Proposed 1: 8/2003

ACUTE EXPOSURE GUIDELINE LEVELS (AEGLs) FOR NITRIC OXIDE

(CAS Reg. No. 10102-43-9)

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PREFACE

Under the authority of the Federal Advisory Committee Act (FACA) P. L. 92-463 of 1972, the National Advisory Committee for Acute Exposure Guideline Levels for Hazardous Substances (NAC/AEGL Committee) has been established to identify, review and interpret relevant toxicologic and other scientific data and develop AEGLs for high priority, acutely toxic chemicals.

AEGLs represent threshold exposure limits for the general public and are applicable to
emergency exposure periods ranging from 10 minutes to 8 hours. Three levels— AEGL-1,
AEGL-2 and AEGL-3— are developed for each of five exposure periods (10 and 30 minutes, 1
hour, 4 hours, and 8 hours) and are distinguished by varying degrees of severity of toxic effects.
The three AEGLs are defined as follows:

AEGL-1 is the airborne concentration (expressed as parts per million or milligrams per cubic meter [ppm or mg/m³]) of a substance above which it is predicted that the general population, including susceptible individuals, could experience notable discomfort, irritation, or certain asymptomatic, non-sensory effects. However, the effects are not disabling and are transient and reversible upon cessation of exposure.

AEGL-2 is the airborne concentration (expressed as ppm or mg/m³) of a substance above which it is predicted that the general population, including susceptible individuals, could experience irreversible or other serious, long-lasting adverse health effects or an impaired ability to escape.

AEGL-3 is the airborne concentration (expressed as ppm or mg/m³) of a substance above which it is predicted that the general population, including susceptible individuals, could experience life-threatening health effects or death.

30 Airborne concentrations below the AEGL-1 represent exposure levels that can produce 31 mild and progressively increasing but transient and nondisabling odor, taste, and sensory irritation, or certain asymptomatic, non-sensory effects. With increasing airborne concentrations 32 33 above each AEGL, there is a progressive increase in the likelihood of occurrence and the 34 severity of effects described for each corresponding AEGL. Although the AEGL values 35 represent threshold levels for the general public, including susceptible subpopulations, such as infants, children, the elderly, persons with asthma, and those with other illnesses, it is recognized 36 37 that individuals, subject to unique or idiosyncratic responses, could experience the effects 38 described at concentrations below the corresponding AEGL.

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SUMMARY

Nitric oxide (NO) is an endogenous molecule that mediates the biological action of endothelium-derived relaxing factor. The chemical is an important regulator of the functions of the cardiovascular, immune, and nervous systems. The toxicity of NO is associated with both methemoglobin formation and oxidation to nitrogen dioxide. Inhaled NO has been used therapeutically to treat adult respiratory distress syndrome, persistent pulmonary hypertension of the newborn, pulmonary hypertension in congenital heart disease and diaphragmatic hernia, pulmonary hypertension following thoracic organ transplantation, idiopathic pulmonary hypertension, and chronic obstructive pulmonary disease. Nitric oxide is also a component of air 10 pollution and is generally measured as part of the total oxides of nitrogen (NO + NO₂) present.

13 Data were insufficient for derivation of AEGL values. Generally, therapeutic levels of 14 20-80 ppm for 24 hours or 100 ppm for 20 minutes have not resulted in adverse effects among 15 treated patients. Methemoglobin levels increased to 9.4-9.6% in lung transplantation or 16 pulmonary hypertension patients after treatment with 80 ppm NO for up to 108 hours (Adatia et 17 al. 1994, Wessel et al. 1994). In all cases, a reduction in NO concentration resulted in a 18 reduction of methemoglobin levels. Methemoglobinemia >7% occurred in 13/37 newborns 19 treated with 80 ppm NO for persistent pulmonary hypertension. The average time to peak level 20 in all patients was 19.6 hours and the highest level of methemoglobin was 11.9% at 8 hours in 21 one patient (Davidson et al. 1998).

23 Most of the experimental animal studies available focused on the therapeutic use of NO 24 in an animal model of human disease. Lethality studies in dogs, rats, and mice lacked complete 25 concentration-response information, were confounded by possible NO₂ contamination, or were secondary citations in which the original source could not be obtained. From these studies, 26 however, it appears that in the absence of lung injury, the mechanism of toxicity of NO is 27 28 methemoglobin formation.

30 NO is unstable in air and undergoes spontaneous oxidation to NO₂ making experimental 31 effects difficult to separate and studies difficult to perform (U.S. EPA 1993). Studies on the 32 conversion of NO to NO₂ in medicinal applications have found the conversion to be significant 33 in an atmospheric concentration of O_2 (20.9%) at room temperatures. While on the laboratory scale, closed system experiments clearly indicate the potential for the production of NO₂, the 34 chemical kinetics of NO conversion during a large-scale atmospheric release and dispersion are 35 36 not well documented. As a result, the conversion of NO to NO₂ during the atmospheric release 37 is of concern to emergency planners, but the NAC is unable to provide any significant guidance, 38 other than to indicate that a significant fraction of the NO will be converted to NO₂ and present 39 an atmospheric hazard that is generally more toxic than just NO.

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41 Therefore, AEGL values for NO are not recommended. Because conversion to NO₂ is expected to occur in the atmosphere, and because NO₂ is more toxic than NO, the AEGL values 42

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for NO₂ are recommended for use with emergency planning for NO. The NAC recognizes, however, that short-term exposures below 80 ppm NO should not constitute a health hazard.

	Summary of AEGL Values for Nitric Oxide ^a							
Classification	10-Minute	30-Minute	1-Hour	4-Hour	8-Hour	Endpoint (Reference)		
AEGL–1 ^b (Non- disabling)	NR	NR	NR	NR	NR			
AEGL–2 ^b (Disabling)	NR	NR	NR	NR	NR			
AEGL-3 ^b (Lethal)	NR	NR	NR	NR	NR			

NR, not recommended

^aAEGL values for nitrogen dioxide should be used for emergency planning.

^bShort-term exposures to below 80 ppm NO should not constitute a health hazard.

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 inhaled nitric oxide in patients with pulmonary hypertension. Crit. Care Med. 22:930 938.

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1. INTRODUCTION

3 Nitric oxide (NO) is an endogenous molecule that mediates the biological action of 4 endothelium-derived relaxing factor. The chemical is an important regulator of the functions of the cardiovascular, immune, and nervous systems (Moncada et al. 1991; Rossaint et al. 1996). 5 One of the most important functions of NO is relaxation of the vascular smooth muscles. 6 7 Because of this action, inhaled NO has been used to treat adult respiratory distress syndrome, 8 persistent pulmonary hypertension of the newborn, pulmonary hypertension in congenital heart 9 disease and diaphragmatic hernia, pulmonary hypertension following thoracic organ transplantation, idiopathic pulmonary hypertension, and chronic obstructive pulmonary disease 10 (Troncy et al. 1997a). 11

13 The major mechanism of toxicity of NO is the binding of hemoglobin (U.S. EPA 1993b). 14 Inhaled NO is absorbed into the bloodstream and binds to hemoglobin forming 15 nitrosylhemoglobin which is rapidly oxidized to methemoglobin (Maeda et al. 1987; Sharrock et al. 1984; U.S. EPA 1993b). The affinity of NO for hemoglobin is about 1500 times greater than 16 that of carbon monoxide (Gibson and Roughton 1957) and the binding and formation of 17 18 methemoglobin is NO concentration- and time-dependent (Maeda et al. 1987; Ripple et al. 1989; 19 Sharrock et al. 1984). Potentiating possible toxicity, NO reacts quantitatively with oxygen in air 20 to form nitrogen dioxide which causes pulmonary edema: $2NO + O_2 \rightarrow 2NO_2$. Nitrogen dioxide then reacts with water to form nitric acid (NIOSH 1976). For this reason, careful monitoring of 21 22 NO₂ concentrations has been suggested when NO is used therapeutically at concentrations ≥ 80 23 ppm especially when coadministered with oxygen (Miller et al. 1994; Foubert et al. 1992).

