

Table J.5. Uranium K_d values selected from literature for development of look-up table.

pH	U Kd (ml/g)	Clay Cont. (wt.%)	CEC (meq/100g)	Surface Area (m ² /g)	Solution	Soil Identification	Reference / Comments
8.3	1.98				Hanford Groundwater	Trench 8 Loamy Sand	Kaplan and Serne (1995, Part. Sat. Column, 40%)
8.3	0.49				Hanford Groundwater	Trench 8 Loamy Sand	Kaplan and Serne (1995, Part. Sat. Column, 40%)
8.3	2.81				Hanford Groundwater	Trench 8 Loamy Sand	Kaplan and Serne (1995, Part. Sat. Column, 38%)
8.3	0.62				Hanford Groundwater	Trench 8 Loamy Sand	Kaplan and Serne (1995, Part. Sat. Column, 22%)
8.3	0.45				Hanford Groundwater	Trench 8 Loamy Sand	Kaplan and Serne (1995, Part. Sat. Column, 30%)
8.3	0.54				Hanford Groundwater	Trench 8 Loamy Sand	Kaplan and Serne (1995, Part. Sat. Column, 23%)
8.3	0.62				Hanford Groundwater	Trench 8 Loamy Sand	Kaplan and Serne (1995, Part. Sat. Column, 25%)
8.3	0.40				Hanford Groundwater	Trench 8 Loamy Sand	Kaplan and Serne (1995, Part. Sat. Column, 17%)
8.3	0.10				Hanford Groundwater	Trench 8 Loamy Sand	Kaplan and Serne (1995, Part. Sat. Column, 7%)
8.3	0.08				Hanford Groundwater	Trench 8 Loamy Sand	Kaplan and Serne (1995, Part. Sat. Column, 7%)
8.3	2.0		5.2		Hanford Groundwater	Trench 8 Loamy Sand	Lindenmeir <i>et al.</i> (1995, Saturated Column 1)
8.3	0.5		5.2		Hanford Groundwater	Trench 8 Loamy Sand	Lindenmeir <i>et al.</i> (1995, Saturated Column 1)
8.3	2.7		5.2		Hanford Groundwater	Trench 8 Loamy Sand	Lindenmeir <i>et al.</i> (1995, Saturated Column 1)
8.3	1.0		5.2		Hanford Groundwater	Trench 8 Loamy Sand	Lindenmeir <i>et al.</i> (1995, Unsat. Column 1, 65%)
8.3	0.5		5.2		Hanford Groundwater	Trench 8 Loamy Sand	Lindenmeir <i>et al.</i> (1995, Unsat. UFA 1, 70%)
8.3	0.2		5.2		Hanford Groundwater	Trench 8 Loamy Sand	Lindenmeir <i>et al.</i> (1995, Unsat. UFA 2, 24%)
8.3	1.1		5.2		Hanford Groundwater	Trench 8 Loamy Sand	Lindenmeir <i>et al.</i> (1995, Unsat. Column 1, 63%)
8.3	1.1		5.2		Hanford Groundwater	Trench 8 Loamy Sand	Lindenmeir <i>et al.</i> (1995, Unsat. Column 2, 43%)
8.3	0.6		5.2		Hanford Groundwater	Trench 8 Loamy Sand	Lindenmeir <i>et al.</i> (1995, Unsat. UFA 1A, 29%)
8.3	0.6		5.2		Hanford Groundwater	Trench 8 Loamy Sand	Lindenmeir <i>et al.</i> (1995, Unsat. UFA 1C, 29%)

pH	U Kd (ml/g)	Clay Cont. (wt.%)	CEC (meq/100g)	Surface Area (m ² /g)	Solution	Soil Identification	Reference / Comments
8.4	0.20		5.3	6.3	Hanford Groundwater	Trench 94	Kaplan <i>et al.</i> (1998, Batch)
8.4	0.15		5.3	6.3	Hanford Groundwater	Trench 94	Kaplan <i>et al.</i> (1998, Batch)
8.4	0.09		5.3	6.3	Hanford Groundwater	Trench 94	Kaplan <i>et al.</i> (1998, Batch)
8.4	0.15		5.3	6.3	Hanford Groundwater	Trench 94	Kaplan <i>et al.</i> (1998, Batch)
8.4	0.14		5.3	6.3	Hanford Groundwater	Trench 94	Kaplan <i>et al.</i> (1998, Batch)
7.92	1.99		6.4	14.8	Hanford Groundwater	Trench AE-3	Kaplan <i>et al.</i> (1998, Batch)
8.05	1.92		6.4	14.8	Hanford Groundwater	Trench AE-3	Kaplan <i>et al.</i> (1998, Batch)
7.99	1.91		6.4	14.8	Hanford Groundwater	Trench AE-3	Kaplan <i>et al.</i> (1998, Batch)
7.99	2.10		6.4	14.8	Hanford Groundwater	Trench AE-3	Kaplan <i>et al.</i> (1998, Batch)
7.98	2.25		6.4	14.8	Hanford Groundwater	Trench AE-3	Kaplan <i>et al.</i> (1998, Batch)
7.97	2.44		6.4	14.8	Hanford Groundwater	Trench AE-3	Kaplan <i>et al.</i> (1998, Batch)
8.48	1.07		6.4	14.8	Hanford Groundwater	Trench AE-3	Kaplan <i>et al.</i> (1998, Batch)
8.26	1.46		6.4	14.8	Hanford Groundwater	Trench AE-3	Kaplan <i>et al.</i> (1998, Batch)
8.44	1.37		6.4	14.8	Hanford Groundwater	Trench AE-3	Kaplan <i>et al.</i> (1998, Batch)
9.12	2.12		6.4	14.8	Hanford Groundwater	Trench AE-3	Kaplan <i>et al.</i> (1998, Batch)
8.46	0.90		6.4	14.8	Hanford Groundwater	Trench AE-3	Kaplan <i>et al.</i> (1996, 100% Unsaturated Batch)
8.46	1.70		5.3	6.3	Hanford Groundwater	Trench 94	Kaplan <i>et al.</i> (1996, 100% Unsaturated Batch)
8.46	1.00		6.0	6.3	Hanford Groundwater	TSB-1	Kaplan <i>et al.</i> (1996, 100% Unsaturated Batch)
8.46	1.10		6.4	14.8	Hanford Groundwater	Trench AE-3	Kaplan <i>et al.</i> (1996, Batch)
8.46	3.50		5.3	6.3	Hanford Groundwater	Trench 94	Kaplan <i>et al.</i> (1996, Batch)
8.46	2.10		6.0	6.3	Hanford Groundwater	TSB-1	Kaplan <i>et al.</i> (1996, Batch)
8.46	0.24		6.4	14.8	Hanford Groundwater	Trench AE-3	Kaplan <i>et al.</i> (1996)
8.46	0.64		6.4	14.8	Hanford Groundwater	Trench AE-3	Kaplan <i>et al.</i> (1996)
8.46	0.51		6.4	14.8	Hanford Groundwater	Trench AE-3	Kaplan <i>et al.</i> (1996)
8.46	0.46		6.4	14.8	Hanford Groundwater	Trench AE-3	Kaplan <i>et al.</i> (1996)
8.46	0.35		6.4	14.8	Hanford Groundwater	Trench AE-3	Kaplan <i>et al.</i> (1996)
8.46	0.53		6.4	14.8	Hanford Groundwater	Trench AE-3	Kaplan <i>et al.</i> (1996)
8.46	0.23		5.3	6.3	Hanford Groundwater	Trench 94	Kaplan <i>et al.</i> (1996)
8.46	0.15		5.3	6.3	Hanford Groundwater	Trench 94	Kaplan <i>et al.</i> (1996)
8.46	0.1		5.3	6.3	Hanford Groundwater	Trench 94	Kaplan <i>et al.</i> (1996)
8.46	0.16		5.3	6.3	Hanford Groundwater	Trench 94	Kaplan <i>et al.</i> (1996)

pH	U Kd (ml/g)	Clay Cont. (wt.%)	CEC (meq/100g)	Surface Area (m ² /g)	Solution	Soil Identification	Reference / Comments
8.46	0.12		5.3	6.3	Hanford Groundwater	Trench 94	Kaplan <i>et al.</i> (1996)
	2	8				Sand	Neiheisel [1983, as listed in Thibault <i>et al.</i> (1990)]
	1	7				Sand	Neiheisel [1983, as listed in Thibault <i>et al.</i> (1990)]
	3	15				Sand	Neiheisel [1983, as listed in Thibault <i>et al.</i> (1990)]
	750	36				Clayey Sand	Neiheisel [1983, as listed in Thibault <i>et al.</i> (1990)]
	770	21				Clayey Sand	Neiheisel [1983, as listed in Thibault <i>et al.</i> (1990)]
	550	19				Clayey Sand	Neiheisel [1983, as listed in Thibault <i>et al.</i> (1990)]
2.00	100					Fine Sandstone and Silty Sand	Haji-Djafari <i>et al.</i> [1981, as listed in Thibault <i>et al.</i> (1990)]
4.50	200					Fine Sandstone and Silty Sand	Haji-Djafari <i>et al.</i> [1981, as listed in Thibault <i>et al.</i> (1990)]
5.75	1,000					Fine Sandstone and Silty Sand	Haji-Djafari <i>et al.</i> [1981, as listed in Thibault <i>et al.</i> (1990)]
7.00	2,000					Fine Sandstone and Silty Sand	Haji-Djafari <i>et al.</i> [1981, as listed in Thibault <i>et al.</i> (1990)]
5.6	25,000					Red-Brown Clayey	Seeley and Kelmers [1984, as listed in Thibault <i>et al.</i> (1990)]
5.6	250					Red-Brown Clayey	Seeley and Kelmers [1984, as listed in Thibault <i>et al.</i> (1990)]
5.20	58.4						Thibault <i>et al.</i> (1990, values determined by coworkers)
5.10	294.9						Thibault <i>et al.</i> (1990, values determined by coworkers)
5.20	160						Thibault <i>et al.</i> (1990, values determined by coworkers)
6.20	45.4						Thibault <i>et al.</i> (1990, values determined by coworkers)

pH	U Kd (ml/g)	Clay Cont. (wt.%)	CEC (meq/100g)	Surface Area (m ² /g)	Solution	Soil Identification	Reference / Comments
7.00	450	36	28.0			Silty Loam Clay	Thibault <i>et al.</i> (1990, values determined by coworkers)
7.30	1.2	15	17.0			Loam	Thibault <i>et al.</i> (1990, values determined by coworkers)
4.90	0.03	2	5.8			Medium Sand	Thibault <i>et al.</i> (1990, values determined by coworkers)
5.50	2900	1	120.0			Organic	Thibault <i>et al.</i> (1990, values determined by coworkers)
7.40	1.9	10	9.1			Fine Sandy Loam	Thibault <i>et al.</i> (1990, values determined by coworkers)
7.40	2.4	11	8.7			Fine Sandy Loam	Thibault <i>et al.</i> (1990, values determined by coworkers)
6.60	590	10	10.8			Fine Sandy Loam	Thibault <i>et al.</i> (1990, values determined by coworkers)
6.50	4500	10	12.6			Fine Sandy Loam	Thibault <i>et al.</i> (1990, values determined by coworkers)
7.10	15	12	13.4			Fine Sandy Loam	Thibault <i>et al.</i> (1990, values determined by coworkers)
7.00	16					Sand	Rancon [1973, as listed in Thibault <i>et al.</i> (1990)]
7.00	33					Organic Peat	Rancon [1973, as listed in Thibault <i>et al.</i> (1990)]
6.50	4400					Clay Fraction	Dahlman <i>et al.</i> [1976, as listed in Thibault <i>et al.</i> (1990)]
2.80	200					Abyssal Red Clay	Erickson (1980)
7.10	790,000					Abyssal Red Clay	Erickson (1980)
8.3	1.70		2.6		Hanford Groundwater	CGS-1 sand (coarse gravel sand)	Serne <i>et al.</i> (1993, Batch)
8.3	2.30		5.2		Hanford Groundwater	Trench 8 Loamy Sand (medium/coarse sand)	Serne <i>et al.</i> (1993, Batch)
8.3	79.30		6.0		Hanford Groundwater	TBS-1 Loamy Sand (Touchet Bed sand)	Serne <i>et al.</i> (1993, Batch)
8.00	56.0				Hanford Groundwater, GR-1	Umtanum Basalt	Salter <i>et al.</i> (1981)
8.00	7.5				Hanford Groundwater, GR-1	Umtanum Basalt	Salter <i>et al.</i> (1981)

