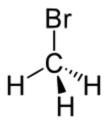
Bromomethane - Wikipedia, the free encyclopedia

Bromomethane



Names

IUPAC name

Identifiers

74-83-9 Y

1209223

<u>Chebi</u>:39275 ✓ Chebi:39275

ChEMBL ChEMBL48339 Y

ChemSpider 6083 Y

EC Number 200-813-2

916

Jmol 3D model <u>Interactive image</u>

KEGG C18447 'Y

MeSH methyl+bromide

PubChem 6323

RTECS number PA4900000

• InChI=1S/CH3Br/c1-2/h1H3 ^Y

Key: GZUXJHMPEANEGY-UHFFFAOYSA-N 'Y

Molar mass 94.94 g·mol⁻¹

Appearance Colorless gas [2]

Odor Chloroform-like

Density 3.97 kg/m³ (gas, 0 °C) 2

1.72 g/mL (liquid, 4 °C)[2]

Melting point -93.66 °C (-136.59 °F; 179.49 K) [2]

Boiling point 4.0 °C (39.2 °F; 277.1 K) [2]

 17.5 g L^{-1}

log P 1.3

Vapor pressure 190 kPa (at 20 °C (68 °F))

Thermochemistry

 $-35.1 - -33.5 \text{ kJ mol}^{-1}$

Hazards





R-phrases R23/25, R36/37/38, R48/20, R50, R59,

R68

S-phrases (S1/2), S15, S27, S36/39, S38, S45,

S59, S61

NFPA 704

<u>Flash point</u> 194 °C (381 °F; 467 K) [2]

535 °C (995 °F; 808 K)[2]

Explosive limits 10%-16%[3]

Lethal dose or concentration (LD, LC):

1200 ppm (mouse, 1 hr)

7316 ppm (rabbit, 30 min)

2833 ppm (rat, 1 hr)

302 ppm (rat, 8 hr)

390 ppm (mouse, 9 hr)[4]

300 ppm (guinea pig, 9 hr) 4

LIC leas libe and a some liberita (NITOCII)

Ca^[3]
Ca [250 ppm]^[3]
Related compounds

Related alkanes

Except where otherwise noted, data are given for materials in their standard state (at 25 °C [77 °F], 100 kPa).

✓Y verify (what is ✓YxN?)
Infobox references

Bromomethane, commonly known as methyl bromide, is an organobromine compound with formula CH₃Br. This colorless, odorless, nonflammable gas is produced both industrially and particularly biologically. It has a tetrahedral shape and it is a recognized ozone-depleting chemical. It was used extensively as a pesticide until being phased out by most countries in the early 2000s.

Occurrence and manufacture[edit]

Bromomethane originates from both natural and human sources. In the <u>ocean</u>, marine organisms are estimated to produce 56,000 tonnes annually. [5] It is also produced in small quantities by certain terrestrial plants, such as members of the <u>Brassicaceae</u> family. It is manufactured for agricultural and industrial use by reacting <u>methanol</u> with hydrogen bromide:

$$CH_3OH + HBr \rightarrow CH_3Br + H_2O$$

Uses[edit]

In 1999, an estimated 71,500 tonnes of synthetic methyl bromide were used annually worldwide. 61 97% of this estimate is used for funication purposes, whilst 3% is used

tne consumption takes place in developed nations, led by

the U.S. (43%) and Europe (24%). Asia and the Middle East combine to use 24% whereas Latin America and Africa have the lowest consumption rates at 9%. [6]



ISPM 15 markings on a pallet from China. The **MB** indicates the wood has been fumigated with methyl bromide to kill pests.

Until its production and use was curtailed by the Montreal Protocol, bromomethane was widely applied as a soil sterilant, mainly for production of seed but also for some crops such as strawberries [7] and almonds. In commercial large-scale monoculture seed production, unlike crop production, it is of vital importance to avoid contaminating the crop with off-type seed of the same species. Therefore, selective herbicides cannot be used. Whereas bromomethane is dangerous, it is considerably safer and more effective than some other soil sterilants. Its loss to the seed industry has resulted in changes to cultural practices, with increased reliance on soil steam sterilization, mechanical <u>roqueing</u>, and <u>fallow</u> seasons. Bromomethane was also used as a general-purpose fumigant to kill a variety of pests including rats and insects. Bromomethane has poor fungicidal properties. Bromomethane is the only fumigant allowed (heat treatment is only other option) under **ISPM 15** regulations when exporting solid wood packaging (forklift pallets, crates, bracing) to ISPM 15 compliant countries. Bromomethane is used to prepare golf courses, particularly to control Bermuda grass. The Montreal Protocol stipulates that bromomethane use be

Bromomethane is also a precursor in the manufacture of other chemicals as a <u>methylating</u> agent, and has been used as a solvent to <u>extract</u> oil from seeds and wool.

Bromomethane was once used in specialty <u>fire</u> <u>extinguishers</u>, prior to the advent of less toxic halons, as it is electrically non-conductive and leaves no residue. It was used primarily for electrical substations, military aircraft, and other industrial hazards. It was never as popular as other agents due to its high cost and toxicity.