Unintended exposure during the therapeutic use of NO may also occur. During inhalation therapy, environmental levels of NO rose to a maximum of 0.462 ppm in a nonventilated room without exhaust, but with normal air conditioning, the levels were 0.075 ppm (Markhorst et al. 1996). Hospital compressed air tanks were found to be contaminated with 5-8 ppm NO on days that corresponded to welding activity near the intake port; otherwise, tank levels were similar to ambient levels (Pinsky et al. 1997).

Nitric oxide is also a component of air pollution and is generally measured as part of the total oxides of nitrogen (NO + NO₂) present. Anthropogenic sources of nitrogen oxides include automobiles, electric utilities, industrial boilers, gas stoves, space heaters, kerosene heaters, wood stoves, and tobacco products (U.S. EPA 1993b). Cigarette smoke contains 400-1000 ppm NO (Fullerton and McIntyre 1996). In the atmosphere, NO reacts with ozone to produce NO₂ and oxygen (U.S. EPA 1993b).

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Selected physicochemical properties of nitric oxide are listed in Table 1.

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TABLE 1. PHYSICOCHEMICAL DATA FOR NITRIC OXIDE				
Parameter	Value	Reference		
Common name	nitric oxide			
Synonyms	nitrogen monoxide	Budavari et al., 1996		
CAS registry no.	10102-43-9			
Chemical formula	NO	Budavari et al., 1996		
Molecular weight	30.01	Budavari et al., 1996		
Physical state	colorless gas	Budavari et al., 1996		
Vapor pressure	26,000 torr at 20°C	ACGIH, 1991		
Vapor density (air = 1)	1.04	Budavari et al., 1996		
Melting/boiling point	-163.6°C/-151.7°C	Budavari et al., 1996		
Solubility in water	4.6 mL/100 mL (20°C) 0.006 g/100 g (24°C)	Budavari et al., 1996 U.S. EPA, 1993a		
Conversion factors in air	1 ppm = 1.25 mg/m^3 1 mg/m ³ = 0.8 ppm	NIOSH, 1976		
Reactivity	combines with O_2 to form NO_2	Budavari et al., 1996		

2. HUMAN TOXICITY DATA

2.1. Acute Lethality

Following induction of anesthesia with nitrous oxide and oxygen, a woman became cyanotic within 2 minutes. Treatment with methylene blue reversed the methemoglobinemia, but she developed severe pulmonary edema several hours later and died of cardiac arrest. A second patient also became cyanotic after initiation of anesthesia and the nitrous oxide was discontinued immediately. Several hours later the second patient developed some respiratory distress but recovered completely after oxygen and steroid therapy. It was determined that the nitrous oxide cylinder had been contaminated with NO (Clutton-Brock 1967). The possible exposure concentration was not determined nor was the contribution of the formation of nitrogen dioxide addressed by the study author. Greenbaum et al. (1967) made several assumptions about retention volume, time to cyanosis, and ventilation rate and estimated that the contamination must have been 1% (10,000 ppm) NO or higher.

2.2. Nonlethal Toxicity

2.2.1. Case Reports

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4 Methemoglobin levels rose to 9.4% in one lung transplantation patient after treatment 5 with 80 ppm NO for 8 hours. A reduction in concentration to 40 ppm over 4 hours resulted in a decrease to 6.6%, and a further reduction to 20 ppm for the next 12 hours reduced 6 methemoglobin levels to 0.9% (Adatia et al. 1994). A Japanese newborn developed a 7 8 methemoglobin level of 40% after 26 hours of exposure to 80 ppm; the level reduced to 3.9% 9 within 20 minutes of infusion with methylene blue and gradual reduction of the NO concentration over one hour then discontinuation. No methemoglobin levels were reported prior 10 to the 26-hour timepoint. The infant survived with no indications of hypoxic brain damage at 4 11 months of age (Nakajima et al. 1997). 12

14 The therapeutic use of NO has been studied extensively in patients with acute respiratory distress syndrome (ARDS). Manktelow et al. (1997) reviewed data collected over 5 years from 15 ARDS patients treated with NO inhalation therapy. In general, patients received 20 ppm NO for 16 48 hours with a reduction to 10 ppm for the next 8 days. No patient had an adverse response to 17 NO and 58% of all patients had clinically significant responses to NO measured as increases in 18 19 the inspiratory fraction of oxygen and decreases in pulmonary vascular resistance. Another 20 review (Troncy et al. 1997b) found that the optimal concentration of NO was between 0.5 and 40 21 ppm for producing the greatest improvement in hypoxia score among ARDS patients. These 22 results were confirmed in a more recent study in which ARDS patients were treated with 1-40 ppm for 30 minutes. Concentration-dependent decreases in pulmonary capillary pressure and 23 24 post-capillary resistance were observed with a maximum effect at 20 ppm (Benzing et al. 1998). 25 Other studies confirm improvements in oxygenation and/or pulmonary artery pressure in ARDS patients treated with 40 ppm for 20 minutes (Doering et al. 1997), 0.1-2 ppm for 15-20 minutes 26 27 (Puvbasset et al. 1994), 100 ppm for 20 minutes (Wenz et al. 1997), and 0.1-100 ppm for 15 28 minutes (Gerlach et al. 1993). Mortality was not affected by NO inhalation in any of these studies. A large increase in cardiac output was reported for one patient with ARDS and acute 29 30 right heart failure treated with 20 ppm NO for 3 days; methemoglobin levels were $\leq 1.7\%$ 31 (Benzing et al. 1997).

33 Newborns and children diagnosed with hypoxemic respiratory failure (Abman et al. 34 1994; Day et al. 1997; Goldman et al. 1997) or persistent pulmonary hypertension (Goldman et 35 al. 1995; Ichida et al. 1997; Kinsella et al. 1997; Nakagawa et al. 1997; Wessel et al. 1997) showed decreased pulmonary artery pressure and/or improved oxygen saturation when treated 36 37 with 10 ppm for up to 24 hours, 20 ppm for up to 4 hours, 60 ppm for 10 minutes, or 80 ppm for 38 up to 12 hours. Two studies reported longer term therapies in which hypoxemic newborns were 39 treated with 10 ppm NO for a duration of 6-331 hours (Biban et al. 1998) and newborns with 40 persistent pulmonary hypertension were treated with 80 ppm for a mean duration of 65.1 hours 41 (Davidson et al. 1998). The large variation in duration of exposure is explained by the fact that

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in most of these trials treatment was continued until success or failure criteria were met as defined by the study protocol.