pH	U Kd (ml/g)	Clay Cont. (wt.%)	CEC (meq/100g)	Surface Area (m ² /g)	Solution	Soil Identification	Reference / Comments
8.00	13.2				Hanford Groundwater, GR-1	Umtanum Basalt	Salter <i>et al.</i> (1981)
8.00	17.8				Hanford Groundwater, GR-1	Umtanum Basalt	Salter <i>et al.</i> (1981)
8.00	20.2				Hanford Groundwater, GR-1	Umtanum Basalt	Salter <i>et al.</i> (1981)
8.00	13.0				Hanford Groundwater, GR-1	Flow E Basalt	Salter <i>et al.</i> (1981)
8.00	2.7				Hanford Groundwater, GR-1	Flow E Basalt	Salter <i>et al.</i> (1981)
8.00	2.2				Hanford Groundwater, GR-1	Flow E Basalt	Salter <i>et al.</i> (1981)
8.00	3.2				Hanford Groundwater, GR-1	Flow E Basalt	Salter <i>et al.</i> (1981)
8.00	2.9				Hanford Groundwater, GR-1	Flow E Basalt	Salter <i>et al.</i> (1981)
8.00	16.0				Hanford Groundwater,GR-1	Pomona Basalt	Salter <i>et al.</i> (1981)
8.00	2.2				Hanford Groundwater,GR-1	Pomona Basalt	Salter <i>et al.</i> (1981)
8.00	3.5				Hanford Groundwater,GR-1	Pomona Basalt	Salter <i>et al.</i> (1981)
8.00	5.2				Hanford Groundwater,GR-1	Pomona Basalt	Salter <i>et al.</i> (1981)
8.00	5.8				Hanford Groundwater,GR-1	Pomona Basalt	Salter <i>et al.</i> (1981)
10.00	2.8				Hanford Groundwater,GR-2	Umtanum Basalt	Salter <i>et al.</i> (1981)
10.00	2.3				Hanford Groundwater,GR-2	Umtanum Basalt	Salter <i>et al.</i> (1981)
10.00	2.8				Hanford Groundwater,GR-2	Umtanum Basalt	Salter <i>et al.</i> (1981)
10.00	2.8				Hanford Groundwater,GR-2	Umtanum Basalt	Salter <i>et al.</i> (1981)
10.00	2.5				Hanford Groundwater,GR-2	Umtanum Basalt	Salter <i>et al.</i> (1981)
10.00	1.0				Hanford Groundwater,GR-2	Flow E Basalt	Salter <i>et al.</i> (1981)
10.00	0.5				Hanford Groundwater,GR-2	Flow E Basalt	Salter <i>et al.</i> (1981)
10.00	0.4				Hanford Groundwater,GR-2	Flow E Basalt	Salter <i>et al.</i> (1981)
10.00	0.8				Hanford Groundwater,GR-2	Flow E Basalt	Salter <i>et al.</i> (1981)
10.00	0.2				Hanford Groundwater,GR-2	Flow E Basalt	Salter <i>et al.</i> (1981)
10.00	0.9				Hanford Groundwater,GR-2	Pomona Basalt	Salter <i>et al.</i> (1981)
10.00	0.6				Hanford Groundwater,GR-2	Pomona Basalt	Salter <i>et al.</i> (1981)
10.00	0.8				Hanford Groundwater,GR-2	Pomona Basalt	Salter <i>et al.</i> (1981)
10.00	0.5				Hanford Groundwater,GR-2	Pomona Basalt	Salter <i>et al.</i> (1981)
10.00	0.4				Hanford Groundwater,GR-2	Pomona Basalt	Salter <i>et al.</i> (1981)
7.66	7.5		1.83	17.7	Hanford Groundwater,GR-1	Umtanum Basalt	Ames <i>et al.</i> (1982)

pH	U Kd (ml/g)	Clay Cont. (wt.%)	CEC (meq/100g)	Surface Area (m ² /g)	Solution	Soil Identification	Reference / Comments
7.66	13		1.83	17.7	Hanford Groundwater,GR-1	Umtanum Basalt	Ames <i>et al.</i> (1982)
7.66	18		1.83	17.7	Hanford Groundwater,GR-1	Umtanum Basalt	Ames <i>et al.</i> (1982)
7.66	20		1.83	17.7	Hanford Groundwater,GR-1	Umtanum Basalt	Ames <i>et al.</i> (1982)
8.38	2.4		1.83	17.7	Hanford Groundwater,GR-2	Umtanum Basalt	Ames <i>et al.</i> (1982)
8.38	2.9		1.83	17.7	Hanford Groundwater,GR-2	Umtanum Basalt	Ames <i>et al.</i> (1982)
8.38	2.9		1.83	17.7	Hanford Groundwater,GR-2	Umtanum Basalt	Ames <i>et al.</i> (1982)
8.38	2.5		1.83	17.7	Hanford Groundwater,GR-2	Umtanum Basalt	Ames <i>et al.</i> (1982)
7.65	2.7		1.5	10.3	Hanford Groundwater,GR-1	Flow E Basalt	Ames <i>et al.</i> (1982)
7.65	2.2		1.5	10.3	Hanford Groundwater,GR-1	Flow E Basalt	Ames <i>et al.</i> (1982)
7.65	3.2		1.5	10.3	Hanford Groundwater,GR-1	Flow E Basalt	Ames <i>et al.</i> (1982)
7.65	2.9		1.5	10.3	Hanford Groundwater,GR-1	Flow E Basalt	Ames <i>et al.</i> (1982)
8.38	0.55		1.5	10.3	Hanford Groundwater,GR-2	Flow E Basalt	Ames <i>et al.</i> (1982)
8.38	0.38		1.5	10.3	Hanford Groundwater,GR-2	Flow E Basalt	Ames <i>et al.</i> (1982)
8.38	0.78		1.5	10.3	Hanford Groundwater,GR-2	Flow E Basalt	Ames <i>et al.</i> (1982)
8.38	0.19		1.5	10.3	Hanford Groundwater,GR-2	Flow E Basalt	Ames <i>et al.</i> (1982)
7.90	2.2		4.84	31.2	Hanford Groundwater,GR-1	Pomona Basalt	Ames <i>et al.</i> (1982)
7.90	3.5		4.84	31.2	Hanford Groundwater,GR-1	Pomona Basalt	Ames <i>et al.</i> (1982)
7.90	5.2		4.84	31.2	Hanford Groundwater,GR-1	Pomona Basalt	Ames <i>et al.</i> (1982)
7.90	5.8		4.84	31.2	Hanford Groundwater,GR-1	Pomona Basalt	Ames <i>et al.</i> (1982)
8.48	0.57		4.84	31.2	Hanford Groundwater,GR-2	Pomona Basalt	Ames <i>et al.</i> (1982)
8.48	0.83		4.84	31.2	Hanford Groundwater,GR-2	Pomona Basalt	Ames <i>et al.</i> (1982)
8.48	0.47		4.84	31.2	Hanford Groundwater,GR-2	Pomona Basalt	Ames <i>et al.</i> (1982)
8.48	0.42		4.84	31.2	Hanford Groundwater,GR-2	Pomona Basalt	Ames <i>et al.</i> (1982)
7.7	27		71.66	646	Hanford Groundwater,GR-1	Smectite, secondary	Ames <i>et al.</i> (1982)
7.7	39		4.84	31.2	Hanford Groundwater,GR-1	Smectite, secondary	Ames <i>et al.</i> (1982)
7.7	127		4.84	31.2	Hanford Groundwater,GR-1	Smectite, secondary	Ames <i>et al.</i> (1982)
7.7	76		4.84	31.2	Hanford Groundwater,GR-1	Smectite, secondary	Ames <i>et al.</i> (1982)
7.7	12		4.84	31.2	Hanford Groundwater,GR-2	Smectite, secondary	Ames <i>et al.</i> (1982)
7.7	42		4.84	31.2	Hanford Groundwater,GR-2	Smectite, secondary	Ames <i>et al.</i> (1982)
7.7	48		4.84	31.2	Hanford Groundwater,GR-2	Smectite, secondary	Ames <i>et al.</i> (1982)
7.7	22		4.84	31.2	Hanford Groundwater,GR-2	Smectite, secondary	Ames <i>et al.</i> (1982)
6.85	477,285				0.01 NaCl	Amor Fe(III) Hydroxide	Ames <i>et al.</i> (1983c)
6.80	818,221				0.01 NaCl	Amor Fe(III) Hydroxide	Ames <i>et al.</i> (1983c)

pH	U Kd (ml/g)	Clay Cont. (wt.%)	CEC (meq/100g)	Surface Area (m ² /g)	Solution	Soil Identification	Reference / Comments
6.90	1,739,877				0.01 NaCl	Amor Fe(III) Hydroxide	Ames <i>et al.</i> (1983c)
6.90	1,690,522				0.01 NaCl	Amor Fe(III) Hydroxide	Ames <i>et al.</i> (1983c)
8.60	4,313				0.01 NaHCO ₃	Amor Fe(III) Hydroxide	Ames <i>et al.</i> (1983c)
8.65	14,098				0.01 NaHCO ₃	Amor Fe(III) Hydroxide	Ames <i>et al.</i> (1983c)
8.65	21,362				0.01 NaHCO ₃	Amor Fe(III) Hydroxide	Ames <i>et al.</i> (1983c)
8.80	26,269				0.01 NaHCO ₃	Amor Fe(III) Hydroxide	Ames <i>et al.</i> (1983c)
7.15	8.4		15.3	1.59	0.01 NaCl	Biotite	Ames <i>et al.</i> (1983b)
7.15	43.9		15.3	1.59	0.01 NaCl	Biotite	Ames <i>et al.</i> (1983b)
7.15	253.5		15.3	1.59	0.01 NaCl	Biotite	Ames <i>et al.</i> (1983b)
7.15	544.3		15.3	1.59	0.01 NaCl	Biotite	Ames <i>et al.</i> (1983b)
7.15	113.7		0.95	1.88	0.01 NaCl	Muscovite	Ames <i>et al.</i> (1983b)
7.15	251.0		0.95	1.88	0.01 NaCl	Muscovite	Ames <i>et al.</i> (1983b)
7.15	459.7		0.95	1.88	0.01 NaCl	Muscovite	Ames <i>et al.</i> (1983b)
7.15	68.2		0.95	1.88	0.01 NaCl	Muscovite	Ames <i>et al.</i> (1983b)
7.15	67.9		1.17	1.22	0.01 NaCl	Phlogopite	Ames <i>et al.</i> (1983b)
7.15	85.4		1.17	1.22	0.01 NaCl	Phlogopite	Ames <i>et al.</i> (1983b)
7.15	95.4		1.17	1.22	0.01 NaCl	Phlogopite	Ames <i>et al.</i> (1983b)
8.65	0.9		15.3	1.59	0.01 NaHCO ₃	Biotite	Ames <i>et al.</i> (1983b)
8.65	3.4		15.3	1.59	0.01 NaHCO ₃	Biotite	Ames <i>et al.</i> (1983b)
8.65	23.0		15.3	1.59	0.01 NaHCO ₃	Biotite	Ames <i>et al.</i> (1983b)
8.65	80.8		15.3	1.59	0.01 NaHCO ₃	Biotite	Ames <i>et al.</i> (1983b)
8.65	2.2		0.95	1.88	0.01 NaHCO ₃	Muscovite	Ames <i>et al.</i> (1983b)
8.65	26.9		0.95	1.88	0.01 NaHCO ₃	Muscovite	Ames <i>et al.</i> (1983b)
8.65	602.5		0.95	1.88	0.01 NaHCO ₃	Muscovite	Ames <i>et al.</i> (1983b)
8.65	3489.6		0.95	1.88	0.01 NaHCO ₃	Muscovite	Ames <i>et al.</i> (1983b)
8.65	0.6		1.17	1.22	0.01 NaHCO ₃	Phlogopite	Ames <i>et al.</i> (1983b)
8.65	1.1		1.17	1.22	0.01 NaHCO ₃	Phlogopite	Ames <i>et al.</i> (1983b)
8.65	0.6		1.17	1.22	0.01 NaHCO ₃	Phlogopite	Ames <i>et al.</i> (1983b)
7	544.5		25	116.1	0.01 NaCl	Illite, only lowest U conc	Ames <i>et al.</i> (1983a)

