Characterizing the Speciation, Distribution, and Correlation of Heavy Metals in Mine Wastes Using Synchrotron Radiation Techniques

Final Technical Report for MRERP Award # 06HQGR0181

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Comparison of Accomplishments with Established Goals

The two primary goals and related hypotheses associated with each goal are presented herein verbatim from the awarded proposal, followed by a general discussion of the work accomplished and its relation to the specific goal.

Goal 1: Using bulk EXAFS spectroscopy, quantitatively characterize speciation trends of Hg and As in mine wastes from different Hg deposit types (silica-carbonate, epithermal hot-spring) as a function of 1) particle size and 2) distance from the mine site.

Hypothesis: The total concentrations of Hg and As will increase as particle size and distance from the host mine decrease, while the proportions of less soluble Hg and As phases will decrease with particle size and distance due to preferential weathering and dissolution of more soluble phases. Hg-chloride phases will be more prevalent among the hot-spring Hg deposits due to elevated chloride concentrations during their formation.

Due to the development of new strategic goals from the USGS and BLM, focus was shifted from Hg mines in the CA Coast Range to the Kelly and Randsburg mining districts in the high desert region of Southern-Central California. These mines were primarily explored for Au and Ag and have associated concentrations of arsenic ranging as high as >10,000 ppm total As. Thus, concentration, distribution, and speciation trends were largely focused on As levels rather than Hg for the duration of this grant period. Nevertheless, the general trends and associations studied were still in keeping with the grant's primary goals. Extensive sampling was conducted throughout the Randsburg mining district, including samples collected of tailings piles, waste rock, soil profiles, and distributed sediment deposits.

Following manual size separation of selected samples using stainless steel sieves to generate 11 discrete size fractions, trends in concentration of heavy metals as a function of particle size were assessed through the development of an Excel spreadsheet template that auto-generates graphs for the concentrations of 48 elements analyzed by ICP-MS. These were combined with the mass distribution of the sample to similarly auto-generate graphs of elemental mass distribution for each of the 48 elements. An example of this process is given in Figure 1, showing as a function of particle size (decreasing from S1 to S11 as described in the initial proposal) the mass distribution of the sample (Figure 1a), the concentration trend of arsenic (Figure 1b), and the mass distribution of arsenic in the samples (Figure 1c).

As shown in Figure 1, the concentration trend for arsenic demonstrates an inverse relationship between particle size and concentration as initially hypothesized; such a trend has been observed for arsenic in the majority of samples and suggests that relatively insoluble and soft As-bearing phases are becoming preferentially concentrated in the fine-grained size fractions. Other trends have been observed and categorized for all elements analyzed for individual samples; a schematic of the different trends observed is shown in Figure 2. This allows the grouping of elements that share similar concentration trends for further analysis using microspectroscopic methods. A list of all samples that have undergone such size separation and analysis is provided in Appendix A.

We have begun bulk analysis using extended X-ray absorption fine structure (EXAFS) spectroscopy to determine the speciation of arsenic in selected samples. This is accomplished by comparing the EXAFS spectrum collected from the sample to a spectral library of well-characterized As species through a linear combination fitting method that identifies both the phases present and their relative proportions to one another. Examples of this fitting method as applied to mine tailings from the Randsburg and Kelly mining districts are shown in Figures 3 and 4, respectively. Compiled results from EXAFS speciation analysis of multiple samples are shown in Figure 5, indicating that the As phase scorodite (FeAsO₄ \cdot 2H₂O) is present in the Kelly samples but not in the Randsburg samples, while arseniosiderite (Ca₂Fe₃(AsO₄)₃O₂ \cdot 3H₂O) and arsenic sorbed to iron oxyhydroxide phases are generally dominant throughout both mine sites.

Goal 2: Apply microspectroscopic methods to low-concentration (<100 ppm) mine wastes, utilizing micro-X-ray fluorescence (uXRF), micro-EXAFS (uEXAFS), and micro-X-ray diffraction (uXRD) techniques to assess the speciation, distribution, and correlation of metals associated with the ore metals in samples that could not otherwise be analyzed by bulk EXAFS spectroscopy.

Hypothesis: Speciation of Hg and As in low-concentration samples will reveal an enrichment in relatively insoluble phases due to their persistence at low levels. Hg is expected to be concentrated as discrete mineralized particles while As may be more widely distributed among individual grains and sorbed species associated with Fe-oxides. The distribution of other significant metals including Co, Cd, Cr, Cu, Ni, Pb, and Zn will be assessed by these techniques to determine whether they are present as discrete metal-rich grains or dispersed as sorbed or amorphous phases.