4 NO inhalation has also been used therapeutically to treat patients with lung or heart 5 disease and following surgery. Decreased pulmonary artery pressure occurred in adult patients 6 with chronic obstructive pulmonary disease treated with 40 ppm for 20 minutes (Roger et al. 7 1997) and with pulmonary fibrosis treated with 2 ppm for 10 minutes (Yoshida et al. 1997). 8 Pulmonary vascular resistance was also significantly reduced in preterm infants treated with 20 9 ppm for 2 hours followed by 5 ppm for 70 hours (Subhedar and Shaw 1997), in patients with heart failure treated with up to 80 ppm for 5 minutes (Semigran et al. 1994), in patients following 10 11 left ventricular assist device implantation treated with 25-40 ppm for up to 48 hours (Wagner et 12 al. 1997), and in patients following lung transplantation treated with 80 ppm for 15 minutes with 13 a decrease to 10 ppm for up to 69 hours (Adatia et al. 1994). Patients with congestive heart 14 failure had increased oxygen uptake and decreased pulmonary hypertension when administered 15 20 ppm NO during light exercise (no duration given) (Matsumoto et al. 1997) and attenuation of excessive increases in tidal volume, which contribute to exercise-induced hyperventilation, when 16 exposed to 30 ppm for about 20 minutes (Bocchi et al. 1997). Decreased pulmonary artery 17 18 pressure, increased cardiac output, and/or increased oxygen arterial saturation occurred in infants 19 treated with 20 ppm NO for 4-250 hours (Journois et al. 1994) or 50 ppm for a mean of 41 hours 20 (methemoglobin, 1.4%) (Schulze-Neick et al. 1997) following surgery for congenital heart 21 defects. 22

NO inhalation of 20 ppm had no effect on PaO_2 during one-lung ventilation in patients undergoing thoracoscopic procedures, although when combined with i.v. almitrine, it limited the decrease of PaO_2 (Moutafis et al. 1997).

2.2.2. Epidemiologic Studies

29 As a component of air pollution, NO levels have been studied in association with various 30 diseases, however, other pollutants such as NO₂ and O₃ were also involved. In Helsinki, Finland, 31 emergency room admissions due to ischemic cardiac diseases were significantly correlated with 32 nitric oxide and ozone levels. NO levels ranged from 7-467 μ g/m³ (5.6-373.6 ppb) during the 33 three year study (Pönkä and Virtanen 1996). In Copenhagen, Denmark, NO and NO_x (NO + 34 NO₂) were significantly associated with the number of emergency medical contacts for children who had respiratory illnesses. The yearly mean concentration of NO was 229 µg/m³ (183.2 ppb) 35 36 and higher NO concentrations correlated with higher NO_x concentrations which were linked to 37 traffic pollution (Keiding et al. 1995). In contrast, no relationship was found between exposure 38 to oxides of nitrogen and respiratory symptoms or decline in FEV₁ among British coal miners 39 exposed to peak NO concentrations in the range of 4-100 ppm (Robertson et al. 1984).

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2.2.3. Experimental Studies

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Seven male and five female healthy volunteers were exposed to 40 ppm NO through a
 tight facial mask for 2 hours (Luhr et al. 1998). Concentrations of nitrogen dioxide were closely
 monitored and did not exceed 2.3 ppm. No changes in blood pressures, heart rates, or peripheral
 oxygen saturation were noted during exposure. Mean methemoglobin concentration increased
 from 0.63% to 1.13% during inhalation of NO.

7 NO was administered by inhalation at 80 ppm for 10 minutes to four groups of 8 volunteers: healthy adults, adults with hyperreactive airways during provocation with methacholine, patients with bronchial asthma, and patients with chronic obstructive pulmonary 9 disease. Bronchodilatory effects were measured as changes in specific airway conductance. No 10 11 unusual smell, taste, or discomfort was noted and no individual reacted with bronchoconstriction during NO exposure. NO did not affect airway conductance in healthy adults or in patients with 12 13 pulmonary disease. However, NO inhalation modulated the methacholine-induced bronchoconstriction toward bronchodilation in individuals with hyperreactive airways and 14 15 increased airway conductance in patients with asthma (Högman et al. 1993a).

Ten healthy volunteers, 8 patients with pulmonary hypertension, and 10 cardiac patients
were exposed to 40 ppm NO for 5 minutes (Pepke-Zaba et al. 1991). No clinical signs of
toxicity were reported by any individual. Pulmonary vascular resistance was significantly
reduced in patients with pulmonary hypertension and in cardiac patients; this was not measured
in the healthy volunteers. Systemic vascular resistance was not affected in any patient or
volunteer. Methemoglobin levels in the volunteer group rose from 0.33% with air to 0.42% after
NO.

25 Eight healthy adult male volunteers were exposed to 1 ppm NO for 2 hours while performing intermittent light exercise consisting of pedaling a stationary bicycle for 15 minutes 26 27 of every half hour (Kagawa 1982). Pulmonary function tests were performed after 1 and 2 hours 28 of exposure and after 1 hour of postexposure recovery. No clinical symptoms in any volunteer 29 were associated with exposure. A small but significant ($p \le 0.05$) decrease in airway 30 conductance was observed in 4/8 individuals during NO exposure that resolved in all but 2 31 subjects 1 hour postexposure; no significant difference in the group mean occurred. As a group, a significant reduction in the percentage increase of maximal expiratory flow at 50% of forced 32 33 vital capacity while breathing a He-O₂ mixture was noted at the end of the exposure period. 34 However, since this reduction was not accompanied by a reduction in forced vital capacity or an 35 increase in the alveolar plateau slope, the author questioned its biological relevance. In a similar study, respiratory resistance was significantly increased (10-12%) in healthy adults and smokers 36 37 exposed to ≥ 20 ppm for 15 minutes (von Nieding et al. 1973).

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39 In another report, specific airway conductance was significantly ($p \le 0.05$) increased in 40 healthy men exposed to 80 ppm NO for 4 minutes following methacholine-induced 41 bronchoconstriction (Sanna et al. 1994). The bronchodilator action of NO described in the 42 second report is consistent with experiments in rabbits and guinea pigs summarized below.

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Pulmonary vasoconstriction was induced in one healthy male volunteer by inhalation of a hypoxic gas mixture (Dupuy et al. 1995). NO was then administered at 10, 20, and 80 ppm for 2 3 15-minute intervals. NO induced a dose-dependent, rapid, consistent, and reversible decrease in pulmonary artery pressure, but no distress, discomfort, or pain were noted from exposure. In a 4 5 similar experiment, healthy volunteers breathed a 12% O₂ atmosphere to induce hypoxic pulmonary vasoconstriction. Addition of 40 ppm NO to the inspired gas resulted in a decrease of 6 7 pulmonary artery pressure to baseline levels within 10 minutes (Frostell et al. 1993).

9 In several studies inhaled NO has been shown to affect bleeding times or platelet aggregation although adverse clinical effects have not been demonstrated. The bleeding-time 10 11 ratio increased to 1.33 in six healthy volunteers exposed to 30 ppm for 15 minutes, but returned to near normal 60 minutes after exposure (Högman et al. 1993b). Platelet aggregation was 12 inhibited after 4 hours in mechanically ventilated neonates treated with 2-80 ppm NO for 13 hypoxic respiratory failure (Cheung et al. 1998). Cardiopulmonary bypass surgery in children 14 15 with congenital heart defects resulted in a decrease in platelet numbers by 50%; with the therapeutic use of 20 ppm NO after surgery (duration not specified), platelet numbers decreased 16 by 70%. However, no prolonged bleeding after withdrawal of indwelling catheters or drainage 17 18 tubes was detected in those patients treated with NO (Breuer et al. 1998).

NO had no effect on left ventricular function in normal healthy adults exposed to 20 ppm for 10 minutes and no increase in methemoglobin levels was found (Hayward et al. 1997).

2.3. Developmental/Reproductive Toxicity

No information was found regarding the developmental or reproductive toxicity of nitric oxide in humans.