pH	U Kd (ml/g)	Clay Cont. (wt.%)	CEC (meq/100g)	Surface Area (m ² /g)	Solution	Soil Identification	Reference / Comments
8.5	90.5		25	116.1	0.01 NaHCO ₃	Illite, only lowest U conc	Ames <i>et al.</i> (1983a)
7	657.8		12.2	68.3	0.01 NaCl	Kaolinite, only lowest U conc	Ames <i>et al.</i> (1983a)
8.5	400.8		12.2	68.3	0.01 NaHCO ₃	Kaolinite, only lowest U conc	Ames <i>et al.</i> (1983a)
7	542.0		120	747	0.01 NaCl	Montmorillonite, only lowest U conc	Ames <i>et al.</i> (1983a)
8.5	1.8		120	747	0.01 NaHCO ₃	Montmorillonite, only lowest U conc	Ames <i>et al.</i> (1983a)
7	299.9		95	861	0.01 NaCl	Nontronite, only lowest U conc	Ames <i>et al.</i> (1983a)
8.5	4.1		95	861	0.01 NaHCO ₃	Nontronite, only lowest U conc	Ames <i>et al.</i> (1983a)
7	138.0		16.03	137.3	0.01 NaCl	Glauconite, only lowest U conc	Ames <i>et al.</i> (1983a)
8.5	114.2		16.03	137.3	0.01 NaHCO ₃	Glauconite, only lowest U conc	Ames <i>et al.</i> (1983a)
7	66.5		140.2	20	0.01 NaCl	Clinoptilolite, only lowest U conc	Ames <i>et al.</i> (1983a)
8.5	0.6		140.2	20	0.01 NaHCO ₃	Clinoptilolite, only lowest U conc	Ames <i>et al.</i> (1983a)
7	225.7		3.18	46.8	0.01 NaCl	Opal, only lowest U conc	Ames <i>et al.</i> (1983a)
8.5	1.7		3.18	46.8	0.01 NaHCO ₃	Opal, only lowest U conc	Ames <i>et al.</i> (1983a)
7	300.5		2.79	626.3	0.01 NaCl	Silica Gel,, only lowest U conc	Ames <i>et al.</i> (1983a)
8.5	639.9		2.79	626.3	0.01 NaHCO ₃	Silica Gel,, only lowest U conc	Ames <i>et al.</i> (1983a)
7.3	4200.0		4.36			Spesutie (silt loam)	Erikson <i>et al.</i> (1993)
6.2	136.0		1.29			Transonic (silt loam)	Erikson <i>et al.</i> (1993)
8.0	44		9.30			Yuma (sandy loam)	Erikson <i>et al.</i> (1993)
6.8	4360		4.36			Spesutie (silt loam)	Erikson <i>et al.</i> (1993)
5.6	328		1.29			Transonic (silt loam)	Erikson <i>et al.</i> (1993)
8.0	54		9.30			Yuma (sandy loam)	Erikson <i>et al.</i> (1993)
	39					River Sediment (Quartz, clay, calcite, organic matter)	Rancon (1973) as cited by Ames and Rai (1978)
	33					River Peat	Rancon (1973) as cited by Ames and Rai (1978)

pH	U Kd (ml/g)	Clay Cont. (wt.%)	CEC (meq/100g)	Surface Area (m ² /g)	Solution	Soil Identification	Reference / Comments
	16					River Sediment (Quartz, clay, calcite)	Rancon (1973) as cited by Ames and Rai (1978)
	270					Soil (Quartz and Clay, from Altered Schist)	Rancon (1973) as cited by Ames and Rai (1978)
	0					Quartz	Rancon (1973) as cited by Ames and Rai (1978)
	7					Calcite	Rancon (1973) as cited by Ames and Rai (1978)
	139					Illite	Rancon (1973) as cited by Ames and Rai (1978)
	27 (0.8-332)				Fresh Water	Gorleben Salt Dome, Sandy Sediment	Warnecke <i>et al.</i> (1984, 1986, 1994), Warnecke and Hild (1988)
	1 (0.3-1.6)				Fresh Water	Gorleben Salt Dome, Sandy Sediment	Warnecke <i>et al.</i> (1984, 1986, 1994), Warnecke and Hild (1988)
	17 (8.5-100)				Saline Water	Gorleben Salt Dome, Clayish Sediment	Warnecke <i>et al.</i> (1984, 1986, 1994), Warnecke and Hild (1988)
	14-1,400				Saline Water	Gorleben Salt Dome, Clayish Sediment	Warnecke <i>et al.</i> (1984, 1986, 1994), Warnecke and Hild (1988)
	4				Quaternary fresh water	Former Konrad Iron Ore Mine	Warnecke <i>et al.</i> (1986), Warnecke and Hild (1988)
	6				Turonian fresh water	Former Konrad Iron Ore Mine	Warnecke <i>et al.</i> (1986), Warnecke and Hild (1988)
	6				Cenomanian saline water	Former Konrad Iron Ore Mine	Warnecke <i>et al.</i> (1986), Warnecke and Hild (1988)
	20				Albian (Hauterivain) saline water	Former Konrad Iron Ore Mine	Warnecke <i>et al.</i> (1986), Warnecke and Hild (1988)
	1.4				Albian (Hils) saline water	Former Konrad Iron Ore Mine	Warnecke <i>et al.</i> (1986), Warnecke and Hild (1988)
	2.6				Kimmeridgian saline water	Former Konrad Iron Ore Mine	Warnecke <i>et al.</i> (1986), Warnecke and Hild (1988)
	3				Oxfordian saline water	Former Konrad Iron Ore Mine	Warnecke <i>et al.</i> (1986), Warnecke and Hild (1988)
	3				Bajocian (Dogger) saline water	Former Konrad Iron Ore Mine	Warnecke <i>et al.</i> (1986), Warnecke and Hild (1988)
3.83	310				Synthetic Groundwater, function of pH	Kaolinite	Giblin (1980)
3.90	235				Synthetic Groundwater, function of pH	Kaolinite	Giblin (1980)

pH	U Kd (ml/g)	Clay Cont. (wt.%)	CEC (meq/100g)	Surface Area (m ² /g)	Solution	Soil Identification	Reference / Comments
3.94	741				Synthetic Groundwater, function of pH	Kaolinite	Giblin (1980)
3.96	211				Synthetic Groundwater, function of pH	Kaolinite	Giblin (1980)
4.03	694				Synthetic Groundwater, function of pH	Kaolinite	Giblin (1980)
4.13	720				Synthetic Groundwater, function of pH	Kaolinite	Giblin (1980)
4.28	898				Synthetic Groundwater, function of pH	Kaolinite	Giblin (1980)
4.33	630				Synthetic Groundwater, function of pH	Kaolinite	Giblin (1980)
4.36	247				Synthetic Groundwater, function of pH	Kaolinite	Giblin (1980)
4.53	264				Synthetic Groundwater, function of pH	Kaolinite	Giblin (1980)
4.58	903				Synthetic Groundwater, function of pH	Kaolinite	Giblin (1980)
4.61	324				Synthetic Groundwater, function of pH	Kaolinite	Giblin (1980)
4.71	522				Synthetic Groundwater, function of pH	Kaolinite	Giblin (1980)
4.81	1,216				Synthetic Groundwater, function of pH	Kaolinite	Giblin (1980)
4.95	1,185				Synthetic Groundwater, function of pH	Kaolinite	Giblin (1980)
4.84	3,381				Synthetic Groundwater, function of pH	Kaolinite	Giblin (1980)
5.00	2,561				Synthetic Groundwater, function of pH	Kaolinite	Giblin (1980)
5.10	2,635				Synthetic Groundwater, function of pH	Kaolinite	Giblin (1980)
5.11	3,807				Synthetic Groundwater, function of pH	Kaolinite	Giblin (1980)
5.19	4,293				Synthetic Groundwater, function of pH	Kaolinite	Giblin (1980)
5.52	4,483				Synthetic Groundwater, function of pH	Kaolinite	Giblin (1980)
5.15	4,574				Synthetic Groundwater, function of pH	Kaolinite	Giblin (1980)
5.24	5,745				Synthetic Groundwater, function of pH	Kaolinite	Giblin (1980)
5.16	7,423				Synthetic Groundwater, function of pH	Kaolinite	Giblin (1980)

pH	U Kd (ml/g)	Clay Cont. (wt.%)	CEC (meq/100g)	Surface Area (m ² /g)	Solution	Soil Identification	Reference / Comments
5.28	3,214				Synthetic Groundwater, function of pH	Kaolinite	Giblin (1980)
5.52	5,564				Synthetic Groundwater, function of pH	Kaolinite	Giblin (1980)
5.44	6,687				Synthetic Groundwater, function of pH	Kaolinite	Giblin (1980)
5.54	6,185				Synthetic Groundwater, function of pH	Kaolinite	Giblin (1980)
5.58	6,615				Synthetic Groundwater, function of pH	Kaolinite	Giblin (1980)
5.85	7,124				Synthetic Groundwater, function of pH	Kaolinite	Giblin (1980)
5.45	8,146				Synthetic Groundwater, function of pH	Kaolinite	Giblin (1980)
5.56	8,506				Synthetic Groundwater, function of pH	Kaolinite	Giblin (1980)
5.74	9,332				Synthetic Groundwater, function of pH	Kaolinite	Giblin (1980)
5.50	10,462				Synthetic Groundwater, function of pH	Kaolinite	Giblin (1980)
5.69	10,681				Synthetic Groundwater, function of pH	Kaolinite	Giblin (1980)
5.54	11,770				Synthetic Groundwater, function of pH	Kaolinite	Giblin (1980)
5.66	13,616				Synthetic Groundwater, function of pH	Kaolinite	Giblin (1980)
5.81	14,675				Synthetic Groundwater, function of pH	Kaolinite	Giblin (1980)
5.86	14,417				Synthetic Groundwater, function of pH	Kaolinite	Giblin (1980)
5.75	20,628				Synthetic Groundwater, function of pH	Kaolinite	Giblin (1980)
6.01	24,082				Synthetic Groundwater, function of pH	Kaolinite	Giblin (1980)
6.20	22,471				Synthetic Groundwater, function of pH	Kaolinite	Giblin (1980)
5.95	26,354				Synthetic Groundwater, function of pH	Kaolinite	Giblin (1980)
6.35	26,078				Synthetic Groundwater, function of pH	Kaolinite	Giblin (1980)
6.40	25,601				Synthetic Groundwater, function of pH	Kaolinite	Giblin (1980)
6.35	27,671				Synthetic Groundwater, function of pH	Kaolinite	Giblin (1980)

