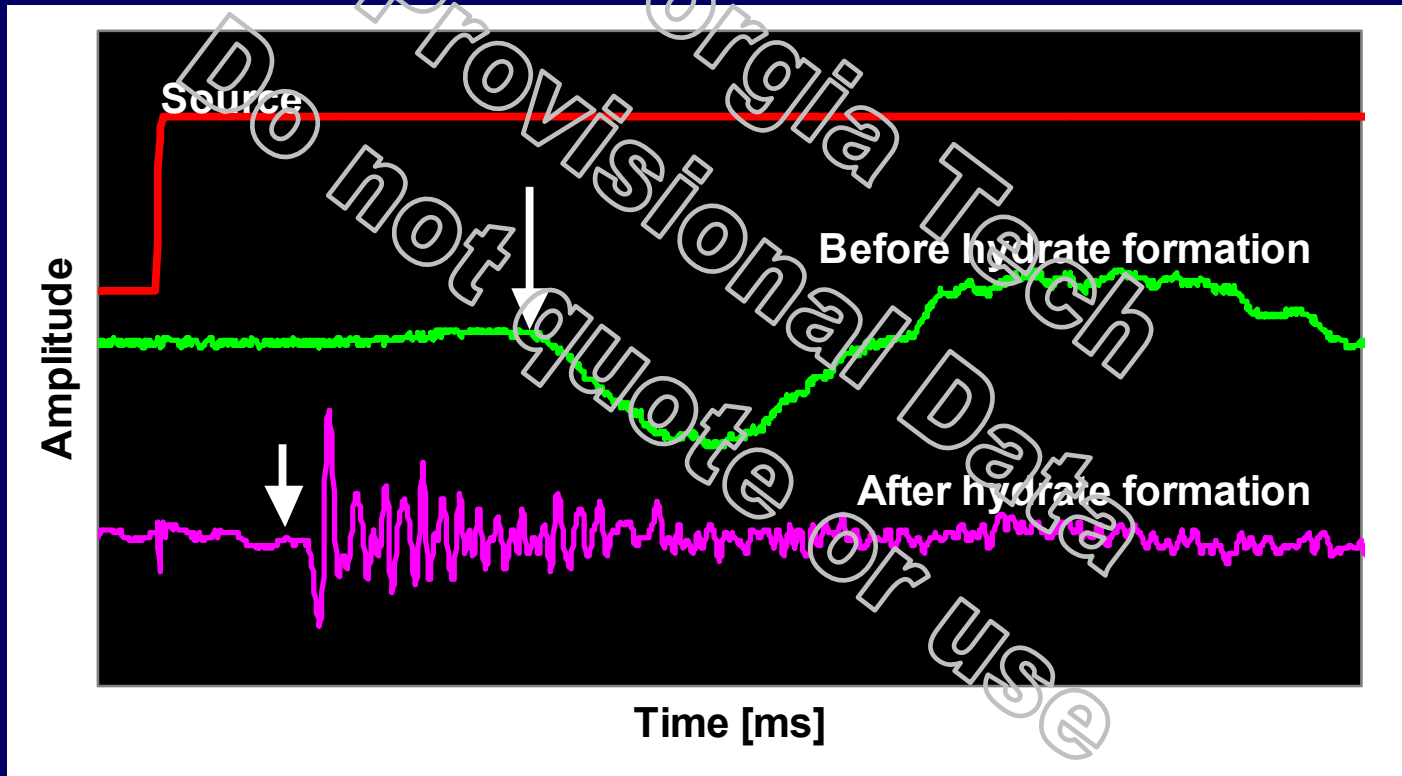


Hydrate affects cohesion and frictional properties

Mechanical Properties

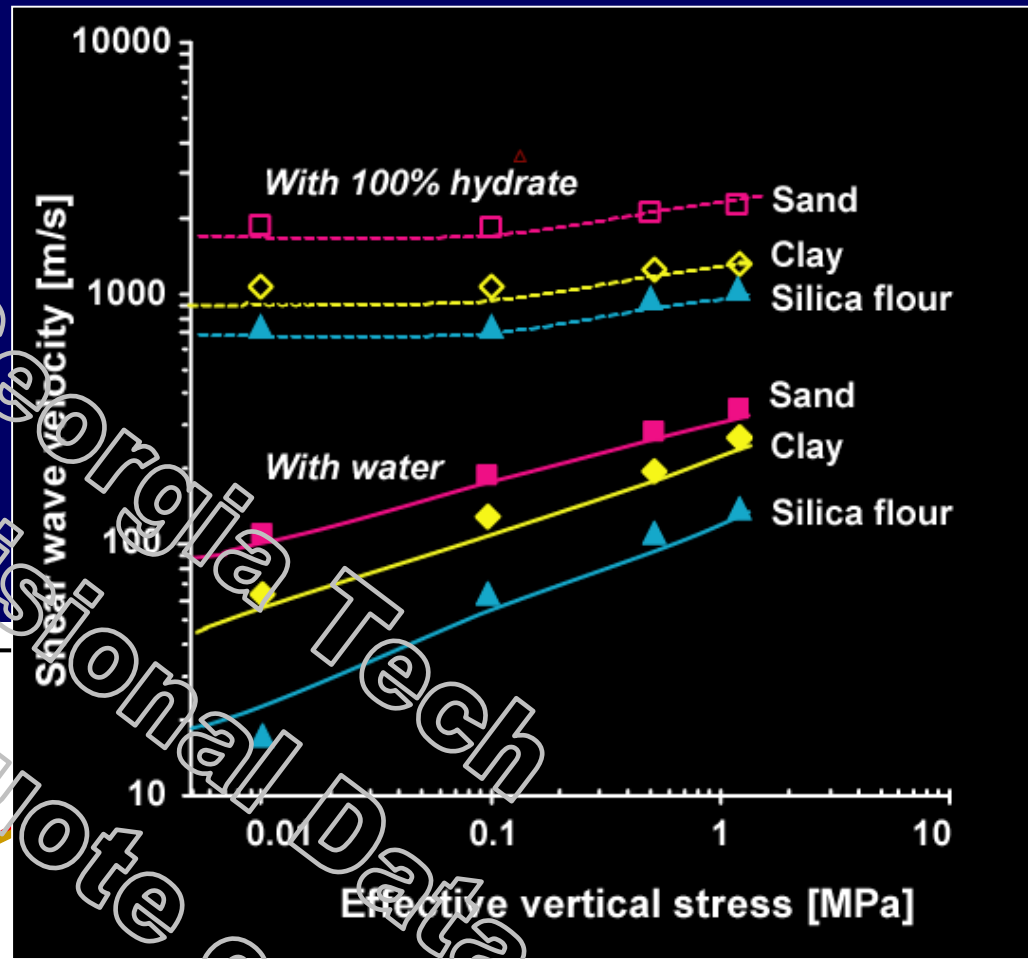
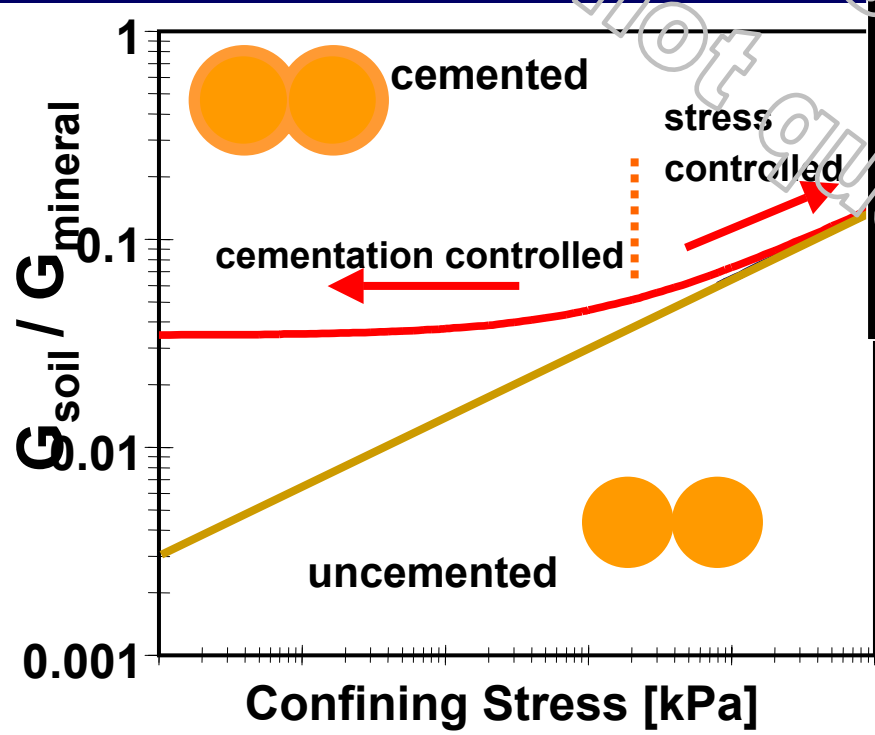
Small strain: deformation can be related to seismic properties and monitored by seismic waves