Bromomethane was used from the 1920s to the 60s, and continued to be used in aircraft engine fire suppression systems into the late 60s.



A selection of stored-pressure bromomethane fire extinguishers, circa 1930s-40s, UK

Regulation[edit]

Bromomethane is readily <u>photolyzed</u> in the atmosphere to release elemental bromine, which is far more destructive to stratospheric ozone than chlorine. As such, it is subject to phase-out requirements of the 1987 <u>Montreal Protocol on Ozone Depleting Substances</u>.

The London Amendment in 1990 added bromomethane to the list of ODS to be phased out. In 2003 the Global

project for inethyr profinae total sector phase out in seven

countries in Central Europe and Central Asia, which was due for completion in 2007. [8]

Alternatives[edit]

Many alternatives for bromomethane in the agricultural field are currently in use and further alternatives are in development, including propylene oxide and furfural. [9]

Australia[edit]

In <u>Australia</u>, bromomethane is the preferred fumigant of the Department of Agriculture and Water Resources for most organic goods imported into Australia. 10 The department conducts methyl bromide fumigation certification for both domestic and foreign fumigators who can then fumigate containers destined for Australia. A list of alternative fumigants is available for goods imported from Europe (known as the BICON database), where methyl bromide fumigation has been banned. Alternatively, the department allows containers from Europe to be fumigated with methyl bromide on arrival to Australia.

New Zealand[edit]

In New Zealand, bromomethane is used as a fumigant for whole logs destined for export. Environmental groups and the Green Party oppose its use. [12][13] In May 2011 the Environmental Risk Management Authority (ERMA) introduced new rules for its use which restrict the level of public exposure to the fumigant, set minimum buffer zones around fumigation sites, provide for notification to nearby residents and require users to monitor air quality during fumigations and report back to ERMA each year. All methyl bromide fumigations must use recapture technology by

4 41

United States[edit]

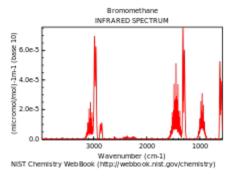
In the <u>United States</u> bromomethane is regulated as a pesticide under the Federal Insecticide, Fungicide, and Rodenticide Act (FIFRA; 7 U.S.C. 136 et seq.) and as a hazardous substance under the Resource Conservation and Recovery Act (RCRA; 42 U.S.C. 6901 et seq.), and is subject to reporting requirements under the **Emergency Planning** and Community Right-to-Know Act (EPCRA; 42 U.S.C. 11001 et seq.). The U.S. Clean Air Act (CAA; 42 U.S.C. 7401 et seq.). A 1998 amendment (P.L. 105-178, Title VI) conformed the Clean Air Act phase out date with that of the Montreal Protocol. [15][16] Whereas the Montreal Protocol has severely restricted the use of bromomethane internationally, the United States has successfully lobbied for critical-use exemptions. [17] Critical use exemptions allow the United States to continue using MeBr until it is scheduled to be completely phased out sometime in 2017.^[18]

In 2004, over 7 million pounds of bromomethane were applied to California. Applications include tomato, strawberry, and ornamental shrub growers, and fumigation of ham/pork products. Also exempt is the treatment of solid wood packaging (forklift pallets, crates, bracing), and the packaged goods, being exported to ISPM 15 countries (to include Canada in 2012).

Chile[edit]

Has to phase out the use in agriculture by 2015. [19]

Health effects[edit]



Brief exposure to high concentrations and prolonged inhalation of lower concentrations are problematic. [20] Exposure levels leading to death vary from 1,600 to 60,000 ppm, depending on the duration of exposure (as a comparison exposure levels of 70 to 400 ppm of carbon monoxide cover the same spectrum of illness/death).

"A TLV-TWA of 1 ppm (3.89 mg/m3) is recommended for occupational exposure to methyl bromide"-ACGIH 8 hour time weighted average. Immediately Dangerous To Life or Health Concentration by NIOSH: "The revised IDLH for methyl bromide is 250 ppm based on acute inhalation toxicity data in humans [Clarke et al. 1945]. This may be a conservative value due to the lack of relevant acute toxicity data for workers exposed to concentrations above 220 ppm. [Note: NIOSH recommends as part of its carcinogen policy that the "most protective" respirators be worn for methyl bromide at any detectable concentration.]"

Detectable concentration by Drager Tube is 0.5 ppm.

Respiratory, kidney, and neurological effects are of the greatest concern. Scientists have investigated whether bromomethane exposure was linked to the death of four New Zealand port workers who died of neuro-degenerative motor neuron disease between 2002 and 2004 and found no connection. [citation needed]

Treatment of wood packaging requires a concentration of up to 16,000 ppm. An Australian Customs officer died of

mandalle un maide maide maide un bila income adina a cabinamana

needed]. Canada Border Services Officers avoided the same fate[citation needed] because a Mexican container of granite was fumigated with 98% methyl bromide and 2% chloropicrin fumigant, the chloropicrin has a tear-gas effect.