pH	U Kd (ml/g)	Clay Cont. (wt.%)	CEC (meq/100g)	Surface Area (m ² /g)	Solution	Soil Identification	Reference / Comments
6.46	30,529				Synthetic Groundwater, function of pH	Kaolinite	Giblin (1980)
6.13	31,477				Synthetic Groundwater, function of pH	Kaolinite	Giblin (1980)
6.26	33,305				Synthetic Groundwater, function of pH	Kaolinite	Giblin (1980)
6.80	37,129				Synthetic Groundwater, function of pH	Kaolinite	Giblin (1980)
6.86	37,657				Synthetic Groundwater, function of pH	Kaolinite	Giblin (1980)
6.81	32,312				Synthetic Groundwater, function of pH	Kaolinite	Giblin (1980)
7.10	29,390				Synthetic Groundwater, function of pH	Kaolinite	Giblin (1980)
7.85	33,583				Synthetic Groundwater, function of pH	Kaolinite	Giblin (1980)
7.67	26,518				Synthetic Groundwater, function of pH	Kaolinite	Giblin (1980)
8.40	30,523				Synthetic Groundwater, function of pH	Kaolinite	Giblin (1980)
8.51	19,632				Synthetic Groundwater, function of pH	Kaolinite	Giblin (1980)
9.45	23,177				Synthetic Groundwater, function of pH	Kaolinite	Giblin (1980)
9.80	17,763				Synthetic Groundwater, function of pH	Kaolinite	Giblin (1980)
9.90	14,499				Synthetic Groundwater, function of pH	Kaolinite	Giblin (1980)
3.8	2				Synthetic Groundwater, function of pH	Quartz	Andersson <i>et al.</i> (1982)
3.5	5				Synthetic Groundwater, function of pH	Quartz	Andersson <i>et al.</i> (1982)
3.7	8				Synthetic Groundwater, function of pH	Quartz	Andersson <i>et al.</i> (1982)
3.7	69				Synthetic Groundwater, function of pH	Quartz	Andersson <i>et al.</i> (1982)
4.0	116				Synthetic Groundwater, function of pH	Quartz	Andersson <i>et al.</i> (1982)
6.4	1,216				Synthetic Groundwater, function of pH	Quartz	Andersson <i>et al.</i> (1982)
6.5	1,824				Synthetic Groundwater, function of pH	Quartz	Andersson <i>et al.</i> (1982)

pH	U Kd (ml/g)	Clay Cont. (wt.%)	CEC (meq/100g)	Surface Area (m ² /g)	Solution	Soil Identification	Reference / Comments
6.6	2,679				Synthetic Groundwater, function of pH	Quartz	Andersson <i>et al.</i> (1982)
7.7	7,379				Synthetic Groundwater, function of pH	Quartz	Andersson <i>et al.</i> (1982)
8.0	2,506				Synthetic Groundwater, function of pH	Quartz	Andersson <i>et al.</i> (1982)
8.3	21,979				Synthetic Groundwater, function of pH	Quartz	Andersson <i>et al.</i> (1982)
8.6	3,999				Synthetic Groundwater, function of pH	Quartz	Andersson <i>et al.</i> (1982)
9.0	14,689				Synthetic Groundwater, function of pH	Quartz	Andersson <i>et al.</i> (1982)
3.4	27				Synthetic Groundwater, function of pH	Biotite	Andersson <i>et al.</i> (1982)
4.4	326				Synthetic Groundwater, function of pH	Biotite	Andersson <i>et al.</i> (1982)
4.4	522				Synthetic Groundwater, function of pH	Biotite	Andersson <i>et al.</i> (1982)
4.7	418				Synthetic Groundwater, function of pH	Biotite	Andersson <i>et al.</i> (1982)
5.1	1,489				Synthetic Groundwater, function of pH	Biotite	Andersson <i>et al.</i> (1982)
5.2	2,512				Synthetic Groundwater, function of pH	Biotite	Andersson <i>et al.</i> (1982)
6.4	2,812				Synthetic Groundwater, function of pH	Biotite	Andersson <i>et al.</i> (1982)
7.3	7,228				Synthetic Groundwater, function of pH	Biotite	Andersson <i>et al.</i> (1982)
7.3	16,634				Synthetic Groundwater, function of pH	Biotite	Andersson <i>et al.</i> (1982)
7.4	9,840				Synthetic Groundwater, function of pH	Biotite	Andersson <i>et al.</i> (1982)
8.1	4,732				Synthetic Groundwater, function of pH	Biotite	Andersson <i>et al.</i> (1982)
9.0	8,337				Synthetic Groundwater, function of pH	Biotite	Andersson <i>et al.</i> (1982)
3.3	207				Synthetic Groundwater, function of pH	Apatite	Andersson <i>et al.</i> (1982)
3.8	324				Synthetic Groundwater, function of pH	Apatite	Andersson <i>et al.</i> (1982)
4.0	726				Synthetic Groundwater, function of pH	Apatite	Andersson <i>et al.</i> (1982)

pH	U Kd (ml/g)	Clay Cont. (wt.%)	CEC (meq/100g)	Surface Area (m ² /g)	Solution	Soil Identification	Reference / Comments
4.0	668				Synthetic Groundwater, function of pH	Apatite	Andersson <i>et al.</i> (1982)
4.4	3,767				Synthetic Groundwater, function of pH	Apatite	Andersson <i>et al.</i> (1982)
4.5	4,732				Synthetic Groundwater, function of pH	Apatite	Andersson <i>et al.</i> (1982)
5.0	16,218				Synthetic Groundwater, function of pH	Apatite	Andersson <i>et al.</i> (1982)
5.3	8,241				Synthetic Groundwater, function of pH	Apatite	Andersson <i>et al.</i> (1982)
6.0	140,605				Synthetic Groundwater, function of pH	Apatite	Andersson <i>et al.</i> (1982)
7.7	24,660				Synthetic Groundwater, function of pH	Apatite	Andersson <i>et al.</i> (1982)
3.6	460				Synthetic Groundwater, function of pH	Attapulгите (Palygorskite)	Andersson <i>et al.</i> (1982)
4.1	1,514				Synthetic Groundwater, function of pH	Attapulгите (Palygorskite)	Andersson <i>et al.</i> (1982)
4.2	7,194				Synthetic Groundwater, function of pH	Attapulгите (Palygorskite)	Andersson <i>et al.</i> (1982)
4.5	6,471				Synthetic Groundwater, function of pH	Attapulгите (Palygorskite)	Andersson <i>et al.</i> (1982)
4.7	4,753				Synthetic Groundwater, function of pH	Attapulгите (Palygorskite)	Andersson <i>et al.</i> (1982)
5.1	23,335				Synthetic Groundwater, function of pH	Attapulгите (Palygorskite)	Andersson <i>et al.</i> (1982)
5.9	12,531				Synthetic Groundwater, function of pH	Attapulгите (Palygorskite)	Andersson <i>et al.</i> (1982)
6.4	266,686				Synthetic Groundwater, function of pH	Attapulгите (Palygorskite)	Andersson <i>et al.</i> (1982)
7.3	645,654				Synthetic Groundwater, function of pH	Attapulгите (Palygorskite)	Andersson <i>et al.</i> (1982)
7.8	82,224				Synthetic Groundwater, function of pH	Attapulгите (Palygorskite)	Andersson <i>et al.</i> (1982)
8.7	46,132				Synthetic Groundwater, function of pH	Attapulгите (Palygorskite)	Andersson <i>et al.</i> (1982)
3.2	1,175				Synthetic Groundwater, function of pH	Montimorillonite	Andersson <i>et al.</i> (1982)
4.4	12,503				Synthetic Groundwater, function of pH	Montimorillonite	Andersson <i>et al.</i> (1982)
6.6	3,917				Synthetic Groundwater, function of pH	Montimorillonite	Andersson <i>et al.</i> (1982)

pH	U Kd (ml/g)	Clay Cont. (wt.%)	CEC (meq/100g)	Surface Area (m ² /g)	Solution	Soil Identification	Reference / Comments
7.0	10,139				Synthetic Groundwater, function of pH	Montimorillonite	Andersson <i>et al.</i> (1982)
7.0	28,054				Synthetic Groundwater, function of pH	Montimorillonite	Andersson <i>et al.</i> (1982)
7.3	10,715				Synthetic Groundwater, function of pH	Montimorillonite	Andersson <i>et al.</i> (1982)
8.2	21,528				Synthetic Groundwater, function of pH	Montimorillonite	Andersson <i>et al.</i> (1982)
8.4	20,370				Synthetic Groundwater, function of pH	Montimorillonite	Andersson <i>et al.</i> (1982)
9.0	18,621				Synthetic Groundwater, function of pH	Montimorillonite	Andersson <i>et al.</i> (1982)
5.1	7,391		45	99	Ca Electrolyte, CO ₂ Free	Kenoma Clay, <2um fraction	Zachara <i>et al.</i> (1992, Fig 6)
5.0	1,177		45	99	Ca Electrolyte, CO ₂ Free	Kenoma Clay, <2um fraction	Zachara <i>et al.</i> (1992, Fig 6)
5.1	2,180		45	99	Ca Electrolyte, CO ₂ Free	Kenoma Clay, <2um fraction	Zachara <i>et al.</i> (1992, Fig 6)
5.4	3,680		45	99	Ca Electrolyte, CO ₂ Free	Kenoma Clay, <2um fraction	Zachara <i>et al.</i> (1992, Fig 6)
5.3	4,437		45	99	Ca Electrolyte, CO ₂ Free	Kenoma Clay, <2um fraction	Zachara <i>et al.</i> (1992, Fig 6)
5.5	7,265		45	99	Ca Electrolyte, CO ₂ Free	Kenoma Clay, <2um fraction	Zachara <i>et al.</i> (1992, Fig 6)
5.5	7,108		45	99	Ca Electrolyte, CO ₂ Free	Kenoma Clay, <2um fraction	Zachara <i>et al.</i> (1992, Fig 6)
5.8	23,603		45	99	Ca Electrolyte, CO ₂ Free	Kenoma Clay, <2um fraction	Zachara <i>et al.</i> (1992, Fig 6)
5.8	22,948		45	99	Ca Electrolyte, CO ₂ Free	Kenoma Clay, <2um fraction	Zachara <i>et al.</i> (1992, Fig 6)
4.7	176		45	99	Ca Electrolyte, CO ₂ Free	Kenoma Clay, <2um fraction	Zachara <i>et al.</i> (1992, Fig 6)
4.8	176		45	99	Ca Electrolyte, CO ₂ Free	Kenoma Clay, <2um fraction	Zachara <i>et al.</i> (1992, Fig 6)
5.0	283		45	99	Ca Electrolyte, CO ₂ Free	Kenoma Clay, <2um fraction	Zachara <i>et al.</i> (1992, Fig 6)
5.0	297		45	99	Ca Electrolyte, CO ₂ Free	Kenoma Clay, <2um fraction	Zachara <i>et al.</i> (1992, Fig 6)
5.4	708		45	99	Ca Electrolyte, CO ₂ Free	Kenoma Clay, <2um fraction	Zachara <i>et al.</i> (1992, Fig 6)
5.7	1,961		45	99	Ca Electrolyte, CO ₂ Free	Kenoma Clay, <2um fraction	Zachara <i>et al.</i> (1992, Fig 6)