Considerable progress has been made on exploring the capabilities of microspectroscopic methods for the analysis of mine waste samples, particularly as a function of particle size. Due to either the absence of necessary instrumentation or the low time investment/quality ratio for data collection, uXRD and uEXAFS were largely eschewed in favor of focusing on uXRF analysis. Still, a wealth of data was generated which provide intriguing insights into the distribution and relationships of metals with one another.

Selected size fractions from samples chosen based on results from the previously-described bulk concentration analyses were investigated using uXRF methods, resulting in finely-detailed (i.e. resolution as small as 2x2 um) spatial maps that could selectively express the relative intensities of over a dozen different elements. A list of samples and size fractions analyzed using microspectroscopic methods is provided in Appendix B. For simplicity's sake this report will describe our findings regarding the observed relationship between As and Fe, which was initially observed in the bulk concentration analyses as discussed earlier.

Comparing the As and Fe elemental distribution maps side-by-side, it becomes apparent that the two elements are generally associated with one another, and furthermore that particles containing both elements appear to fall within one of two categories: those with a relatively high As:Fe ratio and those with a relatively low As:Fe ratio (Figure 6). This can be expressed more readily by generating a correlation plot (Figure 7) that plots the relative fluorescence counts for As and

Fe for every pixel in the map, showing clearly the two populations of particles noted in the elemental distribution maps.

In a number of selected samples a similar association between As and Fe has been observed; this is both consistent with the bulk concentration data and the bulk EXAFS speciation analysis described earlier and shows excellent agreement between the different analytical methods in determining the speciation of arsenic in the mine tailings samples investigated. An additional observation seen among several samples has been an increase in the number of populations (represented as pixels forming lines or "fingers" of constant slope in the correlation plots) with decreasing particle size. This may represent an increasing diversity of As-bearing phases in the finer size fractions and/or the sorption of As to different iron oxide phases resulting in different As:Fe ratios. Ongoing work on these associations and those of other pairs of metals is continuing from the data collected thus far.

Work Products Produced

In addition to this performance report, a manuscript is in preparation entitled "Particle-Size Dependence on Metal Concentrations in Mine Wastes" (authors Kim, C.S., Wilson, K., and Rytuba, J.J.) focusing on the bulk concentration analyses which will be submitted to Environmental Science & Technology by the end of this calendar year. Shortly thereafter we anticipate submitting a second manuscript presenting our findings using bulk and microscale spectroscopic methods by February 2008.

Research funded by the MRERP was presented at the following professional conferences, with students' names provided in bold:

Kim, C.S. and Rytuba, J.J. "Trends in metal speciation and distribution within mine wastes as a function of particle size." Oral presentation at 2007 Geological Society of America Annual Meeting, October 2007, Denver, CO.

Miller S.R., Sugihara E., Petersen N., Mortera H., and Kim C.S. "Leach testing of mine wastes: environmental impact and metal mobility in water supplies and body fluids." Poster presentation at 2007 Geological Society of America Annual Meeting, October 2007, Denver, CO.

Wilson, K.M., Rytuba, J.J., and Kim, C.S. "Trends In metal concentrations and associations as a function Of particle size In mine wastes." Poster presentation at 2007 Geological Society of America Annual Meeting, October 2007, Denver, CO.

Kim, C.S. and Rytuba, J.J. "Characterizing the speciation, distribution, and correlations of heavy metals in mine wastes." Invited oral presentation at 2007 SSRL/LCLS Users Meeting, October 2007, Stanford, CA.

Miller, S., Wilson, K., Sugihara, E. and Kim, C.S. "Speciation of metal-bearing mine wastes using micro X-ray fluorescence." Poster presentation at 2007 SSRL/LCLS Users Meeting, October 2007, Stanford, CA.

Kim, C.S. and Rytuba, J.J. "Particle size effects on heavy metal distribution, speciation, and correlations in mine wastes." Invited oral presentation at 17th V.M. Goldschmidt geochemistry conference, August 2007, Cologne, Germany.

Finally, this project has resulted in the creation of a detailed and sophisticated template for the autogeneration of mass distribution, elemental concentration, and elemental mass distribution trends as a function of particle size that will continue to be used for future sample characterization. An example of one such template is provided in Appendix C.

Concluding Remarks

The insights gained from the research funded by the MRERP have considerably advanced our understanding of trends in concentration, speciation, distribution, and correlations between metals in mine wastes as a function of particle size and will inform or have informed current and future investigations by the grant recipient and his research group. The techniques developed and applied as part of this project will continue to be of use in ongoing characterizations of additional mine sites as part of a formal collaboration between the recipient, the USGS, and the BLM.



Figure 1. Example of bulk mass and concentration trend data generated for each size-separated sample. (a) Sample mass distribution as a function of particle size. (b) Concentration of each element (arsenic shown only) as a function of particle size. (c) Mass distribution of each element (arsenic shown only) within the sample as a function of particle size.