Typical Shear Waveform



Mechanical Properties

Shear Stiffness



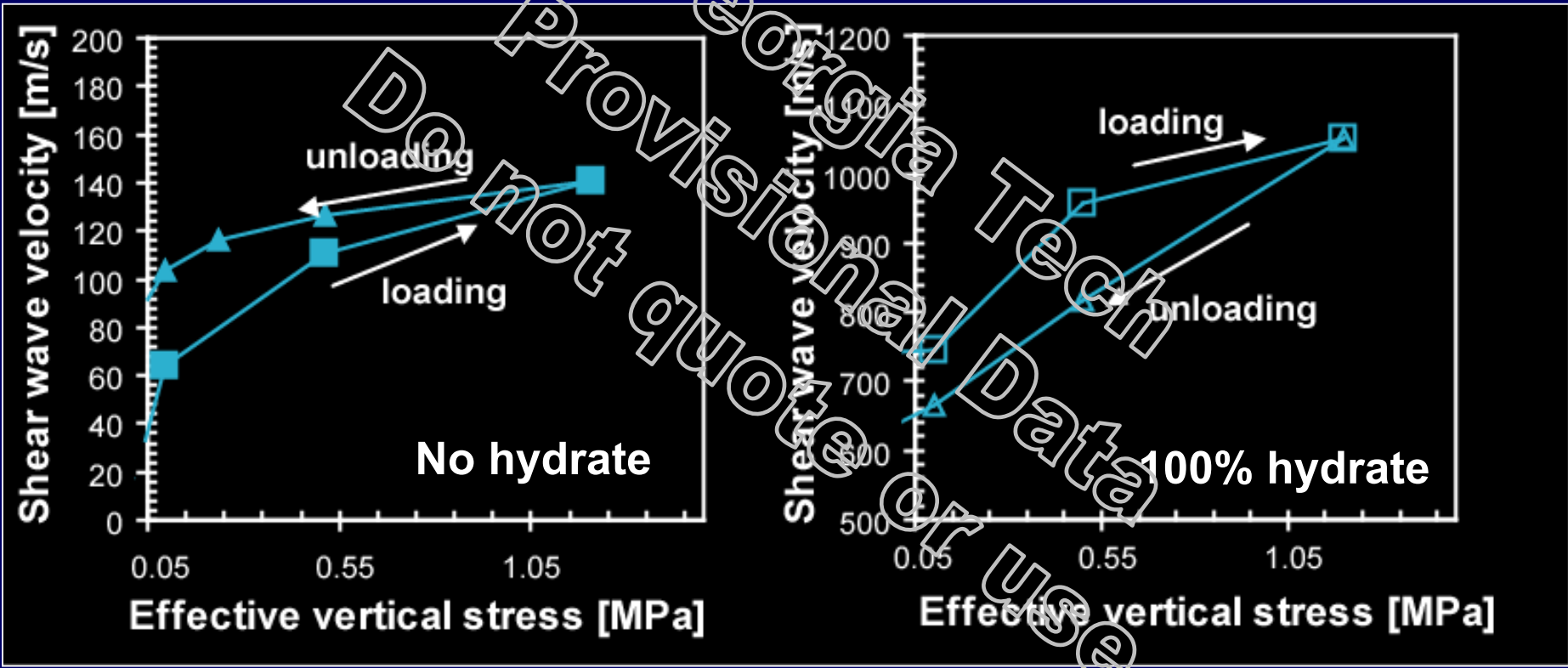
Do not Provisional Data or use

Mechanical Properties

Loading and Unloading Cycles

Relevant for seafloor sampling

Silica flour



Hydrate-bearing sediment behaves like a cemented soil

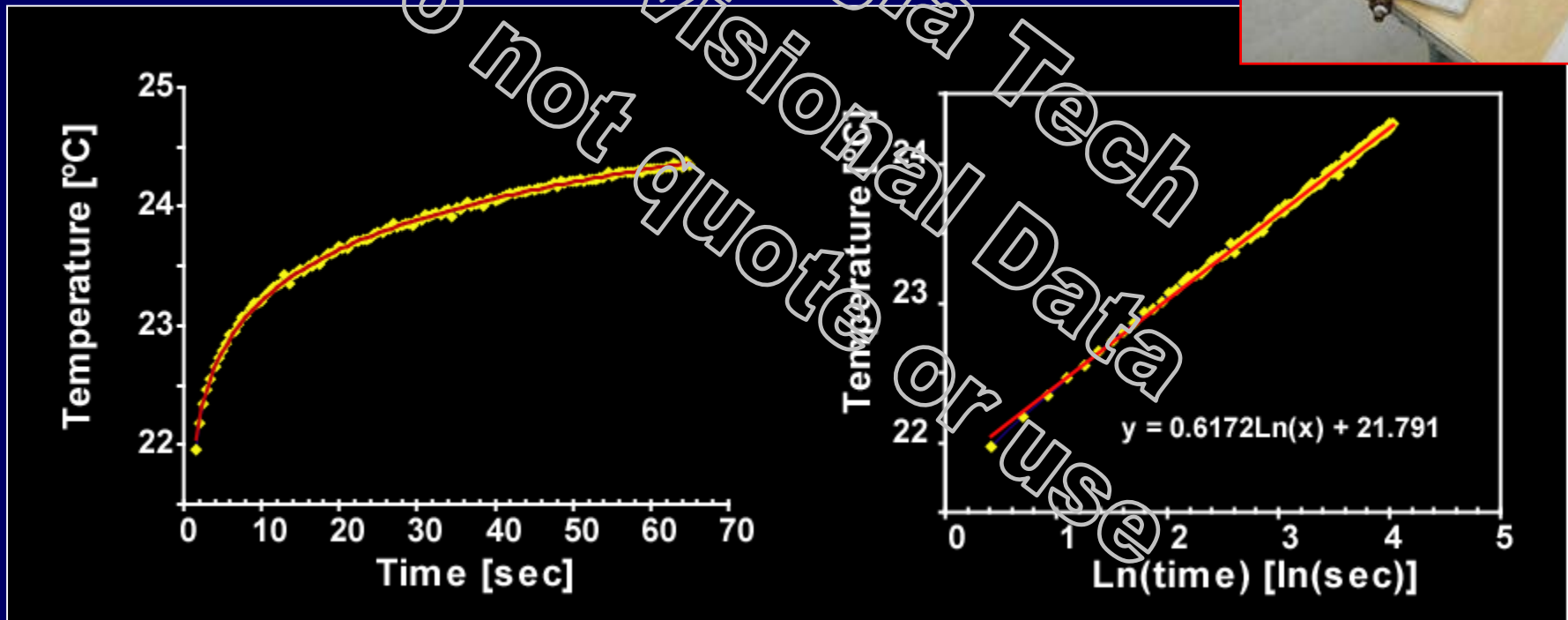