pH	U Kd (ml/g)	Clay Cont. (wt.%)	CEC (meq/100g)	Surface Area (m ² /g)	Solution	Soil Identification	Reference / Comments
5.6	2,367		45	99	Ca Electrolyte, CO ₂ Free	Kenoma Clay, <2um fraction	Zachara <i>et al.</i> (1992, Fig 6)
5.9	4,283		45	99	Ca Electrolyte, CO ₂ Free	Kenoma Clay, <2um fraction	Zachara <i>et al.</i> (1992, Fig 6)
5.9	4,936		45	99	Ca Electrolyte, CO ₂ Free	Kenoma Clay, <2um fraction	Zachara <i>et al.</i> (1992, Fig 6)
6.0	7,936		45	99	Ca Electrolyte, CO ₂ Free	Kenoma Clay, <2um fraction	Zachara <i>et al.</i> (1992, Fig 6)
6.1	8,586		45	99	Ca Electrolyte, CO ₂ Free	Kenoma Clay, <2um fraction	Zachara <i>et al.</i> (1992, Fig 6)
6.2	17,631		45	99	Ca Electrolyte, CO ₂ Free	Kenoma Clay, <2um fraction	Zachara <i>et al.</i> (1992, Fig 6)
6.3	19,553		45	99	Ca Electrolyte, CO ₂ Free	Kenoma Clay, <2um fraction	Zachara <i>et al.</i> (1992, Fig 6)
6.4	30,963		45	99	Ca Electrolyte, CO ₂ Free	Kenoma Clay, <2um fraction	Zachara <i>et al.</i> (1992, Fig 6)
6.5	43,756		45	99	Ca Electrolyte, CO ₂ Free	Kenoma Clay, <2um fraction	Zachara <i>et al.</i> (1992, Fig 6)
5.1	508		59	112	Ca Electrolyte, CO ₂ Free	Ringold Clay Isolate, <2um Fraction	Zachara <i>et al.</i> (1992, Fig 7)
5.2	554		59	112	Ca Electrolyte, CO ₂ Free	Ringold Clay Isolate, <2um Fraction	Zachara <i>et al.</i> (1992, Fig 7)
5.2	676		59	112	Ca Electrolyte, CO ₂ Free	Ringold Clay Isolate, <2um Fraction	Zachara <i>et al.</i> (1992, Fig 7)
5.4	874		59	112	Ca Electrolyte, CO ₂ Free	Ringold Clay Isolate, <2um Fraction	Zachara <i>et al.</i> (1992, Fig 7)
5.4	1,136		59	112	Ca Electrolyte, CO ₂ Free	Ringold Clay Isolate, <2um Fraction	Zachara <i>et al.</i> (1992, Fig 7)
5.6	1,136		59	112	Ca Electrolyte, CO ₂ Free	Ringold Clay Isolate, <2um Fraction	Zachara <i>et al.</i> (1992, Fig 7)
5.7	2,143		59	112	Ca Electrolyte, CO ₂ Free	Ringold Clay Isolate, <2um Fraction	Zachara <i>et al.</i> (1992, Fig 7)
5.8	2,363		59	112	Ca Electrolyte, CO ₂ Free	Ringold Clay Isolate, <2um Fraction	Zachara <i>et al.</i> (1992, Fig 7)
5.9	9,829		59	112	Ca Electrolyte, CO ₂ Free	Ringold Clay Isolate, <2um Fraction	Zachara <i>et al.</i> (1992, Fig 7)
5.9	11,966		59	112	Ca Electrolyte, CO ₂ Free	Ringold Clay Isolate, <2um Fraction	Zachara <i>et al.</i> (1992, Fig 7)
6.0	33,266		59	112	Ca Electrolyte, CO ₂ Free	Ringold Clay Isolate, <2um Fraction	Zachara <i>et al.</i> (1992, Fig 7)
6.1	37,596		59	112	Ca Electrolyte, CO ₂ Free	Ringold Clay Isolate, <2um Fraction	Zachara <i>et al.</i> (1992, Fig 7)

pH	U Kd (ml/g)	Clay Cont. (wt.%)	CEC (meq/100g)	Surface Area (m²/g)	Solution	Soil Identification	Reference / Comments
4.8	377		59	112	Ca Electrolyte, CO ₂ Free	Ringold Clay Isolate, <2um Fraction	Zachara <i>et al.</i> (1992, Fig 7)
4.8	399		59	112	Ca Electrolyte, CO ₂ Free	Ringold Clay Isolate, <2um Fraction	Zachara <i>et al.</i> (1992, Fig 7)
5.1	620		59	112	Ca Electrolyte, CO ₂ Free	Ringold Clay Isolate, <2um Fraction	Zachara <i>et al.</i> (1992, Fig 7)
5.0	637		59	112	Ca Electrolyte, CO ₂ Free	Ringold Clay Isolate, <2um Fraction	Zachara <i>et al.</i> (1992, Fig 7)
5.5	1,476		59	112	Ca Electrolyte, CO ₂ Free	Ringold Clay Isolate, <2um Fraction	Zachara <i>et al.</i> (1992, Fig 7)
5.5	1,603		59	112	Ca Electrolyte, CO ₂ Free	Ringold Clay Isolate, <2um Fraction	Zachara <i>et al.</i> (1992, Fig 7)
5.8	3,091		59	112	Ca Electrolyte, CO ₂ Free	Ringold Clay Isolate, <2um Fraction	Zachara <i>et al.</i> (1992, Fig 7)
6.1	6,047		59	112	Ca Electrolyte, CO ₂ Free	Ringold Clay Isolate, <2um Fraction	Zachara <i>et al.</i> (1992, Fig 7)
6.1	5,823		59	112	Ca Electrolyte, CO ₂ Free	Ringold Clay Isolate, <2um Fraction	Zachara <i>et al.</i> (1992, Fig 7)
6.3	13,713		59	112	Ca Electrolyte, CO ₂ Free	Ringold Clay Isolate, <2um Fraction	Zachara <i>et al.</i> (1992, Fig 7)
6.4	13,341		59	112	Ca Electrolyte, CO ₂ Free	Ringold Clay Isolate, <2um Fraction	Zachara <i>et al.</i> (1992, Fig 7)
4.9	918		59	112	Ca Electrolyte, CO ₂ Free	Ringold Clay Isolate, <2um Fraction	Zachara <i>et al.</i> (1992, Fig 7)
5.1	1,168		59	112	Ca Electrolyte, CO ₂ Free	Ringold Clay Isolate, <2um Fraction	Zachara <i>et al.</i> (1992, Fig 7)
5.1	1,251		59	112	Ca Electrolyte, CO ₂ Free	Ringold Clay Isolate, <2um Fraction	Zachara <i>et al.</i> (1992, Fig 7)
5.6	2,719		59	112	Ca Electrolyte, CO ₂ Free	Ringold Clay Isolate, <2um Fraction	Zachara <i>et al.</i> (1992, Fig 7)
5.7	2,928		59	112	Ca Electrolyte, CO ₂ Free	Ringold Clay Isolate, <2um Fraction	Zachara <i>et al.</i> (1992, Fig 7)
6.7	14,848		59	112	Ca Electrolyte, CO ₂ Free	Ringold Clay Isolate, <2um Fraction	Zachara <i>et al.</i> (1992, Fig 7)
6.8	13,036		59	112	Ca Electrolyte, CO ₂ Free	Ringold Clay Isolate, <2um Fraction	Zachara <i>et al.</i> (1992, Fig 7)
7.0	13,827		59	112	Ca Electrolyte, CO ₂ Free	Ringold Clay Isolate, <2um Fraction	Zachara <i>et al.</i> (1992, Fig 7)
7.0	18,042		59	112	Ca Electrolyte, CO ₂ Free	Ringold Clay Isolate, <2um Fraction	Zachara <i>et al.</i> (1992, Fig 7)
7.0	19,150		59	112	Ca Electrolyte, CO ₂ Free	Ringold Clay Isolate, <2um Fraction	Zachara <i>et al.</i> (1992, Fig 7)
7.1	21,771		59	112	Ca Electrolyte, CO ₂ Free	Ringold Clay Isolate, <2um Fraction	Zachara <i>et al.</i> (1992, Fig 7)

pH	U Kd (ml/g)	Clay Cont. (wt.%)	CEC (meq/100g)	Surface Area (m ² /g)	Solution	Soil Identification	Reference / Comments
7.1	18,097		59	112	Ca Electrolyte, CO ₂ Free	Ringold Clay Isolate, <2um Fraction	Zachara <i>et al.</i> (1992, Fig 7)
7.4	26,008		59	112	Ca Electrolyte, CO ₂ Free	Ringold Clay Isolate, <2um Fraction	Zachara <i>et al.</i> (1992, Fig 7)
7.4	19,488		59	112	Ca Electrolyte, CO ₂ Free	Ringold Clay Isolate, <2um Fraction	Zachara <i>et al.</i> (1992, Fig 7)
7.7	31,032				Ca Electrolyte, CO ₂ Free	Ringold Clay Isolate, <2um Fraction	Zachara <i>et al.</i> (1992, Fig 7)
6.28	3,400				Reducing Conditions	PCE Surface Core, 0-8 cm	Sheppard and Thibault (1988, In Situ)
6.28	2,800				Reducing Conditions	PCE Surface Core, 9-16 cm	Sheppard and Thibault (1988, In Situ)
6.28	3,000				Reducing Conditions	PCE Surface Core, 17-24 cm	Sheppard and Thibault (1988, In Situ)
6.28	11,600				Reducing Conditions	PCE Surface Core, 25-32 cm	Sheppard and Thibault (1988, In Situ)
6.28	18,600				Reducing Conditions	PCE Surface Core, 33-40 cm	Sheppard and Thibault (1988, In Situ)
6.09	3,200				Reducing Conditions	PCE Deep Core, 9-16 cm	Sheppard and Thibault (1988, In Situ)
6.09	8,900				Reducing Conditions	PCE Deep Core, 17-24 cm	Sheppard and Thibault (1988, In Situ)
6.09	9,400				Reducing Conditions	PCE Deep Core, 25-32 cm	Sheppard and Thibault (1988, In Situ)
6.09	12,500				Reducing Conditions	PCE Deep Core, 33-40 cm	Sheppard and Thibault (1988, In Situ)
5.94	3,000				Reducing Conditions	SCE Surface Core, 0-5 cm	Sheppard and Thibault (1988, In Situ)
6.82	8,800				Reducing Conditions	SCE Surface Core, 6-20 cm	Sheppard and Thibault (1988, In Situ)
7.28	2,600				Reducing Conditions	SCE Surface Core, 21-25 cm	Sheppard and Thibault (1988, In Situ)
7.28	1,700				Reducing Conditions	SCE Surface Core, 26-30 cm	Sheppard and Thibault (1988, In Situ)
7.28	700				Reducing Conditions	SCE Surface Core, 31-40 cm	Sheppard and Thibault (1988, In Situ)
	1,300				Reducing Conditions	PCE Surface Core, 0-40 cm	Sheppard and Thibault (1988, Batch)
	2,100				Reducing Conditions	PCE Deep Core, 40-80 cm	Sheppard and Thibault (1988, Batch)
	2,000				Reducing Conditions	SCE Surface Core, 1-10 cm	Sheppard and Thibault (1988, Batch)
	2,900				Reducing Conditions	SCE Surface Core, 10-30 cm	Sheppard and Thibault (1988, Batch)