Figure 2. Schematic of different concentration trends observed within samples (particle size decreases along the X-axis from left to right; Y-axis represents concentration). Analysis of many samples indicates that the "inverse" trend between size and concentration is common for a number of heavy metals including arsenic and mercury.



Figure 3. EXAFS speciation analysis of mine tailings sample from the Rand Mine in the Randsburg mining district.



Figure 4. EXAFS speciation analysis of mine tailings sample from the Clare Mine in the Kelly mining district.



Figure 5. Speciation of samples from the Kelly and Randsburg mine regions using As K-edge EXAFS spectroscopy. (FT) = Fluvially transported tailings; (F) = Collected from air filter.



Figure 6. Iron and arsenic microscale elemental distribution maps showing two populations of particles: those with a high As:Fe ratio and those with a low As:Fe ratio. Regions encompassing particles in each population are circled on both elemental maps to aid comparison.



Figure 7. Correlation plot of relative fluorescence counts for As and Fe for the sample shown in Figure 6, demonstrating clearly the presence of two populations of particles where the two elements are associated with one another. Units on both axes are counts/second.



Figure 8. Comparison of As:Fe correlation plots for the S7 (a) and S11 (b) size fractions, showing an increase in diversity of particle populations with distince As:Fe ratios, suggesting a wider variety of As-bearing phases with decreasing particle size. This results in a "fingering" effect that is more apparent in the finer size fractions. Units on both axes are counts/second.

Appendix A Inventory of size-separated samples

Dates	Sample ID	Location	Sample type	Notes
5/3/05	04RC3	Rinconada		
5/6/05	21CT50	Rinconada		
5/11/05	21CT8C	Rinconada		
7/14/05	22CC22SB	Cache Creek		No Chemex data for S9-S11
7/14/05	23CC11SB	Cache Creek		No Chemex data for S7-S11
9/14/05	05HFGPIS	Hocker Flat Pit		
7/27/06	1-3C	Kelly Mine		No Chemex data for S11
8/3/06	2-B4	Kelly Mine Pile		No Chemex data for S11
12/6/06	MG-W1	Marigold		No Chemex data for S1, S10, S11
12/13/06	06HF1053	Hocker Flat		No Chemex data for S9
12/15/06	06CC12FS	Clear Creek		
12/15/06	RBDCMT_Hill	Descarga	Tailings?	No Chemex data for S11
12/16/06	06HF12S2	Lower Hocker Flat (Randsburg)	Tailings from end of stream (highly weathered	No Chemex data for S2 and S3
12/19/06	06JS3S2	Middle Hocker Flat (Randsburg)	Tailings from end of stream (highly weathered	No Chemex data for S2 and S3
12/19/06	RBMT-DSGA Cliff	Descarga	Tailings?	No Chemex data for S1, S2, S10, S11
12/20/06	RBMT-YAMT1	Descarga	Tailings?	No Chemex data for S1, S2, S3, S10, S11
12/21/06	JBNE-MTW2	Johannesburg	Tailings?	No Chemex data for S10 and S11
4/17/07	07T26	Randsburg	Tailings	No Chemex data for S11
4/25/07	RMBLM-T-2-0	Red Mountain	Tailings	
5/29/07	07RMBK74	Randsburg	Background material	
5/31/07	07T66	Randsburg Solomon Mine	Tailings	
6/15/07	07BK1-0	Randsburg	Background material	
6/15/07	07-OHE-comp	Oat Hill Mine		No Chemex data for S1, S2, S10, S11
6/22/07	07T46	Randsburg	Tailings	No Chemex data for S10 and S11
6/27/07	RMS03C	Ruth Mine		
6/27/07	RMWD1/2	Ruth Mine Composite Mill	Tailings	
9/11/07	07T34	Randsburg Solomon Mine		
9/25/07	GQW-MT3	Golden Queen	Tailings	
10/2/07	GQ-MT1	Golden Queen	Tailings	
10/16/07	GQ-MT4	Golden Queen	Tailings	
10\06	06AU5F	Aurora		No Chemex data for S11
	06BC11T	Bodie Creek	Tailings?	No Chemex data for S10 and S11