2.4. Genotoxicity

No increase in chromosome aberrations was found in human peripheral blood lymphocytes following a 2-hour exposure to 40 ppm NO (Luhr et al. 1998). No other information was found regarding the genotoxicity of nitric oxide in humans.

2.5. Carcinogenicity

No information was found regarding the carcinogenicity of nitric oxide in humans.

2.6. Summary

40 NO has been used extensively in adults and children to lower pulmonary vascular 41 resistance caused by acute respiratory distress syndrome, hypoxemic respiratory failure, persistent pulmonary hypertension, other heart or lung disease, and organ transplantation. The 42

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1 toxicity of NO is associated with methemoglobin formation and oxidation to nitrogen dioxide.

2 Contamination of anesthesia gases has resulted in one fatality, but exposure concentrations were

not measured. Therapeutic levels of 20-80 ppm for 24 hours or 100 ppm for 20 minutes have not
 resulted in adverse effects among treated patients. However, an infant given 80 ppm for 26

5 hours developed clinically significant levels of methemoglobin which were rapidly lowered with

6 infusion of methylene blue and reduction of the NO concentration. Effects of NO on the airways

- 7 are somewhat variable. It appears that NO may have either no effect or cause
- 8 bronchoconstriction in normal subjects, but result in bronchodilation in individuals with

9 chemically-induced bronchoconstriction or asthma.

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3. ANIMAL TOXICITY DATA

13 **3.1.** Acute Lethality

3.1.1. Dogs

17 Greenbaum et al. (1967) exposed dogs to 0.5% NO (5000 ppm) for 25 minutes or to 2% (20,000 ppm) for 7-50 minutes. All dogs died either at the end of exposure or within 16 minutes 18 19 after exposure. Death was associated with a reduction in arterial oxygen content caused by 20 methemoglobinemia, low arterial Po2 due to pulmonary edema, and/or acidemia. Concurrent 21 studies were conducted in which dogs were exposed to nitrogen dioxide. No distinction was seen between the effects of either gas and it is probable that the pulmonary effects observed for 22 nitric oxide were actually from the formation of nitrogen dioxide within the system prior to 23 inhalation by the dogs. This is supported by the authors' observation that considerable oxidation 24 to nitrogen dioxide occurred as indicated by the contents of the reservoir bag of the inhalation 25 system which were visibly brown. Further, methemoglobin levels increased as a function of 26 27 time and concentration of exposure to NO. Administration of methylene blue did not return arterial oxygen content to safe levels in all dogs and the dogs died with methemoglobin levels of 28 3-5% which could not have contributed to mortality. The authors also stated that the cause of 29 30 pulmonary edema was the action of nitrogen dioxide on the alveolar lining fluid forming nitric 31 and nitrous acids which, in turn, cause denaturing of proteins, rupture of lysosomes, and the 32 development of chemical pneumonitis.

3.1.2. Rats

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To assess acute lung injury caused by inhalation of NO, rats were exposed to 500-1500 ppm for 5-30 minutes (Stavert and Lehnert 1990). By the end of the exposure to 1000 ppm for 30 minutes, the animals were cyanotic and 11/20 died within half an hour of termination of 40 exposure. Deaths were attributed to methemoglobin formation although levels were not 41 measured in this study. Exposures of up to 1500 ppm for 15 minutes or 1000 ppm for 30

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minutes produced no increases in lung weight and did not result in any histopathological changes in the lungs.

4 Groups of 5 rats/sex were exposed for 6 hours by nose-only inhalation to 0, 80, 200, 300, 400, or 500 ppm NO (Waters et al. 1998). Concentrations of \geq 300 ppm were lethal and methemoglobin levels were significantly elevated at concentrations of ≥ 200 ppm. No histopathological changes in animals exposed to 200 ppm were observed with light microscopy, but interstitial edema attributed to NO₂ contamination (2.6 ppm) was seen with the electron 8 microscope. Further details of the results or experimental procedures were not available in the 9 10 abstract.

3.1.3. Mice

In a series of experiments, mice were exposed to "predominantly" NO (Pflesser 1935). Exposure to 350 and 3500 ppm resulted in death of all animals while complete survival occurred following exposure to 310 ppm for up to 8 hours. An 8-hour LC_{50} was given as 320 ppm. Death 16 appeared to be due to methemoglobin formation and at necropsy no evidence of lung injury or pulmonary edema was observed.

3.2. Nonlethal Toxicity

3.2.1. Dogs

Anesthetized beagle dogs (3-4 per group) were exposed to 0, 80, 160, 320, or 640 ppm NO for 6 hours (Mihalko et al. 1998, Wilhelm et al. 1998). One animal in the 640 ppm group died. Decreased arterial oxygen concentrations were measured following exposure to 320 and 640 ppm and increased minute volumes and decreased systemic arterial pressures were observed at 640 ppm. Methemoglobin levels were 3, 6.6, 24, and 78%, respectively. Further details of the results and experimental procedures were not available in the abstracts.

30 The pulmonary vasodilating effects of NO have been demonstrated in several canine 31 models of lung injury including hypoxia (Channick et al. 1994, Romand et al. 1994), oleic acid-32 induced injury (Romand et al. 1994, Putensen et al. 1994ab, Zwissler et al. 1995), pulmonary 33 microembolism (Zwissler et al. 1995), cardiac transplant (Chen et al. 1997), and pulmonary 34 shunt (Hopkins et al. 1997). Following lung injury, dogs were given NO at concentrations ranging from 40 to 80 ppm for up to 40 minutes. In all studies, NO significantly decreased 35 pulmonary vascular resistance, decreased pulmonary artery pressure, and/or improved 36 37 ventilation-perfusion mismatch. Where measured, methemoglobin levels did not exceed 1.1% 38 (Channick et al. 1994, Putensen et al. 1994a, Romand et al. 1994).

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40 3.2.2. Rabbits

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1 NO inhalation has been shown to attenuate the effects of experimentally-induced lung 2 injury in the rabbit. Rabbits were given 20 ppm NO for 6 hours with or without prior endotoxininduced lung injury. In control animals, NO had no effect on pulmonary artery pressure, mean 3 4 arterial pressure, heart rate, central venous pressure, or oxygenation. Pulmonary hypertension and deterioration of oxygenation by endotoxin were less pronounced in rabbits receiving NO, but 5 the inflammatory response was not reduced. After 6 hours, methemoglobin levels did not exceed 6 7 1.5% (Nishina et al. 1997). In another study of endotoxin-induced lung injury, increased 8 survival occurred in rabbits treated with 10 ppm for 90 minutes (7/7 vs 5/9 controls), but improvement of pulmonary gas exchange was not demonstrated (Uchida et al. 1996). 9

11 The influence of inhaled NO on airway responsiveness to acetylcholine (ACH) in normal and hyperresponsive rabbits was investigated (Mensing et al. 1997). Following ACH 12 provocation, animals were treated with 150 or 300 ppm for 5-10 minutes. No effects of NO 13 inhalation were seen at ACH concentrations of $\leq 2\%$, however NO significantly reduced airway 14 resistance caused by 4% and 8% ACH. Animals were then made hyperresponsive to ACH by 15 exposure to ammonium persulfate. NO inhalation at 300 ppm significantly decreased the 16 response to ACH to almost the same level as before ammonium persulfate. Similar results were 17 18 obtained with methacholine-induced bronchoconstriction (Högman et al. 1993c). Rabbits were 19 exposed to increasing concentrations of nebulized methacholine with or without inhalation of 80 ppm NO and airway resistance was measured after 5 minutes. During NO inhalation, there was 20 21 no significant increase in methacholine-induced airway resistance. 22

In rabbits exposed to 30 or 300 ppm for 15 minutes bleeding times increased 46% and 72%, respectively, but there were no changes in hematocrit, whole blood or plasma viscosity, erythrocyte aggregation tendency, or erythrocyte deformability (Högman et al. 1993b, 1994).