pH	U Kd (ml/g)	Clay Cont. (wt.%)	CEC (meq/100g)	Surface Area (m ² /g)	Solution	Soil Identification	Reference / Comments
	870				Reducing Conditions	SCE Surface Core, 30-40 cm	Sheppard and Thibault (1988, Batch)
5.7	46		2.3		Site Borehole Groundwater	Clay (Glacial Till, Less Than 5 mm)	Bell and Bates (1988)
5.7	46		3.0		Site Borehole Groundwater	C1:2 (Brown, Slightly Silty, Less Than 5 mm)	Bell and Bates (1988)
5.7	900		2.7		Site Borehole Groundwater	C3 (Dark Brown Coarse Granular Deposit, Less Than 5 mm)	Bell and Bates (1988)
5.7	2,200		2.9		Site Borehole Groundwater	C6 (Brown Coarse Granular Deposit, Less Than 5 mm)	Bell and Bates (1988)
5.7	560		0.8		Site Borehole Groundwater	Sand (Light Brown Coarse Granular Deposit, Less Than 5 mm)	Bell and Bates (1988)
4.16	85.0	0.5	1.11			A12	Serkiz and Johnson (1994)
4.99	170.0	3.3	1.82			A13	Serkiz and Johnson (1994)
3.42	5.3	3	3.74			A13R	Serkiz and Johnson (1994)
3.19	2.1	1.5	1.39			A22	Serkiz and Johnson (1994)
3.01	1.7	4.5	1.4			A23	Serkiz and Johnson (1994)
3.19	3.7	4.4	7.92			A31	Serkiz and Johnson (1994)
3.5	1.4	3.1	1			A32	Serkiz and Johnson (1994)
3.29	1.2	4.7	2.1			A42	Serkiz and Johnson (1994)
5.42	2,200.0	2.5	0.68			A52	Serkiz and Johnson (1994)
3.72	2.3	2	0.42			A53	Serkiz and Johnson (1994)
3.24	2.7	2.8	4.71			B13	Serkiz and Johnson (1994)
3.93	8.5	3.9	3.06			B14	Serkiz and Johnson (1994)
3.86	10.1	4.9				B23	Serkiz and Johnson (1994)
4.02	5.2	2.5	3.8			B23R	Serkiz and Johnson (1994)
3.83	14.0	7.5	5.69			B24	Serkiz and Johnson (1994)
4.62	390.0	6.2	2.5			B32	Serkiz and Johnson (1994)
4.64	180.0	5.5	8.42			B33	Serkiz and Johnson (1994)
4.67	190.0	12.6	21.4			B42	Serkiz and Johnson (1994)
3.66	6.4	1.2	3.02			B43	Serkiz and Johnson (1994)
4.09	39.0	8.2	15.1			B51	Serkiz and Johnson (1994)

pH	U Kd (ml/g)	Clay Cont. (wt.%)	CEC (meq/100g)	Surface Area (m ² /g)	Solution	Soil Identification	Reference / Comments
3.61	5.3					B52	Serkiz and Johnson (1994)
4.69	530.0	3.3	2.39			B52R	Serkiz and Johnson (1994)
3.68	6.4					C13	Serkiz and Johnson (1994)
3.75	23.0	6.4				C14	Serkiz and Johnson (1994)
3.96	30.0		1.28			C22	Serkiz and Johnson (1994)
4.17	980.0	6.4	6.12			C23	Serkiz and Johnson (1994)
5.53	3,600.0	5.5	2.54			C32	Serkiz and Johnson (1994)
4.64	6,300.0	6.1	8.54			C33	Serkiz and Johnson (1994)
5.27	14,000.0	7.9	11.4			C42	Serkiz and Johnson (1994)
4.51	13,000.0	3	5.04			C43	Serkiz and Johnson (1994)
6.78	11,000.0	5.3	1.96			D13	Serkiz and Johnson (1994)
4.14	13.0					D13RA	Serkiz and Johnson (1994)
	9.3	2	2.55			D13RB	Serkiz and Johnson (1994)
4	320.0	10.5	11.4			E13	Serkiz and Johnson (1994)
4.04	310.0	4.5	8.5			E14	Serkiz and Johnson (1994)
5.85	2,700.0	6.4	15.5			E23	Serkiz and Johnson (1994)
4.32	980.0	3.9	13.3			E23R	Serkiz and Johnson (1994)
3.87	290.0	7.3	13.8			E24	Serkiz and Johnson (1994)
4.27	1,500.0	6.5	11.5			E33	Serkiz and Johnson (1994)
4.05	380.0	3.7	10.5			E34	Serkiz and Johnson (1994)
5.27	16,000.0	31.8	20.6			E41	Serkiz and Johnson (1994)
4.87	18,000.0	14.5	20.6			E42	Serkiz and Johnson (1994)
4.3	7,500.0	15.5	16.1			F12	Serkiz and Johnson (1994)
4.9	830.0		8.51			F13	Serkiz and Johnson (1994)
4.69	160.0	8.1	7.48			F22	Serkiz and Johnson (1994)
6.48	16,000.0	13	11.6			F23	Serkiz and Johnson (1994)
4.85	8,700.0	14.2	15.1			F32	Serkiz and Johnson (1994)
4.77	2,900.0	18.3	13.6			F33	Serkiz and Johnson (1994)
5.2	34,000.0	17.2	11.8			F42	Serkiz and Johnson (1994)
4.12	330.0	14.2				F43	Serkiz and Johnson (1994)
5.91	5,500.0	42.2	19.9			F52	Serkiz and Johnson (1994)
5.63	27,000.0	16.3	13.3			F53	Serkiz and Johnson (1994)
4.16	139.0	0.5	1.11			A12	Serkiz and Johnson (1994)
4.99	361.0	3.3	1.82			A13	Serkiz and Johnson (1994)
3.42	9.46	3	3.74			A13R	Serkiz and Johnson (1994)

pH	U Kd (ml/g)	Clay Cont. (wt.%)	CEC (meq/100g)	Surface Area (m ² /g)	Solution	Soil Identification	Reference / Comments
3.19	3.79	1.5	1.39			A22	Serkiz and Johnson (1994)
3.01	1.55	4.5	1.4			A23	Serkiz and Johnson (1994)
3.19	4.43	4.4	7.92			A31	Serkiz and Johnson (1994)
3.5	1.38	3.1	1			A32	Serkiz and Johnson (1994)
3.29	1.19	4.7	2.1			A42	Serkiz and Johnson (1994)
5.42	160.0	2.5	0.68			A52	Serkiz and Johnson (1994)
3.72	16.0	2	0.42			A53	Serkiz and Johnson (1994)
3.24	2.0	2.8	4.71			B13	Serkiz and Johnson (1994)
3.93	10.4	3.9	3.06			B14	Serkiz and Johnson (1994)
3.86	10.7	4.9				B23	Serkiz and Johnson (1994)
4.02	4.0	2.5	3.8			B23R	Serkiz and Johnson (1994)
3.83	11.3	7.5	5.69			B24	Serkiz and Johnson (1994)
4.62	332.0	6.2	2.5			B32	Serkiz and Johnson (1994)
4.64	212.0	5.5	8.42			B33	Serkiz and Johnson (1994)
4.67	180.0	12.6	21.4			B42	Serkiz and Johnson (1994)
3.66	7.1	1.2	3.02			B43	Serkiz and Johnson (1994)
4.09	20.8	8.2	15.1			B51	Serkiz and Johnson (1994)
3.61	2.6					B52	Serkiz and Johnson (1994)
4.69	180.0	3.3	2.39			B52R	Serkiz and Johnson (1994)
3.68	5.6					C13	Serkiz and Johnson (1994)
3.75	28.3	6.4				C14	Serkiz and Johnson (1994)
3.96	27.4		1.28			C22	Serkiz and Johnson (1994)
4.17	823.0	6.4	6.12			C23	Serkiz and Johnson (1994)
5.53	540.0	5.5	2.54			C32	Serkiz and Johnson (1994)
4.64	690.0	6.1	8.54			C33	Serkiz and Johnson (1994)
5.27	1,400.0	7.9	11.4			C42	Serkiz and Johnson (1994)
4.51	460.0	3	5.04			C43	Serkiz and Johnson (1994)
6.78	690.0	5.3	1.96			D13	Serkiz and Johnson (1994)
4.14	26.6					D13RA	Serkiz and Johnson (1994)
	22.6	2	2.55			D13RB	Serkiz and Johnson (1994)
4	650.0	10.5	11.4			E13	Serkiz and Johnson (1994)
4.04	190.0	4.5	8.5			E14	Serkiz and Johnson (1994)
4.32	310.0	3.9	13.3			E23R	Serkiz and Johnson (1994)
3.87	360.0	7.3	13.8			E24	Serkiz and Johnson (1994)
4.27	470.0	6.5	11.5			E33	Serkiz and Johnson (1994)

pH	U Kd (ml/g)	Clay Cont. (wt.%)	CEC (meq/100g)	Surface Area (m²/g)	Solution	Soil Identification	Reference / Comments
4.05	270.0	3.7	10.5			E34	Serkiz and Johnson (1994)
5.27	870.0	31.8	20.6			E41	Serkiz and Johnson (1994)
4.87	630.0	14.5	20.6			E42	Serkiz and Johnson (1994)
4.3	690.0	15.5	16.1			F12	Serkiz and Johnson (1994)
4.9	2,200.0		8.51			F13	Serkiz and Johnson (1994)
4.69	1,200.0	8.1	7.48			F22	Serkiz and Johnson (1994)
6.48	950.0	13	11.6			F23	Serkiz and Johnson (1994)
4.85	660.0	14.2	15.1			F32	Serkiz and Johnson (1994)
4.77	220.0	18.3	13.6			F33	Serkiz and Johnson (1994)
5.2	910.0	17.2	11.8			F42	Serkiz and Johnson (1994)
4.12	700.0	14.2				F43	Serkiz and Johnson (1994)
5.91	600.0	42.2	19.9			F52	Serkiz and Johnson (1994)
5.63	960.0	16.3	13.3			F53	Serkiz and Johnson (1994)