NIOSH considers methyl bromide to be a potential occupational carcinogen as defined by the OSHA carcinogen policy [29 CFR 1990]. "Methyl bromide showed a significant dose-response relationship with prostate cancer risk." [21]

Excessive exposure[edit]

Expression of toxicity following exposure may involve a latent period of several hours, followed by signs such as nausea, abdominal pain, weakness, confusion, pulmonary edema, and seizures. Individuals who survive the acute phase often require a prolonged convalescence. Persistent neurological deficits such as asthenia, cognitive impairment, optical atrophy, and paresthesia are frequently present after moderate to severe poisoning. Blood or urine concentrations of inorganic bromide, a bromomethane metabolite, are useful to confirm a diagnosis of poisoning in hospitalized patients or to assist in the forensic investigation of a case of fatal overdosage. [22]

See also[edit]

- <u>List of highly toxic gases</u>
- Bromoethane
- Bromopropane
- Bromobutane
- Bromopentane
- Bromohexane



- Jump up \(\) "methyl bromide Compound Summary".
 PubChem Compound. USA: National Center for Biotechnology Information. 26 March 2005.
 Identification. Retrieved 26 February 2012.
- 2. ^ <u>Jump up to: a b c d e f g h</u> <u>Record</u> in the GESTIS Substance Database of the IFA
- 3. ^ Jump up to: a b c d "NIOSH Pocket Guide to Chemical Hazards #0400". National Institute for Occupational Safety and Health (NIOSH).
- 4. ^ <u>Jump up to: ^a b "Methyl bromide"</u>. Immediately Dangerous to Life and Health. <u>National Institute for Occupational Safety and Health</u> (NIOSH).
- Jump up ^ Gribble, G. W. (1999). "The diversity of naturally occurring organobromine compounds".
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- 6. ^ Jump up to: a b "Towards Methyl Bromide Phase Out:

 A Handbook for National Ozone Units". UNEP. 1 August
 1999.
- 7. <u>Jump up ^</u> Strawberries hang in the balance, CEN (June 8th, 2015)
- 8. <u>Jump up ^ "Information on Commercially Validated Methyl Bromide Alternative Technologies"</u>. UNEP. 14 September 2009.
- 9. Jump up ^ http://www.epa.gov/ozone/mbr/alts.html
- 10. <u>Jump up ^ "Methyl Bromide Questions and Answers"</u>. department of Agriculture and Water Resources.

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- 11. <u>Jump up ^ "Methyl Bromide Questions and Answers"</u>. the department. Retrieved 2011-02-21.
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- 13. <u>Jump up ^ Kedgley</u>, Sue (2008-02-02). <u>"Picton</u>

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- 16. Jump up ^ CRS Report for Congress: Agriculture: A Glossary of Terms, Programs, and Laws, 2005 Edition -Order Code 97-905
- 17. <u>Jump up ^ Brian Gareau (29 January 2013)</u>. <u>From Precaution to Profit: Contemporary Challenges to Environmental Protection in the Montreal Protocol</u>. Yale University Press. pp. 232–. <u>ISBN 978-0-300-17526-4</u>.
- 18. <u>Jump up ^ "Critical Use Exemption Information"</u>. U.S. Environmental Protection Agency. Retrieved 2015-02-21.
- 19. Jump up ^ Decision XVII/29: Non-compliance with the Montreal Protocol by Chile. United Nations Environment Programme Ozone Secretariat, http://montreal-protocol.org/publications/mp-handbook/ /Section 2 Decisions/Article 8/decs-non-compliance /Decision XVII-29.shtml
- 20. <u>Jump up ^ Muir, GD (ed.)</u> 1971, *Hazards in the Chemical Laboratory*, The Royal Institute of Chemistry, London.
- 21. Jump up ^ http://www.aghealth.org
- 22. <u>Jump up ^</u> R. Baselt, *Disposition of Toxic Drugs and Chemicals in Man*, 8th edition, Biomedical Publications, Foster City, CA, 2008, pp. 982-984.

External links[edit]

- Chemical Alternatives to the agricultural use of Methyl Bromide
- A Dialogical Chamical & Dractice based alternatives to the

- Methyl Bromide Technical Fact Sheet National
 Pesticide Information Center
- Methyl Bromide General Fact Sheet National Pesticide
 Information Center
- Methyl Bromide Pesticide Information Profile Extension
 Toxicology Network
- International Chemical Safety Card 0109
- "NIOSH Pocket Guide to Chemical Hazards #0400".

 National Institute for Occupational Safety and Health
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- <u>Institut national de recherche et de sécurité</u> (1992).
 "<u>Bromométhane.</u>" *Fiche toxicologique n° 67*. Paris:INRS. (French)
- IARC Summaries & Evaluations Vol. 71 (1999)
- The banned pesticide in our soil
- MSDS at Oxford University
- Toxicological profile
- Environmental Health Criteria 166
- OECD SIDS document
- EPA 2010 Critical Use Exemption Nominations
- <u>ChemSub Online (Methyl bromide, Bromomethane)</u>.
- Del. family poisoned <with methyl bromide> in
 Caribbean in grave condition, governor says