Appendix B

Inventory of samples analyzed by micro-X-ray fluorescence

Elements run (channels)	Sample ID	Location	Sample type	Splits
Fe, As, Zn, Cu, Ti, Ca, K, S, Au, Mn	OHV_FILT	Randsburg	Dust from respirator air filter	
	07T66	Randsburg Solomon Mine	Tailings	S5,7,9,11
	07RMBK74	Randsburg	Background	S5,7,9,11
	RMBLM-T-2-0	Red Mountain	Tailings	S5,7,9,11
	RMWD1/2	Ruth Mine Composite Mill	Tailings	S5,9,11
	06BC12T	Bodie Creek	Tailings?	S9,11
Fe, As, Zn, Cu, Ti, Ca, K, S, Mn, Hg, Ni, Se	07T26	Randsburg	Tailings	S7-11
-	07BK59-1	Randsburg	Background	S7-11
	06BC11T	Bodie Creek	Tailings?	S9-11
	06AU5F	Aurora	?	S9-11
07Fe, As, Zn, Cu, Ti, Ca, K, Mn, Hg, Ni, Se, Cl, Si, SP?	06CC12FS	Clear Creek	?	S11
	06HF12S2	Lower Hocker Flat (Randsburg)	Tailings transported to end of stream (highly weathered)	S11
	06JS3S2	Middle Hocker Flat (Randsburg)	Tailings transported to end of stream (highly weathered)	S11
	JBNE_MTW2	Johannesburg	Tailings?	S5, 9, 11
	MG_W1	Marigold	?	S7, 11
	RBDCMT_Hill	Descarga	Tailings?	S7, 9
	RBMT_DSGA Cliff	Descarga	Tailings?	S9, 11
	RBMY_YAMT1	Descarga	Tailings?	S11
	Elements run (channels) Fe, As, Zn, Cu, Ti, Ca, K, S, Au, Mn Fe, As, Zn, Cu, Ti, Ca, K, S, Mn, Hg, Ni, Se 07Fe, As, Zn, Cu, Ti, Ca, K, Mn, Hg, Ni, Se, Cl, Si, SP?	Elements run (channels)Sample IDFe, As, Zn, Cu, Ti, Ca, K, S, Au, MnOHV_FILT 07T66 07RMBK74 RMBLM-T-2-0 RMWD1/2 06BC12TOHV_FILT 07T26 07BK59-1 06BC12TFe, As, Zn, Cu, Ti, Ca, K, S, Mn, Hg, Ni, Se07T26 07BK59-1 	Elements run (channels)Sample IDLocationFe, As, Zn, Cu, Ti, Ca, K, S, Au, MnOHV_FILT 07766Randsburg Solomon Mine 07RMBK74 Randsburg Randsburg RMBLM-T-2-0Red Mountain RMWD1/2 Bodie CreekFe, As, Zn, Cu, Ti, Ca, K, S, Mn, Hg, Ni, Se07T26 07BK59-1 06BC11T 06BC11T 06AU5FRandsburg Bodie Creek07Fe, As, Zn, Cu, Ti, Ca, K, Mn, Hg, Ni, Se, Cl, Si, SP?06CC12FS 06AU5FClear Creek 06HF12S2 Lower Hocker Flat (Randsburg) 06JS3S2 Middle Hocker Flat (Randsburg) JBNE_MTW207Fe, As, Zn, Cu, Ti, Ca, K, Mn, Hg, Ni, Se, Cl, Si, SP?06CC12FS 06CC12FS DescargaClear Creek Dohannesburg MG_MU1	Elements run (channels)Sample IDLocationSample typeFe, As, Zn, Cu, Ti, Ca, K, S, Au, MnOHV_FILT 07766 07RMBK74 RMBLM-T-2-0 RMBLM-T-2-0 RMBLM-T-2-0 RMBLM-T-2-0 RMBLM-T-2-0 RMBLM-T-2-0 RMBLM-T-2-0 RMDD1/2 06BC12TRandsburg Ruth Mine Composite Mill Background Tailings Tailings?Dust from respirator air filter Tailings Background Tailings?Fe, As, Zn, Cu, Ti, Ca, K, S, Mn, Hg, Ni, Se07T26 07BK59-1 06BC11T 06BC11T 06BC11T 06BC11T 06BC11T 06BC11T 06BC11T 06BC2Randsburg Background Tailings?Tailings Background Background Tailings?07Fe, As, Zn, Cu, Ti, Ca, K, Mn, Hg, Ni, Se, Cl, Si, SP?06CC12FS 06CC12FS 06HF12S2 06JC3S2 Middle Hocker Flat (Randsburg) JBNE_MTW2 Johannesburg?07Fe, As, Zn, Cu, Ti, Ca, K, Mn, Hg, Ni, Se, Cl, Si, SP?06CC12FS 06HF12S2 RBM_TOSGA Cliff DEscargaClear Creek RBM_TOSGA Cliff Descarga?07Fe, As, Zn, Cu, Ti, Ca, K, Mn, Hg, Ni, Se, Cl, Si, SP?06CC12FS 06HF12S2 RBM_TOSGA Cliff DEscargaClear Creek RBM_TOSGA?07Fe, As, Zn, Cu, Ti, Ca, K, Mn, Hg, Ni, Se, Cl, Si, SP?06CC12FS 06HF12S2 RBM_TOSGAClear Creek RBM_TOSGA?07Fe, As, Zn, Cu, Ti, Ca, K, Mn, Hg, Ni, Se, Cl, Si, SP?06CC12FS 06HF12S2 RBM_TOSGAClear Creek RBM_TOSGA?07Fe, As, Zn, Cu, Ti, Ca, K, Mn, Hg, Ni, Se, Cl, Si, SP?06CC12FS 06HF12S2 RBM_TOSGAClear Creek RBM_TOSGA?07Fe, As, Zn, Cu, Ti, Ca, K, Mn, Hg, Ni, Se, Cl, Si, SP?06CC12FS RBM_TOSGAClear Creek RBM_TOSGA?07Fe, As, Zn, Cu, Ti,