Thermal Properties

Thermal conductivity

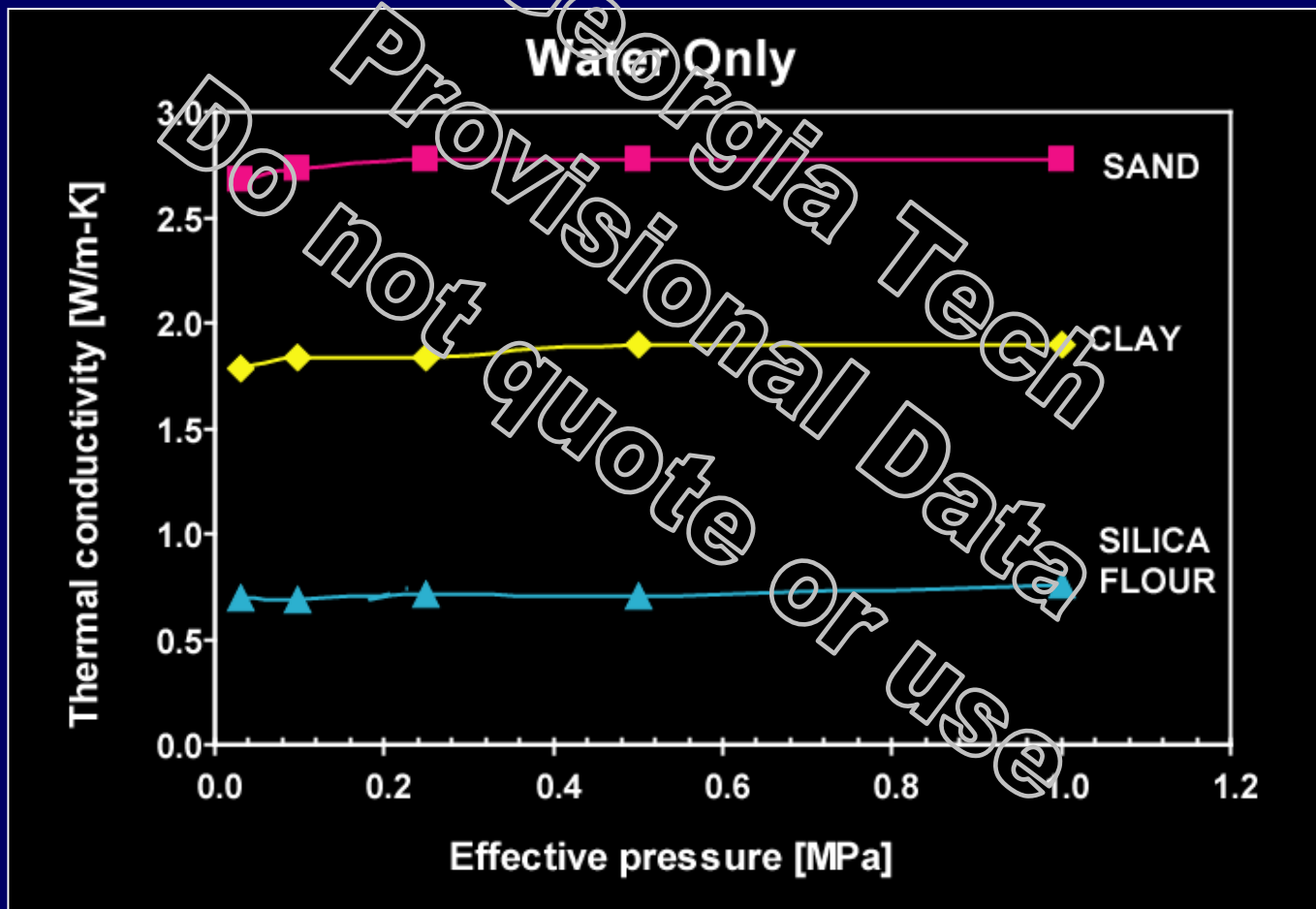


Raw data



Thermal Properties

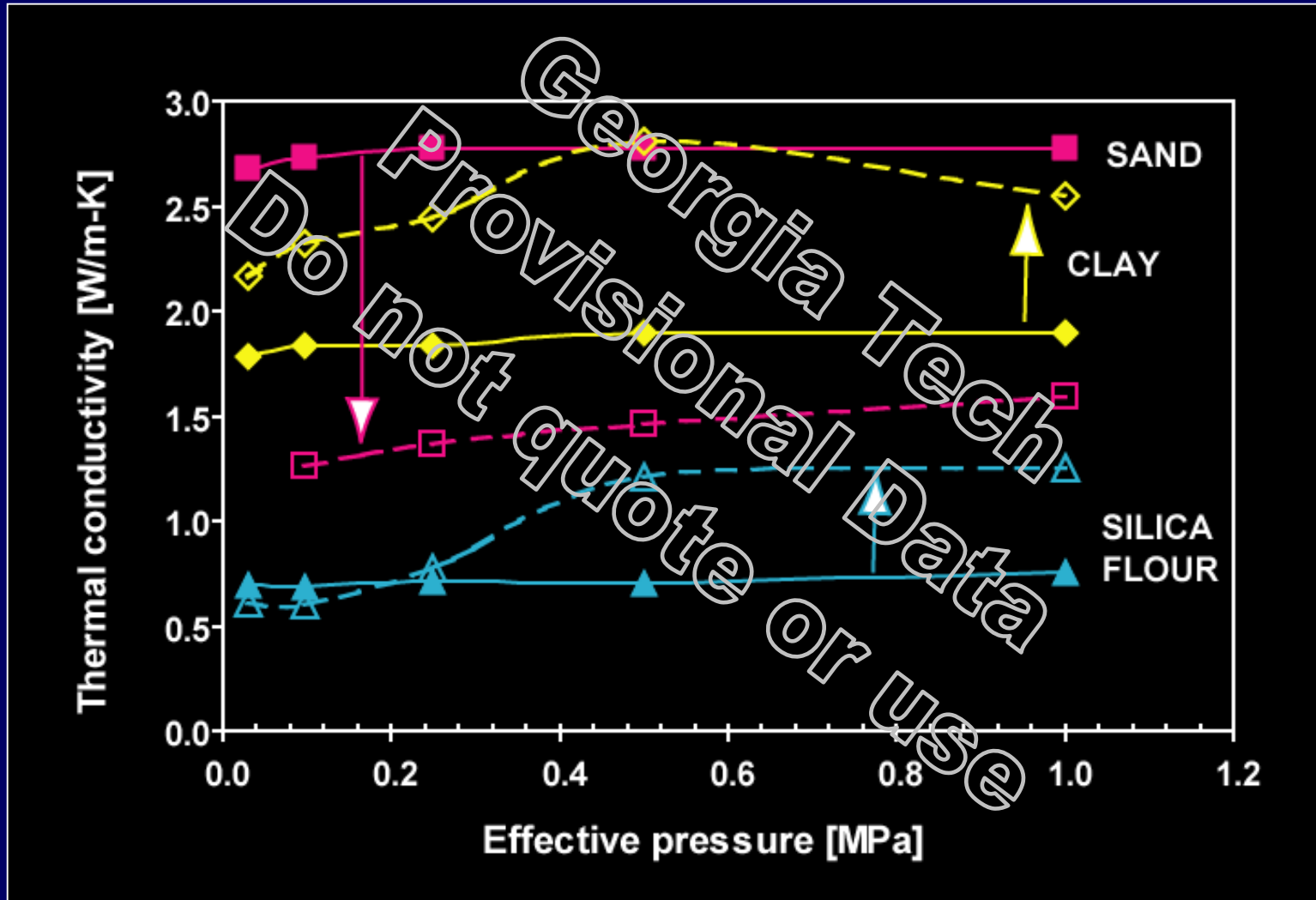
Thermal conductivity (baseline)



Thermal conductivity not dependent on pressure

Thermal Properties

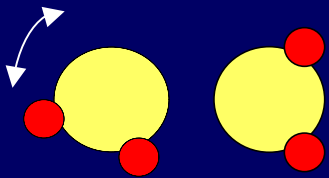
Thermal conductivity (with hydrate)