3.2.3 Pigs

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29 Inhalation of 20, 40, or 80 ppm NO for 5 minutes by healthy pigs resulted in slight, but 30 significant (p = 0.04), reductions in pulmonary artery pressure (Goldstein et al. 1997). The effects of inhaled NO have also been studied in porcine models of adult and neonatal pulmonary 31 hypertension. Dose-related decreases in pulmonary artery pressure and input resistance and 32 33 increases in vascular efficiency have been observed in adult pigs administered 10-80 ppm for up 34 to 20 minutes following vasoconstriction induced by hypoxia (Hillman et al. 1997), thromboxane 35 administration (Goldstein et al. 1997), and oleic acid administration (Shah et al. 1994). No effects on cardiac output, systemic arterial pressure, or left ventricular contractility were 36 37 observed in any study. Inhalation of 40 ppm for 30 minutes by pigs with oleic acid induced lung 38 injury resulted in sustained improvements in pulmonary artery pressure, oxygen partial pressure, and intrapulmonary shunt fraction which deteriorated to control levels following termination of 39 NO inhalation (Shah et al. 1994). NO inhalation did not cause histopathologic changes in the 40 41 lungs and methemoglobin levels were 1.7% following exposure to 80 ppm (Shah et al. 1994). 42

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The effects of inhaled NO were studied in a porcine model of neonatal pulmonary hypertension (Nelin et al. 1994). Pigs, approximately 13 days old, were administered room air, ppm NO, hypoxia, or hypoxia plus 25 ppm NO for 15 minutes each. NO inhalation significantly reduced pulmonary artery pressure both alone and following hypoxia with no changes in dynamic lung compliance, pulmonary resistance, hemoglobin, hematocrit, or methemoglobin. At the end of the experiment, 1000 ppm NO was administered to one animal for 15 minutes which resulted in methemoglobin levels of 20%.

3.2.4. Sheep

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11 Frostell et al. (1991) examined the effects of inhalation of 5-80 ppm NO on the normal and acutely constricted pulmonary circulation in awake lambs. Dose-response data were 12 13 collected for a 6-minute inhalation and toxicity data were collected after 1 and 3 hours. Pulmonary constriction was induced by either infusion of the endoperoxide analoge of 14 15 thromboxane, U46619, or by hypoxia. In normal lambs, 80 ppm for 6 minutes did not affect either pulmonary artery pressure or pulmonary vascular resistance. However, in lambs with 16 constricted pulmonary circulation, a dose-related increase in vasodilation occurred with 17 18 significantly reduced pulmonary artery pressure at 5 ppm and an almost complete vasodilator 19 response at 40 and 80 ppm. Systemic vasodilation did not occur. Inhalation of 80 ppm for 1 and 3 hours did not increase extravascular lung water or methemoglobin levels, or modify lung 20 histology compared with control lambs. 21

23 Decreased pulmonary artery pressure has also been demonstrated in several other ovine 24 models of experimental pulmonary hypertension. The therapeutic effects of NO described by 25 Frostell et al. (1991) were confirmed in another study (DeMarco et al. 1996) in which 80 ppm for 3 hours completely reversed U46619-induced pulmonary hypertension without affecting 26 27 systemic circulation. In this study, maximum methemoglobin levels of 4.7% occurred in the last 28 half hour. A similar dose-dependent reduction in pulmonary artery pressure was shown over a range of concentrations of 4-512 ppm NO with maximum effect at 64 ppm within 5-10 minutes. 29 30 Inhalation of 512 ppm for 20 minutes resulted in methemoglobin levels of 11% (Dyar et al. 31 1993). In newborn lambs with persistent pulmonary hypertension, significantly increased survival occurred in lambs treated with 80 ppm for 23 hours; no evidence of lung injury from 32 33 NO inhalation was observed. Arterial oxygen tension in the NO treated lambs was significantly 34 greater (63 vs 14 mm Hg) within 15 minutes and continued to increase over time. At the end of 35 the study, methemoglobin levels were 3% (Zayek et al. 1993).

Decreased pulmonary artery pressure and increased arterial oxygenation occurred in
 sheep treated with 20 ppm for 48 hours following lung injury from smoke inhalation, but airway
 inflammation was not reduced (Ogura et al. 1994). In premature lambs with hyaline membrane
 disease, exposure to 20 ppm NO for 5 hours did not significantly change oxidative stress
 parameters or induce lung inflammation (Storme et al. 1998).

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3.2.5. Guinea Pigs

Pulmonary resistance was significantly decreased in guinea pigs exposed to 300 ppm for 6 minutes. In the same study, 5-300 ppm for 10 minutes resulted in a dose-related, rapid, consistent, and reversible reduction of pulmonary resistance and an increase in lung compliance following methacholine-induced bronchoconstriction (Dupuy et al. 1992).

3.2.6. Rats

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10 The effects of NO on discrimination learning and brain activity were studied in rats (Groll-Knapp et al. 1988). Rats were exposed to 10 or 50 ppm for 180 minutes and the test 11 atmospheres maintained during behavioral testing and EEG examination. The high 12 13 concentration significantly reduced the number of correct trials and the total number of lever presses in the operant conditioning chamber. Both concentrations resulted in increased 14 15 amplitudes and prolonged peak latencies of the auditory evoked potentials assessed on EEG. Maximum methemoglobin levels were 3.98%. The authors suggested that the effects could be 16 due, in part, to diminished oxygen carrying capacity related to methemoglobin formation. 17 18

19 The effects of inhaled NO on hyperoxic lung injury in rats were investigated (Garat et al. 20 1997). Animals were exposed to 10 or 100 ppm NO while breathing 21% or 100% oxygen for 21 40 hours. No toxic effects on any lung parameter were observed at either NO concentration under normoxic conditions. In hyperoxic conditions, inhalation of 10 ppm prevented increases 22 in thiobarbituric acid reactive substances and wet to dry lung weight ratios, had no effect on the 23 alveolar barrier impermeability to protein, and improved alveolar liquid clearance. These effects 24 did not occur at 100 ppm NO with hyperoxia and the lack of protection may have been due to the 25 formation of nitrogen dioxide in the exposure chambers. 26 27

28 Rats were exposed to an atmosphere of oxides of nitrogen that was produced by mixing nitrogen dioxide and nitric oxide (Brown et al. 1983). The ratio of each chemical was not 29 30 specified or measured in the exposure chambers. Exposures were to 518 ppm for 5 minutes or to 31 1435 ppm for 1 minute. No clinical signs of toxicity were observed during exposure to either concentration but "stertorous respirations" appeared within 24 hours. Histologically, initial lung 32 damage showed thickening and blebbing of the alveolar epithelium followed by a latent period 33 34 of about 6 hours after which development of edema of the interstitium and alveolar septum was 35 observed. The early changes were attributed to a direct oxidant effect. Both clinical signs and histiological findings were more severe following exposure to 518 ppm for 5 minutes. 36

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3.3. Developmental/Reproductive Toxicity

40 No information was found regarding the developmental or reproductive toxicity of
 41 exogenously administered nitric oxide in animals. Growth retardation and hind limb reduction

were found in the offspring of rats given 0.3 and 1.0 mg/mL of N^G-nitro-L-arginine methyl ester, a nitric oxide synthase inhibitor, in the drinking water on gestation days 13-19 (Shepard 1995).