Appendix C

Sample template for size separation/concentration analysis

06HF1053

Sample size distribution by mass

Sample Information						
Sample ID:	06HF1053					
Sample Description:	Hocker Flat					
Date Sieved:	12/13/06					
Name:						

Initial Mass measurement				
Trial	Mass (g)			
1	385.7			
2	384.4			
3	393.7			
4	386.4			
5	397.5			
6	218.7			
7	297			
8	290.5			
9	397.4			
10	901.6			
Total initial mass (g):	4052.9			

Particle Diameter (dp)	Split	% mass
dp >2830 um	S1	29.3%
2830 um>dp>1700 um	S2	8.3%
1700 um>dp>1000 um	S3	11.2%
1000 um>dp>500 um	S4	24.7%
500 um>dp>250 um	S5	18.7%
250 um>dp>125 um	S6	6.0%
125 um>dp>75 um	S7	1.2%
75 um>dp>45 um	S8	0.3%
45 um>dp>32 um	S9	0.1%
32 um>dp>20 um	S10	0.1%
dp<20 um	S11	0.2%

Sample split information					
06HF1053 S1	06HF1053 S6				
dp >2830 um	250 um>dp>125 um				
1161.7 <mark>g</mark>	237.7 g				
580.85 g (split)	118.85 g (split)				
06HF1053 S2	06HF1053 S7				
2830 um>dp>1700 um	125 um>dp>75 um				
328.4 g	46.1 g				
164.2 g (split)	23.05 g (split)				
06HF1053 S3	06HF1053 S8				
1700 um>dp>1000 um	75 um>dp>45 um				
446.7 <mark>g</mark>	13.6 g				
223.35 g (split)	NSS g (split)				
06HF1053 S4	06HF1053 S9				
1000 um>dp>500 um	45 um>dp>32 um				
981.2 <mark>g</mark>	3.7 g				
490.6 g (split)	NSS g (split)				
06HF1053 S5	06HF1053 S10				
500 um>dp>250 um	32 um>dp>20 um				
741.8 <mark>g</mark>	4.2 g				
370.9 g (split)	NSS g (split)				
Sum of splits 3971.2	g 06HF1053 S11				
% error 2.0%	dp<20 um				
	6.1 g				
	NSS g (split)				



	WEI-21	ME-MS61	ME-MS61	ME-MS61	ME-MS61	ME-MS61	ME-MS61
SAMPLE	Recvd Wt.	Ag	AI	As	Ва	Be	Bi
DESCRIPTION	kg	ppm	%	ppm	ppm	ppm	ppm
06HF1053-Bulk	0.08	0.08	5.76	5.5	330	0.58	0.06
06HF1053_S1	0.06	0.09	5.77	4.1	360	0.9	0.06
06HF1053_S2	0.18	0.1	5.59	4.5	310	0.74	0.06
06HF1053_S3	0.06	0.09	5.37	5.9	360	0.83	0.05
06HF1053_S4	0.08	0.07	5.24	5.7	420	0.75	0.05
06HF1053_S5	0.08	0.08	6.03	8	400	0.86	0.05
06HF1053_S6	0.16	0.09	6.25	9.2	330	0.98	0.06
06HF1053_S7	0.04	0.09	6.14	9.7	260	0.77	0.08
06HF1053_S8	0.02	0.11	6.46	11.3	330	0.8	0.1
06HF1053_S10	0.02	0.18	7.08	15.8	460	0.97	0.11
06HF1053_S11	0.02	0.37	7.38	23.6	490	1.05	0.16