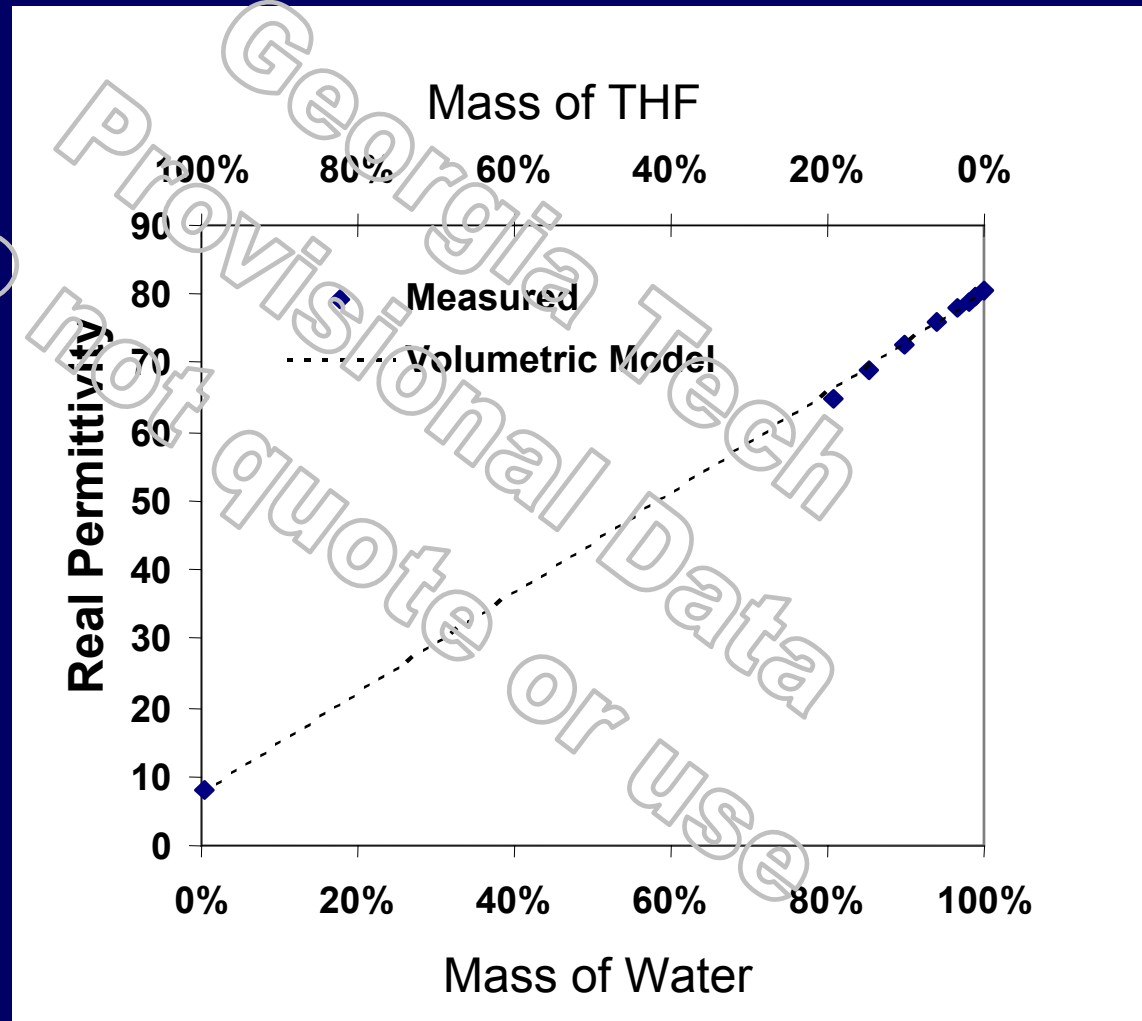
Electrical Properties

Real Permittivity Water-THF mixtures at 200MHz

ORIENTATIONAL



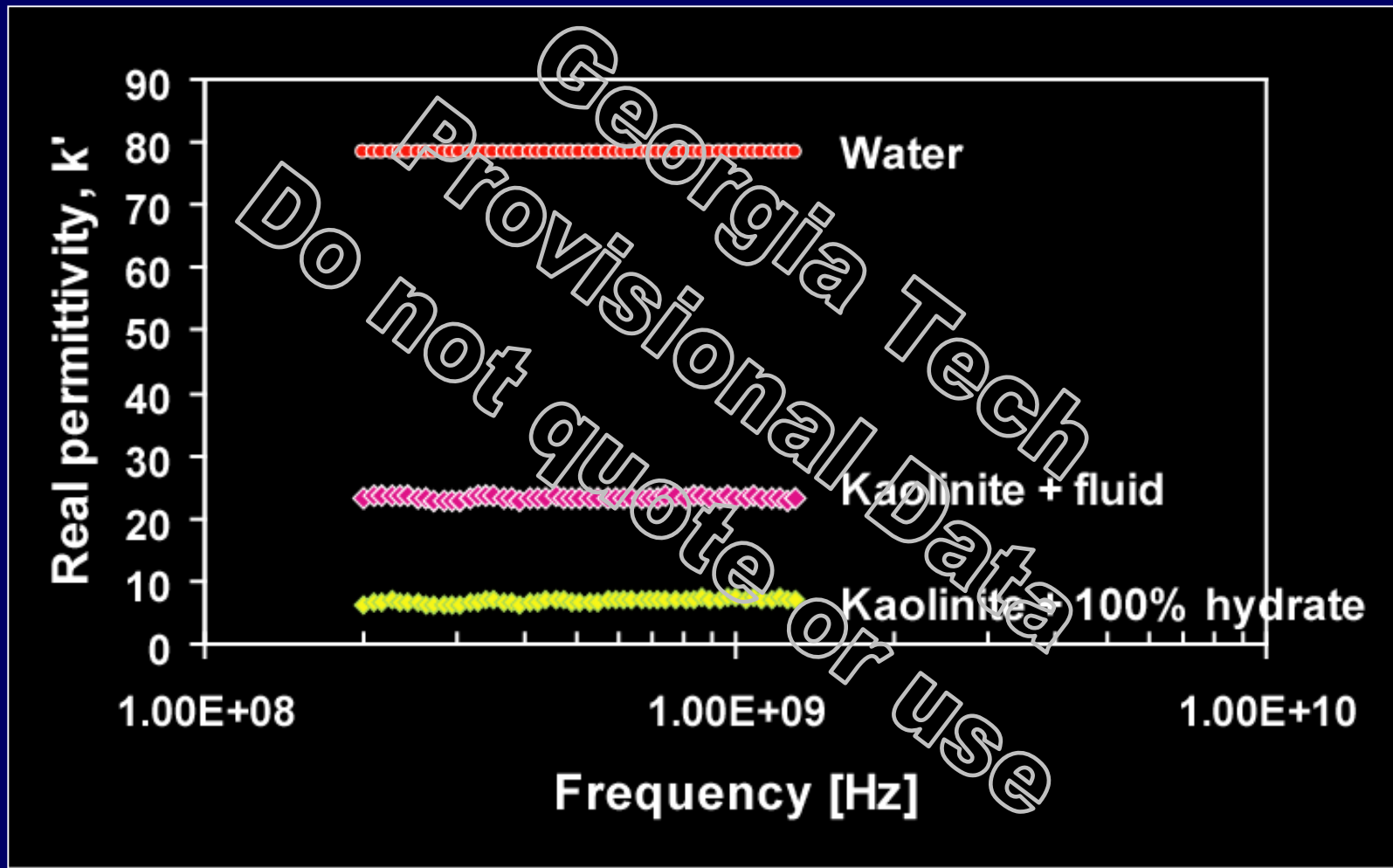
$t = 9 \times 10^{12} \text{ s}$
(Microwave-water)