J.6.0 References

- Ames, L. L., J. E. McGarrah, B. A. Walker, and P. F. Salter. 1982. "Sorption of Uranium and Cesium by Hanford Basalts and Associated Secondary Smectite." *Chemical Geology*, 35:205-225.
- Ames, L. L., J. E. McGarrah, B. A. Walker, and P. F. Salter. 1983c. "Uranium and Radium Sorption on Amorphous Ferric Oxyhydroxide." *Chemical Geology*, 40:135-148.
- Ames, L. L., J. E. McGarrah, and B. A. Walker. 1983a. "Sorption of Trace Constituents from Aqueous Solutions onto Secondary Minerals. I. Uranium." *Clays and Clay Minerals*, 31(5):321-334.
- Ames, L. L., J. E. McGarrah, and B. A. Walker. 1983b. "Sorption of Uranium and Radium by Biotite, Muscovite, and Phlogopite." *Clays and Clay Minerals*, 31(5):343-351.
- Ames, L. L., and D. Rai. 1978. *Radionuclide Interactions with Soil and Rock Media. Volume 1: Processes Influencing Radionuclide Mobility and Retention. Element Chemistry and Geochemistry. Conclusions and Evaluation.* EPA 520/6-78-007 (Volume 1 of 2), U.S. Environmental Protection Agency, Las Vegas, Nevada.
- Amonette, J. E., J. E. Szecsody, H. T. Schaefer, J. C. Templeton, Y. A. Gorby, and J. S. Fruchter. 1994. "Abiotic Reduction of Aquifer Materials by Dithionite: A Promising In-Situ Remediation Technology." In *In-Situ Remediation: Scientific Basis for Current and Future Technologies. Thirty-Third Hanford Symposium on Health and the Environment, November 7-11, 1994, Pasco, Washington*, G. W. Gee and N. R. Wing (eds.). Battelle Press, Richland, Washington.
- Andersson, K., B. Torstenfelt, and B. Allard. 1982. "Sorption Behavior of Long-Lived Radionuclides in Igneous Rock." In *Environmental Migration of Long-Lived Radionuclides Proceedings of an International Symposium on Migration in the Terrestrial Environment of Long-Lived Radionuclides from the Nuclear Fuel Cycle Organized by the International Atomic Energy Agency, the Commission of the European Communities and the OECD Nuclear Energy Agency and held in Knoxville, United States, 27-31 July 1981.*, Knoxville, Tennessee. IAEA-SM-257/20. pp. 111-131. International Atomic Energy Agency, Vienna, Austria.
- Baes, C. F., III, and R. D. Sharp. 1983. "A Proposal for Estimation of Soil Leaching and Leaching Constants for Use in Assessment Models." *Journal of Environmental Quality*, 12:17-28.
- Bates, R. L., and J. A. Jackson (eds.). 1980. *Glossary of Geology*. American Geological Institute, Falls Church, Virginia.

- Barney, G. S. 1982a. *Radionuclide Sorption on Basalt Interbed Materials FY 1981 Annual Report*. RHO-BW-ST-35 P, Rockwell Hanford Operations, Richland, Washington.
- Barney, G. S. 1982b. *Radionuclide Sorption of Columbia River Basalt Interbed Materials*. RHO-BW-SA-198 P, Rockwell Hanford Operations, Richland, Washington.
- Bell, J., and T. H. Bates. 1988. "Distribution Coefficients of Radionuclides Between Soils and Groundwaters and Their Dependence on Various Test Parameters." *The Science of the Total Environment*, 69:297-317.
- Borovec, Z. 1981. "The Adsorption of Uranyl Species by Fine Clay." *Chemical Geology*, 32:45-58.
- Borovec, Z., B. Kribek, and V. Tolar. 1979. "Sorption of Uranyl by Humic Acids." *Chemical Geology*, 27:39-46.
- Brindley, G. W., and M. Bastovanov. 1982. "Interaction of Uranyl Ions with Synthetic Zeolites of Type A and the Formation of Compreignacite-Like and Becquerelite-Like Products." *Clays and Clay Minerals*, 30:135-142.
- Chisholm-Brause, C., S. D. Conradson, C. T. Buscher, P. G. Eller, and D. E. Morris. 1994. "Speciation of uranyl Sorbed at Multiple Binding Sites on Montmorillonite." *Geochimica et Cosmochimica Acta*, 58(17):3625-3631.
- Dahlman, R. C., E. A. Bondietti, and L. D. Eyman. 1976. Biological Pathways and Chemical Behavior of Plutonium and Other Actinides in the Environment. In *Actinides in the Environment*, (ed.) A. M. Friedman, pp. 47-80. ACS Symposium Series 35, American Chemical Society, Washington, D.C.
- Dement'yev, V. S., and N. G. Syromyatnikov. 1968. "Conditions of Formation of a Sorption Barrier to the Migration of Uranium in an Oxidizing Environment." *Geochemistry International*, 5:394-400
- Doi, K., S. Hirono, and Y. Sakamaki. 1975. "Uranium Mineralization by Ground Water in Sedimentary Rocks, Japan." *Economic Geology*, 70:628-646.
- Duff, M. C., and C. Amrhein. 1996. "Uranium(VI) Adsorption on Goethite and Soil in Carbonate Solutions." *Soil Science Society of America Journal*, 60(5):1393-1400.
- Erickson, K. L. 1980. Radionuclide Sorption Studies on Abyssal Red Clays. In *Scientific Basis for Nuclear Waste Management. Volume 2*, (ed.) C. J. M. Northrup, Jr., pp. 641-646. Plenum Press, New York, New York.

- Erikson, R. L., C. J. Hostetler, R. J. Serne, J. R. Divine, and M. A. Parkhurst. 1993. *Geochemical Factors Affecting Degradation and Environmental Fate of Deleted Uranium Penetrators in Soil and Water*. PNL-8527, Pacific Northwest Laboratory, Richland, Washington.
- Fruchter, J. S., J. E. Amonette, C. R. Cole, Y. A. Gorby, M. D. Humphrey, J. D. Isok, F. A. Spane, J. E. Szecsody, S. S. Teel, V. R. Vermeul, M. D. Williams, and S. B. Yabusaki, 1996, *In Situ Redox Manipulation Field Injection Test Report - Hanford 100-H Area*. PNNL-11372, Pacific Northwest National Laboratory, Richland, Washington.
- Giblin, A. M. 1980. "The Role of Clay Adsorption in Genesis of Uranium Ores." *Uranium in the Pine Creek Geosyncline*. In *Proceedings of the International Uranium Symposium on the Pine Creek Geosyncline Jointly Sponsored by the Bureau of Mineral Resources, Geology, and Geophysics and the CSIRO Institute of Earth Resources in Co-operation with the International Atomic Energy Agency and Held in Sydney, Australia 4-8 June, 1979*, eds. J. Ferguson and A. B. Goleby, pp. 521-529. International Atomic Energy Agency, Vienna, Austria.
- Goldsztaub, S. and R. Wey. 1955. "Adsorption of Uranyl Ions by Clays." *Bull. Soc. Franc. Mineral. Crist.*, 78:242.
- Haji-Djafari, S., P. E. Antommaria, and H. L. Crouse. 1981. Attenuation of Radionuclides and Toxic Elements by In Situ Soils at a Uranium Tailings Pond in Central Wyoming. In *Permeability and Groundwater Contaminant Transport*, (eds.) T. F. Zimmie and C. O. Riggs, pp. 221-242. American Society for Testing and Materials, Philadelphia, Pennsylvania.
- Ho, C. H., and N. H. Miller. 1986. "Adsorption of Uranyl Species from Bicarbonate Solution onto Hematite Particles." *Journal of Colloid and Interface Science*, 110:165-171. (Note paper issued under report number AECL-8433, Atomic Energy of Canada Limited, Whiteshell Nuclear Research Establishment, Pinawa, Manitoba, Canada.)
- Ho, C. H., and N. H. Miller. 1985. "Effect of Humic Acid on Uranium Uptake by Hematite Particles." *Journal of Colloid and Interface Science*, 106:281-288. (Note paper issued under report number AECL-8432, Atomic Energy of Canada Limited, Whiteshell Nuclear Research Establishment, Pinawa, Manitoba, Canada.)
- Ho, C. H., and D. C. Doern. 1985. "The Sorption of Uranyl Species on a Hematite Sol." *Canadian Journal of Chemistry*, 63:1100-1104. (Note paper issued under report number AECL-8038, Atomic Energy of Canada Limited, Whiteshell Nuclear Research Establishment, Pinawa, Manitoba, Canada.)

- Horráth, E. 1960. "Investigations of Uranium Adsorption to Peat in Natural Waters Containing U-Traces." *Magyar Tudományos Akad. Atommag Kutató Intézete, Közlemények*, 2:177-183 (in Hungarian).
- Hsi, C-K. D., and D. Langmuir. 1985. "Adsorption of Uranyl Onto Ferric Oxyhydroxides: Application of the Surface Complexation Site-Binding Model." *Geochimica et Cosmochimica Acta*, 49:1931-1941.
- Johnson, W. H., S. M. Serkiz, L. M. Johnson, and S. B. Clark. 1994. *Uranium Partitioning Under Acidic Conditions in a Sandy Soil Aquifer*. WSRC-MS--94-0528, Westinghouse Savannah River Company, Savannah River Site, Aiken, South Carolina.
- Kaplan, R. J. Serne, A. T. Owen, J. Conca, T. W. Wietsma, and T. L. Gervais. 1996. *Radionuclide Adsorption Distribution Coefficient Measured in Hanford Sediments for the Low Level Waste Performance Assessment Project*. PNNL-11385, Pacific Northwest National Laboratory, Richland, Washington.
- Kaplan, D. I., T. L. Gervais, and K. M. Krupka. 1998. "Uranium(VI) Sorption to Sediments Under High pH and Ionic Strength Conditions." *Radiochimica Acta*, 80:201-211.
- Kaplan, D. I., and R. J. Serne. 1995. *Distribution Coefficient Values Describing Iodine, Neptunium, Selenium, Technetium, and Uranium Sorption to Hanford Sediments*." PNL-10379 (Supplement 1), Pacific Northwest Laboratory, Richland, Washington.
- Kent, D. B., V. S. Tripathi, N. B. Ball, J. O. Leckie, and M. D. Siegel. 1988. *Surface-Complexation Modeling of Radionuclide Adsorption in Subsurface Environments*. NUREG/CR-4807, U.S. Nuclear Regulatory Commission, Washington, D.C.
- Kohler, M., G. P. Curtis, D. B. Kent, and J. A. Davis. 1996. "Experimental Investigation and Modeling of Uranium(VI) Transport Under Variable Chemical Conditions." *Water Resources Research*, 32(12):3539-3551.
- Koß, V. 1988. "Modeling of Uranium(VI) Sorption and Speciation in a Natural Sediment Groundwater System." *Radiochimica Acta*, 44/45:403-406.
- Kovalevskii, A. L. 1967. "Dependence of the Content of Some Trace Elements on the Clayiness of Soils." *Mikroelem. Biosfere Ikh Primen. Sci. Khaz. Med. Sib. Dal'nego Vostoka, Dokl. Sib. Knof., 2nd. 1964. O. V. Makew. Buryat. Khizhn. Izd. Ulan-Ude, USSR*.
- Krupka, K. M., D. Rai, R. W. Fulton, and R. G. Strickert. 1985. "Solubility Data for U(VI) Hydroxide and Np(IV) Hydrated Oxide: Application of MCC-3 Methodology," pp. 753-760. In *Scientific Basis for Nuclear Waste Management VIII*, eds. C. M. Jantzen, J. A. Stone, and R. C. Ewing. Materials Research Society Symposium Proceedings, Volume 44, Materials Research Society, Pittsburgh, Pennsylvania.