 \$1> 2830 um
 1000 um>\$4>500 um
 125 um>\$7>75 um
 32 um>\$10>20 um

 2830 um>\$2>1700 um
 500 um>\$5>250 um
 75 um>\$8>45 um
 20 um>\$11

 1700 um>\$3>1000 um
 250 um>\$6>125 um
 45 um>\$9>32 um

ME-MS61	ME-MS61	ME-MS61	ME-MS61	ME-MS61	ME-MS61
Са	Cd	Ce	Co	Cr	Cs
%	ppm	ppm	ppm	ppm	ppm
2.81	0.14	14.1	18.5	223	0.85
3.06	0.14	18.2	22.8	88	0.74
3.33	0.15	16	19.6	127	0.78
2.62	0.14	16.8	18.8	130	1.02
2.15	0.14	16.65	18.1	132	1.16
2.53	0.17	19.75	24.9	219	1.44
3.24	0.21	25.7	29.6	538	1.25
4.3	0.25	35.6	38.9	1320	1
4.17	0.24	34	39.9	1035	1.19
3.66	0.28	31.4	42	411	1.74
3.26	0.34	32.9	47.1	426	2.12



S1 > 2830 um	1000 um> S4 >500 um	125 um> S7 >75 um	32 um> S10 >20 um
2830 um> S2 >1700 um	500 um> S5 >250 um	75 um> S8 >45 um	20 um> S11
1700 um> S3 >1000 um	250 um> S6 >125 um	45 um> S9 >32 um	

ME-MS61	ME-MS61	ME-MS61	ME-MS61	ME-MS61	Hg-CV42
Cu	Fe	Ga	Ge	Hf	Hg
ppm	%	ppm	ppm	ppm	ppm
27.7	4.29	12.6	0.12	0.7	0.07
33.8	5.23	17.9	0.18	1.3	0.02
35.9	4.87	15.2	0.15	0.9	0.02
38.4	4.36	15.25	0.17	1	0.02
34.5	3.71	14	0.14	0.8	0.03
36.9	4.13	16.75	0.18	0.9	0.04
36.7	5.24	18.25	0.17	1	0.1
40.2	8.08	20.1	0.21	1.3	0.29
74	9.39	20.3	0.2	2.1	2.48
174.5	7.21	21.1	0.19	1.6	2.5
250	7.77	20.4	0.18	1.6	0.87



S1 > 2830 um	1000 um> S4 >500 um	125 um> S7 >75 um	32 um> S10 >20 um
2830 um> S2 >1700 um	500 um> S5 >250 um	75 um> S8 >45 um	20 um> S11
1700 um> S3 >1000 um	250 um> S6 >125 um	45 um> S9 >32 um	

ME-MS61	ME-MS61	ME-MS61	ME-MS61	ME-MS61	ME-MS61
In	к	La	Li	Mg	Mn
ppm	%	ppm	ppm	%	ppm
0.043	0.69	8.1	10.8	2.19	789
0.074	0.62	10.1	13.9	2.05	787
0.065	0.54	9	12.1	2.02	843
0.058	0.66	9.8	13.8	1.78	749
0.048	0.76	10	13.9	1.61	678
0.054	0.87	11.9	15.9	1.99	757
0.069	0.7	14.4	14.2	2.64	951
0.102	0.51	19.4	12.7	3.34	1450
0.102	0.57	18.8	14.7	3.31	1570
0.12	0.74	17.9	20.5	3.5	1420
0.188	0.77	19.2	22.9	3.66	1710



S1 > 2830 um	1000 um> S4 >500 um	125 um> S7 >75 um	32 um> S10 >20 um
2830 um> S2 >1700 um	500 um> S5 >250 um	75 um> S8 >45 um	20 um> S11
1700 um> S3 >1000 um	250 um> S6 >125 um	45 um> S9 >32 um	

ME-MS61	ME-MS61	ME-MS61	ME-MS61	ME-MS61	ME-MS61
Мо	Na	Nb	Ni	Р	Pb
ppm	%	ppm	ppm	ppm	ppm
0.38	1.71	3.6	131	390	5.4
1.24	1.74	6.2	82.8	510	3.8
0.46	1.55	4.4	83.9	510	4.2
1.03	1.54	5.2	95.3	460	4.6
0.55	1.58	4.8	102.5	400	6.3
1.21	1.77	5.5	140	370	7.1
0.57	1.68	6.1	188	360	6.6
1.37	1.39	10	204	570	7.7
2.25	1.43	10.4	226	770	8.9
6.12	1.46	9.2	263	840	33.1
10.05	1.44	9.9	353	880	50.3



S1 > 2830 um	1000 um> S4 >500 um	125 um> S7 >75 um	32 um> S10 >20 um
2830 um> S2 >1700 um	500 um> S5 >250 um	75 um> S8 >45 um	20 um> S11
1700 um> S3 >1000 um	250 um> S6 >125 um	45 um> S9 >32 um	