Permittivity is a measure of the degree of polarizability of pore-filling fluid

Electrical Properties

Real Permittivity

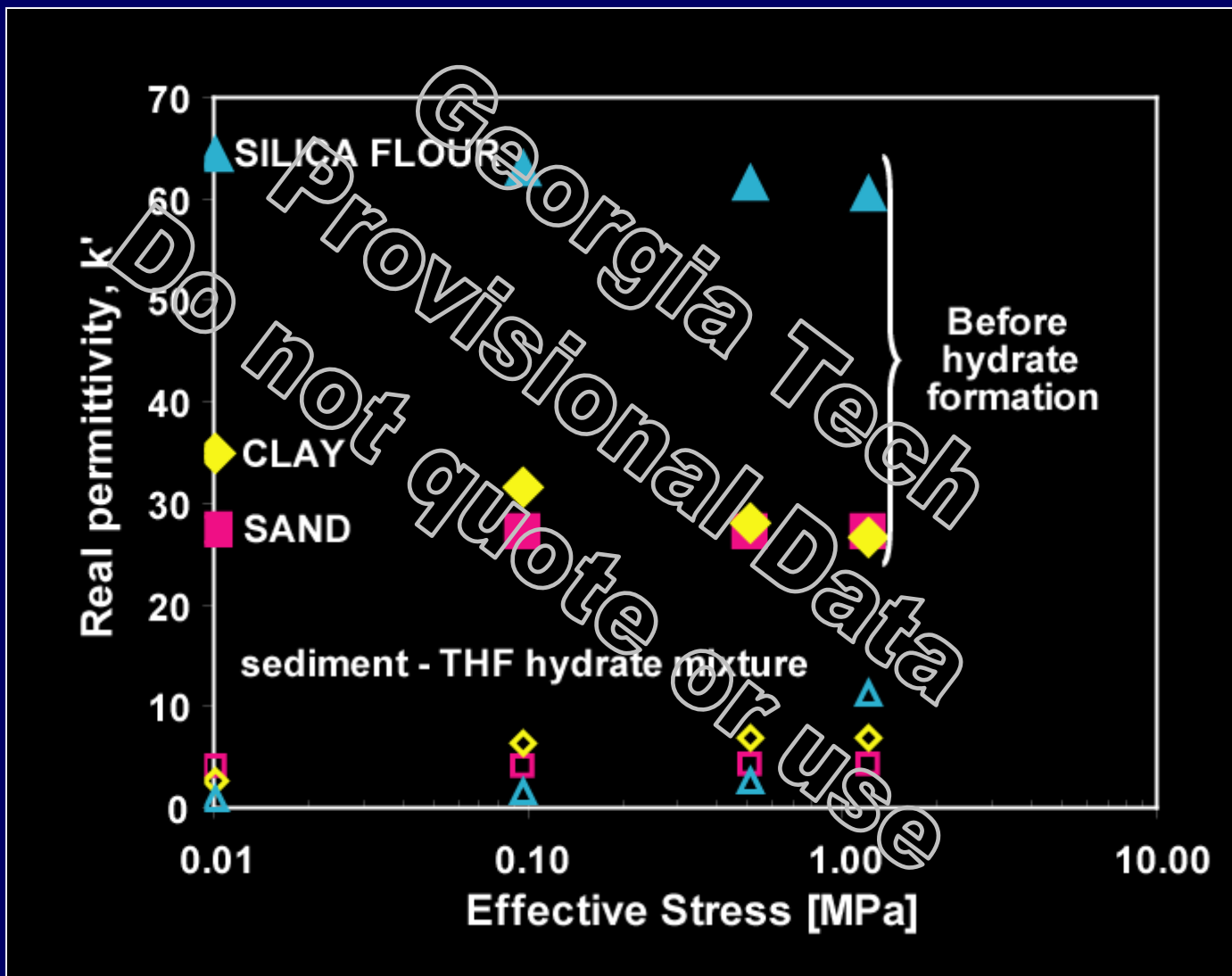


Permittivity is independent of frequency. Note impact of hydrate formation.