3.4. Genotoxicity

Three-week old male Sprague-Dawley rats were exposed by inhalation to 9, 19, or 27 ppm NO for 3 hours, maintained overnight before sacrifice, and lung cells were isolated. At the high dose, there was a significant increase in mutations to ouabain resistance in lung cells. Chromosome aberrations were not observed following exposure to NO but were induced by exposure to NO₂ (Isomura et al. 1984).

A dose-related increase in the number of revertants of *Salmonella typhimurium* (TA1535) occurred when culture dishes were exposed to atmospheres containing 0-20 ppm nitric oxide for 30 minutes. Oxygen was required and mutation was inhibited by antioxidants. Cytotoxicity was seen at 50 ppm nitric oxide (Arroyo et al. 1992).

3.5. Carcinogenicity

No information was found regarding the carcinogenicity of nitric oxide in animals.

3.6. Summary

Most of the experimental animal studies available focused on the therapeutic use of NO in an animal model of human disease. Lethality studies in dogs, rats, and mice lacked complete concentration-response information, were confounded by possible NO_2 contamination, or were secondary citations in which the original source could not be obtained. From these studies, however, it appears that in the absence of lung injury, the mechanism of toxicity of NO is methemoglobin formation.

4. SPECIAL CONSIDERATIONS

4.1. Metabolism and Disposition

Approximately 85-92% of NO is absorbed into the body by humans breathing normally during inhalation exposure to 0.33-5.0 ppm (0.4-6.1 mg/m³) (Yoshida and Kasama 1987). In contrast, about 35% of the total amount of NO delivered is taken up by the lungs in patients with acute lung injury given 5-40 ppm NO as ongoing therapy (Westfelt et al. 1997). Once absorbed, inhaled NO reacts with hemoglobin to form nitrosylhemoglobin from which nitrite and nitrate are generated. Most of the nitrates are excreted in the urine with a small portion secreted into the oral cavity through the salivary glands and transformed to nitrite. Nitrate in the intestine is reduced to ammonia through nitrite, reabsorbed into the body, and converted to urea (Yoshida and Kasama 1987). Pigs given 10-80 ppm sequentially for 10-minute periods followed by 40

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ppm for 30 minutes, showed a concentration-related increase in plasma nitrites and nitrates with a combined concentration of 67 µmol/L at the end of exposure compared to a baseline of 30 µmol/L (Shah et al. 1994). A high ¹⁵N content was found in serum and urine of rats after inhalation of 138-880 ppm ¹⁵NO, and within 24 hours, about 40% of the inhaled ¹⁵N was 4 excreted into the urine. Small amounts of ¹⁵N were found in lung, trachea, liver, kidney, and muscle (Yoshida et al. 1980).

Nitrate (10.4 µmol/L) has been detected in the broncho-alveolar lavage fluid of healthy children from the metabolism of endogenous NO in the lower respiratory tract (Grasemann et al. 1997).

4.2. Mechanism of Toxicity

14 From the available studies, it appears that the major mechanism of toxic action of NO is 15 the binding of hemoglobin (U.S. EPA 1993b). Inhaled NO is absorbed into the bloodstream and binds to hemoglobin forming nitrosylhemoglobin which is rapidly oxidized to methemoglobin 16 (Maeda et al. 1987, Sharrock et al. 1984, U.S. EPA 1993b). The affinity of NO for hemoglobin 17 18 is about 1500 times greater than that of carbon monoxide (Gibson and Roughton 1957) and the 19 binding and formation of methemoglobin is NO concentration- and time-dependent (Maeda et al. 1987, Ripple et al. 1989, Sharrock et al. 1984). Experiments with rats (Maeda et al. 1987) and 20 rabbits (Sharrock et al. 1984) show that NO binding to hemoglobin is rapidly reversible with a 21 half-life of 15-20 minutes when the animals are placed in clean air. 22

24 The signs and symptoms of methemoglobinemia in humans are summarized in Table 2. 25 As can be seen, clinical signs do not appear until methemoglobin levels are 15-20% and toxicity is not evident until levels of about 30%. 26

TABLE 2. SIGNS AND SYMPTOMS IN HUMANS ASSOCIATED WITH METHEMOGLOBIN CONCENTRATIONS		
Methemoglobin Concentration (%)	Signs and Symptoms	
1.1	Normal level	
1-15	None	
15-20	Clinical cyanosis (chocolate brown blood); no hypoxic symptoms	
30	Fatigue; recovery without treatment	
20-45	Anxiety, exertional dyspnea, weakness, fatigue, dizziness, lethargy, headache, syncope, tachycardia	
45-55	Decreased level of consciousness	
55-70, ~60	Hypoxic symptoms: semistupor, lethargy, seizures, coma, bradycardia, cardiac arrhythmias	
>70	Heart failure from hypoxia; high incidence of mortality	
>85	Lethal	

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19 In most of the human and animal experimental studies and the human case reports 20 described previously, methemoglobin levels were <5% even after exposure to as much as 50 21 ppm NO for 41 hours (human infant) or 80 ppm for 23 hours (lamb). Methemoglobin levels rose to 9.4% in one lung transplantation patient after treatment with 80 ppm NO for 8 hours. A 22 23 reduction in concentration to 40 ppm over 4 hours resulted in a decrease to 6.6%, and a further reduction to 20 ppm for the next 12 hours reduced methemoglobin levels to 0.9% (Adatia et al. 24 25 1994). In one patient with pulmonary hypertension, methemoglobin levels rose to 9.6% after 108 hours of treatment with 80 ppm and in another patient levels were 14% after 18 hours 26 27 (Wessel et al. 1994). An American Indian patient with pulmonary hypertension treated with 80 ppm for 6 hours developed levels of 9.4% which decreased rapidly with a reduction in NO to 40 28 29 ppm (Wessel et al. 1994). Methemoglobinemia >7% occurred in 13/37 newborns treated with 80 30 ppm NO for persistent pulmonary hypertension. The average time to peak level in all patients

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was 19.6 hours and the highest level of methemoglobin was 11.9% at 8 hours in one patient (Davidson et al. 1998).

4 Despite the relatively low levels of methemoglobin measured in most studies, clinically 5 significant levels have been reported. A newborn (Japanese) developed a methemoglobin level 6 of 40% after 26 hours of exposure to 80 ppm; the level reduced to 3.9% within 20 minutes of 7 infusion with methylene blue and reduction of the NO concentration (Nakajima et al. 1997). 8 Sheep administered 512 ppm for 20 minutes (Dyar et al. 1993) and pigs given 1000 ppm for 15 9 minutes (Nelin et al. 1994) developed levels of 11% and 20%, respectively. Cyanosis appeared in dogs within 3-8 minutes of exposure to 0.5% or 2% NO (5000 ppm or 20,000 ppm) and 10 methemoglobin levels were 5-25%. However, levels reached 100% in one dog that died after 11 12 exposure to 2% (20,000 ppm) for 50 minutes (Greenbaum et al. 1967). A single 6-hour exposure 13 of dogs to 80, 160, 320, or 640 ppm resulted in methemoglobin levels of 3, 6.6, 24, and 78%, 14 respectively (Wilhelm et al. 1998). Rats exposed to 1000 ppm NO for 30 minutes appeared 15 cyanotic and 11/20 died due to methemoglobin formation but levels were not measured (Stavert 16 and Lehnert 1990).