ME-MS61	ME-MS61	ME-MS61	ME-MS61	ME-MS61	ME-MS61
Rb	Re	s	Sb	Se	Sn
ppm	ppm	%	ppm	ppm	ppm
17.1	<0.002	0.01	0.75	1	0.8
17.7	<0.002	0.03	0.95	2	1.3
17	<0.002	0.02	0.61	2	1.2
21.2	<0.002	0.01	0.71	2	1.3
24	<0.002	0.01	0.68	1	1.5
28.9	<0.002	0.01	0.87	1	1.1
22	<0.002	< 0.01	0.92	2	1.4
16.4	<0.002	0.01	1.13	2	3
18.8	<0.002	0.02	1.29	2	4.3
26.6	<0.002	0.02	1.24	2	36.8
28.9	<0.002	0.02	1.47	3	25



S1 > 2830 um	1000 um> S4 >500 um	125 um> S7 >75 um	32 um> S10 >20 um
2830 um> S2 >1700 um	500 um> S5 >250 um	75 um> S8 >45 um	20 um> S11
1700 um> S3 >1000 um	250 um> S6 >125 um	45 um> S9 >32 um	

ME-MS61	ME-MS61	ME-MS61	ME-MS61	ME-MS61	ME-MS61
Sr	Та	Те	Th	Ti	TI
ppm	ppm	ppm	ppm	%	ppm
158	0.24	0.05	1.8	0.423	0.11
141.5	0.39	< 0.05	2.4	0.577	0.1
147	0.29	<0.05	1.8	0.485	0.11
148.5	0.32	< 0.05	2.1	0.45	0.13
156	0.31	< 0.05	2	0.359	0.16
185	0.36	<0.05	2.4	0.414	0.18
186	0.41	0.05	3	0.587	0.14
171.5	0.69	0.05	7.2	1.12	0.11
170.5	0.68	0.05	6.4	1.14	0.13
164	0.59	0.08	4.2	0.731	0.18
149	0.62	0.12	4.1	0.7	0.22



S1 > 2830 um	1000 um> S4 >500 um	125 um> S7 >75 um	32 um> S10 >20 um
2830 um> S2 >1700 um	500 um> S5 >250 um	75 um> S8 >45 um	20 um> S11
1700 um> \$3 >1000 um	250 um> S6 >125 um	45 um> S9 >32 um	

ME-MS61	ME-MS61	ME-MS61	ME-MS61	ME-MS61	ME-MS61
U	v	w	Y	Zn	Zr
ppm	ppm	ppm	ppm	ppm	ppm
0.6	158	0.7	16.2	62	23.6
0.8	198	1.2	26.6	63	44.9
0.6	186	0.8	23	64	31.8
0.7	161	0.9	20.4	61	34.2
0.7	129	0.7	16.4	52	28.1
0.8	146	1.1	18.8	60	30.2
0.9	197	1	25.8	73	31.6
1.4	303	1.6	41.9	101	41.6
1.5	302	2.1	40.8	110	61.4
1.3	250	2.4	34.4	131	50.5
1.4	249	2	30.9	154	58.4



S1 > 2830 um 2830 um> S2 >1700 um 1700 um> S3 >1000 um	1000 um> S4 >500 um 500 um> S5 >250 um 250 um> S6 >125 um	125 um> S7 >75 um 75 um> S8 >45 um 45 um> S9 >32 um	32 um> S10 >20 um 20 um> S11

TOTAL MASS (mg)		0.34	221430.05	22.47	1424.82	2.94	0.22
SAMPLE	Recvd Wt.	Ag	AI	As	Ba	Be	Bi
DESCRIPTION		%	%	%	%	%	%
06HF1053-Bulk	0.08	28%	30%	28%	27%	23%	32%
06HF1053_S1	0.06	9%	9%	6%	8%	10%	9%
06HF1053_S2	0.18	13%	11%	9%	10%	11%	12%
06HF1053_S3	0.06	26%	24%	26%	25%	28%	22%
06HF1053_S4	0.08	15%	18%	19%	22%	19%	17%
06HF1053_S5	0.08	6%	6%	8%	7%	7%	5%
06HF1053_S6	0.16	1%	1%	2%	1%	2%	1%
06HF1053_S7	0.04	0%	0%	1%	0%	0%	0%
06HF1053_S8	0.02	0%	0%	0%	0%	0%	0%
06HF1053_S10	0.02	0%	0%	0%	0%	0%	0%
06HF1053_S11	0.02	1%	0%	1%	0%	0%	0%





107823.18	0.57	65.16	78.03	673.13	3.88
Ca	Cd	Ce	Co	Cr	Cs
%	%	%	%	%	%
30%	28%	25%	28%	38%	25%
9%	8%	9%	10%	4%	6%
14%	12%	11%	11%	8%	9%
24%	24%	25%	24%	19%	26%
15%	18%	19%	17%	15%	22%
6%	7%	7%	8%	8%	9%
1%	2%	2%	2%	4%	1%
1%	1%	1%	1%	3%	0%
0%	0%	0%	0%	1%	0%
0%	0%	0%	0%	0%	0%
0%	0%	0%	0%	0%	0%