Electrical Properties

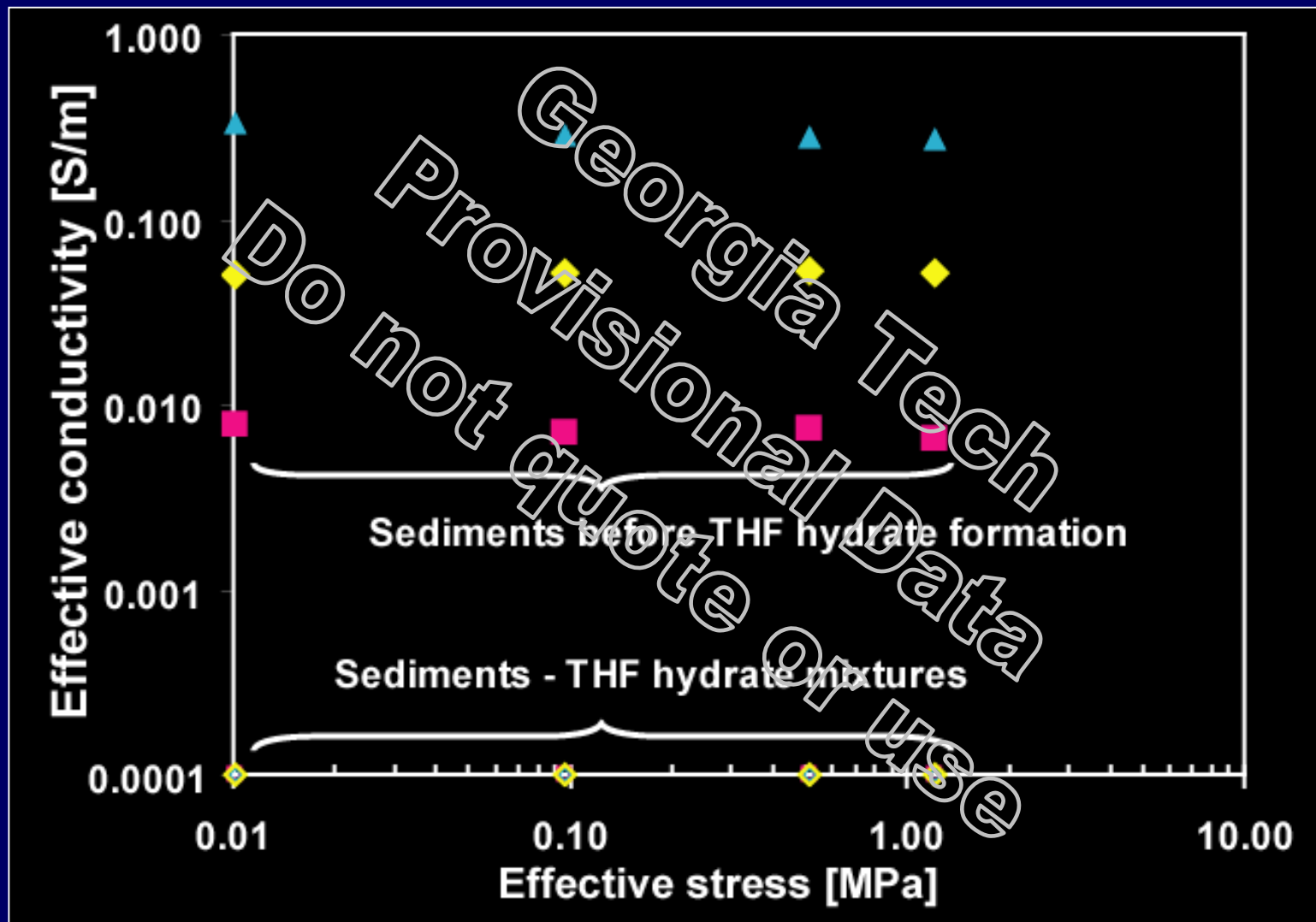
Real Permittivity



Hydrate formation sharply decreases dielectric permittivity

Electrical Properties

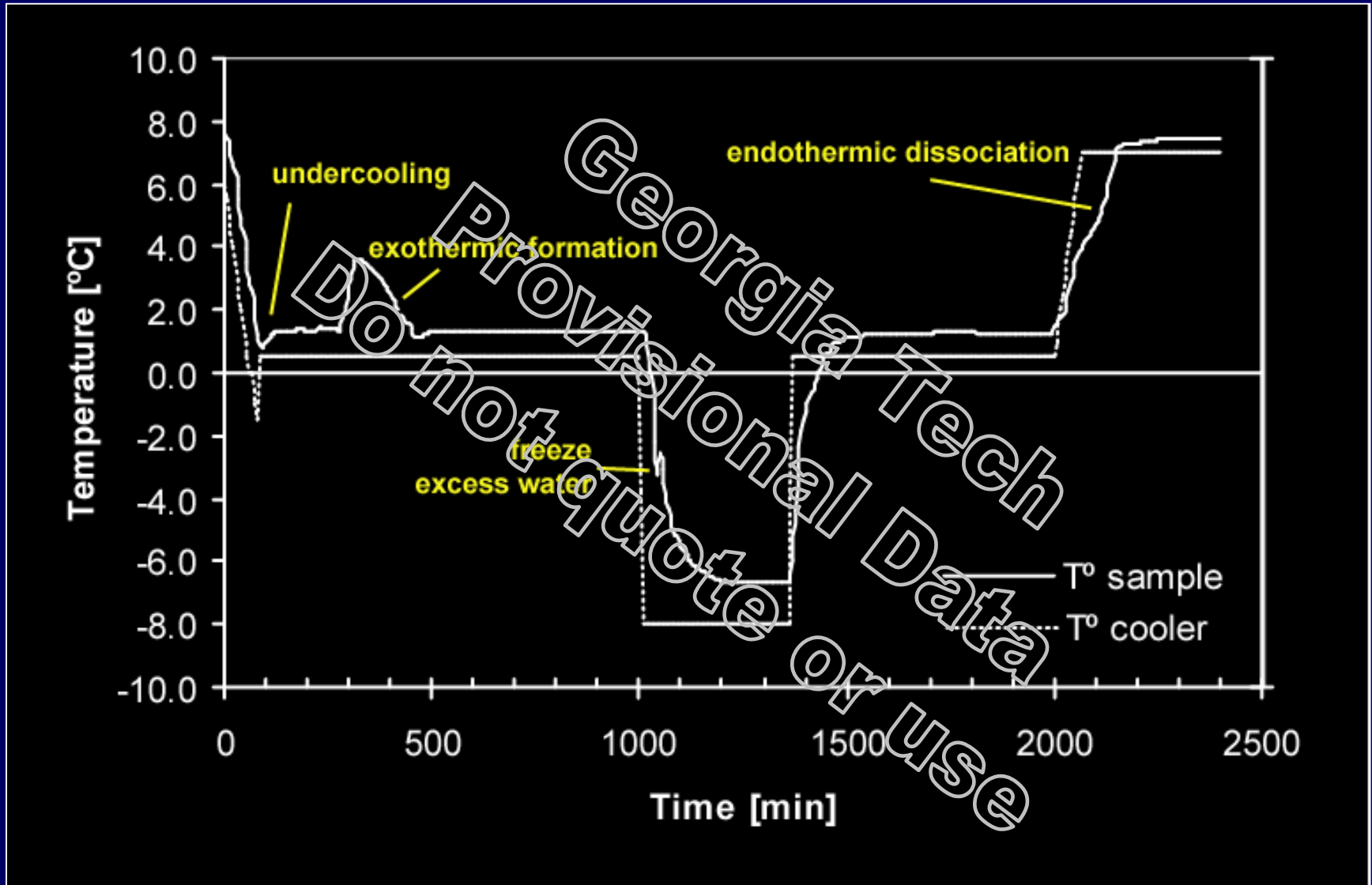
Conductivity



Hydrate formation lowers electrical conductivity to below detection limit



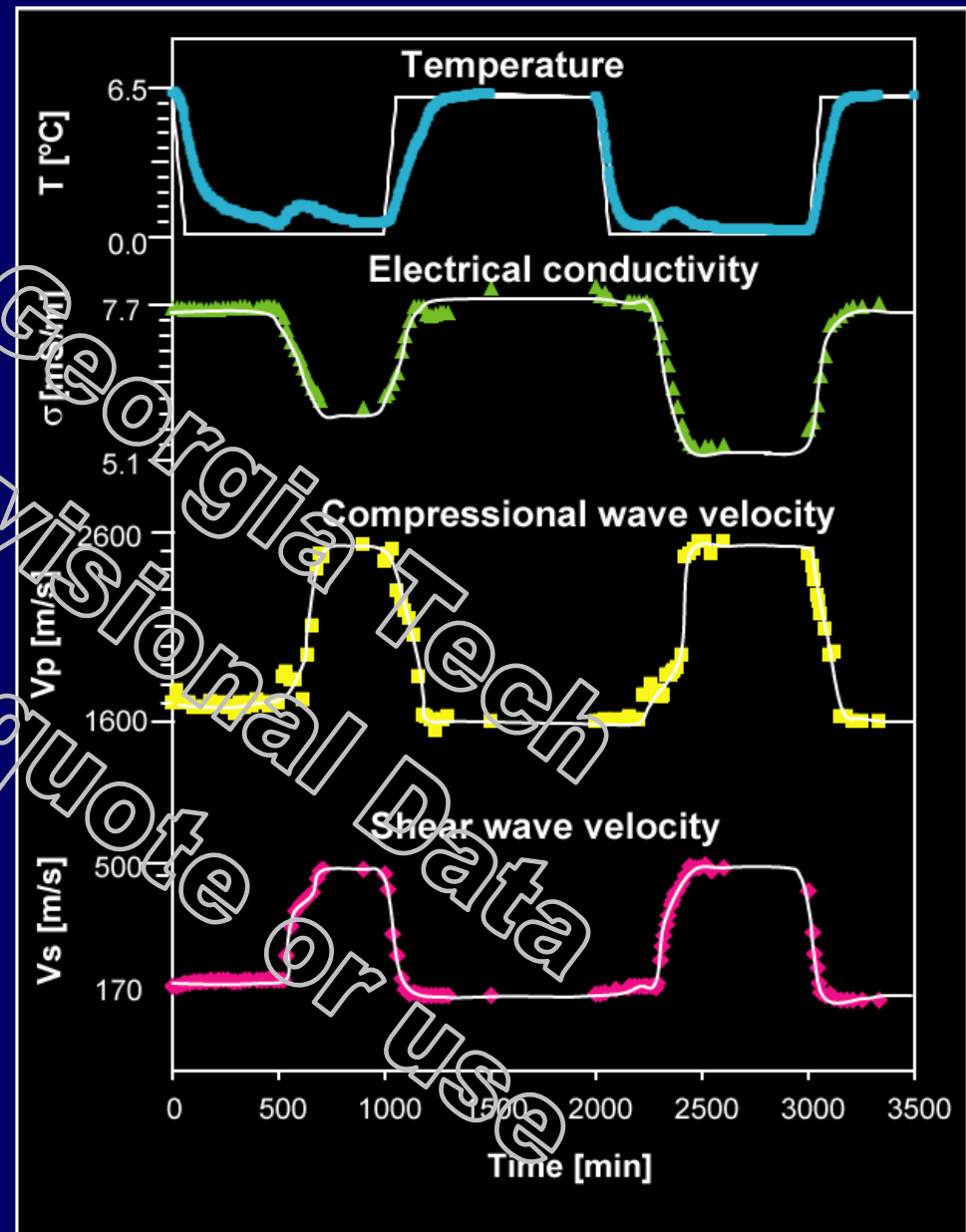
Phase Transformation Studies



DELETED SLIDES (LENSING)

Process Monitoring

Multi-parameter monitoring during phase transformations and multiple cycles



DELETED SLIDES



SAND

Grain size	% Hydrate Pressure		Mechanical Large/intermediate strain							Thermal	Mechanical Low strain					Electrical	Distribution			
			T	T	T	T	T	T	T	T	O	O	O	O	O	O	O	C	S	in
100	0%	0.01	T	T	T	T	T	T	T	T	T	O	O	O	O	O	O	C		
		0.50	T	T	T	T	T	T	T	T	T	O	O	O	O	O	O	C		
		1.00	T	T	T	T	T	T	T	T	T	O	O	O	O	O	O	C		
		2.00										O	O	O	O	O	O	C		
	50%	0.01	T	T	T	T	T	T	T	T	T	O	O	O	O	O	O	C	S	in
		0.50	T	T	T	T	T	T	T	T	T	O	O	O	O	O	O	C	S	in
		1.00	T	T	T	T	T	T	T	T	T	O	O	O	O	O	O	C	S	in