18 While the main toxicological effect of inhaled NO is the induction of methemoglobin, 19 that of NO₂ is the formation of pulmonary edema. Methemoglobin levels did not increase in rats 20 exposed to 40 ppm NO₂ despite a slight elevation (0.2%) in nitrosylhemoglobin levels (Oda et al. 21 1980). Rats exposed to 1000 ppm NO for 30 minutes appeared cyanotic and 11/20 died due to 22 methemoglobin formation but no changes in lung weight or histopathology were observed. In 23 the same study, increased lung weight occurred following exposure to 50 ppm NO₂ for 30 24 minutes and histopathological changes were observed after exposure to 25 ppm for 30 minutes 25 (Stavert and Lehnert 1990). Other studies have failed to show any effect of NO on the respiratory tract of humans (Manktelow et al. 1997, Högman et al. 1993a, Pepke-Zaba et al. 26 27 1991, Kagawa 1982), mice (Pflesser 1935), pigs (Nelin et al. 1994), or lambs (Frostell et al. 28 1991). A concentration of 10 ppm, but not 100 ppm, offered protection against hyperoxic lung 29 injury in rats and it is likely that the higher concentration of NO resulted in significant NO₂ formation (Garat et al. 1997). NIOSH (1976), summarized the effects of NO₂ in humans as 30 31 initial irritation with mild dyspnea during exposure followed by delayed onset of pulmonary 32 edema after several hours of apparent recovery. A similar toxic response, including interstitial 33 fibrosis, has been shown in five species of animals following acute inhalation exposure to NO₂ 34 (Hine et al. 1970) and in rats exposed to mixed oxides of nitrogen (Brown et al. 1983). These 35 results indicate that NO₂ has a direct toxic action on the respiratory tract, but NO does not.

The relative toxicities of NO and NO₂ are complex. NIOSH (1976) summarized experiments by Paribok and Grokholskaya (1962) in mice and guinea pigs. At concentrations >833 ppm for 1 hour, NO was more toxic than NO₂, however, at lower concentrations, NO₂ was more toxic. It appears that for NO, if the concentration is not high enough to be lethal due to methemoglobin formation, the animal recovers completely. On the other hand, concentrations of

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34 35 NO_2 that are not rapidly lethal may cause more persistent effects and in some cases cause death from pulmonary edema after a delay of several days (NIOSH 1976).

4.3. Oxides of Nitrogen

6 NO is unstable in air and undergoes spontaneous oxidation to NO₂ making experimental 7 effects difficult to separate and studies difficult to perform (U.S. EPA 1993b). Studies on the 8 conversion of NO to NO₂ in medicinal applications have found the conversion to be significant 9 in an atmospheric concentration of O_2 (20.9%) at room temperature. The delivery of 100 ppm NO in 21% O₂ through a pediatric tube (d=0.009m l=0.9m) at a flow rate of 2 L/min is calculated 10 11 to produce 1.13 ppm NO₂ (Lindberg and Rydgren 1998). For 80 ppm NO, a concentration 12 commonly used therapeutically, 5 ppm NO₂ is calculated to form by 3 minutes in air (Foubert et 13 al. 1992). NO reacts quantitatively with oxygen in air to form NO₂ which then reacts with water 14 to form nitric acid (NIOSH 1976). For this reason, careful monitoring of NO₂ concentrations has 15 been suggested when NO is used therapeutically at concentrations ≥ 80 ppm especially when 16 coadministered with oxygen (Miller et al. 1994, Foubert et al. 1992). Figure 1 summarizes the 17 reactions of the oxides of nitrogen. 18

$2NO + O_2 \rightarrow 2NO_2$ (minor at ambient temp)

23	$NO + O_3 \rightarrow NO_2 + O_2$
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25	$\text{NO} + \text{HO}_2 \cdot \rightarrow \text{NO}_2 + \text{HO} \cdot$
26	
27	$NO + RO_2 \rightarrow NO_2 + RO \rightarrow NO_2$
28	
29	$NO_2 + HO \cdot \rightarrow HNO_3$
30	
31	$2NO_2 \rightarrow N_2O_4$
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Figure 1: Environmental reactions of the oxides of nitrogen

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1 This family of reaction paths is temperature dependent, but in general favors NO_2 2 production. While on the laboratory scale, closed system experiments clearly indicate the 3 potential for the production of NO₂, the chemical kinetics of NO conversion during a large-scale atmospheric release and dispersion are not well documented. The estimation of the 4 5 concentration isopleths following an accidental release would require the use of a finite element 6 model along with several assumptions as to the chemical rate constants. As a result, the 7 conversion of NO to NO₂ during the atmospheric release is of concern to emergency planners, but the NAC is unable to provide any significant guidance, other than to indicate that a 8 9 significant fraction of the NO will be converted to NO₂ and present an atmospheric hazard that is generally more toxic than just NO. 10

4.4. Other Relevant Information

No information was available to allow comparison of NO toxicity between species or between individuals. Concentrations used in animal models of human diseases are similar to those used therapeutically in humans with no adverse effects. Because the major toxic action of NO is binding to hemoglobin resulting in methemoglobinemia, little inter- or intra-species variation is expected. In addition, NO is administered for extended periods of time to critically ill patients with only slight increases in methemoglobin concentrations.

5. DATA ANALYSIS FOR AEGL-1

AEGL-1 is the airborne concentration (expressed as parts per million or milligrams per cubic meter [ppm or mg/m³]) of a substance above which it is predicted that the general population, including susceptible individuals, could experience notable discomfort, irritation, or certain asymptomatic, non-sensory effects. However, the effects are not disabling and are transient and reversible upon cessation of exposure.

5.1. Summary of Human Data Relevant to AEGL-1

32 Human data relevant to derivation of AEGL-1 are limited to case reports of either slightly 33 increased methemoglobin levels or no adverse effects during NO administration to 34 therapeutically lower pulmonary artery pressure. Methemoglobin levels rose to 9.4% in one 35 lung transplantation patient after treatment with 80 ppm NO for 8 hours. A reduction in concentration to 40 ppm over 4 hours resulted in a decrease to 6.6%, and a further reduction to 36 37 20 ppm for the next 12 hours reduced methemoglobin levels to 0.9% (Adatia et al. 1994). In one 38 patient with pulmonary hypertension methemoglobin levels rose to 9.6% after 108 hours of 39 treatment with 80 ppm and in another patient levels were 14% after 18 hours (Wessel et al. 40 1994). An American Indian patient with pulmonary hypertension treated with 80 ppm for 6 41 hours developed methemoglobin levels of 9.4% which decreased rapidly with a reduction in NO to 40 ppm (Wessel et al. 1994). The average time to peak methemoglobin levels in newborns 42

 treated with 80 ppm was 19.6 hours and the highest level of methemoglobin was 11.9% at 8 hours in one patient (Davidson et al. 1998).

No adverse toxicity was reported for newborns and children diagnosed with hypoxemic respiratory failure (Abman et al. 1994, Day et al. 1997, Goldman et al. 1997) or persistent pulmonary hypertension (Goldman et al. 1995, Ichida et al. 1997, Kinsella et al. 1997, Nakagawa et al. 1997, Wessel et al. 1997) when treated with 10-80 ppm for 10 minutes to 24 hours. Methemoglobin levels were not reported in these cases.

5.2. Summary of Animal Data Relevant to AEGL-1

Animal data relevant to derivation of AEGL-1 were limited to one abstract. A single 6hour exposure of dogs to 80, 160, 320, or 640 ppm resulted in methemoglobin levels of 3, 6.6, 24, and 78%, respectively (Wilhelm et al., 1998). Most experimental animal studies were either on lethality or as models of human disease in which the therapeutic effects of NO were assessed.