- Lindenmeier, C. W., R. J. Serne, J. L. Conca, A. T. Owen, and M. I. Wood. 1995. *Solid Waste Leach Characteristics and Contaminant-Sediment Interactions Volume 2: Contaminant Transport Under Unsaturated Moisture Contents*. PNL-10722, Pacific Northwest Laboratory, Richland, Washington.
- Looney, B. B., M. W. Grant, and C. M. King. 1987. *Estimating of Geochemical Parameters for Assessing Subsurface Transport at the Savannah River Plant*. DPST-85-904, Environmental Information Document, E. I. du pont de Nemours and Company, Savannah River Laboratory, Aiken, South Carolina.
- Manskaya, S. M., G. V. Drozdora, and M. P. Yelmel'yanova. 1956. "Fixation of Uranium by Humic Acids and Melanoidins." *Geokhimiya*, No. 4.
- Masuda, K., and T. Yamamoto. 1971. "Studies on Environmental Contamination by Uranium. II. Adsorption of Uranium on Soil and Its Desorption." *Journal of Radiation Research*, 12:94-99.
- McKinley, J. P., J. M. Zachara, S. C. Smith, and G. D. Turner. 1995. "The Influence of Uranyl Hydrolysis and Multiple Site-Binding Reactions on Adsorption of U(VI) to Montmorillonite." *Clays and Clay Minerals*, 43(5):586-598.
- McKinley, G., and A. Scholtis. 1993. "A Comparison of Radionuclide Sorption Databases Used in Recent Performance Assessments." *Journal of Contaminant Hydrology*, 13:347-363.
- Morris, D. E., C. J. Chisholm-Brause, M. E. Barr, S. D. Conradson, and P. G. Eller. 1994. "Optical Spectroscopic Studies of the Sorption of UO_2^{2+} Species on a Reference Smectite." *Geochimica et Cosmochimica Acta*, 58:3613-3623.
- Neiheisel, J. 1983. Prediction Parameters of Radionuclide Retention at Low-Level Radioactive Waste Sites. EPA 520/1-83-025, U.S. Environmental Protection Agency, Washington, D.C.
- Payne, T. E., and T. D. Waite. 1991. "Surface Complexation Modelling of Uranium Sorption Data Obtained by Isotope Exchange Techniques." *Radiochimica Acta*, 52/53:487-493.
- Puigdomènech, I., and U. Bergström. 1994. *Calculated Distribution of Radionuclides in Soils and Sediments*. SKB Technical Report 94-32, Swedish Nuclear Fuel and Waste Management Company, Stockholm, Sweden.
- Puls, R. W., L. L. Ames, and J. E. McGarrah. 1987. *Sorption and Desorption of Uranium, Selenium, and Radium in a Basalt Geochemical Environment*. WHC-SA-0003-FP, Westinghouse Hanford Company, Richland, Washington.

- Rançon, D. 1973. The Behavior in Underground Environments of Uranium and Thorium Discharge by the Nuclear Industry. In *Environmental Behavior of Radionuclides Released in the Nuclear Industry*, pp. 333-346. IAEA-SM-172/55, International Atomic Energy Agency Proceedings, Vienna, Austria.
- Ritchie, J. C., P. H. Hawks, and J. R. McHenry. 1972. "Thorium, Uranium, and Potassium in Upper Cretaceous, Paleocene, and Eocene Sediments of the Little Tallahatchie River Watershed in Northern Mississippi." *Southeast Geology*, 14:221-231.
- Rozhkova, Ye.V., Ye. G. Razumnaya, M. B. Serebrayakova and O. V. Shchebak. 1959. "Role of Sorption in Concentration of Uranium in Sedimentary Rocks." *Tr. II. Mezhdunar. knof. po miro nmu ispol'z. atom. energii*. 3.
- Rubtsov, D. M. 1972. "Thorium and Uranium Content in the Clay Fraction of Podzolic Mountain Soils of Thin Forests." *Radioekol. Issled Prir. Biogeotsenozakh*, 53-66 (in Russian).
- Salter, P. F., L. L. Ames, and J. E. McGarrah. 1981. *The Sorption Behavior of Selected Radionuclides on Columbia River Basalts*. RHO-BWI-LD-48, Rockwell Hanford Operations, Richland, Washington.
- Seeley, F. G., and A. D. Kelmers. 1984. *Geochemical Information for the West Chestnut Ridge Central Waste Disposal Facility for Low-Level Radioactive Waste*. ORNL-6061, Oak Ridge National Laboratory, Oak Ridge, Tennessee
- Serkiz, S. M. And W. H. Johnson. 1994. Uranium Geochemistry in Soil and Groundwater at the F and H Seepage Basins (U). EPD-SGS-94-307, Westinghouse Savannah River Company, Savannah River Site, Aiken, South Carolina.
- Serne, R. J., J. L. Conca, V. L. LeGore, K. J. Cantrell, C. W. Lindenmeier, J. A. Campbell, J. E. Amonette, and M. I. Wood. 1993. *Solid-Waste Leach Characteristics and Contaminant-Sediment Interactions. Volume 1: Batch Leach and Adsorption Tests and Sediment Characterization*. PNL-8889, Volume 1, Pacific Northwest Laboratory, Richland, Washington.
- Sheppard, M. I., D. I. Beals, D. H. Thibault, and P. O'Connor. 1984. *Soil Nuclide Distribution Coefficients and Their Statistical Distribution*. AECL-8364, Chalk River Nuclear Labs, Atomic Energy of Canada Limited, Chalk River, Canada.
- Sheppard, M. I., and D. H. Thibault. 1988. "Migration of Technetium, Iodine, Neptunium, and Uranium in the Peat of Two Minerotrophic Mires." *Journal of Environmental Quality*, 17:644-653.

- Sheppard, M. I., and D. H. Thibault. 1990. "Default Soil Solid/Liquid Partition Coefficients, K_{ds} , for Four Major Soil Types: A Compendium." *Health Physics*, 59(4)471-482.
- Starik, I. Ye., F. Ye Starik and A. N. Apollonova. 1958. "Adsorption of Traces of Uranium on Iron Hydroxide and Its Desorption by the Carbonate Method." *Zh. Neorgan. Khimii*. 3(1).
- Stenhouse, M. J., and J. Pöttinger. 1994. "Comparison of Sorption Databases Used in Recent Performance Assessments Involving Crystalline Host Rock." *Radiochimica Acta*, 66/67:267-275.
- Stumm, W., and J. J. Morgan. 1981. *Aquatic Chemistry. An Introduction Emphasizing Chemical Equilibria in Natural Waters*. John Wiley and Sons, New York, New York.
- Szalay, A. 1954. "The Enrichment of Uranium in Some Brown Coals in Hungary." *Acta Geol. Acad. Sci. Hungary*, 2:299-311.
- Szalay, A. 1957. "The Role of Humus in the Geochemical Enrichment of U in Coal and Other Bioliths." *Acta Phys. Acad. Sci. Hungary*, 8:25-35.
- Thibault, D. H., M. I. Sheppard, and P. A. Smith. 1990. *A Critical Compilation and Review of Default Soil Solid/Liquid Partition Coefficients, K_{ds} , for Use in Environmental Assessments*. AECL-10125, Whiteshell Nuclear Research Establishment, Atomic Energy of Canada Limited, Pinawa, Canada.
- Tripathi, V. S. 1984. *Uranium(VI) Transport Modeling: Geochemical Data and Submodels*. Ph.D. Dissertation, Stanford University, Stanford, California.
- Tsunashima, A., G. W. Brindley, and M. Bastovanov. 1981. "Adsorption of Uranium from Solutions by Montmorillonite: Compositions and Properties of Uranyl Montmorillonites." *Clays and Clay Minerals*, 29:10-16.
- Turner, D. R. 1993. *Mechanistic Approaches to Radionuclide Sorption Modeling*. CNWRA 93-019, Center for Nuclear Waste Regulatory Analysis, San Antonio, Texas.
- Turner, D. R. 1995. *Uniform Approach to Surface Complexation Modeling of Radionuclide Sorption*. CNWRA 95-001, Center for Nuclear Waste Regulatory Analysis, San Antonio, Texas.
- Turner, D. R., T. Griffin, and T. B. Dietrich. 1993. "Radionuclide Sorption Modeling Using the MINTEQA2 Speciation Code." In *Scientific Basis for Nuclear Waste Management XVI*, (eds.) C. G. Interrante and R. T. Pabalan, Materials Research Society Symposium Proceedings, Volume 294, p. 783-789. Materials Research Society, Pittsburgh, Pennsylvania.

- Turner, G. D., J. M. Zachara, J. P. McKinley, and S. C. Smith. 1996. "Surface-Charge Properties and UO_2^{2+} Adsorption of a Subsurface Smectite." *Geochimica et Cosmochimica Acta*, 60(18):3399-3414.
- Vochten, R. C., L. van Haverbeke, and F. Goovaerts. 1990. "External Surface Adsorption of Uranyl-Hydroxo Complexes on Zeolite Particles in Relation to the Double-Layer Potential." *Journal of the Chemical Society. Faraday Transaction*, 86:4095-4099.
- Waite, T. D., T. E. Payne, J. A. Davis, and K. Sekine. 1992. *Alligators Rivers Analogue Project. Final Report Volume 13. Uranium Sorption*. ISBN 0-642-599394 (DOE/HMIP/RR/92/0823, SKI TR 92:20-13).
- Waite, T. D., J. A. Davis, T. E. Payne, G. A. Waychunas, and N. Xu. 1994. "Uranium(VI) Adsorption to Ferrihydrite: Application of a Surface Complexation Model." *Geochimica et Cosmochimica Acta*, 58(24):5465-5478.
- Warnecke, E., G. Tittel, P. Brennecke, G. Stier-Friedland, and A. Hollman. 1986. "Experimental Investigations of Possible Radionuclide Releases from the Planned Repositories in the Gorleben Salt Dome and Konrad Iron ore Mine as Part of the Long-Term safety Assessment." In *Site, Design and Construction of Underground Repositories for Radioactive Wastes*, IAEA-SM-289/49, p. 401-416, International Atomic Energy Agency, Vienna, Austria.
- Warnecke, E., A. Hollman, G. Tittel, and P. Brennecke. 1994. "Gorleben Radionuclide Migration Experiments: More Than 10 Years of Experience." In *Fourth International Conference on the Chemistry and Migration Behavior of Actinides and Fission Products in the Geosphere*, p. 821-827, R. Oldenbourg Verlag, München, Germany.
- Warnecke, E., and W. Hild. 1988. "German Experience in the Field of Radionuclide Migration in the Geosphere." *Radioactive Waste Management and the Nuclear Fuel Cycle*, 10(1-3):115-144.
- Warnecke, E., A. Hollman, and G. Stier-Friedland. 1984. "Migration of Radionuclides: Experiments Within the Site Investigation Program at Gorleben." In *Scientific Basis for Nuclear Waste Management VII*, (ed.) G. L. McVay, Materials Research Society Symposium Proceedings, Volume 26, p. 41-48. North-Holland, New York, New York.
- Yakobenchuk, V. F. 1968. "Radioactivity and Chemical Properties of Sod-Podzolic Soils in the Ukrainian Western Polesie." *Visn. Sil's Kogosped. Nauki*, 11:45-50 (in Ukrainian).
- Yamamoto, T., E. Yunoki, M. Yamakawa, and M. Shimizu. 1973. "Studies on Environmental Contamination by Uranium. 3. Effects of Carbonate Ion on Uranium Adsorption to and Desorption from Soils." *Journal of Radiation Research*, 14:219-224.

Zachara, J. M., C. C. Ainsworth, J. P. McKinley, E. M. Murphy, J. C. Westall, and P. S. C. Rao. 1992. "Subsurface Chemistry of Organic Ligand-Radionuclide Mixtures." In *Pacific Northwest Laboratory Annual Report for 1991 to the DOE Office of Energy Research. Part 2: Environmental Science*, pp. 1-12. PNL-8000 Pt. 2, Pacific Northwest Laboratory, Richland, Washington.