		2.00										O	O	O	O	O	O	C	S	in
	100%	0.01	T	T	T	T	T	T	T	T	T	O	O	O	O	O	O	C	S	in
		0.50	T	T	T	T	T	T	T	T	T	O	O	O	O	O	O	C	S	in
		1.00	T	T	T	T	T	T	T	T	T	O	O	O	O	O	O	C	S	in
		2.00										O	O	O	O	O	O	C	S	in
20	0%	0.01	T	T	T	T	T	T	T	T	T	O	O	O	O	O	O	C		
		0.50	T	T	T	T	T	T	T	T	T	O	O	O	O	O	O	C		
		1.00	T	T	T	T	T	T	T	T	T	O	O	O	O	O	O	C		
		2.00										O	O	O	O	O	O	C		
	50%	0.01	T	T	T	T	T	T	T	T	T	O	O	O	O	O	O	C	S	in
		0.50	T	T	T	T	T	T	T	T	T	O	O	O	O	O	O	C	S	in
		1.00	T	T	T	T	T	T	T	T	T	O	O	O	O	O	O	C	S	in
		2.00										O	O	O	O	O	O	C	S	in
	100%	0.01	T	T	T	T	T	T	T	T	T	O	O	O	O	O	O	C	S	in
		0.50	T	T	T	T	T	T	T	T	T	O	O	O	O	O	O	C	S	in
		1.00	T	T	T	T	T	T	T	T	T	O	O	O	O	O	O	C	S	in
		2.00										O	O	O	O	O	O	C	S	in
1	0%	0.01	T	T	T	T	T	T	T	T	T	O	O	O	O	O	O	C		
		0.50	T	T	T	T	T	T	T	T	T	O	O	O	O	O	O	C		
		1.00	T	T	T	T	T	T	T	T	T	O	O	O	O	O	O	C		
		2.00										O	O	O	O	O	O	C		
	50%	0.01	T	T	T	T	T	T	T	T	T	O	O	O	O	O	O	C	S	in
		0.50	T	T	T	T	T	T	T	T	T	O	O	O	O	O	O	C	S	in
		1.00	T	T	T	T	T	T	T	T	T	O	O	O	O	O	O	C	S	in
		2.00										O	O	O	O	O	O	C	S	in
	100%	0.01	T	T	T	T	T	T	T	T	T	O	O	O	O	O	O	C	S	in
		0.50	T	T	T	T	T	T	T	T	T	O	O	O	O	O	O	C	S	in
		1.00	T	T	T	T	T	T	T	T	T	O	O	O	O	O	O	C	S	in
		2.00										O	O	O	O	O	O	C	S	in

SILT

CLAY



SUMMARY

- Shown < 20% of results acquired to date
- Significant analysis required to place results in context
- **Processes**
 - Hydrate formation initiates at particle surfaces
 - Loading-unloading cycles reveal cementation behavior
 - P- and S-waves monitor phase transformation in lab and possibly during core recovery
 - Lensing under specific laboratory conditions
- **Properties**
 - **Mechanical**: Hydrate-bearing sediments behave like cemented soils (strength parameters, moduli, P- and S-wave) and hydrate impact is similar at both large and small strains
 - **Electrical**: Hydrate radically lowers both real permittivity and conductivity
 - **Thermal**: Hydrate increases pressure-dependence of conductivity