5.3. Derivation of AEGL-1

AEGL-1 values were not derived. The NAC recommends the use of the AEGL values for NO₂ for emergency planning. The proposed AEGL values for NO₂ are given in Appendix A. In several human case reports a concentration of 80 ppm for 6-108 hours resulted in methemoglobin levels of approximately 10% without adverse toxicity (Adatia et al. 1994, Wessel et al. 1994). Therefore, concentrations below 80 ppm NO should not constitute a health hazard.

6. DATA ANALYSIS FOR AEGL-2

AEGL-2 is the airborne concentration (expressed as ppm or mg/m³) of a substance above which it is predicted that the general population, including susceptible individuals, could experience irreversible or other serious, long-lasting adverse health effects or an impaired ability to escape.

34 6.1. Summary of Human Data Relevant to AEGL-2

Human data relevant to AEGL-2 were not found.

38 6.2. Summary of Animal Data Relevant to AEGL-2

Animal data relevant to AEGL-2 were not found.

6.3. Derivation of AEGL-2

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AEGL-2 levels were not derived because no human or animal data were available upon which to base the calculations. The NAC recommends the use of the AEGL values for NO_2 for emergency planning. The proposed AEGL values for NO_2 are given in Appendix A.

7. DATA ANALYSIS FOR AEGL-3

AEGL-3 is the airborne concentration (expressed as ppm or mg/m³) of a substance above which it is predicted that the general population, including susceptible individuals, could experience life-threatening health effects or death.

7.1. Summary of Human Data Relevant to AEGL-3

The one report of human fatality due to NO contamination of anesthetic gases does not contain concentration or duration information. However, an infant developed a methemoglobin level of 40% after 26 hours of exposure to 80 ppm (Nakajima et al., 1997). This level is clearly life-threatening in a newborn.

7.2. Summary of Animal Data Relevant to AEGL-3

Animal data relevant to derivation of AEGL-3 were not found. Lethality studies in dogs (Greenbaum et al. 1967), rats (Stavert and Lehnert 1990), and mice (Pflesser 1935) lacked complete dose-response information, were confounded by possible NO₂ contamination, or were secondary citations in which the original source could not be obtained. An approximate 30-minute LC₅₀ of 1000 ppm was reported by Stavert and Lehnert (1990) at which 11/20 rats died. These studies were valuable, however, in determining the mechanism of toxicity of NO, i.e., methemoglobin formation in the absence of lung injury. Sheep developed methemoglobin levels of only 11% after exposure to 512 ppm for 20 minutes (Dyar et al. 1993) and pigs developed levels of 20% after exposure to 1000 ppm for 15 minutes (Nelin et al. 1994), although these levels are not life-threatening.

7.3. Derivation of AEGL-3

Data are insufficient for derivation of AEGL-3 values. A methemoglobin level of 40% in an infant is clearly life-threatening, but this level did not occur until after 26 hours of continuous exposure to 80 ppm. The concentration of 80 ppm has been safely used in many clinical applications for durations less than 26 hours, although higher concentrations for shorter durations would be expected to result in critical levels of methemoglobin formation. Lethality studies in animals lacked sufficient concentration-response information to calculate either an LC₀₁ value or the concentration of NO which results in formation of critical methemoglobin levels

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Importantly, in the event of an accidental release of NO, attention should be given to the formation of NO₂ in the atmosphere. Careful monitoring of the concentrations of both NO and NO₂ is warranted. Therefore, AEGL-3 values are not proposed because of conflicting data and lack of sound toxicity information. The NAC recommends the use of the AEGL values for NO₂ for emergency planning. The proposed AEGL values for NO₂ are given in Appendix A.

8. SUMMARY OF AEGLS

8.1. AEGL Values and Toxicity Endpoints

AEGL values for NO are not recommended. Because conversion to NO₂ is expected to occur in the atmosphere, and because NO₂ is more toxic than NO, the AEGL values for NO₂ are recommended for use with emergency planning for NO (Appendix A). The NAC recognizes, however, that short-term exposures below 80 ppm NO should not constitute a health hazard.

8.2. Comparison with Other Standards and Guidelines

Standards and guidance levels for workplace and community exposures are listed in
Table 3. An occupational TWA of 25 ppm has been adopted by several groups (ACGIH 2003,
NIOSH 2003, OSHA 1999). International standards are also 25 ppm for a workday (Ministry of
Social Affairs and Employment 2000, Swedish National Board of Occupational Safety and
Health 1996). In addition, Sweden has adopted 50 ppm as a short-term exposure limit and the
immediate danger to life and health (IDLH) of 100 ppm (NIOSH 1996) is based on human and
animal data for oxides of nitrogen due to lack of useful data on nitric oxide.

TABLE 3. Extant Standards and Guidelines for Nitric Oxide						
	Exposure Duration					
Guideline	10 minute	30 minute	1 hour	4 hour	8 hour	
AEGL-1	NR	NR	NR	NR	NR	
AEGL-2	NR	NR	NR	NR	NR	
AEGL-3	NR	NR	NR	NR	NR	
IDLH (NIOSH) ^a		100 ppm				
REL-TWA (NIOSH) ^b					25 ppm	
PEL-TWA (OSHA) ^c					25 ppm	
TLV-TWA (ACGIH) ^d					25 ppm	
MAC (The Netherlands) ^e					25 ppm	
OEL-LLV (Sweden) ^f					25 ppm	
OEL-CLV (Sweden) ^g	50	ppm				

NR: not recommended

- **aIDLH (Immediately Dangerous to Life and Health, National Institute of Occupational Safety and Health)** (NIOSH 1996) represents the maximum concentration from which one could escape within 30 minutes without any escape-impairing symptoms, or any irreversible health effects.
- ^bNIOSH REL-TWA (National Institute of Occupational Safety and Health, Recommended Exposure Limits Time Weighted Average) (NIOSH 2003) is defined analogous to the ACGIH-TLV-TWA.
- ^cOSHA PEL-TWA (Occupational Health and Safety Administration, Permissible Exposure Limits Ceiling) (OSHA 1999) is defined analogous to the ACGIH-TLV-TWA, but is for exposures of no more than 10 hours/day, 40 hours/week.
- ^dACGIH TLV-TWA (American Conference of Governmental Industrial Hygienists, Threshold Limit Value Time Weighted Average) (ACGIH 2003) is the time-weighted average concentration for a normal 8-hour workday and a 40-hour workweek, to which nearly all workers may be repeatedly exposed, day after day, without adverse effect.
- ^eMAC (Maximaal Aanvaaarde Concentratie [Maximal Accepted Concentration]) (SDU Uitgevers [under the auspices of the Ministry of Social Affairs and Employment], The Hague, The Netherlands 2000) is defined analogous to the ACGIH-TLV-TWA.
- ^f**OEL-LLV (Occupational Exposure Limits Level Limit Value)** (Swedish National Board of Occupational Safety and Health, 1996) is an occupational exposure limit value for exposure during one working day.
- ^gOEL-CLV (Occupational Exposure Limits Ceiling Limit Value) (Swedish National Board of Occupational Safety and Health, 1996) is an occupational exposure limit value for exposure during a reference period of fifteen minutes.

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8.3. Data Adequacy and Research Needs

Data were not available for derivation of AEGL values. The available studies did not contain full concentration-response information, and many did not account for conversion of NO to NO_2 . NO_2 is more toxic than NO and the rate of conversion in the event of an accident is unknown.

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