136.13	173523.39	58.04	0.59	3.52	0.18
Cu	Fe	Ga	Ge	Hf	Hg
%	%	%	%	%	%
24%	29%	25%	24%	23%	45%
8%	10%	10%	10%	12%	4%
12%	13%	12%	11%	11%	5%
28%	25%	26%	28%	28%	11%
19%	16%	18%	18%	17%	12%
6%	6%	7%	7%	6%	5%
1%	1%	1%	1%	1%	3%
0%	1%	0%	0%	1%	2%
0%	0%	0%	0%	0%	5%
1%	0%	0%	0%	0%	6%
1%	0%	0%	0%	0%	3%





0.22	27136.78	37.80	51.25	77499.35	3055.16
In	к	La	Li	Mg	Mn
%	%	%	%	%	%
23%	30%	25%	24%	33%	30%
11%	8%	9%	9%	9%	8%
13%	9%	11%	11%	12%	12%
26%	24%	25%	26%	23%	24%
17%	21%	20%	20%	15%	16%
6%	8%	7%	7%	6%	6%
1%	1%	2%	1%	2%	1%
1%	0%	1%	0%	1%	1%
0%	0%	0%	0%	0%	0%
0%	0%	0%	0%	0%	0%
1%	0%	0%	0%	0%	0%





2.90	64706.88	18.71	435.21	1720.48	21.16
Мо	Na	Nb	Ni	Р	Pb
%	%	%	%	%	%
15%	31%	22%	35%	26%	30%
14%	9%	11%	6%	10%	6%
7%	11%	11%	9%	13%	9%
35%	23%	27%	21%	26%	21%
14%	18%	19%	17%	17%	22%
10%	7%	7%	8%	5%	8%
1%	1%	2%	2%	1%	1%
1%	0%	1%	1%	0%	0%
0%	0%	0%	0%	0%	0%
1%	0%	0%	0%	0%	1%
2%	0%	0%	0%	0%	1%





80.34	0.00	504.26	2.94	5.81	4.97
Rb	Re	S	Sb	Se	Sn
%	%	%	%	%	%
25%	0%	23%	30%	20%	19%
7%	0%	20%	11%	11%	9%
9%	0%	18%	9%	15%	11%
26%	0%	19%	24%	34%	26%
22%	0%	15%	17%	13%	22%
9%	0%	5%	7%	4%	5%
1%	0%	0%	1%	2%	1%
0%	0%	0%	1%	0%	1%
0%	0%	0%	0%	0%	0%
0%	0%	0%	0%	0%	3%
0%	0%	0%	0%	0%	3%



614.22	1.20	0.06	8.10	17576.40	0.51
Sr	Та	Те	Th	Ti	TI
%	%	%	%	%	%
30%	23%	93%	26%	28%	25%
8%	11%	0%	10%	11%	6%
11%	11%	0%	10%	12%	10%
24%	26%	0%	25%	25%	25%
19%	19%	0%	18%	15%	23%
7%	7%	0%	7%	6%	8%
1%	2%	4%	2%	2%	1%
0%	1%	1%	1%	1%	0%
0%	0%	0%	0%	0%	0%
0%	0%	1%	0%	0%	0%
0%	0%	1%	0%	0%	0%





2.70	636.92	3.33	76.72	240.63	120.76
U	v	w	Y	Zn	Zr
%	%	%	%	%	%
26%	29%	24%	25%	30%	23%
10%	10%	12%	11%	9%	12%
10%	13%	11%	13%	12%	12%
25%	25%	27%	26%	25%	28%
19%	15%	16%	16%	16%	17%
7%	5%	8%	6%	6%	6%
2%	1%	1%	2%	1%	1%
1%	1%	1%	1%	1%	0%
0%	0%	0%	0%	0%	0%
0%	0%	0%	0%	0%	0%
0%	0%	0%	0%	0%	0%





Sample									
06HF1053									
	Direct	Inverse	Unimodal (max)	Bimodal (2 max)	Trimodal (3 max)	Reverse Unimodal (min)	Constant	Poor	Below Detection
		Ag Al As Bi Cd E Co Cs Cu G Hg La Mo Nb Ib Sb Ta U W Zn	Cr(S7) Fe(S8) Sn(S10) Sr(S6) Th(S7) Ti(S8)	Ba(S4,11) Ca(S2,7) Hf(S1,8) K(S5,11) Li(S5,11) RB(S5,11) TI(S5,11)	Zr(S1,8,11)	In(S4) Mg(S4) P(S6) V(S4) Y(S4)	Na S Se	Ge Te	Re

Elemental Concentration Trend Categorization